

Georgian Technical University  
Institute TECHINFORMI

E ISSN 2667-9221  
<https://doi.org/10.36073/2667-9221>

*Caucasus Abstracts Journal of*  
**Nanoscience *and***  
**Nanotechnology**

**N 3, 2021**

Tbilisi, Georgia

Georgian Technical University  
Institute TECHINFORMI

E ISSN 2667-9221

<https://doi.org/10.36073/2667-9221>

*Caucasus Abstracts Journal of*  
**Nanoscience** *and*  
**Nanotechnology**

**N 3, 2021**

Tbilisi, Georgia

**Editors-in-Chief:**

Levan Chkhartishvili – GTU, Department of Engineering Physics, Georgia

Madona Kopaleishvili – GTU, Institute Techninformi, Georgia

**Editorial Board:**

Irina Bedinashvili – GTU, Institute Techninformi, Georgia

Giorgi Bibileishvili – GTU, Membrane Technologies Engineering Institute, Georgia

Nikoloz Chikhradze – LEPL G.Tsulukidze Mining Institute, Georgia

Grigol Tatishvili – TSU, R. Agladze Institute of Inorganic Chemistry and Electrochemistry, Georgia

Teimuraz Chubinishvili – GTU, Institute Techninformi, Georgia

Mzia Ghogheliani – GTU, Institute Techninformi, Georgia

Nelly Makhviladze – GTU, Institute Techninformi, Georgia

**Text Editor:**

Valeri Sarjveladze - GTU, Institute Techninformi, Georgia

**Editorial Office:** Institute TECHINFORMI, 47 Kostava St., Tbilisi, Georgia  
Tel.: 233-53-15; 233-59-03  
E-mail: tech@gtu.ge  
<https://techninformi.ge>

© GTU, TECHINFORMI 2021

<https://techninformi.ge/>

All contents in this website including website layout and design, images, text, programs and all other information are the property of American Scientific Publishers and is protected by copyright and other intellectual property laws. All materials including names, trademarks, and logos posted on the website are subject to copyrights owned by American Scientific Publishers. The contents of any American Scientific Publishers website pages should not be reproduced, copied, displayed, distributed, modified, published, printed, created derivative works from or sell or license all or any part of the contents, products or services obtained from this website in any form or medium to anyone, created or compiled as database, collection or compilation, stored in a retrieval system, transmitted, translated into any foreign languages or otherwise used in form whatsoever by any means electronic, mechanical, photocopying, scanning, recording, or otherwise without the written permission of the American Scientific Publishers. All registered trade marks, names, logo and similar related materials used on any American Scientific Publishers website pages, even when not marked as such, are not to be considered unprotected by law. All rights are reserved.

## Contents

<b>1. NANOPHYSICS .....</b>	<b>5</b>
<b>1.1. Phenomena and Effects .....</b>	<b>5</b>
<b>1.2. Properties of Materials and Structures .....</b>	
<b>2. NANOCHEMISTRY .....</b>	<b>14</b>
<b>2.1. Inorganic Materials .....</b>	<b>14</b>
<b>2.2. Organic Materials .....</b>	<b>14</b>
<b>3. NANOBIOLOGY .....</b>	
<b>4. NANOTECHNOLOGY .....</b>	<b>16</b>
<b>4.1. Materials and Structures .....</b>	<b>16</b>
<b>4.2. Obtaining Technologies .....</b>	<b>17</b>
<b>4.3. Processing Technology .....</b>	<b>28</b>
<b>4.4. Nanobiotechnology .....</b>	<b>29</b>
<b>5. NANOENGINEERING .....</b>	<b>31</b>
<b>5.1. Devices and Sensors .....</b>	<b>31</b>
<b>6. NANOMEDICINE .....</b>	<b>33</b>
<b>6.1. Medical Physics .....</b>	<b>33</b>
<b>6.2. Medical Chemistry .....</b>	<b>35</b>
<b>Authors Search .....</b>	<b>40</b>

## **List of Publications Reflected in the Present Issue**

### **GEORGIA**

1. Annals of Agrarian Science. – 2020. – vol. 18. – #1; – 2021. – vol. 19. – #3.
2. Bulletin of the Georgian National Academy of Sciences. – 2021. – vol. 15. – #1, #2, #3.
3. Ceramics and Advanced Technologies. – 2020. – vol. 22. – #2(44).
4. Collection of Scientific Works of Tbilisi State Medical University (TSMU). – 2020. – vol. 54.
5. Georgian Medical News (GMN). – 2021. – #2(311), #3(312).
6. 6th International Conference – Nanotechnology. Book of Abstracts. – Tbilisi. – 4-7 October. – 2021. – pp. 116.

### **ARMENIA**

1. Armenian Journal of Physics. – 2021. – vol. 14. – #1, #3.
2. Proceedings of NAS RA. Physics. – 2020. – vol. 55. – #4; – 2021. – vol. 56. – #1, #2, #3.

### **AZERBAIJAN**

1. Azerbaijan Chemical Journals. – 2020. – #4; – 2021. – #1, #2, #3.

## 1. NANOPHYSICS

### 1.2. Properties of Materials and Structures

**3.1.2.1. The effect of high-amplitude deformation and high-frequency magnetic field exposure on the elastic/inelastic properties of PTFE-based hybrid nanocomposite filled with Fe cluster-doped CNTs.** /E. Kutelia, G. Darsavelidze, T. Dzigrashvili, D. Gventsadze, I. Kurashvili, O. Tsurtsunia, L. Gventsadze, L. Nadaraia, L. Rukhadze, T. Kukava, S. Bakhtiyarov/. Bulletin of the Georgian National Academy of Sciences. – 2021. – v. 15. – #1. – pp. 38-44. – eng.; abs.: eng., geo.

The influence of high-amplitude torsional deformation ( $\epsilon \sim 10^{-1} \div 10^{-2}$ ) and high-frequency (2.4 GHz) magnetic field treatment on elastic/inelastic properties of PTFE-based new hybrid nanocomposites modified with a two-component filler (2.5 wt% Fe-cluster-doped CNT nanopowder + 5wt% chalcopryrite micropowder) was studied using low-frequency amplitude-independent (AIF) and amplitude-dependent (ADIF) internal friction measurements. The behavior of elastic/inelastic properties of the new trial PTFE-based hybrid nanocomposite modified by a two-component filler (Fe-cluster-doped CNTs + chalcopryrite micro-particles) was investigated in dependence on highamplitude torsional deformation ( $\epsilon \sim 10^{-1} \div 10^{-2}$ ) and post-deformation high-frequency (2.4 GHz) magnetic field exposure and additional thermal treatment, using AIF and ADIF measurements. It is shown that self-healing of micro/nano-cracks nucleated in the deformed samples of the nanocomposite may be properly performed via their exposure to high-frequency magnetic field and the additional annealing at 200°C that leads to the recovery of the values of microplastic deformation beginning critical amplitude ( $\epsilon_c$ ) to the values even exceeding its initial magnitude by ~38%. Fig. 3, Tab. 2, Ref. 10.

**Keywords:** PTFE, hybrid nanocomposite, Fe-cluster-doped CNTs, chalcopryrite, magnetic field, internal friction

#### References:

1. Biswas S.K. (1992). Friction and wear of PTFE – Review. Wear. 158: 193-211.
2. Aly A.A., Zeidan El.B., Alshennawy A.A., El-Masry A.A., Wasel W.A. (2012). Friction and wear of polymer composites filled by nano-particles: a review. World Journal of Nano Science and Engineering. 2:32-39.
3. Jiabin Ye., David L. Burris and Ting Xie (2016). A review of transfer film and their role in ultra-low-wear sliding of polymers. Lubricants 4(4): 1-15.
4. Gandotra H., Mahajan S., Jandival S., Gupta S. (2018). Effect of fillers on tribological properties of PTFE: a review. International Journal of Scientific and Technical Advancements. 4 (1): 147-150.
5. Gventsadze D., Kutelia E., Nadaraia L., Padgurskas J., Gventsadze L., Tsurtsunia O. (2017). The tribological properties of PTFE modified with chalcopryrite. Proceedings of BALTRIB, 62-65, edited by J. Padgurskas. ISSN 2424-5089/e ISBN 978-609-449-093-4.
6. Kutelia E., Gventsadze D., Tsurtsunia O., Rukhadze L., Jalabadze N., Kukava T., Dzigrashvili T. (2018). Investigation of new antifrictional/frictional nanocomposites based on PTFE matrix filled with Fe-doped carbon nanoparticles. Advanced Materials Letters. 9(5): 320-325 www.iaamonline.org
7. Kutelia E., Darsavelidze G., Dzigrashvili T., Kukava T., Rukhadze L., Gventsadze L., Tsurtsunia O., Nadaraia L., Kurashvili I., Bakhtiyarov S. (2018). Internal Friction in PTFE-based nanocomposite materials filled with Fe cluster-doped CNTs. Georgian Engineering News. 87 (3): 5-13.
8. Shi Y., Feng X., Wang H., Lu X. (2007). Tribological and mechanical properties of PTFE composites filled with the combination of short carbon fiber and carbon nano-fiber. Key Engineering Materials. 334-335. 689-692.
9. Ahmed A.S., Ramanujan R.V. (2015). Curie temperature controlled self-healing magnet-polymer composites. Journal of Materials Research. 30 (7): 946-958. 10.
10. McCrum N.G. (1959). An internal friction study of polytetrafluoroethylene. Journal of Polymer Science. 34: 355-369.

**3.1.2.2. Nonmonotonic carrier dispersion in dimensionally quantized nanostructures with broken symmetries.** /H.S. Nikoghosyan, S.L. Harutyunyan, V.F. Manukyan, G.H. Nikoghosyan/. Proceedings of NAS RA. Physics. – 2020. – vol. 55. – #4. – pp. 537-549. – rus.; abs.: rus., arm., eng.

The features of the energy spectrum and carrier motion in one-dimensional asymmetric semiconductor nanosystems are considered. Asymmetric structures with an additional dip in the potential profile in quantum wells make it possible to vary the positions of size-quantized levels. And the additional influence of an external magnetic field leads to non-monotonic dispersion and a transverse local carrier drift. Such violations of fundamental symmetries will entail corresponding changes in the nature of electronic transitions. The dynamic properties of the electronic system of the nanostructure with the asymmetry of the quantum profile are analyzed, depending on the strength of the external magnetic field. Fig. 4, Ref. 12.

**Keywords:** energy spectrum, one-dimensional asymmetric semiconductor nanosystems, fundamental symmetries, quantum profile, external magnetic field

**References:**

1. В.В. Филиппов, А.А. Заворотный, Е.Н. Бормонтов. Конденсированные среды и межфазные границы. 13, 363 (2011).
2. М.А. Ремнев, И.Ю. Катеев, В.Ф. Елесин. ФТП. 44 1068 (2010).
3. В.П. Драгунов, И.Г. Неизвестный, В.А. Гридчин. Основы наноэлектроники. Москва. Физматкнига. 2006.
4. H.S. Nikoghosyan, S.L. Harutyunyan, V.F. Manukyan, G.H. Nikoghosyan. Physica B. 575. 411710 (2019).
5. H.S. Nikoghosyan, S.L. Harutyunyan, V.F. Manukyan, G.H. Nikoghosyan. Journal of Contemporary Physics. 54, 345, (2019).
6. Ю.А. Артамонов, А.А. Горбачевич, Ю.В. Копаев. ЖЭТФ. 101, 557 (1992).
7. А.А. Горбачевич. ЖЭТФ. 95, 1467 (1989).
8. А.А. Горбачевич, В.В. Копаев, Ю.В. Копаев. Письма в ЖЭТФ. 57, 565 (1993).
9. Л.Д. Ландау, Е.М. Лифшиц. Квантовая механика. Нерелятивистская теория. Москва. Наука. 1989.
10. А. Анималу. Квантовая теория кристаллических твердых тел. Москва. Мир. 1981.
11. В.Я. Демиховский, Г.А. Вугальтер. Физика квантовых низкоразмерных структур. Москва. Логос. 2000.
12. О.В. Кибис. Письма в ЖЭТФ. 66, 551 (1997).

**3.1.2.3. Physical adsorption of single-stranded DNA on carbon nanotube.** /D.G. Khechoyan, V.F. Morozov/. Proceedings of NAS RA. Physics. – 2021. – vol. 56. – #1. – pp. 100-105. – rus.; abs.: rus., arm., eng.

The results of simulation modeling of single-stranded DNA adsorption on the surface of a single-walled carbon nanotube performed by the molecular dynamics method using the GROMACS software package are presented. The dependence of the timescale and degree of DNA adsorption was studied in dependence on the chirality of the carbon nanotube and the DNA sequence. Fig. 4, Ref. 16.

**Keywords:** simulation, single-stranded DNA, single-walled carbon nanotube, GROMACS software package, DNA adsorption

**References:**

1. J. Hu, J. Shi, S. Li, Y. Qin, Z.X. Guo, Y. Song, D. Zhu. Chem. Phys. Lett. 401, 352 (2005).
2. V.N. Khabashesku, J.L. Margrave, E.V. Barrera. Diamond Relat. Mater. 14, 859 (2005).
3. D.A. Heller, et al. Science 311, 508 (2006).
4. C.L. Chen, et al. Nanotechnology 21, 095504 (2010).
5. X. Dong, D. Fu, Y. Xu, J. Wei, Y. Shi, P. Chen, L.J. Li. J. Phys. Chem. C 112, 9891 (2008).
6. M. Zheng, et al. Nat. Mater. 2, 338 (2003).
7. W. Martin, W. Zhu, G. Krilov. J. Phys. Chem. B, 112, 16076 (2008).
8. D. Pramanik, P.K. Maiti. ACS Appl. Mater. Interfaces, 9, 35287 (2017).
9. W. Cheung, et al. J. Mater. Chem., 19, 6465 (2009).

10. P.G. He, M. Bayachou. *Langmuir*, 21, 6086 (2005).
11. R.R. Johnson, A.T.C. Johnson, M.L. Klein. *Nano Lett.*, 8, 69 (2008).
12. D. Van Der Spoel, et al. *J. Comput. Chem.* 26, 1701 (2005).
13. K. Lindorff-Larsen, et al. *Proteins*. 78, 1950 (2010).
14. W.L. Jorgensen, D.S. Maxwell, J. Tirado-Rives. *J. Am. Chem. Soc.* 118, 11225 (1996).
15. H.R. Drew, et al. *Proc. Natl. Acad. Sci. USA*. 78, 2179 (1981).
16. F. Albertorio, M.E. Hughes, J.A. Golovchenko, D. Branton. *Nanotechnology*. 20 395101 (2009).

**3.1.2.4. Geometric features and numerical analysis of InAsSbP composition micro- and nano-structures shape transformation at nucleation from liquid phase.** /K.M. Gambaryan, V.M. Aroutiounian/. *Proceedings of NAS RA. Physics.* – 2021. – vol. 56. – #2. – pp. 208-217. – rus.; abs.: rus., eng.

Results of the characterization and numerical analysis of InAsSbP composition strain-induced micro- and nanostructures shape transition are presented. Nucleation is performed from In–As–Sb–P quaternary composition liquid phase in Stranski–Krastanow growth mode. Geometric features and the shape transformation chronology of truncated pyramidal islands, lens-shape and pyramidal quantum dots (QDs) are under consideration, which opens up new possibilities at nanoscale engineering and nanoarchitecture of several types of nanostructures. High-resolution scanning electron (HR-SEM) and transmission electron (TEM) microscopes are used for micro- and nanostructures characterization. We show that as the islands volume decreases, the following succession of shape transitions are detected: truncated pyramid, {111} faceted pyramid, {111} and partially {105} faceted pyramid, completely unfaceted “prepyramid”, which gradually evolve to hemisphere and then again to pyramidal QD but with higher facet indexes. Critical sizes of islands shape transformation from “pre-pyramid” to hemisphere (500–550 nm) and then from lens-shape again to pyramidal QDs (5–7 nm) are experimentally detected and theoretically evaluated. It is shown that theoretically calculated values coincide with experimentally obtained data. Fig. 2, Tab. 1, Ref. 21.

**Keywords:** numerical analysis, InAsSbP, Stranski–Krastanow growth mode, pyramidal quantum dots (QDs), truncated pyramid, faceted pyramid, pyramidal QDs

#### References:

1. D. Bimberg, M. Grundmann, N.N. Ledentsov. *Quantum Dot Heterostructures*. New York: Wiley. 1998.
2. P. Bhattacharya, X.H. Su, S. Chakrabarti, et al. *Appl. Phys. Lett.* 86, 191106 (2005).
3. A. Rogalski. *Acta Phys. Pol. A*. 116, 389 (2009).
4. V.M. Aroutiounian, S.G. Petrosian, A. Khachatryan, K. Touryan. *J. Appl. Phys.* 89, 2268 (2001).
5. A.M. Rudin, L.J. Guo, et al. *Appl. Phys. Lett.* 73, 3429 (1998).
6. O. Marquardt, T. Hickel, J. Neugebauer, K.M. Gambaryan, V.M. Aroutiounian. *J. Appl. Phys.* 110, 043708 (2011).
7. K.M. Gambaryan. *Nanoscale Res. Lett.* 5, 587 (2010).
8. V.M. Aroutiounian, K.M. Gambaryan, P.G. Soukiassian. *Surface Science*. 604, 1127 (2010).
9. H. Ishikuro, T. Hiramoto. *Appl. Phys. Lett.* 71, 3691 (1997).
10. K.M. Gambaryan, V.M. Aroutiounian, V.G. Harutyunyan. *Infrared Phys. & Tech.* 54, 114 (2011).
11. I. Stranski, L. Krastanow. *Math.-Naturwiss.* 146, 797 (1938).
12. K.M. Gambaryan, V.M. Aroutiounian, T. Boeck, M. Schulze. *Phys. Status Solidi C*, 6, 1456 (2009).
13. I. Daruka, J. Tersoff, A.-L. Barabasi. *Phys. Rev. Lett.* 82, 2753 (1999).
14. J. Tersoff, F.K. LeGoues. *Phys. Rev. Lett.* 72, 3570 (1994).
15. M. Zinke-Allmang, L.C. Feldman, M.H. Grabow. *Surf. Sci. Rep.* 16, 377 (1992).
16. F.M. Ross, J. Tersoff, R.M. Tromp. *Phys. Rev. Lett.* 80, 984 (1998).
17. J. Tersoff, B.J. Spencer, A. Rastelli, H. von Kanel. *Phys. Rev. Lett.* 89, 196104 (2002).



18. N. Liu, J. Tersoff, O. Baklenov, A.L. Holmes, Jr., C.K. Shih. Phys. Rev. Lett. 84, 334 (2000).
19. K.M. Gambaryan, V.M. Aroutiounian, T. Boeck, et al. J. Phys. D: Appl. Phys. 41, 162004 (2008).
20. J. Tersoff, R.M. Tromp. Phys. Rev. Lett. 70, 2782 (1993).
21. K.L. Safonov, V.G. Dubrovskii, N.V. Sibirev, Yu.V. Trushin. Technical Physics Letters 33, 490 (2007).

**3.1.2.5. Monovalent and divalent impurity states in a semiconductor nanoplatelets.** /V.A. Harutyunyan, H.A. Sarkisyan/. Proceedings of NAS RA. Physics. – 2021. – vol. 56. – #3. – pp. 348-355. – rus.; abs.: rus., arm., eng.

Within the framework of the variational method, hydrogen-like impurity states in the semiconductor nanoplatelets with the shape of a rectangular parallelepiped of a small thickness are investigated. Due to the small thickness of the nanostructure, it is shown that, the impurity can be considered two-dimensional. In the case of a divalent impurity, the electron-electron interaction is also considered two-dimensional and taken into account as a perturbation. By the analogy with the theory of the helium atom, the electron-electron interaction energy is determined for the para-state. Fig. 4, Ref. 16.

**Keywords:** variational method, semiconductor nanoplatelets, nanostructure, divalent impurity, helium atom, electron-electron interaction

## References:

1. A.O. Bochkarev, M.A. Grekov. Physical Mesomechanics. 22, 209 (2019).
2. B. Karami, S. Karami. Advances in nano research. 7, 51 (2019).
3. E.V. Shornikova et al. Nano letters, 20, 1370 (2020).
4. B. Jin, H. Wang, M.L. Sushko, C. Jin, R. Tang. Nanoscale. 12, 19592 (2020).
5. D. Xiang, Y. Li, L. Wang, Y. Zhao, K. Wu. ACS Photonics. 8, 745 (2021).
6. D.A. Baghdasaryan, E.M. Kazaryan, H.A. Sarkisyan, K.D. Moiseev. Physica E. 90, 170 (2017).
7. Y. Yanget al. IEEE Transactions on Nanotechnology. 18, 220 (2019).
8. A.G. Vitukhnovsky et al. Chem. Phys. Letters. 619, 185 (2015).
9. J. Planelles.Theor. Chem. Acc. 136, 81 (2017).
10. Л.Г. Мардоян, Г.С. Погосян, А.Н. Сисакян, В.М. Тер-Антонян. Теоретическая и математическая физика. 66, 99 (1984).
11. Д.Б. Айрапетян, Э.М. Казарян, О.Х. Тевосян. Изв. НАН Армении. Физика. 49, 190 (2014).
12. О.Х. Тевосян. Изв. НАН Армении. Физика. 47, 427 (2012).
13. Р.Г. Погосян. Изв. НАН Армении. Физика. 49, 114 (2014).
14. A.K. Manaselyan, A.V. Ghazaryan, A.A. Kirakosyan. J. Contemp. Phys. 43, 211 (2008).
15. A.K. Manaselyan, A.V. Ghazaryan, A.A. Kirakosyan. J. Contemp. Phys. 45, 269 (2010).
16. T.A. Sargsian. J. Contemp. Phys. 54, 168 (2019).

**3.1.2.6. Electrical characteristics and photoresponse of the "carbon nanofilm on silicon" heterostructure.** /G. A. Dabaghyan, L.M. Matevosyan, K.E. Avjyan/. Proceedings of NAS RA. Physics. – 2021. – vol. 56. – #3. – pp. 374-383. – rus.; abs.: rus., eng.

The electrical characteristics and photoresponse of the “carbon nanofilm on silicon” heterostructure obtained by pulsed-laser deposition, where the thickness of the carbon nanofilm is selected from the condition of the maximum antireflection effect of the substrate, have been investigated. It was found that the obtained heterostructure is rectifying with a rectifying coefficient of 35 at 1 V. The direct current-voltage characteristic from 0.1 V to 0.35 V is in satisfactory agreement with the expression  $J = J_0 \exp(eU/\eta kT)$ . An increase in the voltage in the forward direction leads to the appearance of the space charge-limited currents ( $J = AU^2$ ). Linearization of the  $C^{-2} - U$  dependence indicates the sharpness of the impurity distribution in the space charge region. The mechanism of the photoresponse of the

heterostructure is similar to the photoresponse of anisotype heterostructures with the “window” effect. The longwavelength edge (1.1  $\mu\text{m}$ ) of the photosensitivity is determined by the silicon substrate, and absorption in the carbon nanofilm leads to an additional expansion of the photosensitivity region. The heterostructure has uniform photosensitivity at the level 0.8 of a relative photoresponse in the wavelength range of 0.55–1.1  $\mu\text{m}$ . The short-wavelength tail reaches up to 0.4  $\mu\text{m}$ . Fig. 8. Ref. 17.

**Keywords:** “carbon nanofilm on silicon” heterostructure, maximum antireflection effect of the substrate, space charge-limited currents, short-wavelength tail

#### References:

1. A.G. Milnes, D.L. Feucht. Heterojunctions and Metal-Semiconductor Junctions. New York. London: Academic press. 1972, 418 p.
2. L.S. Lunin, M.L. Lunina, O.V. Devitsky, I.A. Sysoev. Semiconductors. 51(3), 387 (2017).
3. D.S. Da Silva, A.D.S. Côrtes, M.H. Oliveira Jr., E.F. Motta, G.A. Viana, P.R. Mei, F.C. Marques. Journal of Applied Physics. 110, 043510 (2011).
4. R. Hauert. Diamond and Related Materials. 12, 583 (2003)
5. A.C. Ferrari. Surface and Coatings Technology. 180-181, 190 (2004)
6. K.E. Avjyan, L.A. Matevosyan, K.S. Ohanyan, L.G. Petrosyan. Instruments and Experimental Techniques. 59(1), 60 (2016).
7. Z.Q. Ma, B.X. Liu. Solar Energy Materials & Solar Cells. 69, 339 (2001)
8. S.V. Gaponov, A.A. Gudkov, E.V. Klunokov, M.D. Strikovskij. Sov. Electronnaja Promishlennost. 5-6, 110 (1981).
9. J.C. Miller. Laser Ablation – Principles and Applications, Berlin: Springer-Verlag. 1994
10. R. Eason. Pulsed Laser Deposition of Thin Films. John Wiley & Sons. 2006
11. A.G. Alexanian, N.S. Aramyan, K.E. Avjyan, A.M. Khachatryan, R.P. Grigoryan, A.S. Yeremyan. Technology of PLD for photodetector materials. in: Combinatorial and High-Throughput Discovery and Optimization of Catalysts and Materials. edited by R.A. Potirailo and W.F. Maier. CRC/Taylor & Francis. 2006.
12. S. Nakano, T. Matsuoka, S. Kiyama, et al. Jpn. J. Appl. Phys. 25(12), 1936 (1986)
13. A.D. Compaan, I. Matulionis, S. Nakade. Opt. Lasers Eng. 34(1), 15 (2000).
14. D. Bäuerle. Laser Processing and Chemistry. 3rd ed. Berlin: Springer. 2000.
15. Y. Miyajima, S.J. Henley, G. Adamopoulos, V. Stolojan, E. Garcia-Caurel, B. Drévillon, J.M. Shannon, S.R.P. Silva. Journal of Applied Phys. 105, 073521 (2009).
16. H. Yasuda, R. Matsuno, N. Koito, H. Hosoda, T. Tani, M. Naya. Appl. Phys Lett. 111, 231105 (2017). 17. B. Fan, K. Nose, D. Diao, T. Yoshida. Appl. Surface Science. 273, 816 (2013).

**3.1.2.7. Investigation of properties of graphene quantum dots and carbon nanotubes synthesized in a colloid solution.** /N.B. Margaryan, N.E. Kokanyan, E.P. Kokanyan/. Proceedings of NAS RA. Physics. – 2021. – vol. 56. – #3. – pp. 392-398. – rus.; abs.: rus., eng.

In this paper, a simple and effective method for the synthesis of carbon nanotubes, graphene-based quantum dots is described. The topological properties of these nanostructures are studied by atomic force and scanning electron microscopes. The potential of quantum dots is investigated by the Kelvin probe method. To study the formed bonds and for a detailed structural analysis, Raman spectroscopy is performed. Other self-organized structures based on graphene are also disclosed using Raman spectroscopy. The effect of photon-phonon scattering on the Raman scattering spectrum is discussed. Fig. 4, Ref. 26.

**Keywords:** synthesis of carbon nanotubes, graphene-based quantum dots, Kelvin probe method, Raman spectroscopy, photon-phonon scattering

## References:

1. N.B. Margaryan, N.E. Kokanyan, E.P. Kokanyan. Journal of Saudi Chemical Society. 23, 13 (2019).
2. I. Levchenko, S. Xu, G. Teel, D. Mariotti, M.L.R. Walker, M. Keidar. Nat. Commun. 9, art. numb. 879 (2018).
3. J. Chen, C. Li, G. Shi. Phys. Chem. Lett. 4, 1244 (2013).
4. I. Levchenko, K. Bazaka, Y. Ding, et al. Appl. Phys. Rev. 5, 011104 (2018).
5. F. Perrozzi, S. Prezioso, L. Ottaviano. J. Phys.: Condens. Matter. 27, 013002 (2014).
6. L.A. Ponomarenko, F. Schedin, M.I. Katsnelson, et. al. Science. 320, 356 (2008).
7. K. Geim, K.S. Novoselov. Nat. Mater. 6, 183 (2007).
8. M.Y. Han, B. Ozyilmaz, Y. Zhang, P. Kim. Phys. Rev. Lett. 98, 206805 (2007).
9. D. Pan, J. Zhang, Z. Li, M. Wu. Adv. Mater. 22, 734 (2010).
10. W. Qian, T. Liu, F. Wei, H. Yuan. Carbon. 41. 1851 (2003).
11. M.S. Dresselhaus, G. Dresselhaus, R. Saito, A. Jorio. Phys. Rep. 409, 47 (2005).
12. J. Zhao, M. Shaygan, J. Eckert, M. Meyyappan, M.H. Rummeli. Nano Lett. 14, 306 (2014).
13. S. Haar, M. Bruna, J.X. Lian. J. Phys. Chem. Lett. 7, 2714 (2016).
14. A.S. Pavlova, E.A. Obratsova. J. Nanophotonics. 10(1), 012525 (2016).
15. J. Chen et al. FlatChem. 5, 25 (2017).
16. K. Safarova, A. Dvorak, R. Kubinek, M. Vujtek, A. Rek. Modern Research and Educational Topics in Microscopy. 513, Formatex. 2007.
17. N. Margaryan. J. Phys. Sci. Appl. 7(2), 46 (2017).
18. M. Massicotte et al. Nanotechnology. 24, 325601 (2013).
19. S. Costa, E. Borowiak-Palen, M. Kruszyńska, A. Bachmatiuk, R.J. Kaleńczuk. Materials Science Poland. 26(2), 433 (2008).
20. J. Wu, P. Wang, F. Wang, Y. Fang. Nanomaterials. 8, 864 (2018).
21. L. Bokobza, J. Zhang. eXPRESS Polymer Letters. 6(7), 601 (2012).
22. N. Lang, W. Kohn. Physical Review B. 3(4). 1215 (1971).
23. A. Nazarov et al. Phys. Status Solidi C. 10 (7–8). 1172 (2013).
24. R. Arenal, A.C.Y. Liu. Appl. Phys. Lett. 91. 211903 (2007).
25. J.L. Koenig. Analytical Chemistry. 65(1). 207 (1999).
26. A. Niilisk et al. Carbon. 98, 658 (2016).

**3.1.2.8. Physicomechanical properties of nanocomposites based on copolymers of ethylene with  $\alpha$ -olefins and clinoptilolite.** /N.T. Kakhramanov, I.V. Bayramova, V.S. Osipchik, A.D. Ismayilzade, S.R. Abdalova, I.A. Ismayilov, U.V. Namazli/. Azerbaijan Chemical Journal. – 2020. – #4. – pp. 22-27. – eng.; abs.: eng., az., rus.  
DOI: <https://doi.org/10.32737/0005-2531-2020-4-22-27>

The results of studying the effect of clinoptilolite concentration on the properties of nanocomposites based on of ethylene with butylene and of ethylene with hexene copolymer are presented. The effect of clinoptilolite particle size on ultimate tensile stress, elongation at break, flexural modulus, heat resistance, and melt flow index of composites was studied. It is shown that nanocomposites based on ethylene copolymers are characterized by higher values of physicomechanical properties. The additional use of ingredients such as alizarin and calcium stearate contributes to a significant improvement in the complex of properties of nanocomposites based on ethylene copolymers and clinoptilolite. Tab. 3, Ref. 12.

**Keywords:** ultimate tensile stress, elongation at break, heat resistance, flexural modulus, composite, copolymer

## References:

1. Berlin A.A., Volfson S.A., Oshman V.G. Principy sozdaniya kompozicionnyh materialov. M.: Himija. 1990. 240 s.

2. Ermakov S.N., Kerber M.L., Kravchenko T.P. Himicheskaja modifikacija i smeshenie polimerov pri reakcionnoj jekstruzii. *Plasticheskie massy*. 2007. No 10. S. 32–41.
3. Kakhramanov N.T., Ismailzade A.D., Arzumano N.B., Mammadli U.M., Martinova Q.S. Filled composites based on polyolefins and clinoptilolite. *American Scientific Journal*. 2016. No 4. V. 4. P. 60–65.
4. Stegno E.V., Lalajan V.M., Grachev A.V., Vladimirov L.V., Berlin A.A. Svojstva gibridnyh smesey polioksida bora i sopolimera jetilena s vinilacetatom. *Enciklopedicheskij slovar*. 2018. No 5. S. 1–7.
5. Kakhramanov N.T., Kurbanova R.V., Koseva N.S., Gahramanly Ju.N., Mamedli U.M. Gibridnye nanokompozity na osnove polipropilena i klinoptilolita. *Plast. massy*. 2019. No 3. V. 4. S. 32–34.
6. Kakhramanov N.T., Kurbanova R.V. Gibridnye nanokompozity na osnove funkcionalizirovannogo polijetilena vysokoj plotnosti i appretirovannogo bentonita. *Enciklopedicheskij spravocnik*, 2019. No 7. S. 17–25.
7. Kalinchev Je.L., Sakovceva M.B., Pavlova I.V., Kavokin E.I., Sakovich D.A. Effektivnyj podhod k sozdaniju sovremennyh polimernyh kompozicionnyh materialov. *Polimer. materialy*. 2008. No 3. S. 4–14.
8. Sirota A.G., Bugorkova V.S. Ob effektivnosti poljarnyh modificirujushhih dobavok k polietilenu. *Plasticheskie massy*. 2010. No 5. S. 6–11.
9. Kakhramanov N.T., Azizov A.G., Osipchik V.S. Nanostrukturirovannye kompozity polimerno materialovedenie. *Plasticheskie massy*. 2016. No 1. V. 2. S. 49–57.
10. Cherdynceva S.V., Belousov S.I., Krashennnikov S.V. Vlijanie vida organicheskogo modifikatora montmorillonita na fiziko-himicheskie svojstva nanokompozitov na osnove poliamida-6, poluchennyh smesheniem v rasplave. *Plast. massy*. 2013. No 5. S. 39–43.
11. Kakhramanov N.T., Gasimova G.Sh., Pesetsriy S.S., Kakhramanly Y.N., Gurbanova R.V., Hajiyeve R.Sh., Suleymanova E.I. Physical and mechanical properties of nanocomposites based on block copolymer propylene with ethylene and graphite. *Chemical problems*. 2019. No 1. S. 72–80.
12. Kakhramanov N.T., Bajramova I.V., Koseva N.S., Gadzhieva R.Sh. Fiziko-mehaniicheskie svojstva kompozitov na osnove vezuviana i sopolimera jetilena s butilenom. *Perspektivnye materialy*. 2019. No 3. S. 47–53.

**3.1.2.9. Copper-containing nanocomposites on the basis of isotactic polypropylene and butadiene-nitrile rubber.** /T.M. Gulieva/. *Azerbaijan Chemical Journal*. – 2021. – #3. – pp. 38-42. – eng.; abs.: eng., az., rus.

DOI: <https://doi.org/10.32737/0005-2531-2021-3-38-43>

The influence of additives of nanofillers containing nanoparticles of copper oxides stabilized by a polymer matrix of maleinized high-pressure polyethylene obtained by the mechano-chemical method on the structure and properties features of metal-containing nanocomposites based on isotactic polypropylene and butadienenitrile rubber by X-ray phase and differential thermal analyses is studied. The improvement of strength, deformation and rheological parameters, as well as thermal-oxidative stability of the obtained nanocomposites was revealed, that is probably due to the synergistic effect of interaction of copper-containing nanoparticles with maleic groups of maleinated high-pressure polyethylene. It is shown that nanocomposites based on isotactic polypropylene and butadiene-nitrile rubber can be processed both by pressing method and by injection molding and extrusion methods. The prospects of using a nanofiller containing NPs of copper oxide stabilized by a high-pressure polyethylene matrix obtained by the mechanochemical method as an additive to TPE based on PP/BNR is shown to contribute to the creation of a fine-crystalline structure of the composition, and therefore its properties are improved and thereby expand scope of the obtained nanocomposite. Fig. 4, Tab. 1, Ref. 18.

**Keywords:** isotactic polypropylene, butadiene-nitrile rubber, metal-containing nanocomposites, copper oxide nanoparticles, maleinized high-pressure polyethylene, thermal properties, XRD, SEM analyses

**References:**

1. Polymer blends / Ed. D. Paul and S. Newman. M.: Mir, 1981.V. 2. P. 312–338.

2. Abdou-Sabet S., Datta S. *Polymer Blends* / Ed. by D.R. Paul, C.B. Bucknall. New York; Chichester; Weinheim; Brisbane; Singapore; Toronto: Wiley, 2000. 1224 p.
3. Holden G. *Elastomers, thermoplastic*. Encyclopedia of polymer science and technology. 12 volumes. Ed. by H.F. Mark. V. 6. John Wiley & Sons, 2004. P. 63–88.
4. Ashpina O. *TEP Trends The Chemical J.* 2011. N 1. P. 58–61.
5. Magerramov A.M., Kurbanova N.I., Bayramov M.N., Alimirzoyeva N.A., Ragimova S.K., Nabiyeva A.N. Osobennosti radiotermolyuminesentsii kompozitsiy polipropilena i etilenpropilendiyenovogo elastomera SKEPT-4044 s nanorazmernymi metallosoderzhashchimi napolnitelyami. *Fizika i khimiya obrabotki materialov*. 2020. № 3. S. 66–73.
6. Kakhramanov N.T., Guliyev A.D., Pesetskiy S.S. Dinamicheski vulkanizovannye nanokompozity na osnove random polipropilena, butadiyen-nit- ril'nogo kauchuka i kaolina. *Kompozity i nanostruktury*. 2019. № 4. S. 131–136.
7. Guseynova Z.N., Kakhramanov N.T., Mamedov B.A., Osipchik V.S., Mamedli U.M. Termoelastoplasty na osnove termoplastichnykh poliolefinov i butilkauchuka. *Perspektivnyye materialy*. 2018. № 7. S. 33–42.
8. Kurbanova N.I., Alimirzoyeva N.A., Kuliyeu A.M., Guseynova Z.N., Ishenko N.Ya. Metal-containing nanocomposites on the basis of isotactic polypropylene, *Inorganic Materials: Applied Research*. 2019. V. 10. No. 2. P. 411–415.
9. Perestoronina Z.A., Ableyev R.I., Baranets I.V., Kurlyand S.K. Vliyaniye polimernykh dobavok na usileniye mezhfaznogo vzaimodeystviya v smesevykh termoelastoplastakh Kauchuk i rezina. 2012. № 2. S. 13–16.
10. Zaikin, A.E., Bobrov G.B. Compatibilization of polypropylene and butadiene–acrylonitrile rubber using an organic peroxide and an oligoether acrylate. *Russian J. Applied Chemistry*. 2015. V. 88. No. 5. P. 800–807.
11. Wolfson S.I., Okhotina N.A., Nigmatullina A.I., Sabirov R.K., Kuznetsova O.A., Akhmerova L.Z. Elastic-hysteretic properties of dynamic thermoplastic elastomers modified by nanofiller. *Plast. masses*. 2012. No 4. P. 42–45.
12. Mikhaylin Yu.A. Polymer nanocomposite materials. *Polymer materials*. 2009. No. 7. P. 10–13.
13. Pomogailo A.D., Rosenberg A.S., Uflyand I.E. *Metal nanoparticles in polymers*. Moscow: Chemistry, 2000. 672 p.
14. Gubin S.P., Yurkov G.Yu., Kosobudsky I.D. Nanomaterials based on metal-containing nanoparticles in polyethylene and other carbon-chain polymers. *International. J. Materials and Product Technology*. 2005. V. 23. N 1-2. P. 2–25.
15. Guliyeva T.M., Kurbanova N.I. Obtaining and study of the structure and properties of metalcontaining nanoparticles in the matrix of maleinized polyethylene. *Gənc tədqiqatçı*. 2019. V cild. № 2. P. 93–98.
16. Pomogailo A.D. Molecular polymer-polymer composition. *Synthetic aspects Advances in chemistry*. 2002. V. 71. № 1. P. 5–38.
17. Kuleznev V.N. *Smesi i splavy polimerov. Konspekt lektsiy*. SPb.: Nauchnyye osnovy tekhnologii. 2013. 216 s.
18. Kakhramanly Yu.N. *Nesovmestimyye polimernyye smesi i kompozitsionnyye materialy na ikh osnove*. Baku: Elm, 2013. 152 s.

**3.1.2.10. Copper-containing nanocomposites based on isotactic polypropylene and high pressure polyethylene.** /S.K. Ragimova/. *Azerbaijan Chemical Journal*. – 2021. – #3. – pp. 49-53. – eng.; abs.: eng., az., rus.

DOI: <https://doi.org/10.32737/0005-2531-2021-3-49-53>

The effect of nanofilles additives containing copper oxide nanoparticles stabilized by a polymer matrix of high-pressure polyethylene obtained by the mechanochemical method on features of the structure and properties of metal-containing nanocomposites based on isotactic polypropylene and high-pressure polyethylene was studied using differential thermal (DTA) and X-ray phase (XRD) analyzes. The improvement of strength, deformation and rheological parameters, as well as thermal-oxidative stability of the obtained nanocomposites was revealed, that apparently, is associated with the synergistic effect of

interfacial interaction of copper-containing nanoparticles in the PE matrix with the components of the PP/PE polymer composition. Fig. 2, Tab. 2, Ref. 19.

**Keywords:** isotactic polypropylene, high pressure polyethylene, copper oxide nanoparticles, thermal properties, DTA and XRD analysis

#### References:

1. Ermakov S.N., Kravchenko T.P. Polymer compatibility. Thermodynamic and chemical aspects. Plast. Mass. 2012. No. 4. P. 32–38.
2. Novokshonov V.V., Musin I.N., Kimbelblat V.I. Dependence of the properties of PP/EPR mixtures on the composition and characteristics of polymers. Plast. mass. 2009. No 5. P. 7–10.
3. Kuchmenova L.Kh., Slonov A.L., Zhansitov A.A., Shelgaev V.N., Khashirova S.Yu., Mikitaev A.K. The study of the thermal properties of polymerpolymer compositions based on PP. Plast. mass. 2014. No 7–8. P. 7–9.
4. Nigmatulina A.I., Volfson S.I., Okhotina N.A., Shaldybina M.S. Svoystva dinamicheskikh termoelastoplastov, soderzhashchiy modifitsirovanny polipropilen i sloisty napolnitel. Vestn. Kazan. tekhnol. un-ta. 2010. № 9. S. 329–331.
5. Perestoronina Z.A., Ableyev R.I., Baranets I.V., Kurlyand S.K. Vliyaniye polimernykh dobavok na usileniye mezhfaznogo vzaimodeystviya v smesevykh termoelastoplastakh. Kauchuk i rezina. 2012. № 2. S. 13–16.
6. Zaikin A.Ye., Bobrov G.B. Maslostoykiy termoelastoplast na osnove smesi polipropilena i sopolimeraetilena s vinilatsetatom, vulkanizirovanny 53 po reaktsii gidrosilirovaniya. Vest. Kazan. Tekhnol. un-ta. 2013. T. 16. № 2. S. 105–108.
7. Kakhramanov N.T., Guliyev A.D., Pesetskiy S.S. Dinamicheski vulkanizovannyye nanokompozity na osnove random poliproplena, butadiyen-nitril'nogo kauchuka i kaolina. Kompozity i nanostruktury. 2019. № 4. S. 131–136.
8. Sevastyanov D.V., Doriomedov M.S., Daskovsky M.I., Skripachev S.Yu. Self-reinforced polymer composites classification, preparation, mechanical properties and applications (review). Electronic scientific journal "VIAM Works". 2017. No 4. P. 104–118.
9. Mikhaylin Yu.A. Polymer nanocomposite materials. Polymer materials. 2009. No 7. P. 10–13.
10. Tretyakov A.O. Polymer nanocomposites are materials of the 21st century. Equipment and tools for professionals. 2003. No 2 (37). P. 18–20.
11. Foster L. Nanotechnology. Science, innovation and opportunity. M.: Technosphere. 2008. 352 p.
12. Pomogailo A.D., Rosenberg A.S., Uflyand I. E. Metal nanoparticles in polymers. M.: Chemistry. 2000. 672 p.
13. Gubin S.P., Yurkov G.Yu., Kosobudsky I.D. Nanomaterials based on metal-containing nanoparticles in polyethylene and other carbon-chain polymers. Int. J. Materials and Product Technology. 2005. V. 23. № 1–2. P. 2–25.
14. Kurbanova N.I., Kuliyeu A.M., Alimirzoeva N.A., Aliyev A.T., Ishenko N.Ya., Nurullayeva D.R. Preparation of copper-containing nanoparticles in polyethylene matrix without use of solvents. Science and Technology of Polymers and Advanced Materials: Applied Research Methods. Editor(s): O.V. Mukbaniani, T.N. Tatrishvili, M.J.M. Abadie to be published by Apple Academic Press. 2019. P. 57–65.
15. Ragimova S.K. Obtaining the metal-containing nanoparticles in polyethylene matrix by mechanochemical method and study of their properties. Azerb. Chem. J. 2020. No 2. P. 20–25.
17. Tekhnicheskiye svoystva polimernykh materialov: Uchebno-spravochnoye posobiye. Pod obshchey red. prof. V.K. Kryzhanovskogo. SPb: Professiya. 2007. 240 s.
18. Pomogailo A.D. Molecular polymer-polymer compositions. Synthetic aspects. Advances in chemistry. 2002. V. 71. № 1. P. 5–38.
19. Kuleznev V.N. Smesi i splavy polimerov. Konspekt lektsiy. SPb.: Nauchnyye osnovy iologii. 2013. 216 s.
20. Kakhramanly Yu.N. Nesovmestimyye polimernyye smesi i kompozitsionnyye materialy na ikh osnove. Baku: Elm, 2013. 152 s.

## 2. NANO CHEMISTRY

### 2.1. Inorganic Materials

**3.2.1.1. Microwave synthesis of ZnO/Ag nanocomposite.** /A.A. Sargsyan, V.V. Baghramyan, N.B. Knyazyan, R.K. Hovsepyan, N.R. Aghamalyan, G.R. Badalyan/. Proceedings of NAS RA. Physics. – 2020. – vol. 55. – #4. – pp. 559-565. – rus.; abs.: rus., arm., eng.

A microwave (MW) method has been developed for the production of ZnO/Ag nanocomposites using chemical precipitation and decomposition of thermally unstable compounds. Chemical co-precipitation is a simple and effective method compared to other methods for producing ZnO/Ag nanocomposites. The characteristics of the synthesized product were determined by differential thermal analysis (DTA), X-ray phase analysis (XRD) and scanning electron microscopy (SEM). The studies show the effectiveness of MW processing for the preparation of ZnO/Ag nanocomposites. Fig. 4, Ref. 15.

**Keywords:** microwave (MW) method, ZnO/Ag nanocomposites, chemical precipitation and decomposition, differential thermal analysis (DTA), X-ray phase analysis (XRD), scanning electron microscopy (SEM)

#### References:

1. Z. Liu, W. Hou, P. Pavaskar, M. Aykol, S.B. Cronin. Nano Letters. 11, 1111 (2011).
2. A.J. Esswein, D.G. Nocera. Chem. Rev. 107, 4022 (2007).
3. М.В. Евстафьева, А.Н. Редькин, Е.Е. Якимов. Нано- и Микросистемная техника. 18, 729 (2016).
4. М.В. Рыжова, А.Н. Редькин, Е.Е. Якимов. Межд. научно-техн. конф. «Технологии микро- и нанoeлектро-ники в микро- и наносистемной технике». Сб. материалов конференции. с. 231 (2016).
5. Е.И. Бурыйлин, А.Г. Веселов, А.С. Джумалиев, О.А. Кирясова, Т.А. Пушкарева, С.Л. Рябушкин. ЖТФ. 77(5), 130 (2007).
6. Н.А. Шабанова. Химия и технология нанодисперстных оксидов. Москва. Академкнига. 2007. с. 565.
7. V. Briois, C. Giorgetti. J. Sol–Gel Sci. Techn. 39, 25 (2006).
8. A. Aimable, M.T. Buscaglia, V. Buscaglia, P. Bowen. J. European Ceramic Society. 30, 591 (2010).
9. Y. Xia, Y. Xiong, B. Lim, S.E. Skrabalak. Angew. Chem. Int. Ed. 23, 60 (2009).
10. Р.К. Овсепян, Н.Р. Агамалян, Е.А. Кафадарян, Г.Г. Мнацаканян, А.А. Аракелян, С.И. Петросян, Г.Р. Бадалян. Изв. НАН Армении. Физика. 53, 477 (2018).
11. C. Meng, L. Ying, J.T. Han, J.Y. Zhang, Z.Y. Li, D.L. Qian. J. Fudan University. 45, 34 (2006).
12. E.T. Thostenson, T.W. Chou. Composites Part A: Applied Science and Manufacturing. 30, 1055 (1999).
13. Н. Brittany. Microwave Synthesis Chemistry at the Speed of Light. CEM Publishing. USA. 2002.
14. Д.Л. Рахманкулов, И.Х. Бикбулатов, Н.С. Улаев, С.Ю. Шавшукова. Микроволновое излучение и интенсификация химических процессов. Химия. Москва. 2003.
15. Н.С. Николаева, В.В. Иванов, А.А. Шубин. J. Siberian Federal University Chemistry. 2, 153 (2010).

### 2.2. Organic Materials

**3.2.2.1. Photochemical degradation of phenol with the participation of TiO<sub>2</sub> nanoparticles and ethyl-3,3,5,5-tetraciano-2-hydroxide-2-metil-4,6-diphenyl cyclohexane carboxylate.** /E.M. Gadirova/. Azerbaijan Chemical Journal. – 2021. – #2. – pp. 101-105. – eng.; abs.: eng., az., rus.

DOI: <https://doi.org/10.32737/0005-2531-2021-2-101-105>

The photochemical decomposition of phenol with the participation of TiO<sub>2</sub> nano-particles and ethyl-3,3,5,5-tetraciano-2-hydroxide-2-metil-4,6-diphenyl cyclohexane carboxylate by UV spectroscopy was studied for the first time. It has been shown, that UV irradiation of this mixture during 1 hour brings to 52% decomposition of phenol. Fig. 5, Ref. 12.

**Keywords:** UV decomposition, photocatalysis, ethyl-3,3,5,5-tetracyano-2-hydroxy-2-methyl-4,6-diphenyl cyclohexane carboxylate, phenol, waste water solution

**References:**

1. Gümüş D., Akbal F. Photocatalytic degradation of textile dye and wastewater. *Water, Air, and Soil Pollution*. 2011. V. 216. No 1–4. P. 117–124.
2. De Luis A.M., Lombrana J.I., Menendez A., Sanz J. Analysis of the toxicity of phenol solutions treated with  $H_2O_2$ /UV and  $H_2O_2$ /Fe oxidative systems. *Industrial and Engineering Chemistry Research*. 2011. V. 50. No 4. P. 1928–1937.
3. Hu Xuebing, Yu Yun, Ren Shuang, Lin Na, Wang Yongqing, Zhou Jianer. Highly efficient removal of phenol from aqueous solutions using graphene oxide/ $Al_2O_3$  composite membrane. *J. Porous Materials*. 2018. V. 25. No 3. P. 719–726.
4. Santhosh C., Velmurugan V., Jacob G., Jeong S.K., Grace A.N., Bhatnagar A. Role of nanomaterials in water treatment applications: a review. *J. Chem. Eng.* 2016. 306. P. 1116–1137.
5. Wang F. Novel high performance magnetic activated carbon for phenol removal: equilibrium, kinetics and thermodynamics. *J. Porous Mater.* 2017. 24. P. 1–9.
6. Mohammadi S., Kargari A., Sanaeepur H., Abbassian K., Najafi A., Mofarrah E. Phenol removal from industrial wastewaters: a short review. *Desalin. Water Treat.* 2015. 53. P. 2215–2234.
7. Kurbanova M.M., Sadygova A.Z., Gadirova E.M. First Synthesis and Structure of Ethyl 3,3,5,5-Tetracyano-2-hydroxy-2-methyl-4,6-diphenylcyclohexane-1-carboxylate. *Russian J. Organ. Chem.* 2019. V. 55. No 3. P. 381–383.
8. Elmina M. Gadirova. Photochemical degradation of phenol in the presence of titanium dioxide nanoparticles. *Proceedings of universities. Appl. Chem. Biotechn.* 2019. V. 9. No 2. P. 176–182.
9. Neumann B., Bogdanoff P., Tributsch H., Sakthivel S., Kisch H. Electrochemical mass spectroscopic and surface photovoltage studies of catalytic water photooxidation by un-doped and carbon-doped titania. *J. Phys. Chem. B*. 2005. V. 109. No 35. P. 16579–16586.
10. Qiu X., Burda C. Chemically synthesized nitrogen-doped metal oxide nanoparticles. *Chem. Physics*. 2007. V. 339. No 1–3. P. 1–10.
11. Li Y., Cao W., Ran F., Zhang X. Photocatalytic degradation of methylene blue aqueous solution under visible light irradiation by using N-doped titanium dioxide. *Key Engineering Materials*. 2007. V. 336–338. P. 1972–1975.
12. Fujishima A., Rao T. N., Tryk D.A. Titanium dioxide photocatalysis. *J. Photochemistry and Photobiology. C*. 2000. V. 1. No 1. P. 1–21.



## 4. NANOTECHNOLOGY

### 4.1. Materials and Structures

**3.4.1.1. Research of the possibility of nanostructuring functional materials by pre-recrystallization heat treatment.** /O. Dubovyy, A. Karpechenko, T. Makryha, M. Bobrov, A. Labartkava, A. Labartkava/. Bulletin of the Georgian National Academy of Sciences. – 2021. – v. 15. – #1. – pp. 45-51. – eng.; abs.: eng., geo.

The paper is devoted to the research of the possibility of nanostructuring of functional materials, steels and ceramic sprayed coatings by pre-recrystallization heat treatment. The effect of the size of coherent X-ray scattering regions, the number of nanostructured elements and the subgrains misorientation angle on the physical and mechanical properties of technically pure iron and steel were experimentally studied. The possibility of thermal stabilization of a 62% polygonization nanoscale substructure during pre-recrystallization heat treatment at 500°C after combined plastic deformation is shown. A combination of 30% dynamic and 30% static deformations makes it possible to use such treatment in industry. Fig. 2, Tab. 3, Ref. 16.

**Keywords:** physical and mechanical properties, thermal sprayed coatings, deformation of steels, pre-recrystallization heat treatment

#### References:

1. Zhdanov O.O. (2015). Zakonomirnosti vplyvu peredrekryalizacijnoi termichnoi obrobky na fizyko-mehanichni vlastyvoli deformovanyh stalej. Avtoref. dys. na zdobuttja nauk. stupenja kand. tehn. nauk.: spec. 05.02.01 – “Materialoznavstvo”. Herson (in Ukrainian).
2. Das D., Samanta A., Chattopadhyay P.P. (2006). Deformation behavior of bulk ultrafine grained copper prepared by sub-zero rolling and controlled recrystallization. *Materials & Manufacturing Processes*. 21 (7): 698-702.
3. Sun S.L., Huang Q.X., He W.W., Zhang M.G. (2014). Workability behavior of 9% Cr ferritic/martensitic steel. *Materials & Manufacturing Processes*. 29 (10): 1190-1196.
4. Jurkova O.I. (2011). Strukturnyj stan i mehanichni vlastyvoli plastychno deformovanogo zaliza. *Metaloznnavstvo ta obrobka metaliv*. 1: 3-9 (in Ukrainian).
5. Valiev R.Z. (2007) Ob”emnye nanostrukturnye metallicheskie materialy: poluchenie. struktura i svoistva. M. (in Russian).
6. Valiev R.Z., Aleksandrov I. V. (2000). Nanostrukturnye materialy, poluchennye intensivnoï plasticheskoï deformatsiei. M. (in Russian).
7. Jurkova O.I., Karpov R.V., Kljagin Je.O. (2010). Osoblyvosti formuvannja nanokrystalichnoi struktury v  $\alpha$ -zalizi pry deformacii tertjam. *Metaloznnavstvo ta obrobkametaliv*. 1:12-16 (in Ukrainian).
8. Alymov M.I., Averin S.I., Shustov V.S., Gordopolova L.V. (2013) Rol’ mekhanizmov plasticheskoï deformatsii pri vysokotemperaturnom spekanii chastits. Pis’ma o materialakh, 3 (4): 315-317 (in Russian).
9. Dubovyy O.M., Lebedeva N.Ju., Jankovec T.A. (2010). Vplyv peredrekryalizacijnoi termichnoi obrobky na fizyko-mehanichni vlastyvoli napylenyh pokryttiv ta deformovanyh metaliv ta splaviv. *Metaloznnavstvo ta obrobka metaliv*, 3: 7-10 (in Ukrainian).
10. Dubovyy O.M., Bondarenko O.V., Zhdanov O.O., Zhyzhko O.V., Bobrov M.M., Galkina T.S. (2010). Doslidzhennja mozhlyvostej pidvyshhennja fizyko-mehanichnyh vlastyvostej deformovanyh metaliv i splaviv termichnoju obrobkoju. *Obrobka materialiv u mashynobuduvanni*. Mykolaiv: Admiral Makarov National University of Shipbuilding: 69-79 (in Ukrainian).
11. Dubovyy O.M., Karpechenko A.A., Bobrov M.M., Zhdanov O.O., Makruha T.O., Nedelko Ju.Je. (2017). Formuvannja nanorozmirnoi poligonizacijnoi substruktury ta ii vplyv na fizyko-mehanichni vlastyvoli metaliv, stopiv i naporoshenyh pokryttiv. *Metallofizyka y novejshe tehnology*, 39(2): 209-243 (in Ukrainian).

12. Patent №95378 UA МПК (2009) S21D8/00, C22F 1/00. Sposib deformacijno-termichnoi obrobky metaliv ta splaviv. Dubovyy O.M., Jankovec T.A., Lebedjeva N.Ju., Kazymyrenko Ju.O., Zhdanov O.O., Bobrov M.M. Bulletin №14, July, 2011.
13. Dubovyy O.M., Karpechenko A.A., Bobrov M.M., Labartkava Al.V., Nedelko Ju.Je., Lyman O. O. (2019). Pidvyshhennja fizyko-mehanichnyh ta ekspluatacijnyh vlastyvostej elektrodugovyh ta plazmovykh pokryttiv formuvannjam termichno stabilnoi zdribnenoj i nanorozmirnoj substrukturny. Metalofiz. novitni tehnologii, 41(4): 461-480 (in Ukrainian).
14. Gorelik S.S., Dobatkin S.V., Kaputkina L.M. (2005). Rekristallizatsiia metallov i splavov. M. MISIS (in Russian).
15. Dubovyy O.M., Lju Shen, Makruha T.O. (2017). Vplyv kombinovanogo deformuvannja na termichnu stabilnist poligonizacijnoi substrukturny zaliza, nikelju j stalej 20; 45. Zbirnik naukovih prac NUK. Mykolaiv: Admiral Makarov National University of Shipbuilding. 1: 39-47 (in Ukrainian).
16. Dubovyy O.M., Makruha T.O. (2018). Vplyv vydu kombinovanogo deformuvannja na poligonizacijnu substrukturny zaliza ta stali U8. Zbirnik naukovih prac NUK. Mykolaiv: Admiral Makarov National University of Shipbuilding. 3-4(474): 66-74 (in Ukrainian).

**3.4.1.2. 6th International Conference - Nanotechnology. Book of Abstracts.** – Tbilisi. – 4-7 October. – 2021. – pp. 116.

Book of Abstracts contains more than 100 abstracts of papers submitted to the 6th International Conference “Nanotechnology”, 4–7 October 2021, Tbilisi, Georgia (GTUnano2021) organized by the Georgian Technical University (GTU). The GTUnano2021 is held in memory of Prof. Alex Gerasimov, the initiator of GTU’s regular series of nanoconferences in Georgia. The 6th conference participants represent universities, institutes, research centers, etc. leading in the field of nanotechnology and nanosciences from 20 countries (Armenia, Azerbaijan, Belarus, China, Czech Republic, Georgia, Germany, Hungary, India, Iraq, Japan, Kazakhstan, Mexico, Poland, Russia, Serbia, Spain, Turkey, Ukraine, and United States of America). Fig. 43. Tab. 3, Ref. 197.

**Keywords:** 6th International Conference “Nanotechnology”, abstracts of papers, GTU, nanotechnology and nanosciences, conference participants

## 4.2. Obtaining Technologies

**3.4.2.1. Receipt and technology of aloe floating tablets.** /M. Bakuradze, L. Bakuridze, E. Mosidze, D. Berashvili, I. Tsurtsunia/. Collection of Scientific Works of TSMU. – 2020. – v. 54. – pp. 29-31. – geo.; abs.: geo., eng.

In regenerative medicine, plants may be used as an alternative transplantation source. There is plenty of structural similarity between the vegetative and animal tissues and similarly to the human blood vessels, the plant fibers may be used as nutrient carriers. In tissue engineering, a decellularized plant may become a unique, multifunctional medical device, a plant matrix, the so-called scaffold. The performed studies resulted in the optimal composition of plant decellularization agents. It was determined that the duration of the process significantly depends on the morphological, anatomic and histological peculiarities of the plant. The biopharmaceutical studies proved that the absorption method is the most appropriate for plant decellularization for it shortens the process and is easier to perform than the perfusion. Employment of the 0.1% methylene blue made it evident that the decellularization technology does not affect the integrity and conductivity of the matrix fibers. On the grounds of the biopharmaceutical studies, we provided the plant matrix made up of the silver bionanoparticles. The properties of the silver bionanoparticles in the matrix were studied by the transmission electron microscope. The size of silver bionanoparticles is 1000 nm. and those are distributed across the matrix. We studied the silver separation dynamics from the nanodesigned

plant matrix and determined that 70% of silver was separated after 12h exposition of the matrix. Tab. 1, Fig. 4, Ref. 5.

**Keywords:** regenerative medicine, alternative transplantation source, decellularization agents, plant matrix, silver, bionanoparticles, aloe tablets, morphological, anatomic and histological peculiarities

### References:

1. Ржеусский С.Э. Валидация спектрофотометрической методики количественного определения наночастиц серебра в водных растворах // Вестник Фармации. 2019. N1. N. 21-25.
2. Vacanti J. Tissue engineering and regenerative medicine: from first principles to state of the art. *Journal of Pediatric Surgery*. 2010. 45(2):291–294.
3. Guyette J, et al. Bioengineering Human Myocardium on Native Extracellular Matrix. *Circulation Research*. 2016. 118(1):56–72.
4. Modulevsky D, Cuerrier C, Pelling A. Biocompatibility of Subcutaneously Implanted Plant-Derived Cellulose Biomaterials. *PLoS One*. 2016. 11(6):e0157894.
5. Gershlak J, et al. Crossing kingdoms: Using decellularized plants as perfusable tissue engineering scaffolds. *Biomaterials*. 2017. 125:13–22.

**3.4.2.2. Microwave synthesis, characterization and testing of acute toxicity of boron nitride nanoparticles by monitoring of behavioral and physiological parameters.** /A. Chirakadze, N. Mitagvaria, D. Jishiashvili, G. Petriashvili, N. Dvali, Z. Shiolashvili, K. Chubinidze, N. Makhatadze, A. Jishiashvili, Z. Buachidze, I. Khomeriki/. *Bulletin of the Georgian National Academy of Sciences*. - 2021. – v. 15. – #2. – pp. 120-126. – eng.; abs.: eng., geo.

Hexagonal boron nitride nanoparticles, nanosheets and nanotubes (BNNPs) are even more promising materials for biomedical application than carbon nanotubes (CNTs) and nanoparticles (CNPs) due to their negligible cytotoxicity. The reported research yielded in development and testing of two distinctive microwaves enhanced comparatively low-temperature methods of synthesis of the hexagonal boron nitride nanoparticles and nanosheets with reduced distortion of the crystal lattice, and an improved method of general toxicity testing of the developed nanomaterials utilizing continuous observation of behavioral effects in white rats in combination with blood oxygen saturation, systolic blood pressure and body temperature measurements in full agreement with the 4R principles of animal welfare in scientific research. The obtained results allow us to expect that the developed materials can be a good basis for developing highly effective modalities for anticancer (in combination with chemotherapy, hyperthermia and radiotherapy) and antiviral (in combination with chemotherapy and hyperthermia) treatment. Fig. 2, Ref. 14.

**Keywords:** cancer, low-temperature synthesis, microwave radiation, boron nitride, nanoparticles, turbostratic effect, behavioral testing

### References:

1. Chirakadze A., Jishiashvili D., Shiolashvili Z., Petriashvili G., Chubinidze K. (2020). Development and testing of combined nano-based liquids for treatment of cancer cells based on nanoparticles with a therapeutic Curie temperature and liquid crystals: Georgian Experience. *Abstracts of International Conferences & Meetings (AICM)*. Krispon Advancing Science. Edinburgh.
2. Mitagvaria N., Lazrshvili I., Devdariani M., Davlianidze L., Nebieridze M., Saginadze N., Kvachakidze I., Gumberidze L., Sikharulidze N. (2015). Hormesis – a basis for homeostasis in the presence of stressors. An example of hyperthermic stress. *Journal of Biological Physics and Chemistry*. 15: 187-193.
3. Chirakadze A., Jishiashvili D., Mitagvaria N., Lazrshvili I., Shiolashvili Z., Jishiashvili A., Makhatadze N., Buachidze Z., Khuskivadze N. (2019). Studies of the comparatively low-temperature synthesis and

- preliminary toxic characteristics of silver doped lanthanum manganite nanoparticles using conventional and microwave heating. *Proceedings of MTP: Modern Trends in Physics*. 47-51, Baku.
4. Siegel R., Miller K. D., Jemal A. (2020). Cancer statistics, 2020. *CA: A Cancer Journal for Clinicians*. 21590, <https://doi.org/10.3322/caac.21590>. Accessed on April. 2, 2021.
  5. Jemal A., Bray F., Center M.M., Ferlay J., Ward E., Forman D. (2011). Global cancer statistics. *CA, Cancer Journal for Clinicians*. A 1, 61: 69–90.
  6. Petousis P., Naeim A., Mosleh A., Hsu W., Geffen D. John B. (2018). Evaluating the impact of uncertainty on risk prediction: towards more robust prediction models. *AMIA Annu. Symp. Proc.* 1: 1461–1470.
  7. Quante A.S., Ming C., Rottmann M., Engel J., Boeck S., Heinemann V., Westphale C.B. (2016). Projections of cancer incidence and cancer-related deaths in Germany by 2020 and 2030. *Cancer Med.* 5, 9: 2649–2656.
  8. Merlo A., Mokkapati V.R., Pandit S. and Mijakovic I. (2018). Boron nitride nanomaterials: biocompatibility and bio-applications. *Biomater. Sci.* 6: 2298-2311.
  9. Huang C., Chen X., Ye W. Ye, Hu J., Xu C., Qiu X. (2013). Stable colloidal boron nitride nanosheet dispersion and its potential application in catalysis. *Journal of Materials Chemistry A*. 1: 2192-2197.
  10. Shen T., Liu S., Yan W., Wang J. (2019). Highly efficient preparation of hexagonal boron nitride by direct microwave heating for dye removal. *Journal of Materials Science*. 54, 3: 8852-8859.
  11. Silly M.G., Jaffrennou P., Barjon J., Rosencher E. (2007). Luminescence properties of hexagonal boron nitride: cathodoluminescence and photoluminescence spectroscopy measurements. *Physical Review B*. 75, 8; DOI: <https://doi.org/10.1103/PhysRevB.75.085205>. Accessed on April, 2, 2021.
  12. Sun C., Guo C., Ma X., Xu L., Qian Y. (2009). A facile route to prepare boron nitride hollow particles at 4500C. *Journal of Crystal Growth*. 31: 3682-3686.
  13. Alkoy S., Toy C., Gönül T., Tekin A. (1997). Crystallization behavior and characterization of turbostratic boron nitride. *Journal of the European Ceramic Society*. 17: 1415-1422.
  14. Turkez H., Arslan M.E., Soenmez E., Acjikyildiz M., Tatar A., Geyikoglu F. (2019). Synthesis, characterization and cytotoxicity of boron nitride nanoparticles: emphasis on toxicogenomics. *Cytotechnology*. 71: 351–361, <https://doi.org/10.1007/s10616-019-00292-8>, Accessed on April, 2, 2021.

**3.4.2.3. Obtaining of  $\beta$ -SiAlON nanocomposite with aluminothermal and nitrogen processes.** /Z. Kovziridze, N. Darakhvelidze, N. Nijaradze, G. Tabatadze, M. Mshvildadze, Z. Mestvirishvili, M. Balakhashvili, V. Kinkladze/. *Ceramics and Advanced Technologies*. – 2020. – vol. 22. – #2(44). pp. 21-36. – geo.; abs.: geo., eng.

To obtain a composite in SiAlON- $\text{Al}_2\text{O}_3$  system and to study its properties. Obtaining the composite by metallothermic and nitrogenation methods. In the present work, the composite containing SiAlON is obtained through alum-thermal process, by the reactive sintering method in nitrogen medium, from the mixture of aluminosilicate raw material (Prosyanaya kaolin and Polog refractory clay - Ukraine), nanopowder of aluminum oxide (German company "ALCOA"), and metallic silicium with small additives of glass perlite Aragac (Armenia). The advantage of this method is that the aluminosilicate raw material decomposes during the heat treatment process and the alum-thermal-nitration process takes place at the same time, making it easier to open AlN and  $\text{Al}_2\text{O}_3$  in the newly formed  $\beta$ - $\text{Si}_3\text{N}_4$  crystal lattice, which provides  $\beta$ -SiAlON generation at a relatively low temperature, 1250-13000°C. Corundum-SiAlON composite material is obtained by reactive sintering process at a temperature of 14500C. The corundum and sialon phases in the composite are confirmed by X-ray phase, spectral and electronmicroscopic analyzes. To obtain consolidated samples, the material obtained by reactive sintering was grounded in the attritor and hot pressed at 30 MPa and 1620° C and was kept at the final temperature - 7 min. The phase composition of the obtained samples remained unchanged after hot pressing, the density increased and the porosity dropped below 1%, accordingly the numerical values of the mechanical properties were increased:  $\sigma_{\text{press.}}$  - 1600 MPa;  $\sigma_{\text{bend.}}$  - 460 MPa; HV - 19.7 GPa. Obtained corundum-SiAlON composite with its physical-technical properties: porosity – 0-1%; density - 3.21 g/cm<sup>3</sup>;  $\sigma_{\text{press.}}$  - 1923 MPa;  $\sigma_{\text{bend.}}$  - 470

MPa; HV - 19.7 GPa, elasticity modulus - 22 GPa; dynamic hardness - 3214 N/mm<sup>2</sup>; chemical stability to sulfuric acid (density - 1.84 g/cm<sup>3</sup>) - 99.3%, to water - 99.8%. The obtained materials may be recommended in armor engineering, when measuring temperature in metals molten as protective coatings for the thermocouple, as well as in high-temperature furnace linings, as well as in clean processing operations as a metalworking cutting material. Fig. 5, Tab. 6, Ref. 29.

**Keywords:**  $\beta$ -SiAlON, corundum, reactive sintering, composite, properties, metalworking cutting material

## References:

- Ekstrom T., Kall P.O., Nygren M., Olsson P.O. Dense Single-Phase Beta-SiAlON Ceramics by Glass Encapsulated Hot Isostatic Pressing. J. of mat. Sci. 1989. V. 24. p. 1853-1862.
- Rosenflanz A., I-Wei-Chen. Phase Relationships and Stability of  $\alpha$ -SiAlON. J. Am. Ceram. Soc. 1999. V.82. №4. P. 25-28.
- Стрелов К. К. Теоретические основы технологии огнеупорных материалов: учеб. пособие для вузов / К. К. Стрелов, И. Д. Кашеев. 2-е изд. перераб. и доп. М. 1996. с. 608.
- Чухolina Л.Н. Способ получения порошка сиалона. <http://bd.patent.su/2378000> (18. 11.2012.)
- Zheng G, Zhao J., Gao Z., Cao Q. Cutting performance and wear mechanisms at Sialon-Si<sub>3</sub>N<sub>4</sub> graded nano-composite ceramic cutting tools/ The International Journal of advanced Manufacturing Technology. 2012 V.58 , I. 1-4. P. 19-28.
- Tressler R. E. Theory and Experiment in Corrosion of Advanced Ceramics//Corrosion of Advanced Ceramics/ NATO ASI Series E: Applied Sciences/Ed. K.G. Nickel. The Netherlands. 1994. N267. p. 3-22.
- Piekarczyk J., Lis J., Bialoskorski J. Elastic Properties, Hardness and Indentation Fracture Toughness of beta-Sialons/ Key Engineering Materials. 1990. V. 89-91. p. 542-546.
- ზ. კოვზირიძე, ნ. ნიჟარაძე, ნ. დარახველიძე, გ. ტაბატაძე, ზ. მესტვირიშვილი. გეოპოლიმერის ბაზაზე აზოტის გარემოში მიმდინარე კარბო- და ალუმინთერმული პროცესები. საქართველოს კერამიკოსთა ასოციაციის ჟურნალი „კერამიკა“ ტ. 16. N1(31). 2014. გვ. 32-36
- ზ. კოვზირიძე, ნ. ნიჟარაძე, ნ. დარახველიძე, გ. ტაბატაძე, თ. ჭეიშვილი, ზ. მესტვირიშვილი, მ. მშვილდაძე, ე. ნიკოლეიშვილი. ნიტროალუმინთერმული პროცესებით სიალონების მიღება. საქართველოს კერამიკოსთა ასოციაციის ჟურნალი „კერამიკა“ ტ. 16. N2(32). 2014. გვ. 23-31.
- Kovziridze Z., Nijaradze N., Tabatadze G., Cheishvili T., Mestvirishvili Z., Nikoleishvili E., Mshvildadze M., Daraxvelidze N. Obtaining of Nanocomposites in SiC-SiAlON and Al<sub>2</sub>O<sub>3</sub>-SiAlON System by Alumothermal Processes. Journal of Electronics Cooling and Thermal Control, 2014, 4, Published Online December 2014 in SciRes. Pp.1-13 <http://www.scirp.org/journal/jectc> USA.
- Kovziridze Z., Nijaradze N., Tabatadze G., Daraxvelidze N., Mestvirishvili Z. Obtaining of SiAlONs via alumothermal and nitrogen processes. 14th International Conference of European Ceramic Society. 21- 25 June. Toledo. Spain. Poster 2348. 2015.
- Kovziridze Z., Nijaradze N., Tabatadze G., Daraxvelidze N., Mestvirishvili Z. Smart Materials in the SiAlON-SiC-Al<sub>2</sub>O<sub>3</sub> System. Journal of Material Science and Engineering, International Conference and Expo on Ceramics. August 17-18. 2015. Chicago. USA.
- ზ. კოვზირიძე, ნ. ნიჟარაძე, ნ. დარახველიძე, გ. ტაბატაძე, ზ. მესტვირიშვილი. სიალონ-შემცველი კომპოზიტის მიღება ნიტროალუმინთერმული პროცესებით, რეაქციული შეცხოვისა და ცხელი დაწნეხის მეთოდით. საქართველოს კერამიკოსთა ასოციაციის ჟურნალი „კერამიკა“ ტ. 18. 1(35). 2016. გვ. 9-19.
- Kovziridze Z., Nijaradze N., Darakhvelidze N., Mestvirishvili Z. Smart Materials in the SiAlON-SiCAl<sub>2</sub>O<sub>3</sub>-TiB<sub>2</sub>-ZrB<sub>2</sub> System. 2nd Annual world Congress of Smart Materials (WCSM-2016) 4-6 March. 2016. Singapore
- Kovziridze Z., Nijaradze N., Darakhvelidze N., Tabatadze G., Mestvirishvili Z., Nikoleishvili E., Mshvildadze M., Preparation of Composites by Nitro Aluminothemic Processes, over  $\beta$ -SiAlON Matrix in the SiAlON-SiC-

- Al<sub>2</sub>O<sub>3</sub> System. Journal of Electronics Cooling and Thermal Control, Vol. 6 No. 2. Pub. Date: June 15. 2016. PP. 62-77.
16. Kovziridze Z., Nijaradze N., Darkhvelidze N., Tabatadze G., Mestvirishvili Z. Obtaining of nano composites via alum-thermal and nitrogen processes in the SiC-Si<sub>3</sub>N<sub>4</sub>-AlN-Al<sub>2</sub>O<sub>3</sub>-SiO<sub>2</sub>. System. 15th Conference & Exhibition of the European Ceramic Society. Ecers 2017. July 9-13. 2017 / Budapest. Hungary.
  17. ზ. კოვზირიძე, ნ. ნიჟარაძე, ნ. დარახველიძე, გ. ტაბატაძე, თ. ჭეიშვილი, ზ. მესტვირიშვილი, მ. მშვილდაძე. კომპოზიტის მიღება მეტალოთერმული და აზოტირების პროცესებით Si-SiC-Al გეოპოლიმერის სისტემებში. საქართველოს კერამიკოსთა ასოციაციის ჟურნალი „კერამიკა და მოწინავე ტექნოლოგიები“ ტ. 19. 2(38). 2017. გვ. 33-52.
  18. Kovziridze Z., Nijaradze N., Tabatadze G., Cheishvili T., Mestvirishvili Z., Mshvildadze M., Darakhvelidze N. Kinkladze V. Obtaining Of SiAlON Composite via Metal-Thermal and Nitrogen Processes in the SiC-Si-Al-Geopolymer System. Journal of Electronics Cooling and Thermal Control. 2017. 7. 103- 122.
  19. Kovziridze Z., Nijaradze N., Darakhvelidze N., Tabatadze G., Mestvirishvili Z. Obtaining of composite via metal-thermal and nitrogen processes in the SiC-SiAl-geopolymer System. 7th International Congress on Ceramics – ICC7. Foz de Iguaçu. PR. Brazil. June 17-21. 2018.
  20. ზ. კოვზირიძე, ნ. ნიჟარაძე, ნ. დარახველიძე, გ. ტაბატაძე, მ. ბალახაშვილი. რეაქციული შეცხოების მეთოდით სიალონშემცველი კომპოზიტების მიღება SiC-B<sub>4</sub>C-Si-Al-Al<sub>2</sub>O<sub>3</sub> სისტემაში მეტალოთერმული და აზოტირების პროცესებით. საქართველოს კერამიკოსთა ასოციაციის ჟურნალი „კერამიკა და მოწინავე ტექნოლოგიები“ ტ. 20. 2(40). 2018. გვ. 13-17.
  21. ზ. კოვზირიძე, ნ. ნიჟარაძე, ნ. დარახველიძე, გ. ტაბატაძე, ზ. მესტვირიშვილი. კომპოზიტების ფაზური შედგენილობის შესწავლა SiC-B<sub>4</sub>C-SiAl-Al<sub>2</sub>O<sub>3</sub> სისტემაში. საქართველოს კერამიკოსთა ასოციაციის ჟურნალი „კერამიკა და მოწინავე ტექნოლოგიები“ ტ. 21. 1(41). 2019. გვ. 44-51.
  22. Z. Kovziridze, N. Nijaradze, N. Darakhvelidze, G. Tabatadze, Z. Mestvirishvili, M. Balakhashvili, M. Mshvildadze. Obtaining of the Composite of βSiAlON Matrix via Metal-Thermal and Nitrogen Processes in the B<sub>4</sub>C-SiC-Al<sub>2</sub>O<sub>3</sub>-Si-Al-Carbon Fiber Geopolymer. System. XVI ECERS Conference and Exhibition of the European Ceramic Society. Torino. Italy. 16-20 June 2019. P.680. Abstract Book.
  23. Kovziridze Z., Nijaradze N., Darkhvelidze N., Tabatadze G., Mestvirishvili Z. Ceramic Composite in the SiC-SiAlON System. Euro Global Congress on Tychonix Nanotech. 2019 11-12 November. Valencia, Spain 2019.
  24. Kovziridze Z., Nijaradze N., Darakhvelidze N., Tabatadze G., Mestvirishvili Z. Composite in the SiAl<sub>2</sub>O<sub>3</sub>-BN-SiAlON System. 8th International Congress on Ceramics. August 23-28. Bexco. Busan. Korea.
  25. ზ. კოვზირიძე. მაკრომექანიკური მახასიათებლების ფორიან ფაზაზე დამოკიდებულების ფორმულა. Journal of the Georgian Ceramists Association. Ceramics and Advanced Technologies Vol. 20. 1(39). 2018. pp. 38-44. [www.ceramics.gtu.ge](http://www.ceramics.gtu.ge)
  26. Ковзиридзе З.Д. Разработка научных основ и технологии получения цельзиановой и алюмосиликатной керамики с использованием барита и перлита. Диссертация на соискание ученой степени доктора технических наук. Тбилиси 1993. Стр. 41-50.
  27. Z. Kovziridze. The Formula of Dependence of Mechanical Characteristics of Materials on Crystalline Phase Composition in the Matrix. Advances in Materials Physics and Chemistry. Vol. 10. No. 8. August 2020. ISSN: 2331-1959. DOI: 10.4236/ampc.2020.108013.
  28. ზ. კოვზირიძე. კერამიკულ მასალათა და კომპოზიციების მექანიკური მოდულის ფორმულა. საქართველოს ინტელექტუალური საკუთრების ეროვნული ცენტრი „საქპატენტი“ / ზ. კოვზირიძე. მოწმობა 7136 2017/10/11.
  29. Griffith A.A. Phil. Trans. Roy. Soc. London A221. 1920.

**3.4.2.4. Synthesis, characteristic and activity of nanosized Cu–Me (Me–Co, Zn, Ni) oxide systems in CO oxidation in the presence of H<sub>2</sub>.** /S.T. Jafarova/. Azerbaijan Chemical Journal. – 2021. – #1. – pp. 48-54. – eng.; abs.: eng., az., rus.

DOI: <https://doi.org/10.32737/0005-2531-2021-1-48-54>

Nanooxides of Cu–Me composition (Me–Co, Zn, Ni) were synthesized by hydrothermal reduction of metal salts with subsequent calcination and the influence of their properties (size, morphology, structure) on catalytic activity of deep CO oxidation reaction in the presence of H<sub>2</sub> was considered. The nanooxides have been characterized by XRD and SEM methods. It was revealed that particles of Cu–Co–O are nanoplates (30–35 nm), and Cu–Zn–O (12.5–20 nm) are nanorods. The SEM method revealed a higher structural organization of the Cu–Co–O particles than Cu–Zn–O; the growth of nanocrystals is shown by varying the magnification of the scale grid of images. The highest activity of the Cu–Co–O system was found among the mentioned and corresponding individual oxides. The effect of metal (Cu/Co) ratio on the dispersibility and morphology of nanoparticles and their activity has been studied. The non-additive increase in activity is explained by the redox properties of cobalt oxides and the contribution of copper to electronic state of this element. The variation of composition, as well as high dispersibility (30–35 nm) make it possible to reduce the temperature of oxidation beginning (T50%) of CO to less than 115° C. Fig. 7, Ref. 17.

**Keywords:** nanooxides, nanoplates, nanorods, modification, morphology, structure, CO oxidation, electron microscopy

#### References:

1. Yaidelin A. Manrique, Carlos V. Miguel, Diogo Mendes, Adelio Mendes. Modeling and Simulation of a Packed-bed Reactor for Carrying out the Water-Gas Shift Reaction. *International J. Chemical Reactor Engineering*. 2012. V. 10. Issue 1. P. 1542–6580.
2. Laurent Piccolo, Salim Nassreddine, Franck Morfin Surface study of the hydrogen-free or preferential oxidation of CO: Iridium vs. platinum. *Catalysis Today*. 2012. V. 189. Issue 1. P. 42–48.
3. Zong Hu, Xiaofei Liu, Dongmei Meng, Yun Guo, Yunglong Guo, Guanzhong Lu. Effect of Ceria Crystal Plane on the Physicochemical and Catalytic Properties of Pd/Ceria for CO and Propane Oxidation. *ACS Catal.* 2016. V. 6. No 4. P. 2265– 2279.
4. Xinli Zhu Min Shen Lance L. Lobban Richard G. Mallinson. Structural effects of Na promotion for high water gas shift activity on Pt–Na/TiO<sub>2</sub>. *J. Catalysis*. 2011. V. 278. Issue 1. P. 123–132.
5. Carabineiro S.A.C., Bogdanchikova N., Tavares P.B., Figueiredo J.L. Nanostructured iron oxide catalysts with gold for the oxidation of carbon monoxide. *RSC Advances*. 2012. V. 2. No 7. P. 2957.
6. Centeno M.Á., Reina T.R., Ivanova S., Laguna Ó.H., Odriozola J.A. Au/CeO<sub>2</sub> Catalysts: Structure and CO Oxidation Activity. *Catalysts*. 2016. 6. 158. DOI: 10.3390/catal6100158.
7. Amini, E., Rezaei, M. Preparation of mesoporous Fe–Cu mixed metal oxide nanopowder as active and stable catalyst for low-temperature CO oxidation. *Chinese Journal of Catalysis*. 2015. V. 36. No 10. P. 1711–1718.
8. Jing Wanga, Caiyun Hana Xiaoya Gaoa Jichang Lua Gengpin Wanab. Rapid synthesis of Fe-doped CuO–Ce<sub>0.8</sub>Zr<sub>0.2</sub>O<sub>2</sub> catalysts for CO preferential oxidation in H<sub>2</sub>-rich streams: Effect of iron source and the ratio of Fe/Cu. *J. Power Sources*. 2017. V. 343. P. 437–445.
9. Kanaparthi Ramesh, Luwei Chen, Fengxi Chen, Yan Liu, Zhan Wang, Yi-Fan Han. Re-investigating the CO oxidation mechanism over unsupported MnO, Mn<sub>2</sub>O<sub>3</sub> and MnO<sub>2</sub> catalysts. *Catalysis Today*. 2008. V. 131. No 1. P. 477–482.
10. Sajad Mobini, Fereshteh Meshkani, Mehran Rezaei. Synthesis and characterization of nanocrystalline copper–chromium catalyst and its application in the oxidation of carbon monoxide. *Process Safety and Environmental Protection*. 2017. V. 107. P. 181–189.
11. Ch Anil, Giridhar Madras. Kinetics of CO oxidation over Cu doped Mn<sub>3</sub>O<sub>4</sub>. *J. Molecular Catalysis A: Chemical*. 2016. V. 424. P. 106–114.

12. Tabakova T., Avgouropoulos G., Papavasiliou J., Manzi M., Bokuzzi H., Tenchev K., Vinidigni F., Jaoannide T. CO-free hydrogen production over Au/CeO<sub>2</sub>-Fe<sub>2</sub>O<sub>3</sub> catalysts: Part 1. Impact of the support composition on the performance for the preferential CO oxidation reaction. *Applied. Catalysis B: Environmental*. 2011. V. 101. Issues 3–4. P. 256–265.
13. Mahmood Andache, Ali Nemati Kharat, Mehran Rezaei. Preparation of mesoporous nanocrystalline CuO–ZnO–Al<sub>2</sub>O<sub>3</sub> catalysts for the H<sub>2</sub> purification using catalytic preferential oxidation of CO (COPROX). In. *J. Hydrogen Energy*. 2019. V. 44. Issue 50. P. 27401–27411.
14. Dey S., Dhal G.C. Deactivation and regeneration of hopcalite catalyst for carbon monoxide oxidation: a review. *Materials today chemistry*. 2019. V. 14. 100180.
15. Yafei Guo, Jin Lin, Jian Sun, Jubing Zhang, Changhai Li, Shouxiand Lu. Precursor Effects on Catalytic Behaviors of Copper–Manganese–Cerium Ternary Oxides Pellets for Low-Temperature CO Oxidation. *Catalysis Letters*. 2019. pp 1–13.
16. Dzhaferova S.T., Medzhidov A.A., Akhmedov M.M., Ialchin B., Fatullaeva P.A., Agaeva S.A., Abbasov M.G. Poluchenie nanorazmernykh poroshkov metodom gidrotermalnogo sovmestnogo razlozheniia nitratoov Cu, Co i Al v poliino srede. *Azer. him. zhurn*. 2018. № 2. S. 20–26.
17. Kathleen Mingle, Jochen A. Lauterbach, Synthesis-Structure-Activity Relationships in Co<sub>3</sub>O<sub>4</sub> Catalyzed CO Oxidation / *Front. Chem*. 25 May 2018 /<https://doi.org/10.3389/fchem.2018.00185>.

**3.4.2.5. Synthesis and characterization of cobalt oxide nanostructures a brief review.** /S.J. Mammadyarova/. *Azerbaijan Chemical Journal*. – 2021. – #2. – pp. 80-93. – eng.; abs.: eng., az., rus.

DOI: <https://doi.org/10.32737/0005-2531-2021-1-48-54>

The newest achievement in the synthesis of cobalt oxide nanoparticles are considered. Cobalt oxide nanoparticles have attracted a great attention due to their uncommon properties and application in a supercapacitor, optoelectronic device, Li-ion battery gas sensor and electrochromic devices. Recently, nanostructured transition metal oxides with valuable properties have become a new class of materials for many technological fields. Cobalt oxide nanoparticles obtained from various precursors show different size distribution as well as different optical, electrical, magnetic, and electrochemical properties. A reduction in particle size to nanometer-scale leads to changes in properties compared to bulk ones due to quantum size effects. Depending on the application area, the choice of an appropriate synthesis method for nanoparticles with desirable properties is a crucial factor. This work aims to provide additional information on the synthesis methods and properties of cobalt oxide nanoparticles. Fig. 4, Tab. 1, Ref. 96.

**Keywords:** cobalt oxide, crystallite size, supercapacitor, synthesis method

#### References::

1. Raghavender T., Ramesh K.G., Pravansu S.M. Nanostructured Co<sub>3</sub>O<sub>4</sub> electrodes for supercapacitor applications from plasma spray technique. *J. Power Sources*. 2012. V. 209. P. 44–51.
2. Xu M., Wang F., Zhao M., Yang S., Song X. Molten hydroxides synthesis of hierarchical cobalt oxide nanostructure and its application as anode material for lithium ion batteries. *Electrochimica Acta*. 2011. V. 56. P. 4876–4881.
3. Vetter S., Haffer S., Wagner T., Tiemann M. Nanostructured Co<sub>3</sub>O<sub>4</sub> as a CO gas sensor: Temperature-dependent behavior. *Sensors and Actuators B: Chemical*. 2015. V. 206. P. 133–138.
4. Ronan B., Gregory C., Sabine V. Sonochemical oxidation of vanillyl alcohol to vanillin in the presence of a cobalt oxide catalyst under mild conditions. *Ultrasonics Sonochemistry*. 2017. V. 36. P. 27–35.
5. Haldorai Y., Kim J.Y., Ezhil A.T., Vilian, Heo N.S., Huh Y.S., Han Y. An enzyme-free electrochemical sensor based on reduced graphene oxide Co<sub>3</sub>O<sub>4</sub> nanospindle composite for sensitive detection of nitrite. *Sensors and Actuators B: Chemical*. 2016. V. 227. P. 92–99.



6. Moon J., Kim T.K., VanSaders B., Choi C., Liu Z., Jin S., Chen R. Black oxide nanoparticles as durable solar absorbing material for hightemperature concentrating solar power system. *Solar Energy Material s& Solar Cells*. 2015. V. 134. P. 417–424.
7. Wang L., Song X., Zheng Y. Electrochromic properties of nanoporous  $\text{Co}_3\text{O}_4$  thin films prepared by electrodeposition method. *Micro & Nano Letters*. 2012. V. 7. P. 1026–1029.
8. Nethravathi C., Sen S., Ravishankar N., Rajamathi M., Pietzonka C., Harbrecht B. Ferrimagnetic Nanogranular  $\text{Co}_3\text{O}_4$  through Solvothermal Decomposition of Colloidally Dispersed Monolayers of  $\alpha$ -Cobalt Hydroxide. *J. Phys. Chem. B*. 2005. V. 109. P. 11468–11472.
9. Dong Q., Wang X., Willis W.S., Song D., Huang Y., Zhao J., Li B., Lei Y. Nitrogen-doped Hollow  $\text{Co}_3\text{O}_4$  Nanofibers for both Solid-state pH Sensing and Improved Non-enzymatic Glucose Sensing. *Electroanalysis*. 2019. V. 31. P. 1– 11.
10. Mishra S., Yogi P., Sagdeo P.R., Kumar R.  $\text{TiO}_2$ – $\text{Co}_3\text{O}_4$  core–shell nanorods: bifunctional role in better energy storage and electrochromism. *ACS Appl. Energy Mater*. 2018. V. 1. P. 790–798.
11. Kupfer B., Majhi K., Keller D.A. Thin film  $\text{Co}_3\text{O}_4/\text{TiO}_2$  heterojunction solar cells. *Adv. Energy Mater*. 2015. V. 5. P. 1401007.
12. Shi X., Han S., Sanedrin R.J., Zhou F. and Selke M. Synthesis of Cobalt Oxide Nanotubes from Colloidal Particles Modified with a Co(III)- Cysteinato Precursor. *Chem. Mater*. 2002. V. 14. P. 1897–1902.
13. Ding Y., Wang Y., Su L., Bellagamba M., Zhang H., Electrospun Y. Lei.  $\text{Co}_3\text{O}_4$  nanofibers for sensitive and selective glucose detection. *Biosensors and Bioelectronics*. 2010. V. 26. P. 542–548.
14. Vickers D., Archer L.A., Floyd-Smith T. Synthesis and characterization of cubic cobalt oxide nanocomposite fluids. *Colloids and Surfaces A: Physicochem. Eng. Aspects*. 2009. V. 348. P. 39–44.
15. Ozkaya T., Baykal A., Toprak M.S., Koseoğlu Y., Durmuş Z. Reflux synthesis of  $\text{Co}_3\text{O}_4$  nanoparticles and its magnetic characterization. *Journal of Magnetism and Magnetic Materials*. 2009. V. 321. P. 2145–2149.
16. Bhatte K.D., Bhanage B.M. Synthesis of cobalt oxide nanowires using a glycerol thermal route. *Materials Letters*. 2013. V. 96. P. 60–62.
17. Zhang Y., Chen Y., Wang T., Zhou J., Zhao Y. Synthesis and magnetic properties of nanoporous  $\text{Co}_3\text{O}_4$  nanoflowers. *Microporous and Mesoporous Materials*. 2008. V. 114. P. 257–261.
18. Wu J., Dai Y., Pan Z., Huo D., Wang, T. Zhang H., Hu J., Yan S.  $\text{Co}_3\text{O}_4$  hollow microspheres on polypyrrole nanotubes network enabling long-term cyclability sulfur cathode. *Applied Surface Science*. 2020. V. 510. P. 145529.
19. Wadekar K.F., Nemade K.R. and Waghuley S.A. Chemical synthesis of cobalt oxide ( $\text{Co}_3\text{O}_4$ ) nanoparticles by co-precipitation method. *Research J. Chemical Sciencies*. 2017. V. 7. P. 53–55.
20. Viljoen E.L., Moloto M.J., Thabede P.M. Impact of acetate ions on the shape of  $\text{Co}_3\text{O}_4$  nanoparticles. *Digest Journal of Nanomaterials and Biostructures*. 2017. V. 12. P. 571–577.
21. Allaedini G. and Muhammad A. Study of influential factors in synthesis and characterization of cobalt oxide nanoparticles. *J. Nanostructure in Chemistry*. 2013. V. 3. P. 1–16.
22. Sharifi S.L., Shakur H.R., Mirzaei A. Characterization of Cobalt Oxide  $\text{Co}_3\text{O}_4$  Nanoparticles Prepared by Various Methods: Effect of Calcination Temperatures on Size, Dimension and Catalytic Decomposition of Hydrogen Peroxide. *Int. J. Nanosci. Nanotechnol*. 2013. V. 9. P. 51–58.
23. Xu J.M., Zhang J., Wang B.B., Liu F. Shaperegulated synthesis of cobalt oxide and its gassensing property. *Journ. of Alloys and Compounds*. 2015. V. 619. P. 361–367.
24. Luisetto I., Pepe F., Bemporad E. Preparation and characterization of nano cobalt oxide. *J Nanopart Res*. 2008. V. 10. P. 59–67.
25. Abdelhak L., Bedhif B., Amar B., Cherifa D., Benhebal H. Tuning of the physical properties by various transition metal doping in  $\text{Co}_3\text{O}_4$ : TM (TM = Ni, Mn, Cu) thin films: A comparative study. *Chinese J. Physics*. 2018. V. 56. P. 1845–1852.
26. Katalin Sinko, Geza Szabo, and Miklos Zrinyi. Liquid-Phase Synthesis of Cobalt Oxide Nanoparticles. *Journal of Nanoscience and Nanotechnology*. 2011. V. 11. P. 1–9.

27. Shadrokh S., Farahmandjou M. and Firozabadi T.P. Fabrication and Characterization of Nanoporous Co Oxide ( $\text{Co}_3\text{O}_4$ ) Prepared by Simple Solgel Synthesis. *Phys. Chem. Res.* 2016. V. 4. P. 153–160.
28. Devi V.S., Athika M., Duraisamy E., Prasath A., Sharma A.S., Elumalai P. Facile sol-gel derived nanostructured spinel  $\text{Co}_3\text{O}_4$  as electrode material for high-performance supercapattery and lithiumion storage. *J. Energy Storage.* 2019. V. 25. P. 100815.
29. Wang G., Shen X., Horvat J., Wang B., Liu H., Wexler D., Yao J. Hydrothermal Synthesis and Optical, Magnetic, and Supercapacitance Properties of Nanoporous Cobalt Oxide Nanorods. *J. Phys. Chem. C.* 2009. V. 113. P. 4357–4361.
30. Hashemi Amiri S.E., Vaezi M.R. and Kandjani A.E. A comparison between hydrothermally prepared  $\text{Co}_3\text{O}_4$  via  $\text{H}_2\text{O}_2$  assisted and calcination methods. *J. Ceramic Processing Res.* 2011. V. 12. P. 327–331.
31. Nassar M.Y. Size-controlled synthesis of  $\text{CoCO}_3$  and  $\text{Co}_3\text{O}_4$  nanoparticles by free-surfactant hydrothermal method. *Materials Letters.* 2013. V. 94. P. 112–115.
32. Nugroho A., Kim J. Effect of KOH on the continuous synthesis of cobalt oxide and manganese oxide nanoparticles in supercritical water. *J. Ind. Eng. Chem.* 2014. V. 20. P. 4443–4446.
33. Elhag S., Ibupoto Z.H., Nour O., Willander M. Synthesis of  $\text{Co}_3\text{O}_4$  Cotton-Like Nanostructures for Cholesterol Biosensor. *Materials.* 2015. V. 8. P. 149–161.
34. Sun S., Zhao X., Yang M., Ma L. and Shen X. Facile and Eco-Friendly Synthesis of Finger-Like  $\text{Co}_3\text{O}_4$  Nanorods for Electrochemical Energy Storage. *Nanomaterials.* 2015. V. 5. P. 2335–2347.
35. Ma J., Zhang S., Liu W., Zhao Y. Facile preparation of  $\text{Co}_3\text{O}_4$  nanocrystals via a solvothermal process directly from common  $\text{Co}_2\text{O}_3$  powder. *J. Alloys and Compounds.* 2010. V. 490. P. 647–651.
36. Nassar M.Y., Mohamed T.Y., Ahmed I.S. One-pot solvothermal synthesis of novel cobalt salicylaldimine-urea complexes: A new approach to  $\text{Co}_3\text{O}_4$  nanoparticles. *J. Molecular Structure.* 2013. V. 1050. P. 81–87.
37. Pang M., Long G., Jiang S., Ji Y., Han W., Wang B., Liu X., Xi Y., Wang D., Xu F. Ethanolassisted solvothermal synthesis of porous nanostructured cobalt oxides ( $\text{CoO}/\text{Co}_3\text{O}_4$ ) for high-performance supercapacitors. *Chem. Eng. J.* 2015. V. 280. P. 377–384.
38. Al-Qirby L.M., Radiman S., Siong C.W., Ali A.M. Sonochemical synthesis and characterization of  $\text{Co}_3\text{O}_4$  nanocrystals in the presence of the ionic liquid  $[\text{EMIM}][\text{BF}_4]$ , *Ultrasonics Sonochem.* 2017. V. 38. P. 640–651.
39. Askarinejad A. and Morsali A. Direct ultrasonicassisted synthesis of sphere-like nanocrystals of spinel  $\text{Co}_3\text{O}_4$  and  $\text{Mn}_3\text{O}_4$ . *Ultrasonics Sonochem.* 2009. V. 16. P. 124–131.
40. Kamar E.M. and Reda S.M. Sonochemical method for synthesizing  $\text{Co}_3\text{O}_4$ /graphene nanocomposites for use as counter electrode in dye-sensitized solar Cells. *Appl. Phys. A.* 2016. V. 122. P. 688.
41. Xia X.H., Tu J.P., Zhang J., Huang X.H., Wang X.L., Zhang W.K., Huang H. Enhanced electrochromics of nanoporous cobalt oxide thin film prepared by a facile chemical bath deposition. *Electrochemistry Communications.* 2008. V. 10. P. 1815–1818.
42. Li Y., Huang K., Yao Z., Liu S., Qing X.  $\text{Co}_3\text{O}_4$  thin film prepared by a chemical bath deposition for electrochemical capacitors. *Electrochimica Acta.* 2011. V. 56. P. 2140–2144.
43. Kandalkar S.G., Gunjekar J.L., Lokhande C.D., Joo O. Synthesis of cobalt oxide interconnected flacks and nano-worms structures using low temperature chemical bath deposition. *J. Alloys and Compounds.* 2009. V. 478. P. 594–598.
44. Kandalkar S.G., Dhawale D.S., Chang-Koo Kim. Chemical synthesis of cobalt oxide thin film electrode for supercapacitor application. *Synthetic Metals.* 2010. V. 160. P. 1299–1302.
45. Kung C., Lina C., Vittal R., Ho K. Synthesis of cobalt oxide thin films in the presence of various anions and their application for the detection of acetaminophen. *Sensors and Actuators. B.* 2013. V. 182. P. 429–438.
46. Obodo R.M., Nwanya A.C., Ekwealor A.B.C., Ahmad I., Zhao T., Osuji R.U., Maaza M., Ezema F.I. Influence of pH and annealing on the optical and electrochemical properties of cobalt (III) oxide ( $\text{Co}_3\text{O}_4$ ) thin films. *Surfaces and Interfaces.* 2019. V. 16. P. 114–119.
47. Martínez-Gil M., Cabrera-German D., PintorMonroy M.I., García-Valenzuela J.A., Cota-Leal M., De W. la Cruz, Quevedo-Lopez M.A., PérezSalas R., Sotelo-Lerma M. Effect of annealing temperature on the thermal

- transformation to cobalt oxide of thin films obtained via chemical solution deposition. *Materials Science in Semiconductor Processing*. 2020. V. 107. P. 104825.
48. Valanarasu S., Dhanasekaran V., Karunakaran M., Chandramohan R., Mahalingam T. Role of Solution pH on the Microstructural Properties of Spin Coated Cobalt Oxide Thin Films. *J. of Nanosci. Nanotechnol.* 2013. V. 13. P. 1–6.
  49. Abdelhak L., Bedhief B., Amar B., Cherifa D., Benhebal H. Tuning of the physical properties by various transition metal doping in  $\text{Co}_3\text{O}_4$ : TM (TM = Ni, Mn, Cu) thin films: A comparative study. *Chin. J. Physics*. 2018. V. 56. P. 1845–1852.
  50. Patil V., Joshi P., Chougule M., S. Sen. Synthesis and Characterization of  $\text{Co}_3\text{O}_4$  Thin Film. *Soft Nanosci. Letters*. 2012. V. 2. P. 1–7.
  51. Khansari A., Salavati-Niasari M., Gholamrezaei S. Solid State Synthesis of Cobalt Oxide Nanohexagonals. *Synthesis and Reactivity in Inorganic, Metal-Organic, and Nano-Metal Chemistry*. 2015. V. 45. P. 1063–1068.
  52. Khalaji A.D. Synthesis, Characterization and Optical Properties of  $\text{Co}_3\text{O}_4$  Nanoparticles. *Asian J. Nanosci. Materials*. 2019. V. 2. P. 186–190.
  53. Farhadi S., Safabakhsh J. and Zaringhadam P. Synthesis, characterization, and investigation of optical and magnetic properties of cobalt oxide ( $\text{Co}_3\text{O}_4$ ) nanoparticles. *J. Nanostruct in Chem.* 2013. V. 3. P. 1–9.
  54. Zhang Y., Chen Y., Wang T., Zhou J., Zhao Y. Synthesis and magnetic properties of nanoporous  $\text{Co}_3\text{O}_4$  nanoflowers. *Micropor. Mesopor. Mater.* 2008. V. 114. P. 257–261.
  55. Feng C., Wang H., Zhang J., Hu W., Zou Z. & Deng Y. One-pot facile synthesis of cobalt oxide nanocubes and their magnetic properties. *J. Nanopart Res.* 2014. V. 16. P. 2413.
  56. Turan E., Zeybekoğlu E., Kul M. Effects of bath temperature and deposition time on  $\text{Co}_3\text{O}_4$  thin films produced by chemical bath deposition. *Thin Solid Films*. 2019. V. 692. P. 137632.
  57. Elaakib H., Pierson J.F., Chaik M., Samba C., Ait Dads H., Narjis A., Outzourhit A. Evolution of the structural, morphological, optical and electrical properties of reactively RF-sputtered cobalt oxide thin films with oxygen pressure. *Vacuum*. 2019. V. 159. P. 346–352.
  58. Ambika S., Gopinath S., Saravanan K., Sivakumar K., Ragupathi C., Sukantha T.A. Structural, morphological and optical properties and solar cell applications of thioglycolic routed nano cobalt oxide material. 2019. V. 5. P. 305–309.
  59. Alla S.K., Duvuru H.B., Shaw S.K., Vara Prasad B.B.V.S., Kumar M.K., Meena S.S., Gupta N., Prasad N.K. Zr-substituted cobalt oxide nanoparticles: structural, magnetic and electrical properties. *Journal of Materials Science: Materials in Electronics*. 2019. V. 30. P. 20088–20098.
  60. Duvuru H.B., Alla S.K., Shaw S.K., Meena S.S., Gupta N., Vara Prasad B.B.V.S., Kothawale M.M., Kumar M.K., Prasad N.K. Magnetic and dielectric properties of Zn substituted cobalt oxide nanoparticles. *Ceramics Int.* 2019. V. 45. P. 16512–16520.
  61. Kalam K., Seemen H., Mikkor M., Jõgiaas T., Ritslaid P., Tamm A., Kukli K., Kasikov A., Link J., Stern R., Dueñas S., Castán H. Electrical and magnetic properties of atomic layer deposited cobalt oxide and zirconium oxide nanolaminates. 2019. V. 669. P. 294–300.
  62. El A.M. Sayed and El-Gamal S. Synthesis and investigation of the electrical and dielectric properties of  $\text{Co}_3\text{O}_4$ /(CMC+PVA) nanocomposite films. *J. Polym Res.* 2015. V. 22. P. 97.
  63. Durano M.M., Tamboli A.H., Kim H. Cobalt oxide synthesized using urea precipitation method as catalyst for the hydrolysis of sodium borohydride. *Colloids and Surfaces A: Physicochem. Eng. Aspects*. 2017. V. 520. P. 355–360.
  64. Ashraf Janjua M.R.S. Synthesis of  $\text{Co}_3\text{O}_4$  Nano Aggregates by Co-precipitation Method and its Catalytic and Fuel Additive Applications. *Open Chem.* 2019. V. 17. P. 865–873.
  65. Pudukudy M., Yaakob Z. Sol-gel synthesis, characterisation, and photocatalytic activity of porous spinel  $\text{Co}_3\text{O}_4$  nanosheets *Chemical Papers*. 2014. V. 68. P. 1087–1096.

66. Farhadi S., Javanmard M. and Nadri G. Characterization of Cobalt Oxide Nanoparticles Prepared by the Thermal Decomposition of  $[\text{Co}(\text{NH}_3)_5(\text{H}_2\text{O})](\text{NO}_3)_3$  Complex and Study of Their Photocatalytic Activity. *Acta Chim. Slov.* 2016. V. 63. P. 335–343.
67. Kang M. and Zhou H. Facile Synthesis and Structural Characterization of  $\text{Co}_3\text{O}_4$  Nanocubes. *AIMS Mater. Science.* 2015. V. 2. P. 16–27.
68. Ren Q., Feng Z., Mo S., Huang C., Li S., Zhang W., Chen L., Fu M., Wu J., Ye D. 1D- $\text{Co}_3\text{O}_4$ , 2D- $\text{Co}_3\text{O}_4$ , 3D- $\text{Co}_3\text{O}_4$  for catalytic oxidation of toluene. *Catalysis Today.* 2019. V. 332. P. 160–167.
69. Kung C., Lin C., Li T., Vittal R., Ho K. Synthesis of  $\text{Co}_3\text{O}_4$  thin films by chemical bath deposition in the presence of different anions and application to  $\text{H}_2\text{O}_2$  sensing. *Procedia Eng.* 2011. V. 25. P. 847–850.
70. Behling R., Chatel G., Valange S. Sonochemical oxidation of vanillyl alcohol to vanillin in the presence of a cobalt oxide catalyst under mild conditions. *Ultrasonics Sonochem.* 2017. V. 36. P. 27–35.
71. Dong Q., Wang X., Willis W.S., Song D., Huang Y., Zhao J., Li B. and Lei Y. Nitrogen-doped Hollow  $\text{Co}_3\text{O}_4$  Nanofibers for both Solid-state pH Sensing and Improved Non-enzymatic Glucose Sensing. *Electroanal.* 2019. V. 31. P. 1–11.
72. Dai G., Xie J., Li C., Liu S. Flower-like  $\text{Co}_3\text{O}_4$ /graphitic carbon nitride nanocomposite based electrochemical sensor and its highly sensitive electrocatalysis of hydrazine. *J. Alloys and Compounds.* 2017. V. 727. P. 43–51.
73. Palani S. and Arumugam S. Nano  $\text{Co}_3\text{O}_4$  as Anode Material for Li-Ion and Na-Ion Batteries: An Insight into Surface Morphology. *Chemistry Select.* 2018. V. 3. P. 5040–5049.
74. Fan L., Zhang W., Zhu S., and Lu Y. Enhanced Lithium Storage Capability in Li-Ion Batteries Using Porous 3D  $\text{Co}_3\text{O}_4$  Nanofiber Anodes. *Ind. Eng. Chem. Res.* 2017. V. 56. P. 2046–2053.
75. Wang Y., Guo R., Liu W., Zhu L., Huang W., Wang W., Zheng H.  $\text{Co}_3\text{O}_4$  nanospheres composed of highly interconnected nanoparticles for boosting Li-Ion storage. *J. Power Sources.* 2019. V. 444. P. 227260.
76. Wang D., Yu Y., He H., Wang J., Zhou W., Abruña H.D. Template-Free Synthesis of Hollow Structured  $\text{Co}_3\text{O}_4$  Nanoparticles as High-Performance Anodes for Lithium-Ion Batteries. *ACS Nano.* 2015. V. 9. P. 1775–1781.
77. Zheng F., Wei L. Synthesis of ultrafine  $\text{Co}_3\text{O}_4$  nanoparticles encapsulated in nitrogen-doped porous carbon matrix as anodes for stable and long-life lithium ion battery. *J. Alloys and Compounds.* 2019. V. 790. P. 955–962.
78. Park J.S., Shin D.O., Lee C.S., Lee Y., Kim J.Y., Kim K.M., Shin K. Mesoporous perforated  $\text{Co}_3\text{O}_4$  nanoparticles with a thin carbon layer for high performance Li-ion battery anodes. *Electrochimica Acta.* 2018. V. 264. P. 376–385.
79. Keshmarzi M.K., Daryakenari A.A., Omidvar H., Javanbakht M., Ahmadi Z., Delaunay J., Badrnezhad R. Pulsed electrophoretic deposition of nanographitic flakenanostructured  $\text{Co}_3\text{O}_4$  layers for efficient lithium-ion-battery anode. *J. Alloys and Compounds.* 2019. V. 805. P. 924–933.
80. Hu R., Zhang H., Bu Y., Zhang H., Zhao B., Yang C. Porous  $\text{Co}_3\text{O}_4$  nanofibers surface-modified by reduced graphene oxide as a durable, high-rate anode for lithium ion battery. *Electrochim. Acta.* 2017. V. 228. P. 241–250.
81. Wang B., Wang S., Tang Y., Ji Y., Liu W., Lu X. Hydrothermal Synthesis of Mesoporous  $\text{Co}_3\text{O}_4$  Nanorods as High Capacity Anode Materials for Lithium Ion Batteries. *Energy Procedia.* 2019. V. 158. P. 5293–5298.
82. Yang J., Gao M., Lei J., Jin X., Yu L., Ren F. Surfactant-assisted synthesis of ultrathin two dimensional  $\text{Co}_3\text{O}_4$  nanosheets for applications in lithium-ion batteries and ultraviolet photodetector. *J. Solid State Chem.* 2019. V. 274. P. 124–133.
83. Dwivedi P.K., G. Parte, M. Thripuranthaka, M.V. Shelke. High efficiency lithium storage in 3D composite foam of  $\text{Co}_3\text{O}_4$  nanoparticles integrated carbon nanohorns. *Mater. Sci. Eng.* 2021. V. 263. P. 114839.
84. Wang H., Zhu Y., Yuan C., Li Y., Duan Q. Cobalt-phthalocyanine-derived ultrafine  $\text{Co}_3\text{O}_4$  nanoparticles as high-performance anode materials for lithium ion batteries. *Appl. Surface Sci.* 2017. V. 414. P. 398–404.
85. Liu Y., Wan H., Jiang N., Zhang W., Zhang H., Chang B., Wang Q., Zhang Y., Wang Z., Luo S., Sun H. Chemical reduction-induced oxygen deficiency in  $\text{Co}_3\text{O}_4$  nanocubes as advanced anodes for lithium ion batteries. *Solid State Ionics.* 2019. V. 334. P. 117–124.

86. Deng J., Lv X., Zhong J., Sun X. Carbon coated porous  $\text{Co}_3\text{O}_4$  nanosheets derived from cotton fibers as anodes for superior lithium ion batteries. *Appl. Surface Sci.* 2019. V. 475. P. 446–452.
87. Zhang L., Li H., Li K., Li L., Wei J., Feng L., Fu Q. Morphology-controlled fabrication of  $\text{Co}_3\text{O}_4$  nanostructures and their comparative catalytic activity for oxygen evolution reaction. *J. Alloys and Compounds.* 2016. V. 680. P. 146–154.
88. Hou Y., Hou C., Zhai Y., Li H., Chen T., Fan Y., Wang H., Wang W. Enhancing the electrocatalytic activity of 2D micro-assembly  $\text{Co}_3\text{O}_4$  nanosheets for  $\text{Li-O}_2$  batteries by tuning oxygen vacancies and  $\text{Co}^{3+}/\text{Co}^{2+}$  ratio. *Electrochim. Acta.* 2019. V. 324. P. 134884.
89. Dhas C.R., Venkatesh R., Kirubakaran D.D., Merlin J.P., Subramanian B., Ezhil Raj A.M. Electrochemical sensing of glucose and photocatalytic performance of porous  $\text{Co}_3\text{O}_4$  films by nebulizer spray technique. 2017. V. 186. P. 561–573.
90. Philippot K., De Tovar J., Romero N., Denisov S.A. Light-driven water oxidation using hybrid photosensitizer-decorated  $\text{Co}_3\text{O}_4$  nanoparticles. 2018. V. 9. P. 506–515.
91. Atique Ullah A.K.M., Amin F.B., Hossain A. Tailoring surface morphology and magnetic property by precipitants concentrations dependent synthesis of  $\text{Co}_3\text{O}_4$  nanoparticles. *Ceramics Int.* 2020. V. 46. P. 27892–27896.
92. Yetim N.K. Hydrothermal synthesis of  $\text{Co}_3\text{O}_4$  with different morphology: Investigation of magnetic and electrochemical properties. *J. Molecular Structure.* 2021. V. 1226. P. 129414.
93. Zdorovets M.V., Shumskaya A.E., Kozlovskiy A.L. Investigation of the effect of phase transformations on the magnetic and electrical properties of  $\text{Co}/\text{Co}_3\text{O}_4$  nanowires. *J. Magnetism and Magnetic Mater.* 2020. V. 497. P. 166079.
94. Ramamoorthy C., Rajendran V. Effect of surfactants assisted  $\text{Co}_3\text{O}_4$  nanoparticles and its structural, optical, magnetic and electrochemical properties. *Optik.* 2017. V. 145. P. 330–335.
95. Anandhababu G., Ravi G. Facile synthesis of quantum sized  $\text{Co}_3\text{O}_4$  nanostructures and their magnetic properties. *Nano-Structures & Nano Objects.* 2018. V. 15. P. 1–9.
96. Yin K., Ji J., Shen Y., Xiong Y., Bi H., Sun J., Xu T., Zhu Z., Sun L. Magnetic properties of  $\text{Co}_3\text{O}_4$  nanoparticles on graphene substrate. *J. Alloys and Compounds.* 2017. V. 720. P. 345–351.

### 4.3. Processing Technologies

**3.4.3.1. Anionic zeolite nanomaterial – environmentally safe complex fertilizer with prolonged action.** /G. Tsintskaladze, T. Sharashenidze, M. Zautashvili, M. Burdjanadze, G. Antia, N. Mumladze/. *Bulletin of the Georgian National Academy of Sciences.* – 2021. – v. 15. – #3. – pp. 59-64. – eng.; abs.: eng., geo.

The development and implementation of effective and cost-effective environmental technologies is one of the priority problems in Georgia for the rehabilitation of soil fertility and natural vegetation cover. The paper proposes a new method for nanomodification of natural zeolite - clinoptilolite, based on the introduction of the appropriate salt into the structure of the zeolite so that the resulting material does not lose its zeolite structure and acquires both cation-exchange and anionexchange properties. Some amount of ammonium dihydrogen phosphate ( $\text{NH}_4\text{H}_2\text{PO}_4$ ), potassium nitrate ( $\text{KNO}_3$ ) and cations mixed with them (Fe, Ca, Mn, Zn, Mg, Cu, Mo, Co, Sn) were introduced by fusion method into the clinoptilolite structure. Only the amount of ammonium dihydrogen phosphate changed, while the amount of potassium nitrate ( $\text{KNO}_3$ ) and cations remained unchanged. Accordingly, zeolite nanomaterials of various composition, structure and properties were obtained, which were studied by the methods of chemical, IR spectroscopic and X-ray diffractometric analyses. The obtained zeolite nanomaterials as fertilizers of complex composition and long-term action were used to study their effect on wheat productivity both in the open field and in laboratory conditions. Zeolite nanomaterials of three different compositions were studied. Tab. 3, Ref. 14.

**Keywords:** natural zeolite, zeolite-anionic form, nanomodified, I.R. spectroscopic, agriculture

#### References:

1. Andronikashvili T., Urushadze T. (2008). Ispolzovanie tseolitsoderzhashchikh porod v rastenievodstve. J. Agrochemistry. 12: 63-79. M. (in Russian).
2. Sheudzhen A. (2005). Soderzhanie mikroelementov v pochvakh i dostupnostikh rasteniiam. Materials of the regional scientific-practical conference: Fertilizers and harvest. 238-269. Maykop (in Russian).
3. Marshania I. (1991). Agrochemistry. 741 p. Tbilisi (in Georgian).
4. Zavalin A., Blagoveshchenskaya G., Chernova L., Shmyreva N. (2012) Upravlenie azotnym pitaniem rastenii v pochve. J. Agrochemical Bulletin. 4:38-40. Moscow (in Russian).
5. Volynkin V., Volynkin O. (2013). Deistvie sostava udobreniia i doz azota pri sistematicheskom primenenii v sevooborote i na monokulture pshenitsy. J. Fertility. 2:20-22. M. (in Russian).
6. Kudashkin M. (2011). Vliianie azotnykh i mikroudobrenii na urozhainostozimoi pshenitsy razlichnykh srokov seva v sevooborotakh agrolandshaftov iuga Nechernozemya. J. Agrochemistry. 5:26-34. M. (in Russian).
7. Tsitsishvili G., Tsintskaladze G., Tsitsishvili V., Chipashvili D., Tsintskaladze Z. (2006). New form of phosphorous containing clinoptilolite. Azerbaijan Chemical Journal. 3:100-102. Baku.
8. Tsintskaladze G., Eprikashvili L., Zautashvili M., Sharashenidze T., Burdjanadze M., Burkiashvili N., Gabunia V. (2013). New biotechnological material based on natural zeolite acts as a fertilizer of a prolonged action and prospects of its application. 559-562. II International Scientific-Practical Conference: Bioeconomy and Sustainable Development of Agriculture. Tbilisi (in Georgian).
9. Flanigen E., Khatami H., Szymanski H. (1974). Infrared structural studies of zeolites frameworks. Chapter 16:201-229. In: Flanigen, E.M. and Sand, L.B., Eds., molecular sieve zeolites. Advances in Chemistry. 101:460-488, American Chemical Society. Washington DC.
10. Tsitsishvili G., Andronikashvili T., Kardava M. (1993). Prirodnye tseolity v zemledelii, 128p., Tbilisi (in Russian).
11. Andronikashvili T., Urushadze T., Eprikashvili L. (2009). Tseolitoderzhashchie substraty – novyi put ot rastenievodstva k rasteniyeirozvodstvu. J. Annals of Agrarian Science. 7. 4:14-45. Tbilisi (in Russian).
12. Tsintskaladze G., Tsitsishvili V., Sharashenidze T., Zautashvili M., Iluridze G., Tsintskaladze P. (2016) Possibilities of agricultural application of zeolite nanomaterials enriched by nitrate ions. Proceedings of the Georgian National Academy of Sciences. Chemical Series. 42. 1:78-80. Tbilisi (in Georgian).
13. Tsintskaladze G., Eprikashvili L., Urushadze T., Kordzakhia T., Sharashenidze T., Zautashvili M., Burjanadze M. (2016) Nanomodified natural zeolite as a fertilizer of prolog acting. J. Annals of Agrarian Science. 14:163-168. Tbilisi.
14. Tsintskaladze G., Eprikashvili L., Mumladze N., Gabunia V., Sharashenidze T., Zautashvili M., Kordzakhia T., Shatakishvili T. (2017). Nitrogenous zeolite nanomaterial and the possibility of its application in agriculture. J. Annals of Agrarian Science. 15:365-369. Tbilisi.

#### 4.4. Nanobiotechnology

**3.4.4.1. Growing technology for soybeans with nanoherbicides.** /A. Korakhashvili, T. Kacharava, L. Korakhashvili/. Annals of Agrarian Science. – 2021. – v. 19. - #3. – pp. 199-203. – eng.; abs.: eng.

Modern herbicide market in agriculture is about 2 billion tons and about 73 billion dollars industry with sophisticated multi-impact problems with food safety and human health, with increasing of weed resistance with every passing year at the topmost. Nanoherbicides under development in the current decade of our century could be a new strategy to address all the issues caused by the conventional non-nanoherbicides. From the beginning of 21 century group of Georgian scientists with farmers associations have begun development nanoherbicides (experimental name “Nanocooper 076”, which is under registration) in soybean experimental pilot plots and farmer’s fields, which will allow farmers to clear

their soybean plantings from weeds without using toxic chemicals, like Glyphosate. As the potential use of nanostructured nanomaterials enables the use of nanoherbicides effectively and rules out the emergence of various weed-resistant population at an early stage of growing agricultural crops (first weeks after sowing), these very desirable nano technological methods and practices in general agriculture are reviewed by this article. Fig. 2, Tab. 1, Ref. 18.

**Keywords:** soya seed pilling, Nanocooper 076, soil contamination, friendly nanotechnologies, nanoherbicides, agriculture

### References:

1. FAO. Country Programming Framework for Georgia. 2016 to 2020. Italy. 2015.
2. Gullner G., Komivec N, Rennenberg H. Detoxification of Chloroacetinilide by Transgenic Poplars. In: Phytoremediation: environmental and molecular biological aspects. OECD workshop. Hungary. Abstr. 2004. 24 pp.
3. Korakhashvili A. Soybean Seed Inoculation Method. Georgia State Patent # 1180. Tbilisi. Georgia. 1996. 5 pp. (in Georgian).
4. Korakhashvili A. New Growing Technologies of Grain Legumes and Their role in Farmers Economics. "Caravan". Aleppo. Syria. 2001. pp 23-29.
5. Korakhashvili A., Annual Management Plan for Farming by Computer Program BARMEX, Third European Conference on Precision Agriculture. Montpellier. France. 2001. pp 47-51.
6. Agladze G., Korakhashvili A. Grass landraces of Georgian arid pastures. Report of a Working Group on Forages. Elvas. Portugal. 199. 97 pp.
7. Korakhashvili A., Teo Urushadze. Growing of Oldest Legumes by Advance Technologies in Georgia. "Grain Production". # 3. Moscow. Russia. 2002. pp. 34-35 (in Russian).
8. Korakhashvili A., D. Kirvalidze, T. Kvrivishvili, R. Vaismiller, E. Sanadze. Research of Cinnamonic Calcareous Soil Fertilizing Systems for Pastures of Akhaltsikhe District. Communications in Soil Sciences and Plant Analysis. Taylor and Francis. USA. vol. 42. #7. (2011) 767-786.
9. Korakhashvili A. Regeneration and Conservation of Chickpea Genetic Resources of Georgia. International Conference on Enhanced. Genepool Utilization. Cambridge. United Kingdom. 2014. pp. 43-44.
10. Korakhashvili A. Seed registration, development and certification, in Enabling the Business of Agriculture, WB/EBRD. Washington. USA. 2016. pp. 126-131.
11. Korakhashvili A., T. Urushadze, D. Kirvalidze. Endemic and Released Legume Crops Sustainable Production in Georgia. Lam LAMBERT Academic Publication. Germany. USA. UK. 2018. 55 pp.
12. Mahendra Shah, Strong Maurice. Food in the 21st Century: from Science to Sustainable Agriculture. Washington. USA. 1999. 72 pp.
13. Njoroge W. John. Indicators of Sustainable Farming. IFOAM. Imsbac. Germany. 1997. 124 pp.
14. Ronald D. Knutson, J.B. Penn, Barry L., Flinch Baugh. Agricultural and Food Policy. New Jersey. USA. 1998. 521 pp.
15. Zaalishvili G., Khatiashvili G., Ugrekheldze D., Gordeziani M., Kvesitadze G. Plant potential for detoxification (Review). Appl. Biochem Microbial 36. 2000. pp. 443- 451.
16. Korakhashvili A., Kirvalidze D. Chickpea Genetic Resources Regeneration and Safety Duplication in Georgia, Universal J. of Agricultural Research, USA, Vol. 4(3) (2016) 67-70.
17. Korakhashvili A., Chickpea Genetic Resources Regeneration and Safety Duplication in Georgia. CABI Oxfordshire-Boston. UK-USA. 2016. pp. 210-221.
18. Korakhashvili A., T. Sanikidze, L. Korakhashvili. Adaptation of Food Safety Communication Systems RASFF and INFOSAN in Georgian Cheese Production. Workshop of AASSA comity. Academies of Sciences. New Delhi. India. 2017. pp. 18-21.

## 5. NANOENGINEERING

### 5.1. Devices and Sensors

**3.5.1.1. Single wall carbon nanotube gas sensors.** /V.M. Aroutiounian/. Armenian Journal of Physics. – 2021. – vol. 14. – #1. – pp. 74-84. – eng.; abs.: eng.

DOI: <https://doi.org/10.52853/18291171-2021.14.1-74V>

Excellent physical properties of carbon nanotubes (CNTs) are used for manufacturing of many electronic devices. Single wall version of CNTs is promising for detection many important gases including gases exhaled by the organism. The most promising is the realization of gas sensors based on metal oxides doped with CNTs. Application of CNT-based sensors to breathe analysis, properties of the SWCNTs gas sensors with metal nanoparticles and metal oxides and CNTs biosensors are reviewed in this paper. Fig. 1. Tab. 2, Ref. 83.

**Keywords:** carbon nanotubes CNTs, electronic devices, gas sensors, CNT-based sensors, biosensors

#### References:

1. M.L. Geier, P.L. Prabhumirashi, J.J. McMorro et al. Nano Lett. 2013. 13, 4810.
2. J. Zou, K. Zhang, J. Li et al. Sci. Rep. 2015. 5, 11755.
3. Z. Cao, B.B.Q. Wei. Energy Environ. Sci. 2013. 6, 3183.
4. N. Yang, X. Chen, T. Ren et al. Sens. Actuators B 2015. 207, 690.
5. V.M. Aroutiounian In: Dekker Encyclopedia of Nanoscience and Nanotechnol. Second Edition. Taylor and Francis: New York. 2012. p. 1.
6. V.M. Aroutiounian In: Semiconductor gas sensors, Woodhead Publishing Series in Electronic and Optical Materials N 38. 2013. chapter 12. p. 408.
7. V.M. Arourounian Int. J. Hydrogen Energy 32. N 9. p. 1145. 2007.
8. V.M. Aroutiounian J.Cont. Physics 54. p. 356, 2019.
9. V. Aroutiounian Armenian J. Phys. 2018. 11. 1. p. 39.
10. Harris P. Carbon nanotubes and related structures, Cambridge University Press 1999. 279 pp.
11. E.L. Hopley, Sh. Salmasi, D. M. Kalaskar, A. M. Seifalian Biotechnology Advances 2014.
12. J. E. Ellis and A. Star Chem Plus Chem 2016. 81. 1248.
13. Adamyan Z., Aroutiounian V., Sayunts A.et al. Sens. Sens. Syst., 2018, 7, p. 31.
14. Aroutiounian V.M. Int. Sci. J. Alternative Energy and Ecology, 2018, 249-251, p. 38.
15. Aroutiounian V.M. Lithuanian J. Phys. 2015, 55, 4, p. 319.
16. Aroutiounian V.M., V. Arakelyan V.M., Shahnazaryan G.E. Advances in Nano Research 2015, p. 1.
17. Aroutiounian V.M. Journal of Nanomedicine. Nanotechnology and Nanomaterials. 2020, 1, 1. p. 1.
18. Aroutiounian V.M. Медицинская наука Армении. 2021(in press).
19. Aroutiounian V.M. Медицинская наука Армении. 2020, LX, 1, pp. 3.
20. V. . Aroutiounian Biomedical J. Sci. & Tech. Research, 30, 2, p. 23211, 2020
21. Aroutiounian V.M. Journal of Nanomedicine & Nanotechnology, 2020, 11, 3, p. 1.
22. Aroutiounian V.M. J. Cont. Phys. 2021,56, 2 (in press)
23. Z. Zanolli, R. Leghrib, A. Felten et al ACS Nano 2011, 5, 4592.
24. M. Penza, R. Rossi, M. Alvisi et al Sens. Actuators B 2009, 140, 176.
25. M. Shao, J. Power Sources 2011, 196, 2433.
26. S. Mubeen, T. Zhang, B. Yoo et al J. Phys. Chem. C 2007, 111, 6321
27. I.Lundström, M. S. Shivaraman, C. Svensson, Surf. Sci. 1977, 64, 497.
28. Abdelhalim, A. Abdellah, G. Scarpa, P. Lugli, Nanotechnology 2014, 25, 055208.
29. S. Mubeen, T. Zhang, N. Chartuprayoon et al Anal. Chem. 2010, 82, 250.
30. B.R. Goldsmith, J.G. Coroneus, V.R. Khalap et al Science 2007, 315, 77.
31. J. Gong, J. Sun, Q. Chen, Sens. Actuators B 2008, 130, 829.
32. F. Rigoni, G. Drera, S. Pagliara et al Carbon 2014, 80, 356.
33. J.E. Ellis, U. Green, D.C. Sorescu et al J. Phys. Chem. Lett. 2015, 6, 712.



34. Y. Sun, H. H. Wang, *Adv. Mater.* 2007, 19, 2818
35. S. Cui, H. Pu, G. Lu et al *ACS Appl. Mater. Interfaces* 2012, 4, 4898.
36. M. Ding, D. C. Sorescu, A. Star, *J. Am. Chem. Soc.* 2013, 135, 9015.
37. L. Dang, G. Zhang, K. Kan et al *J. Mater. Chem. A* 2014, 2, 4558.
38. H. Liu, W. Zhang, H. Yu et al *ACS Appl. Mater. Interfaces* 2016, 8, 840.
39. G. Peng, S. Wu, J. E. Ellis et al *J. Mater. Chem. C* 2016, 4, 6575.
40. P.C. Jurs, G.A. Bakken, H.E. McClelland, *Chem. Rev.* 2000, 100, 2649.
41. N.J. Kybert, M.B. Lerner, J.S. Yodh et al *ACS Nano* 2013, 7, 2800.
42. S. Chatterjee, M. Castro, J. F. Feller, *Sens. Actuators B* 2015, 220, 840.
43. G. Peng, E. Trock, H. Haick, *Nano Lett.* 2008, 8, 3631.
44. G. Peng, U. Tisch, H. Haick, *Nano Lett.* 2009, 9, 1362.
45. H. P. Hong, J. H. Kim, C. J. Lee, N. K. Min, *Sens. Actuators B* 2015, 220, 27.
46. J. P. Novak, E. S. Snow, E. J. Houser et al. *APL*, 83, 19, ??
47. N. Yang, X.Chen, T. Ren, et al. *Sens. Actuators B* 207 (2015) 690.
48. L.C. Clark, C. Lyons *Ann. N. Y. Acad. Sci.* 102 (1) (1962) 29.
49. M. Raicopol, A. Pruna, C. Damian, L. Pila Nanoscale *Res. Lett.* 8 (1) (2013) 1–8.
50. X. Guo *Adv. Mater.* 25 (25) (2013) 3397–3408.
51. B.L. Allen, P.D. Kichambare, A. Star *Adv. Mater.* 19 (11) (2007) 1439–1451.
52. D.R. Thévenot, K. Toth, R. A. Durst, G. S. Wilson *Biosens. Bioelectron.* 16 (1) (2001) 121–131.
53. H.R. Byon, H. C. Choi *J. Am. Chem. Soc.* 128 (7) (2006) 2188–2189.
54. X.W. Tang, S. Bansaruntip, N. Nakayama et al *E. Yenilmez, Y.L. Chang, Q. Wang Nano Lett.* 6 (8)(2006) 1632–1636.
55. M.-J. B. Kim, S. Woo, J. W. Kim et al *J. Pharm. Pharmacol.* 1 (2013) 18–24.
56. A.Star, J.-C.P. Gabriel, K. Bradley, G. Grüner *Nano Lett.* 3 (4) (2003) 459–463.
57. A.Nie, L. Pan, H. Li et al *J. Electroanal. Chem.* 666 (2012) 85–88.
58. Anker, J.N., Hall, W.P., Lyandres, O. et al *In Nanoscience and Technology*. Co-Published with Macmillan Publishers Ltd.: London. UK. 2009. pp. 308–319. ISBN 9789814287005.
59. Farrera, C. Torres Andón, F. Feliu *ACS Nano* 2017, 11, 10637–10643.
60. Shao, L., Gao, Y., Yan, F. *Sensors* 2011, 11, 11736–11751.
61. Ramgir, N.S., Yang, Y., Zacharias, M. *Small* 2010, 6, 1705–1722.
62. Chen, Z., Zhang, X., Yang, R et al *Nanoscale* 2011, 3, 1949.
63. Liu, Y., Dong, X., Chen, P. *Chem. Soc. Rev.* 2012, 41, 2283.
64. Shen, J., Zhu, Y., Yang, X., Li *Chem. Commun.* 2012, 48, 3686–3699.
65. Gao, C., Guo, Z., Liu, J.-H., Huang, X.-J. *Nanoscale* 2012, 4, 1948.
66. Liu, Z., Tabakman, S., Welsher, K., Dai, H. *Nano Res.* 2009, 2, 85–120.
67. Yang, W., Ratnac, K.R., Ringer, S.R. et al *Angew. Chem. Int. Ed.* 2010, 49, 2114.
68. Boghossian, A.A., Zhang, J., Barone et al *ChemSusChem* 2011.
69. Kim, S.-J.J., Choi, S.-J.J., Jang, J.-S.S et al *Acc. Chem. Res.* 2017, 50, 1587.
70. Alvarez, M.M., Aizenberg, J., Analoui, M. et al *ACS Nano* 2017, 11, 5195–5214.
71. Kruss, S., Hilmer, A.J., Zhang, J. et al *Adv. Drug Deliv. Rev.* 2013, 65, 1933.
72. Eatemadi, A., Daraee, H., Karimkhanloo et al *Nanoscale Res. Lett.* 2014.
73. Iverson, N.M., Barone, P.W., Shandell, M. et al *Nat. Nanotechnol.* 2013, 8, 873.
74. Ménard-Moyon, C., Kostarelos, K. Prato, M. 2010, 17, 107.
75. Wu, Y., Phillips, J.A., Liu, H. et al *ACS Nano* 2008, 2, 2023.
76. Bisker, G., Dong, J., Park, H.D. et al *Nat. Commun.* 2016, 7, 1.
77. Zhang, J., Landry, M.P., Boghossian, A.A. *Nat. Nanotechnol.* 2013, 8, 959.
78. Beyene, A.G., Delevich, K., Yang, S.J., Landry, M.P. *Biochemistry.* 2018, 57, 6379.
79. Dinarvand, M., Neubert, E., Meyer et al *Nano Lett.* 2019, 19, 6604.
80. Hendler-Neumark A. and Bisker G. *Sensors* 2019, 19, 5403.
81. Ahn, J.-H., Kim, J.-H., Boghossian, A.A. et al *Nano Lett.* 2011, 11, 19.
82. Reuel, N.F., Ahn J.-H., Boghossian et al *J.-H.J. Am. Chem. Soc.* 2011, 133, 17923.
83. Pan, J., Li, F., Choi, J.H.J. *Mater. Chem. B* 2017, 5, 6511.

## 6. NANOMEDICINE

### 6.1. Medical Physics

**3.6.1.1. Development and testing of nanoparticles for treatment of cancer cells by Curie temperature controlled magnetic hyperthermia.** /A. Chirakadze, N. Mitagvaria, D. Jishiashvili, M. Devdariani, G. Petriashvili, L. Davlianidze, N. Dvali, K. Chubinidze, A. Jishiashvili, Z. Buachidze, I. Khomeriki/. Bulletin of the Georgian National Academy of Sciences. – 2021. – v. 15. – #1. – pp. 91-98. – eng.; abs.: eng., geo.

A vast amount of nanoparticles has been developed and proposed for the local hyperthermia of cancer during the last decades, but only a few of them correspond to the mandatory requirements of having therapeutic range Curie temperature ( $T_C = 41-45^\circ\text{C}$ ), high-rate crystallinity and “strong” magnetic properties, strictly controlled homogeneity and dispersion of the nanoparticles, good biocompatibility and harmless decomposition products. Among them are the nickel-copper (Ni-Cu) and silver doped lanthanum manganite ( $\text{Ag}_x\text{La}_{1-x}\text{MnO}_3$ ) nanoparticles. The developed research showed that the materials obtained at lower than usual temperatures using microwave enhanced synthesizes and annealing can be successfully used for local hyperthermia revealing high magnetic properties. Behavioral toxicity testing of the developed nanoparticles was enhanced by blood oxygen saturation measurements using noninvasive oximetry in white rats. Both of the developed nanomaterials revealed a lower toxicity level than the commercially available  $\text{Fe}_2\text{O}_3$  nanoparticles. Fig. 4, Ref. 14.

**Keywords:** Cancer, magnetic hyperthermia, behavioral tests, toxicity, nanoparticles, synthesis, magnetic properties, microwave, blood oxygen saturation

#### References:

1. Report on behalf of the Children, Teenagers and Young Adults Expert Advisory Group, National Cancer Registration and Analysis Service. (2018). Childhood Cancer Statistics, England. Annual report 2018: Incidence of childhood cancer, 2001 to 2015. PHE publications, Gateway number: 2018135, 31 pp. [https://file:///C:/Users/593%2022-11-01/Downloads/Childhood\\_Cancer\\_Statistics\\_England\\_FINAL\\_updated.20\(6\).pdf](https://file:///C:/Users/593%2022-11-01/Downloads/Childhood_Cancer_Statistics_England_FINAL_updated.20(6).pdf).
2. Ward Z.J., Yeh J.M., Bhakta N., Frazier A.L., Atun R. (2019). Estimating the total incidence of global childhood cancer: a simulation-based analysis. *The Lancet Oncology*: 20. 4: 483-493.
3. Ban I., Stergar J., Maver U. (2018). Ni-Cu magnetic nanoparticles: review of synthesis methods. surface functionalization approaches, and biomedical applications. *Nanotechnology Review*. 7. 2: 187–28.
4. Katz E. (2019). Synthesis, properties and applications of magnetic nanoparticles and nanowires - a brief introduction. *Magnetochemistry*. 5. 61. doi:10.3390/magnetochemistry5040061.
5. Mary J.A., Manukandani A., Kennedt L.J., Buoudina M., Sundaram R., Vijaya J.J. (2014). Structure and magnetic properties of Cu–Ni alloy nanoparticles prepared by rapid microwave combustion method. *Trans. Nonferrous Met. Soc. China*. 24: 1467–1479.
6. Ehi-Eromosele C.O., Ita B.I., Edobor-Osoha A., Ehi-Eromosele F.E. (2016). Low-temperature solution combustion synthesis and magneto-structural characterization of polycrystalline  $\text{La}_{1-x}\text{Ag}_x\text{MnO}_3$  ( $y \leq x$ ) Manganites. *International Journal of Self-Propagating High-Temperature Synthesis*. 25, 1: 23–29.
7. Chirakadze A., Jishiashvili D., Shiolasvili Z., Petriashvili G., Chubinidze K. (2020). Development and testing of combined nano-based liquids for treatment of cancer cells based on nanoparticles with a therapeutic Curie temperature and liquid crystals: Georgian Experience. *Abstracts of International Conferences & Meetings (AICM)*. Krispon Advancing Science. Edinburgh.
8. Chirakadze A., Jishiashvili D., Petriashvili G., Mitagvaria N. (2020). Development and toxicity testing of nanoliquids for cancer treatment utilizing the phosphatized nanoparticles and liquid crystals with controlled release. *Abstracts of International Conferences & Meetings (AICM)*. Krispon Advancing Science. Edinburgh.
9. Chirakadze A., Jishiashvili D., Mitagvaria N., Lazrishvili I., Shiolasvili Z., Jishiashvili A., Makhatadze N., Buachidze Z., Khuskivadze N. (2019). Studies of the comparatively low-temperature synthesis and

preliminary toxic characteristics of silver doped lanthanum manganite nanoparticles using conventional and microwave heating. Proceedings of MTP: Modern Trends in Physics. Baku. 01-03 May. 47-51.

10. Mitagvaria N., Lazrshvili I., Devdariani M., Davlianidze L., Nebieridze M., Saginadze N., Kvachakidze I., Gumberidze L., Sikharulidze N. (2015). Hormesis - a basis for homeostasis in the presence of stressors. An example of hyperthermic stress. Journal of Biological Physics and Chemistry. 15: 187-193.
11. Decker M.J., Conrad K., Strohl P. (1989). Noninvasive oximetry in the rat. Biomed Instrum. Technol. 23, 3: 222-228.
12. Gurunathan S., Kang M.H., Qasim M., Kim J.H. (2018). Nanoparticle-mediated combination therapy: two-in-one approach for cancer. Int. J. Mol. Sci. 19, 3264; doi: 10.3390/ijms19103264.
13. Petriashvili G., Devadze L., Zurabishvili T., Sepashvili N., Chubinidze K. (2016). Light controlled drug delivery containers based on spiropyran doped liquid crystal microspheres. Biomedical Optics Express. 7, 2; 442-447.
14. Yagawa Y., Tanigawa K., Kobayashi Y., Yamamoto M. (2017). Cancer immunity and therapy using hyperthermia with immunotherapy, radiotherapy, chemotherapy and surgery. J. Cancer Metastasis Treat. 3: 218-230.

**3.6.1.2. Information about artificially intelligent nanoarray for detection of various diseases in nanomedicine.** /V.M. Aroutiounian/. Armenian Journal of Physics. – 2021. – vol. 14. – #3. – pp. 148-150. – eng.; abs.: eng.

DOI: <https://doi.org/10.52853/18291171-2021.14.3-148>

In brief is summarized information about artificially intelligent nanoarray for detection of various diseases in nanomedicine. Ref. 20.

**Keywords:** artificially intelligent nanoarray, detection of various diseases, nanomedicine

**References:**

1. V.M. Aroutiounian, Journal of Nanomedicine & Nanotechnology 11 (2020) 1.
2. Ji.W. Yoon, J.H. Lee, Lab on a chip. 1 (2017) 8.
3. V.M. Aroutiounian, J. Cont. Phys.-Arm. Acad. Sci. 54 (2009) 324.
4. N. Nasiri, Ch. Clark, Biosensors 9 (2019) 43.
5. M. Nakhleh, Y. Broza, H. Haick, Nanomedicine 9 (2014) 1991.
6. Y.Y. Broza, H. Haick, Nanomedicine 8 (2013) 785.
7. R. Vishinkin, H. Haick, Small 11 (2015) 6142.
8. G. Konvalina, H. Haick, Acc. Chem. Res. 47 (2014) 66.
9. P.I. Gouma, K. Kalyanasundaram, Appl. Phys. Lett. 93 (2008) 244102.
10. W. pel, Sens. Actuators B 4 (1991) 7.
11. N. Shehada, J.C. Cancilla, J.S. Torrecilla et al, ACS Nano 10 (2016) 7047.
12. B. Wang, J.C. Cancilla, J.S. Torrecilla et al, Nano Lett. 14 (2014) 933.
13. M.K. Nakhleh, S. Baram, R. Jerjes et al, Adv. Mater. Technologies 1 (2016) 1600132.
14. E. Homede, M. Abo Jabal, R. Ionescu et al, Adv. Funct. Mater. 26 (2016) 6359.
15. N. Shehada, G. r nstrup, K. Funka et al, Nano Lett. 15 (2015) 1288.
16. B. Wang, T.-P. Huynh, W. Wu et al, Adv. Mater. 28 (2016), 4012.
17. R. Ionescu, Y. Broza, H. Shaltieli et al, ACS Chem. Neurosci. 2 (2011) 687.
18. M.K. Nakhleh, H. Amal, R. Jerjes et al, ACS Nano 11 (2017) 112.
19. G. Peng, U. Tisch, O. Adams et al, Nat. Nanotechnol. 4 (2009) 669.
20. E. Dovgolevsky, G. Konvalina, U. Tisch, H. Haick, J. Phys. Chem. C 114 (2010) 14042.

## 6.2. Medical Chemistry

**3.6.2.1. Pseudoprotein-based nanoparticles show promise as carriers for ophthalmic drug delivery.** /Tem. Kantaria, Ten. Kantaria, S. Kobauri, W. Zhang, N. Eter, P. Heiduschka, A. Kezeli, G. Chichua, D. Tugushi, R. Katsarava/. *Annals of Agrarian Science*. – 2020. – v. 18. – #1. – pp. 43-53. – eng.; abs.: eng.

Drug delivery used to treat ocular disease still poses a challenge to modern ophthalmology. Well-established intravitreal injections imply discomfort to the patients and risk of ocular complications. Therefore, opportunities to deliver drugs by topical administration are investigated thoroughly. Despite its seemingly easy accessibility, the eye is well protected by efficient mechanisms that rapidly remove drugs after instillation on the eye surface. Hence, eye drops are less effective for the treatment of various diseases, which necessitates a risk-containing procedure of intravitreal injection. One of the rational ways to overcome the problem is the application of drug-loaded polymeric nanoparticles (NPs) that are able to penetrate through ocular barriers when administered topically. Pseudo-proteins (PPs) - amino acid-based biodegradable polymers are one of the most suitable materials for the design of drug delivering NPs. One of the most important features of such kind of nanovehicles is "disappearance" from the body after their function is fulfilled. We have prepared biodegradable NPs of various types by nanoprecipitation of the PEA-class of PP composed of L-leucine, 1,6-hexanediol and sebacic acid (8L6). The originally designed arginine-based cationic PEA and comb-like PEA containing lateral PEG-2000 chains along with 8L6 anchoring fragments in the backbones were used to construct positively charged and PEGylated NPs. The NPs were loaded with fluorescein diacetate (FDA) as a fluorescent probe to detect if the NP penetrated through the ocular barriers. A preliminary in vivo study on intraocular infiltration of the NPs has been done using wild-type C57BL/6 mice. After penetrating into the cellular lysosomes, FDA probes became visible due to the hydrolysis of the diacetate groups, thus allowing for the detection of the NPs as tiny fluorescent spots inside the tissues. One day after administration, fluorescent dots were found at various sites - always in the peripheral cornea and the sclera, and in different layers of the outer retina depending on the type of NPs used. Four days after administration, fluorescent dots were still visible in the peripheral cornea and the sclera with some of the NPs. These results show that the new type of NPs infiltrate the ocular tissues after topical administration and are taken up by the cells. This raises hope that the NPs may be useful carriers for ocular delivery of therapeutic agents. Fig. 4, Tab. 3, Ref. 35.

**Keywords:** biodegradable polymers, pseudo-proteins, nanoparticles, biodegradable surfactant, PEGylation, ocular penetration

### References:

1. S. Resnikoff, D. Pascolini D. Etya'ale et al. Global data on visual impairment in the year 2002. *Bull. World Health Organ.* 82 (2004) 844–851.
2. R. Gaudana, H.K. Ananthula, A. Parenky et al. Ocular drug delivery. *AAPS J.* 12 (2010) 348–360.
3. A. Urtti, Challenges and obstacles of ocular pharmacokinetics and drug delivery, *Adv. Drug Deliv. Rev.* 58 (2006) 1131–1135.
4. R. Herrero-Vanrell, M. Vicario-de-la-Torre, V. Andrés-Guerrero et al. Nano and microtechnologies for ophthalmic administration, an overview. *J. Drug Deliv. Sci. Technol.* 23 (2013) 27.
5. E.M. Del Amo A. Urtti, Current and future ophthalmic drug delivery systems. A shift to the posterior segment. *Drug Discov. Today.* 13 (2008) 135–143.
6. M. Kropp K.M., Morawa G., Mihov A.K., Salz N., Harmening A. Franken, A. Kemp A.A., Dias, J. Thies, S. Johnen, G. Thuman. Biocompatibility of Poly(ester amide) (PEA) Microfibrils in Ocular Tissues. *Polymers* 6 (2014) 243-260.
7. V. Andrés-Guerrero M. Zongc, E. Ramsay et al., Novel biodegradable polyesteramide microspheres for controlled drug delivery in Ophthalmology. *J. Contr. Release* 211 (2015) 105–117.

8. L.L. Zhang, C.D. Xiong, X. M. Deng. Biodegradable polyester blends for biomedical application. *J. Appl. Polym. Sci.* 56 (1995) 103–112.
9. D. Eglin, D. Mortisen, M. Alini, Degradation of synthetic polymeric scaffolds for bone and cartilage tissue repairs. *Soft Matter* 5 (2009) 938–947.
10. K. Hemmrich, J. Salber M., Meersch U., Wiesemann T., Gries N., Pallua D. Klee, Three dimensional nonwoven scaffolds from a novel biodegradable poly(ester amide) for tissue engineering applications. *J. Mater. Sci. Mater. Med.* 19 (2008) 257–267.
11. P. Karimi A.S., Rizkalla K., Mequanint, Versatile Biodegradable Poly(ester amide)s Derived from  $\alpha$ -Amino Acids for Vascular Tissue Engineering. *Materials* 3 (2010) 2346–2368.
12. D.K. Knight, E.R. Gillies, K. Mequanint, Strategies in Functional Poly(ester amide) Syntheses to Study Human Coronary Artery Smooth Muscle Cell Interactions. *Biomacromolecules*. 12 (2011) 2475–2487.
13. S.P. Higgins, A.K. Solan, L.E. Niklason. Effects of polyglycolic acid on porcine smooth muscle cell growth and differentiation. *J Biomed Mater Res A.* 67 (2003) 295–302.
14. S. Kobauri, Tem. Kantaria, N. Kupatadze, N. Kutsiava, D. Tugushi, R. Katsarava. Pseudo-proteins: A new family of biodegradable polymers for sophisticated biomedical applications. *Nano technology & nano science J.* 1 (2019). 37-42.
15. R. Katsarava, Ten. Kantaria, S. Kobauri. Pseudo-proteins and related synthetic amino acid based polymers (Review). *Journal of Materials Education* (in press).
16. N. Zavrashvili, J. Puiggali, R. Katsarava. Synthetic analogues of proteins – poly (amino acid)s, pseudo-poly(amino acid)s. poly(depsipeptide)s, and pseudoproteins. *Current Pharmaceutical Design* (in press).
17. G. Jokhadze, M. Machaidze, H. Panosyan, C.C. Chu, R. Katsarava. Synthesis and characterization of functional elastomeric poly (ester amide) co-polymers. *J. Biomater. Sci. Polym. Ed.*, 18 (2007). 411. C.C. Chu/. R. Katsarava. Elastomeric functional biodegradable copolyester amides and copolyester urethanes. (accessed on 5 August 2008) <http://www.google.tl/patents/US7408018>.
18. A. Zimmer, J. Kreuter. Microspheres and nanoparticles used in ocular delivery systems. *Adv. Drug Delivery Reviews.* 16 (1995). 61-73.
19. S.K. Sahoo, F. Dilnawaz, S. Krishnakumar. Nanotechnology in ocular drug delivery. *Drug Discovery Today.* 13 (2008). 144-151. A. Bochot, E. Fattal, V. Boutet, J.R. Deverre, et al. Intravitreal delivery of oligonucleotides by sterically stabilized liposomes. *Invest. Oph-Annals of Agrarian Science* 18 (2020) 43–53. T. Kantaria et al. 53thalmol. Vis. Sci. 43 (2002). 253–259. J.M. Irache, M. Merodio, A. Arnedo et al. Albumin nanoparticles for the intravitreal delivery of anticytomegaloviral drugs. *Mini Rev. Med. Chem.* 5 (2005). 293–305. R. Pignatello, C. Bucolo, G. Spedalieri et al. Flurbiprofen-loaded acrylate polymer nanosuspensions for ophthalmic application. *Biomaterials*, 23 (2002) 3247–3255. V. Vidmar, S. Pepelnjak, J. Jalseniak. The in vivo evaluation of poly(lactic acid) microcapsules of pilocarpine hydrochlo-ride. *J. Microencapsulation.* 2 (1985). 289-292.
20. R. Katsarava, Z. Gomurashvili. Biodegradable polymers composed of naturally occurring  $\alpha$ -amino acids. in: A. Lendlein, A. Sisson (Eds.), *Handbook of Biodegradable Polymers—Isolation. Synthesis. Characterization and Applications.* Wiley-VCH Verlag GmbH & Co. KGaA: Weinheim. Germany. 2011.
21. Tem. Kantaria, Teng. Kantaria, S. Kobauri, M. Ksovreli, T. Kachlishvili, N. Kulikova, D. Tugushi, R. Katsarava. Biodegradable nanoparticles made of amino acid based ester polymers: preparation, characterization, and in vitro bio-compatibility study. *Appl. Sci.* 6 (2016). 444. doi:10.3390/app6120444
22. T. Memanishvili, N. Zavrashvili, N. Kupatadze, D. Tugushi, M. Gverdtsiteli, V.P. Torchilin, C. Wandrey, L. Baldi, S.S. Manoli, R. Katsarava. Arginine-based biodegradable ether-ester polymers of low cytotoxicity as po-tential gene carriers. *Biomacromolecules* 15 (2014). 2839-2848.
23. Tem. Kantaria, Ten. Kantaria, S. Kobauri, M. Ksovreli, T. Kachlishvili, N. Kulikova, D. Tugushi, R. Katsarava. A new generation of biocompatible nanoparticles made of resorbable poly(ester amide)s, *J. Annals of Agrarian Science.* 17 (2019). 49-58.

24. C. Le Boultais, L. Acar H., Zia P.A., Sado T., Needham, R. Leverage. Ophthalmic Drug Delivery Systems—Recent Advances. *Prog. Retin. Eye Res.* 17 (1998). 33–58.
25. M. Mudgil, N. Gupta M., Nagpal, P. Pawar. Nanotechnology: a new approach for ocular drug delivery system. *Int. J. Pharm. Pharm. Sci.* 4 (2012) 105–112.
26. S. Laurent, L.H. Yahia. Protein corona: Applications and challenges. Springer Series in Biophysics. Martinac B. editor. 2013. pp 45-63.
27. B. Sahoo, M. Goswami, S. Nag et al. Spontaneous formation of a protein corona prevents the loss of quantum dot fluorescence in physiological buffers. *Chem Phys Lett.* 445 (2007) 217.
28. C. Giannavola, C. Bucolo, A. Maltese et al. Influence of Preparation Conditions on Acyclovir-Loaded Poly-d,l-Lactic Acid Nanospheres and Effect of PEG Coating on Ocular Drug Bioavailability. *Pharmaceutical Research.* 20 (2003) 584-590.
29. P. Breeuwer, J.L. Drocourt, N. Bunschoten, M.H. Zwietering, F.M. Rombouts, T. Abee. Characterization of Uptake and Hydrolysis of Fluorescein Diacetate and Carboxy fluorescein Diacetate by Intracellular Esterases in *Saccharomyces cerevisiae*. Which Result in Accumulation of Fluorescent Product. *Appl. Environ. Microbiol.* 61 (1995) 1614–1619.
30. R. Katsarava, N. Kulikova, J. Puiggali, Amino acid based biodegradable polymers—promising materials for the applications in regenerative medicine. *J.J. Regener. Med.* 1 (2016) 012.
31. S. Laurent L.H. Yahia. Protein corona: Applications and challenges. Springer Series in Biophysics. Martinac B. editor. 2013. pp 45-63.
32. B. Sahoo, M. Goswami, S. Nag et al. Spontaneous formation of a protein corona prevents the loss of quantum dot fluorescence in physiological buffers. *Chem Phys Lett.* 445 (2007) 217.
33. C. Giannavola, C. Bucolo, A. Maltese et al. Influence of Preparation Conditions on Acyclovir-Loaded Poly-d,l-Lactic Acid Nanospheres and Effect of PEG Coating on Ocular Drug Bioavailability. *Pharmaceutical Research.* 20 (2003) 584-590.
34. P. Breeuwer, J.L. Drocourt, N. Bunschoten, M.H. Zwietering, F.M. Rombouts, T. Abee. Characterization of Uptake and Hydrolysis of Fluorescein Diacetate and Carboxy fluorescein Diacetate by Intracellular Esterases in *Saccharomyces cerevisiae*. Which Result in Accumulation of Fluorescent Product. *Appl. Environ. Microbiol.* 61 (1995) 1614–1619.
35. R. Katsarava, N. Kulikova, J. Puiggali, Amino acid based biodegradable polymers—promising materials for the applications in regenerative medicine. *J.J. Regener. Med.* 1 (2016) 012.

### 3.6.2.2. Biopharmaceutical understanding of formulation preparation variability of PLGA nanoparticles loaded with erysimum extract. /L. Ebralidze, A. Tsertsvadze, L. Bakuridze, D. Berashvili, A. Bakuridze/. Georgian Medical News (GMN). – 2021. – #2(311). – pp. 173-177. – eng., abs.: eng., geo., rus.

The purpose of this study was to evaluate effect of process and formulation variables on the preparation of Erysimum extract loaded PLGA nanoparticles. The influence of the various biopharmaceutical factors such as type of organic solvent, type and concentration of surfactant, polymer concentration in the organic phase, ratio of organic phase and water phase were studied. Modified emulsification solvent evaporation method was used for preparation of nanoparticles. Based on the performed experiments optimal formulation of nanocomposite is suggested. Nanoparticle size, size distribution and entrapment efficiency were determined. Among five non-ionic surfactants polyvinyl alcohol provided more stable nanocomposite. Influence mechanisms of different surfactants on nanoparticle formation are provided. Water miscible organic solvent, acetone obtained 232 nm nanoparticles with improved size distribution. Entrapment efficiency was increased to 73% by reducing ratio of organic and water phases. Based on experiments nanoparticles with stable, reproducible properties are fabricated. Fig. 9, Tab. 4, Ref. 9.

**Keywords:** polymeric nanoparticle, PLGA, formulation variables, endemic plant species

## References:

1. E. Sah and H. Sah. Recent Trends in Preparation of Poly (lactide-coglycolide) Nanoparticles by Mixing Polymeric Organic Solution with Antisolvent. vol. 2015. 2015.
2. B. L. Banik, P. Fattahi, and J. L. Brown. Polymeric nanoparticles: The future of nanomedicine. Wiley Interdiscip. Rev. Nano-medicine Nanobiotechnology. vol. 8. no. 2. pp. 271–299, 2016.
3. W. McCarthy. Polymeric Drug Delivery Techniques. Aldrich Mater. Sci. no. 5. pp. 2–5, 2016.
4. C.I.C. Crucho and M.T. Barros. Polymeric nanoparticles: A study on the preparation variables and characterization methods. Mater. Sci. Eng. C. vol. 80. pp. 771–784. 2017.
5. S.M.I. Mors. Review Article Role of Surfactants in Nanotechnology and Their Applications. vol. 3. no. 5. pp. 237–260. 2014.
6. M. Silva, A. Santini, and E.B. Souto. Polymeric Nanoparticles: Production. 2020.
7. Rao J.P., Geckeler K. E. Polymer nanoparticles: Preparation techniques and size-control parameters. Prog. Polym. Sci. vol.36. no. 7. pp. 887–913. 2011.
8. R. Article. International Journal of Pharmacology and Review Article Recent biomedical applications and patents on biodegradable polymer. pp. 30–42. 2014.
9. Türk C.T.Ş., Bayindir Z.S., Badilli U. Preparation of polymeric nanoparticles using different stabilizing agents 2009; vol. 38. no. 4. 257–268.

**3.6.2.3. Formulation thermoresponsive nanocomposite hydrogel with embedded PLGA nanoparticles containing cytotoxic agent.** /L. Ebralidze, A. Tsertsvadze, L. Bakuridze, D. Berashvili, A. Bakuridze/. Georgian Medical News (GMN). – 2021. – #3(312). – pp.133-138. – eng., abs.: eng., geo., rus.

The aim of the study was to develop and characterize the nanocomposite in-situ hydrogel as local drug delivery system of cytotoxic agent. In-situ hydrogel consisting of 15% thermosensitive (Poloxamer 407) and 1% mucoadhesive (sodium alginate) polymers was selected as the optimal formulation by the conducted studies. The influence of nanoparticle concentration on gelation time and temperature has been experimentally established. As a result, the optimum concentration of nanoparticles (5%) is selected, which does not alter the gel forming process. The resulting nanocomposite hydrogel was characterized through Fourier transform infrared spectroscopy (FT-IR), scanning electron microscopy (SEM), rotational viscometer (LV DV- 1T). FT-IR spectra confirmed the PLGA nanoparticles presence within the hydrogel matrix through the absorption peak located at  $1750\text{ cm}^{-1}$ . SEM images allowed observing the nanoparticles to be homogeneously dispersed. The release pattern of the active substance from the nanocomposite hydrogel is following: at 72 h, 64% and 78% of the active substance were released into the phosphate buffer and cell culture area, respectively. Irritation test on hen's egg model revealed that formulated nanocomposite hydrogel did not show damage of vascular system. Fig. 10, Tab. 1, Ref. 9.

**Keywords:** Nanocomposite, thermosensitive hydrogel, PLGA nanoparticles

## References:

1. Lu Y., Mahato R.I. Pharmaceutical perspectives of cancer therapeutics. 2009.
2. GLOBOCAN. The Global Cancer Observatory - All cancers. Int. Agent Res. Cancer – WHO 2020. 419: 199–200.
3. Vashist A. et al. Nanocomposite Hydrogels: Advances in Nanofillers Used for Nanomedicine. Gels 2018. 4(3): 75. doi: 10.3390/gels4030075.
4. Rafieian S., Mirzadeh H., Mahdavi H., Masoumi M.E. A review on nanocomposite hydrogels and their biomedical applications. IEEE J. Sel. Top. Quantum Electron. 2019. 26(1): 154–174, 2019. doi: 10.1515/secm-2017-0161.

5. Matsumura K. Special issue: Nanocomposite hydrogels for biomedical applications. Appl. Sci. 2020. 10(1): 15–16, 2020. doi: 10.3390/app10010389.
6. Nirmal H.B., Bakliwal S.R., Pawar S.P. In-Situ gel: New trends in controlled and sustained drug delivery system. Int. J. PharmTech Res. 2010. 2(2): 1398–1408.
7. Bakuridze A., Beridze D., Jokhadze M., Metreveli M. The Study of Ajara and Ajara-Lazica Endemics on the Content of Biologically Active Compound Coumarin 2016. 4: 76–83.
8. Beridze D., Jokhadze M., Bakuridze A., Metreveli M., Manvelidze Z. Gas chromatography - mass spectrometry (GC-MS) analysis of bioactive compounds of Ajara and Ajara-Lazica endemic species. International Journal of Current Research 2016. vol. 8. issue 09: 38939-38944.
9. Metreveli M. North American Multi-Purpose Dear Introducent Plants in Western Georgia Humid Subtropical Conditions. Am. J. Environ. Prot. 2015. 4(3): 168. 2015, doi: 10.11648/j. ajep.s.2015040301.36.



### Author Search

Abdalova S.R. 3.1.2.8.  
Aghamalyan N.R. 3.2.1.1.  
Antia G. 3.4.3.1.  
Aroutiounian V.M. 3.1.2.4., 3.5.1.1., 3.6.1.2.  
Avjyan K.E. 3.1.2.6.  
Badalyan G.R. 3.2.1.1.  
Baghramyan V.V. 3.2.1.1.  
Bakhtiyarov S. 3.1.2.1.  
Bakuradze M. 3.4.2.1.  
Bakuridze A. 3.6.2.2., 3.6.2.3.  
Bakuridze L. 3.4.2.1., 3.6.2.2., 3.6.2.3.  
Balakhashvili M. 3.4.2.3.  
Bayramova I.V. 3.1.2.8.  
Berashvili D. 3.4.2.1., 3.6.2.2., 3.6.2.3.  
Bobrov M. 3.4.1.1.  
Buachidze Z. 3.4.2.2., 3.6.1.1.  
Burdjanadze M. 3.4.3.1.  
Chichua G. 3.6.2.1.  
Chirakadze A. 3.4.2.2., 3.6.1.1.  
Chubinidze K. 3.4.2.2., 3.6.1.1.  
Dabaghyan G.A. 3.1.2.6.  
Darakhvelidze N. 3.4.2.3.  
Darsavelidze G. 3.1.2.1.  
Davlianidze L. 3.6.1.1.  
Devdariani M. 3.6.1.1.  
Dubovyy O. 3.4.1.1.  
Dvali N. 3.4.2.2., 3.6.1.1.  
Dzigrashvili T. 3.1.2.1.  
Ebralidze L. 3.6.2.2., 3.6.2.3.  
Eter N. 3.6.2.1.  
Gadirova E.M. 3.2.2.1.  
Gambaryan K.M. 3.1.2.4.  
Gulieva T.M. 3.1.2.9.  
Gventsadze D. 3.1.2.1.  
Gventsadze L. 3.1.2.1.  
Harutyunyan S.L. 3.1.2.2.  
Harutyunyan V.A. 3.1.2.5.  
Heiduschka P. 3.6.2.1.  
Hovsepyan R.K. 3.2.1.1.  
Ismayilov I.A. 3.1.2.8.  
Ismayilzade A.D. 3.1.2.8.  
Jafarova S.T. 3.4.2.4.  
Jishiashvili A. 3.4.2.2., 3.6.1.1.  
Jishiashvili D. 3.4.2.2., 3.6.1.1.

Kacharava T. 3.4.4.1.  
Kakhramanov N.T. 3.1.2.8.  
Kantaria Tem. 3.6.2.1.  
Kantaria Ten. 3.6.2.1.  
Karpechenko A. 3.4.1.1.  
Katsarava R. 3.6.2.1.  
Kezeli A. 3.6.2.1.  
Khechoyan D.G. 3.1.2.3.  
Khomeriki I. 3.4.2.2., 3.6.1.1.  
Kinkladze V. 3.4.2.3.  
Knyazyan N.B. 3.2.1.1.  
Kobauri S. 3.6.2.1.  
Kokanyan E.P. 3.1.2.7.  
Kokanyan N.E. 3.1.2.7.  
Korakhashvili L. 3.4.4.1.  
Korakhashvili A. 3.4.4.1.  
Kovziridze Z. 3.4.2.3.  
Kukava T. 3.1.2.1.  
Kurashvili I. 3.1.2.1.  
Kutelia E. 3.1.2.1.  
Labartkava A. 3.4.1.1.  
Makhatadze N. 3.4.2.2.  
Makryha T. 3.4.1.1.  
Mammadyarova S.J. 3.4.2.5.  
Manukyan V.F. 3.1.2.2.  
Margaryan N.B. 3.1.2.7.  
Matevosyan L.M. 3.1.2.6.  
Mestvirishvili Z. 3.4.2.3.  
Mitagvaria N. 3.4.2.2., 3.6.1.1.  
Morozov V.F. 3.1.2.3.  
Mosidze E. 3.4.2.1.  
Mshvildadze M. 3.4.2.3.  
Mumladze N. 3.4.3.1.  
Nadaraia L. 3.1.2.1.  
Namazli U.V. 3.1.2.8.  
Nijaradze N. 3.4.2.3.  
Nikoghosyan G.H. 3.1.2.2.  
Nikoghosyan H.S. 3.1.2.2.  
Osipchik V.S. 3.1.2.8.  
Petriashvili G. 3.4.2.2., 3.6.1.1.  
Ragimova S.K. 3.1.2.10.  
Rukhadze L. 3.1.2.1.  
Sargsyan A.A. 3.2.1.1.  
Sarkisyan H.A. 3.1.2.5.  
Sharashenidze T. 3.4.3.1.  
Shiolashvili Z. 3.4.2.2.

## Authors Search

---

Tabatadze G. 3.4.2.3.

Tsertsvadze A. 3.6.2.2., 3.6.2.3.

Tsintskaladze G. 3.4.3.1.

Tsursumia I. 3.4.2.1.

Tsursumia O. 3.1.2.1.

Tugushi D. 3.6.2.1.

Zautashvili M. 3.4.3.1.

Zhang W. 3.6.2.1.