

4. NANOTECHNOLOGY

4.1. Materials and Structures

3.4.1.1. Research of the possibility of nanostructuring functional materials by pre-recrystallization heat treatment. /O. Dubovyy, A. Karpechenko, T. Makryha, M. Bobrov, A. Labartkava, A. Labartkava/. Bulletin of the Georgian National Academy of Sciences. – 2021. – v. 15. – #1. – pp. 45-51. – eng.; abs.: eng., geo.

The paper is devoted to the research of the possibility of nanostructuring of functional materials, steels and ceramic sprayed coatings by pre-recrystallization heat treatment. The effect of the size of coherent X-ray scattering regions, the number of nanostructured elements and the subgrains misorientation angle on the physical and mechanical properties of technically pure iron and steel were experimentally studied. The possibility of thermal stabilization of a 62% polygonization nanoscale substructure during pre-recrystallization heat treatment at 500°C after combined plastic deformation is shown. A combination of 30% dynamic and 30% static deformations makes it possible to use such treatment in industry. Fig. 2, Tab. 3, Ref. 16.

Keywords: physical and mechanical properties, thermal sprayed coatings, deformation of steels, pre-recrystallization heat treatment

References:

1. Zhdanov O.O. (2015). Zakonomirnosti vplyvu peredrekryalizacijnoi termichnoi obrobky na fizyko-mehanichni vlastyvoli deformovanyh stalej. Avtoref. dys. na zdobuttja nauk. stupenja kand. tehn. nauk.: spec. 05.02.01 – “Materialoznavstvo”. Herson (in Ukrainian).
2. Das D., Samanta A., Chattopadhyay P.P. (2006). Deformation behavior of bulk ultrafine grained copper prepared by sub-zero rolling and controlled recrystallization. Materials & Manufacturing Processes. 21 (7): 698-702.
3. Sun S.L., Huang Q.X., He W.W., Zhang M.G. (2014). Workability behavior of 9% Cr ferritic/martensitic steel. Materials & Manufacturing Processes. 29 (10): 1190-1196.
4. Jurkova O.I. (2011). Strukturnyj stan i mehanichni vlastyvoli plastychno deformovanogo zaliza. Metaloznavstvo ta obrobka metaliv. 1: 3-9 (in Ukrainian).
5. Valiev R.Z. (2007) Ob”emnye nanostrukturnye metallicheskie materialy: poluchenie. struktura i svoïstva. M. (in Russian).
6. Valiev R.Z., Aleksandrov I. V. (2000). Nanostrukturnye materialy, poluchennye intensivnoï plasticheskoï deformatsiei. M. (in Russian).
7. Jurkova O.I., Karpov R.V., Kljagin Je.O. (2010). Osoblyvosti formuvannja nanokrystalichnoi struktury v α -zalizi pry deformacii tertjam. Metaloznavstvo ta obrobkametaliv. 1:12-16 (in Ukrainian).
8. Alymov M.I., Averin S.I., Shustov V.S., Gordopolova L.V. (2013) Rol’ mekhanizmov plasticheskoï deformatsii pri vysokotemperaturnom spekanii chastits. Pis’ma o materialakh, 3 (4): 315-317 (in Russian).
9. Dubovyy O.M., Lebedeva N.Ju., Jankovec T.A. (2010). Vplyv peredrekryalizacijnoi termichnoi obrobky na fizyko-mehanichni vlastyvoli napylenyh pokryttiv ta deformovanyh metaliv ta splaviv. Metaloznavstvo ta obrobka metaliv, 3: 7-10 (in Ukrainian).
10. Dubovyy O.M., Bondarenko O.V., Zhdanov O.O., Zhyzhko O.V., Bobrov M.M., Galkina T.S. (2010). Doslidzhennja mozhlyvostej pidvyshhennja fizyko-mehanichnyh vlastyvostej deformovanyh metaliv i splaviv termichnoju obrobkoju. Obrobka materialiv u mashynobuduvanni. Mykolaiv: Admiral Makarov National University of Shipbuilding: 69-79 (in Ukrainian).
11. Dubovyy O.M., Karpechenko A.A., Bobrov M.M., Zhdanov O.O., Makruha T.O., Nedelko Ju.Je. (2017). Formuvannja nanorozmirnoi poligonizacijnoi substruktury ta ii vplyv na fizyko-mehanichni vlastyvoli metaliv, stopiv i naporoshenyh pokryttiv. Metallofizyka y novejshe tehnology, 39(2): 209-243 (in Ukrainian).

12. Patent №95378 UA МПК (2009) S21D8/00, C22F 1/00. Sposib deformacijno-termichnoi obrobky metaliv ta splaviv. Dubovyy O.M., Jankovec T.A., Lebedjeva N.Ju., Kazymyrenko Ju.O., Zhdanov O.O., Bobrov M.M. Bulletin №14, July, 2011.
13. Dubovyy O.M., Karpechenko A.A., Bobrov M.M., Labartkava Al.V., Nedelko Ju.Je., Lyman O. O. (2019). Pidvyshhennja fizyko-mehanichnyh ta ekspluatacijnyh vlastyvostej elektrodugovyh ta plazmovykh pokryttiv formuvannjam termichno stabilnoi zdribnenoj i nanorozmirnoj substrukturny. Metalofiz. novitni tehnologii, 41(4): 461-480 (in Ukrainian).
14. Gorelik S.S., Dobatkin S.V., Kaputkina L.M. (2005). Rekristallizatsiia metallov i splavov. M. MISIS (in Russian).
15. Dubovyy O.M., Lju Shen, Makruha T.O. (2017). Vplyv kombinovanogo deformuvannja na termichnu stabilnist poligonizacijnoi substrukturny zaliza, nikelju j stalej 20; 45. Zbirnik naukovih prac NUK. Mykolaiv: Admiral Makarov National University of Shipbuilding. 1: 39-47 (in Ukrainian).
16. Dubovyy O.M., Makruha T.O. (2018). Vplyv vydu kombinovanogo deformuvannja na poligonizacijnu substrukturny zaliza ta stali U8. Zbirnik naukovih prac NUK. Mykolaiv: Admiral Makarov National University of Shipbuilding. 3-4(474): 66-74 (in Ukrainian).

3.4.1.2. 6th International Conference - Nanotechnology. Book of Abstracts. – Tbilisi. – 4-7 October. – 2021. – pp. 116.

Book of Abstracts contains more than 100 abstracts of papers submitted to the 6th International Conference “Nanotechnology”, 4–7 October 2021, Tbilisi, Georgia (GTUnano2021) organized by the Georgian Technical University (GTU). The GTUnano2021 is held in memory of Prof. Alex Gerasimov, the initiator of GTU’s regular series of nanoconferences in Georgia. The 6th conference participants represent universities, institutes, research centers, etc. leading in the field of nanotechnology and nanosciences from 20 countries (Armenia, Azerbaijan, Belarus, China, Czech Republic, Georgia, Germany, Hungary, India, Iraq, Japan, Kazakhstan, Mexico, Poland, Russia, Serbia, Spain, Turkey, Ukraine, and United States of America). Fig. 43. Tab. 3, Ref. 197.

Keywords: 6th International Conference “Nanotechnology”, abstracts of papers, GTU, nanotechnology and nanosciences, conference participants

4.2. Obtaining Technologies

3.4.2.1. Receipt and technology of aloe floating tablets. /M. Bakuradze, L. Bakuridze, E. Mosidze, D. Berashvili, I. Tsurtsunia/. Collection of Scientific Works of TSMU. – 2020. – v. 54. – pp. 29-31. – geo.; abs.: geo., eng.

In regenerative medicine, plants may be used as an alternative transplantation source. There is plenty of structural similarity between the vegetative and animal tissues and similarly to the human blood vessels, the plant fibers may be used as nutrient carriers. In tissue engineering, a decellularized plant may become a unique, multifunctional medical device, a plant matrix, the so-called scaffold. The performed studies resulted in the optimal composition of plant decellularization agents. It was determined that the duration of the process significantly depends on the morphological, anatomic and histological peculiarities of the plant. The biopharmaceutical studies proved that the absorption method is the most appropriate for plant decellularization for it shortens the process and is easier to perform than the perfusion. Employment of the 0.1% methylene blue made it evident that the decellularization technology does not affect the integrity and conductivity of the matrix fibers. On the grounds of the biopharmaceutical studies, we provided the plant matrix made up of the silver bionanoparticles. The properties of the silver bionanoparticles in the matrix were studied by the transmission electron microscope. The size of silver bionanoparticles is 1000 nm. and those are distributed across the matrix. We studied the silver separation dynamics from the nanodesigned

plant matrix and determined that 70% of silver was separated after 12h exposition of the matrix. Tab. 1, Fig. 4, Ref. 5.

Keywords: regenerative medicine, alternative transplantation source, decellularization agents, plant matrix, silver, bionanoparticles, aloe tablets, morphological, anatomic and histological peculiarities

References:

1. Ржеусский С.Э. Валидация спектрофотометрической методики количественного определения наночастиц серебра в водных растворах // Вестник Фармации. 2019. N1. N. 21-25.
2. Vacanti J. Tissue engineering and regenerative medicine: from first principles to state of the art. *Journal of Pediatric Surgery*. 2010. 45(2):291–294.
3. Guyette J, et al. Bioengineering Human Myocardium on Native Extracellular Matrix. *Circulation Research*. 2016. 118(1):56–72.
4. Modulevsky D, Cuerrier C, Pelling A. Biocompatibility of Subcutaneously Implanted Plant-Derived Cellulose Biomaterials. *PLoS One*. 2016. 11(6):e0157894.
5. Gershlak J, et al. Crossing kingdoms: Using decellularized plants as perfusable tissue engineering scaffolds. *Biomaterials*. 2017. 125:13–22.

3.4.2.2. Microwave synthesis, characterization and testing of acute toxicity of boron nitride nanoparticles by monitoring of behavioral and physiological parameters. /A. Chirakadze, N. Mitagvaria, D. Jishiashvili, G. Petriashvili, N. Dvali, Z. Shiolashvili, K. Chubinidze, N. Makhatadze, A. Jishiashvili, Z. Buachidze, I. Khomeriki/. *Bulletin of the Georgian National Academy of Sciences*. - 2021. – v. 15. – #2. – pp. 120-126. – eng.; abs.: eng., geo.

Hexagonal boron nitride nanoparticles, nanosheets and nanotubes (BNNPs) are even more promising materials for biomedical application than carbon nanotubes (CNTs) and nanoparticles (CNPs) due to their negligible cytotoxicity. The reported research yielded in development and testing of two distinctive microwaves enhanced comparatively low-temperature methods of synthesis of the hexagonal boron nitride nanoparticles and nanosheets with reduced distortion of the crystal lattice, and an improved method of general toxicity testing of the developed nanomaterials utilizing continuous observation of behavioral effects in white rats in combination with blood oxygen saturation, systolic blood pressure and body temperature measurements in full agreement with the 4R principles of animal welfare in scientific research. The obtained results allow us to expect that the developed materials can be a good basis for developing highly effective modalities for anticancer (in combination with chemotherapy, hyperthermia and radiotherapy) and antiviral (in combination with chemotherapy and hyperthermia) treatment. Fig. 2, Ref. 14.

Keywords: cancer, low-temperature synthesis, microwave radiation, boron nitride, nanoparticles, turbostratic effect, behavioral testing

References:

1. Chirakadze A., Jishiashvili D., Shiolashvili Z., Petriashvili G., Chubinidze K. (2020). Development and testing of combined nano-based liquids for treatment of cancer cells based on nanoparticles with a therapeutic Curie temperature and liquid crystals: Georgian Experience. *Abstracts of International Conferences & Meetings (AICM)*. Krispon Advancing Science. Edinburgh.
2. Mitagvaria N., Lazrshvili I., Devdariani M., Davlianidze L., Nebieridze M., Saginadze N., Kvachakidze I., Gumberidze L., Sikharulidze N. (2015). Hormesis – a basis for homeostasis in the presence of stressors. An example of hyperthermic stress. *Journal of Biological Physics and Chemistry*. 15: 187-193.
3. Chirakadze A., Jishiashvili D., Mitagvaria N., Lazrshvili I., Shiolashvili Z., Jishiashvili A., Makhatadze N., Buachidze Z., Khuskivadze N. (2019). Studies of the comparatively low-temperature synthesis and

- preliminary toxic characteristics of silver doped lanthanum manganite nanoparticles using conventional and microwave heating. *Proceedings of MTP: Modern Trends in Physics*. 47-51, Baku.
4. Siegel R., Miller K. D., Jemal A. (2020). Cancer statistics, 2020. *CA: A Cancer Journal for Clinicians*. 21590, <https://doi.org/10.3322/caac.21590>. Accessed on April. 2, 2021.
 5. Jemal A., Bray F., Center M.M., Ferlay J., Ward E., Forman D. (2011). Global cancer statistics. *CA, Cancer Journal for Clinicians*. A 1, 61: 69–90.
 6. Petousis P., Naeim A., Mosleh A., Hsu W., Geffen D. John B. (2018). Evaluating the impact of uncertainty on risk prediction: towards more robust prediction models. *AMIA Annu. Symp. Proc.* 1: 1461–1470.
 7. Quante A.S., Ming C., Rottmann M., Engel J., Boeck S., Heinemann V., Westphale C.B. (2016). Projections of cancer incidence and cancer-related deaths in Germany by 2020 and 2030. *Cancer Med.* 5, 9: 2649–2656.
 8. Merlo A., Mokkapati V.R., Pandit S. and Mijakovic I. (2018). Boron nitride nanomaterials: biocompatibility and bio-applications. *Biomater. Sci.* 6: 2298-2311.
 9. Huang C., Chen X., Ye W. Ye, Hu J., Xu C., Qiu X. (2013). Stable colloidal boron nitride nanosheet dispersion and its potential application in catalysis. *Journal of Materials Chemistry A*. 1: 2192-2197.
 10. Shen T., Liu S., Yan W., Wang J. (2019). Highly efficient preparation of hexagonal boron nitride by direct microwave heating for dye removal. *Journal of Materials Science*. 54, 3: 8852-8859.
 11. Silly M.G., Jaffrennou P., Barjon J., Rosencher E. (2007). Luminescence properties of hexagonal boron nitride: cathodoluminescence and photoluminescence spectroscopy measurements. *Physical Review B*. 75, 8; DOI: <https://doi.org/10.1103/PhysRevB.75.085205>. Accessed on April, 2, 2021.
 12. Sun C., Guo C., Ma X., Xu L., Qian Y. (2009). A facile route to prepare boron nitride hollow particles at 4500C. *Journal of Crystal Growth*. 31: 3682-3686.
 13. Alkoy S., Toy C., Gönül T., Tekin A. (1997). Crystallization behavior and characterization of turbostratic boron nitride. *Journal of the European Ceramic Society*. 17: 1415-1422.
 14. Turkez H., Arslan M.E., Soenmez E., Acjikyildiz M., Tatar A., Geyikoglu F. (2019). Synthesis, characterization and cytotoxicity of boron nitride nanoparticles: emphasis on toxicogenomics. *Cytotechnology*. 71: 351–361, <https://doi.org/10.1007/s10616-019-00292-8>, Accessed on April, 2, 2021.

3.4.2.3. Obtaining of β -SiAlON nanocomposite with aluminothermal and nitrogen processes. /Z. Kovziridze, N. Darakhvelidze, N. Nijaradze, G. Tabatadze, M. Mshvildadze, Z. Mestvirishvili, M. Balakhashvili, V. Kinkladze/. *Ceramics and Advanced Technologies*. – 2020. – vol. 22. – #2(44). pp. 21-36. – geo.; abs.: geo., eng.

To obtain a composite in SiAlON- Al_2O_3 system and to study its properties. Obtaining the composite by metallothermic and nitrogenation methods. In the present work, the composite containing SiAlON is obtained through alum-thermal process, by the reactive sintering method in nitrogen medium, from the mixture of aluminosilicate raw material (Prosyanaya kaolin and Polog refractory clay - Ukraine), nanopowder of aluminum oxide (German company "ALCOA"), and metallic silicon with small additives of glass perlite Aragac (Armenia). The advantage of this method is that the aluminosilicate raw material decomposes during the heat treatment process and the alum-thermal-nitration process takes place at the same time, making it easier to open AlN and Al_2O_3 in the newly formed β - Si_3N_4 crystal lattice, which provides β -SiAlON generation at a relatively low temperature, 1250-13000°C. Corundum-SiAlON composite material is obtained by reactive sintering process at a temperature of 14500C. The corundum and sialon phases in the composite are confirmed by X-ray phase, spectral and electronmicroscopic analyzes. To obtain consolidated samples, the material obtained by reactive sintering was grounded in the attritor and hot pressed at 30 MPa and 1620° C and was kept at the final temperature - 7 min. The phase composition of the obtained samples remained unchanged after hot pressing, the density increased and the porosity dropped below 1%, accordingly the numerical values of the mechanical properties were increased: $\sigma_{\text{press.}}$ - 1600 MPa; $\sigma_{\text{bend.}}$ - 460 MPa; HV - 19.7 GPa. Obtained corundum-SiAlON composite with its physical-technical properties: porosity – 0-1%; density - 3.21 g/cm³; $\sigma_{\text{press.}}$ - 1923 MPa; $\sigma_{\text{bend.}}$ - 470

MPa; HV - 19.7 GPa, elasticity modulus - 22 GPa; dynamic hardness - 3214 N/mm²; chemical stability to sulfuric acid (density - 1.84 g/cm³) - 99.3%, to water - 99.8%. The obtained materials may be recommended in armor engineering, when measuring temperature in metals molten as protective coatings for the thermocouple, as well as in high-temperature furnace linings, as well as in clean processing operations as a metalworking cutting material. Fig. 5, Tab. 6, Ref. 29.

Keywords: β -SiAlON, corundum, reactive sintering, composite, properties, metalworking cutting material

References:

- Ekstrom T., Kall P.O., Nygren M., Olsson P.O. Dense Single-Phase Beta-SiAlON Ceramics by Glass Encapsulated Hot Isostatic Pressing. J. of mat. Sci. 1989. V. 24. p. 1853-1862.
- Rosenflanz A., I-Wei-Chen. Phase Relationships and Stability of α -SiAlON. J. Am. Ceram. Soc. 1999. V.82. №4. P. 25-28.
- Стрелов К. К. Теоретические основы технологии огнеупорных материалов: учеб. пособие для вузов / К. К. Стрелов, И. Д. Кашеев. 2-е изд. перераб. и доп. М. 1996. с. 608.
- Чухolina Л.Н. Способ получения порошка сиалона. <http://bd.patent.su/2378000> (18. 11.2012.)
- Zheng G, Zhao J., Gao Z., Cao Q. Cutting performance and wear mechanisms at Sialon-Si₃N₄ graded nano-composite ceramic cutting tools/ The International Journal of advanced Manufacturing Technology. 2012 V.58 , I. 1-4. P. 19-28.
- Tressler R. E. Theory and Experiment in Corrosion of Advanced Ceramics//Corrosion of Advanced Ceramics/ NATO ASI Series E: Applied Sciences/Ed. K.G. Nickel. The Netherlands. 1994. N267. p. 3-22.
- Piekarczyk J., Lis J., Bialoskorski J. Elastic Properties, Hardness and Indentation Fracture Toughness of beta-Sialons/ Key Engineering Materials. 1990. V. 89-91. p. 542-546.
- ზ. კოვზირიძე, ნ. ნიჟარაძე, ნ. დარახველიძე, გ. ტაბატაძე, ზ. მესტვირიშვილი. გეოპოლიმერის ბაზაზე აზოტის გარემოში მიმდინარე კარბო- და ალუმინთერმული პროცესები. საქართველოს კერამიკოსთა ასოციაციის ჟურნალი „კერამიკა“ ტ. 16. N1(31). 2014. გვ. 32-36
- ზ. კოვზირიძე, ნ. ნიჟარაძე, ნ. დარახველიძე, გ. ტაბატაძე, თ. ჭეიშვილი, ზ. მესტვირიშვილი, მ. მშვილდაძე, ე. ნიკოლეიშვილი. ნიტროალუმინთერმული პროცესებით სიალონების მიღება. საქართველოს კერამიკოსთა ასოციაციის ჟურნალი „კერამიკა“ ტ. 16. N2(32). 2014. გვ. 23-31.
- Kovziridze Z., Nijaradze N., Tabatadze G., Cheishvili T., Mestvirishvili Z., Nikoleishvili E., Mshvildadze M., Daraxvelidze N. Obtaining of Nanocomposites in SiC-SiAlON and Al₂O₃-SiAlON System by Alumothermal Processes. Journal of Electronics Cooling and Thermal Control, 2014, 4, Published Online December 2014 in SciRes. Pp.1-13 <http://www.scirp.org/journal/jectc> USA.
- Kovziridze Z., Nijaradze N., Tabatadze G., Daraxvelidze N., Mestvirishvili Z. Obtaining of SiAlONs via alumothermal and nitrogen processes. 14th International Conference of European Ceramic Society. 21- 25 June. Toledo. Spain. Poster 2348. 2015.
- Kovziridze Z., Nijaradze N., Tabatadze G., Daraxvelidze N., Mestvirishvili Z. Smart Materials in the SiAlON-SiC-Al₂O₃ System. Journal of Material Science and Engineering, International Conference and Expo on Ceramics. August 17-18. 2015. Chicago. USA.
- ზ. კოვზირიძე, ნ. ნიჟარაძე, ნ. დარახველიძე, გ. ტაბატაძე, ზ. მესტვირიშვილი. სიალონ-შემცველი კომპოზიტის მიღება ნიტროალუმინთერმული პროცესებით, რეაქციული შეცხოვისა და ცხელი დაწნეხის მეთოდით. საქართველოს კერამიკოსთა ასოციაციის ჟურნალი „კერამიკა“ ტ. 18. 1(35). 2016. გვ. 9-19.
- Kovziridze Z., Nijaradze N., Darakhvelidze N., Mestvirishvili Z. Smart Materials in the SiAlON-SiCAl₂O₃-TiB₂-ZrB₂ System. 2nd Annual world Congress of Smart Materials (WCSM-2016) 4-6 March. 2016. Singapore
- Kovziridze Z., Nijaradze N., Darakhvelidze N., Tabatadze G., Mestvirishvili Z., Nikoleishvili E., Mshvildadze M., Preparation of Composites by Nitro Aluminothemic Processes, over β -SiAlON Matrix in the SiAlON-SiC-

- Al₂O₃ System. Journal of Electronics Cooling and Thermal Control, Vol. 6 No. 2. Pub. Date: June 15. 2016. PP. 62-77.
16. Kovziridze Z., Nijaradze N., Darkhvelidze N., Tabatadze G., Mestvirishvili Z. Obtaining of nano composites via alum-thermal and nitrogen processes in the SiC-Si₃N₄-AlN-Al₂O₃-SiO₂. System. 15th Conference & Exhibition of the European Ceramic Society. Ecers 2017. July 9-13. 2017 / Budapest. Hungary.
 17. ზ. კოვზირიძე, ნ. ნიჟარაძე, ნ. დარახველიძე, გ. ტაბატაძე, თ. ჭეიშვილი, ზ. მესტვირიშვილი, მ. მშვილდაძე. კომპოზიტის მიღება მეტალოთერმული და აზოტირების პროცესებით Si-SiC-Al გეოპოლიმერის სისტემებში. საქართველოს კერამიკოსთა ასოციაციის ჟურნალი „კერამიკა და მოწინავე ტექნოლოგიები“ ტ. 19. 2(38). 2017. გვ. 33-52.
 18. Kovziridze Z., Nijaradze N., Tabatadze G., Cheishvili T., Mestvirishvili Z., Mshvildadze M., Darakhvelidze N. Kinkladze V. Obtaining Of SiAlON Composite via Metal-Thermal and Nitrogen Processes in the SiC-Si-Al-Geopolymer System. Journal of Electronics Cooling and Thermal Control. 2017. 7. 103- 122.
 19. Kovziridze Z., Nijaradze N., Darakhvelidze N., Tabatadze G., Mestvirishvili Z. Obtaining of composite via metal-thermal and nitrogen processes in the SiC-SiAl-geopolymer System. 7th International Congress on Ceramics – ICC7. Foz de Iguaçu. PR. Brazil. June 17-21. 2018.
 20. ზ. კოვზირიძე, ნ. ნიჟარაძე, ნ. დარახველიძე, გ. ტაბატაძე, მ. ბალახაშვილი. რეაქციული შეცხოვის მეთოდით სიალონშემცველი კომპოზიტების მიღება SiC-B₄C-Si-Al-Al₂O₃ სისტემაში მეტალოთერმული და აზოტირების პროცესებით. საქართველოს კერამიკოსთა ასოციაციის ჟურნალი „კერამიკა და მოწინავე ტექნოლოგიები“ ტ. 20. 2(40). 2018. გვ. 13-17.
 21. ზ. კოვზირიძე, ნ. ნიჟარაძე, ნ. დარახველიძე, გ. ტაბატაძე, ზ. მესტვირიშვილი. კომპოზიტების ფაზური შედგენილობის შესწავლა SiC-B₄C-SiAl-Al₂O₃ სისტემაში. საქართველოს კერამიკოსთა ასოციაციის ჟურნალი „კერამიკა და მოწინავე ტექნოლოგიები“ ტ. 21. 1(41). 2019. გვ. 44-51.
 22. Z. Kovziridze, N. Nijaradze, N. Darakhvelidze, G. Tabatadze, Z. Mestvirishvili, M. Balakhashvili, M. Mshvildadze. Obtaining of the Composite of βSiAlON Matrix via Metal-Thermal and Nitrogen Processes in the B₄C-SiC-Al₂O₃-Si-Al-Carbon Fiber Geopolymer. System. XVI ECERS Conference and Exhibition of the European Ceramic Society. Torino. Italy. 16-20 June 2019. P.680. Abstract Book.
 23. Kovziridze Z., Nijaradze N., Darkhvelidze N., Tabatadze G., Mestvirishvili Z. Ceramic Composite in the SiC-SiAlON System. Euro Global Congress on Tychonix Nanotech. 2019 11-12 November. Valencia, Spain 2019.
 24. Kovziridze Z., Nijaradze N., Darakhvelidze N., Tabatadze G., Mestvirishvili Z. Composite in the SiAl₂O₃-BN-SiAlON System. 8th International Congress on Ceramics. August 23-28. Bexco. Busan. Korea.
 25. ზ. კოვზირიძე. მაკრომექანიკური მახასიათებლების ფორიან ფაზაზე დამოკიდებულების ფორმულა. Journal of the Georgian Ceramists Association. Ceramics and Advanced Technologies Vol. 20. 1(39). 2018. pp. 38-44. www.ceramics.gtu.ge
 26. Ковзиридзе З.Д. Разработка научных основ и технологии получения цельноановой и алюмосиликатной керамики с использованием барита и перлита. Диссертация на соискание ученой степени доктора технических наук. Тбилиси 1993. Стр. 41-50.
 27. Z. Kovziridze. The Formula of Dependence of Mechanical Characteristics of Materials on Crystalline Phase Composition in the Matrix. Advances in Materials Physics and Chemistry. Vol. 10. No. 8. August 2020. ISSN: 2331-1959. DOI: 10.4236/ampc.2020.108013.
 28. ზ. კოვზირიძე. კერამიკულ მასალათა და კომპოზიციების მექანიკური მოდულის ფორმულა. საქართველოს ინტელექტუალური საკუთრების ეროვნული ცენტრი „საქპატენტი“ / ზ. კოვზირიძე. მოწმობა 7136 2017/10/11.
 29. Griffith A.A. Phil. Trans. Roy. Soc. London A221. 1920.

3.4.2.4. Synthesis, characteristic and activity of nanosized Cu–Me (Me–Co, Zn, Ni) oxide systems in CO oxidation in the presence of H₂. /S.T. Jafarova/. Azerbaijan Chemical Journal. – 2021. – #1. – pp. 48-54. – eng.; abs.: eng., az., rus.

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Nanooxides of Cu–Me composition (Me–Co, Zn, Ni) were synthesized by hydrothermal reduction of metal salts with subsequent calcination and the influence of their properties (size, morphology, structure) on catalytic activity of deep CO oxidation reaction in the presence of H₂ was considered. The nanooxides have been characterized by XRD and SEM methods. It was revealed that particles of Cu–Co–O are nanoplates (30–35 nm), and Cu–Zn–O (12.5–20 nm) are nanorods. The SEM method revealed a higher structural organization of the Cu–Co–O particles than Cu–Zn–O; the growth of nanocrystals is shown by varying the magnification of the scale grid of images. The highest activity of the Cu–Co–O system was found among the mentioned and corresponding individual oxides. The effect of metal (Cu/Co) ratio on the dispersibility and morphology of nanoparticles and their activity has been studied. The non-additive increase in activity is explained by the redox properties of cobalt oxides and the contribution of copper to electronic state of this element. The variation of composition, as well as high dispersibility (30–35 nm) make it possible to reduce the temperature of oxidation beginning (T50%) of CO to less than 115° C. Fig. 7, Ref. 17.

Keywords: nanooxides, nanoplates, nanorods, modification, morphology, structure, CO oxidation, electron microscopy

References:

1. Yaidelin A. Manrique, Carlos V. Miguel, Diogo Mendes, Adelio Mendes. Modeling and Simulation of a Packed-bed Reactor for Carrying out the Water-Gas Shift Reaction. International J. Chemical Reactor Engineering. 2012. V. 10. Issue 1. P. 1542–6580.
2. Laurent Piccolo, Salim Nassreddine, Franck Morfin Surface study of the hydrogen-free or preferential oxidation of CO: Iridium vs. platinum. Catalysis Today. 2012. V. 189. Issue 1. P. 42–48.
3. Zong Hu, Xiaofei Liu, Dongmei Meng, Yun Guo, Yunglong Guo, Guanzhong Lu. Effect of Ceria Crystal Plane on the Physicochemical and Catalytic Properties of Pd/Ceria for CO and Propane Oxidation. ACS Catal. 2016. V. 6. No 4. P. 2265– 2279.
4. Xinli Zhu Min Shen Lance L. Lobban Richard G. Mallinson. Structural effects of Na promotion for high water gas shift activity on Pt–Na/TiO₂. J. Catalysis. 2011. V. 278. Issue 1. P. 123–132.
5. Carabineiro S.A.C., Bogdanchikova N., Tavares P.B., Figueiredo J.L. Nanostructured iron oxide catalysts with gold for the oxidation of carbon monoxide. RSC Advances. 2012. V. 2. No 7. P. 2957.
6. Centeno M.Á., Reina T.R., Ivanova S., Laguna Ó.H., Odriozola J.A. Au/CeO₂ Catalysts: Structure and CO Oxidation Activity. Catalysts. 2016. 6. 158. DOI: 10.3390/catal6100158.
7. Amini, E., Rezaei, M. Preparation of mesoporous Fe–Cu mixed metal oxide nanopowder as active and stable catalyst for low-temperature CO oxidation. Chinese Journal of Catalysis. 2015. V. 36. No 10. P. 1711–1718.
8. Jing Wanga, Caiyun Hana Xiaoya Gaoa Jichang Lua Gengpin Wanab. Rapid synthesis of Fe-doped CuO–Ce_{0.8}Zr_{0.2}O₂ catalysts for CO preferential oxidation in H₂-rich streams: Effect of iron source and the ratio of Fe/Cu. J. Power Sources. 2017. V. 343. P. 437–445.
9. Kanaparthi Ramesh, Luwei Chen, Fengxi Chen, Yan Liu, Zhan Wang, Yi-Fan Han. Re-investigating the CO oxidation mechanism over unsupported MnO, Mn₂O₃ and MnO₂ catalysts. Catalysis Today. 2008. V. 131. No 1. P. 477–482.
10. Sajad Mobini, Fereshteh Meshkani, Mehran Rezaei. Synthesis and characterization of nanocrystalline copper–chromium catalyst and its application in the oxidation of carbon monoxide. Process Safety and Environmental Protection. 2017. V. 107. P. 181–189.
11. Ch Anil, Giridhar Madras. Kinetics of CO oxidation over Cu doped Mn₃O₄. J. Molecular Catalysis A: Chemical. 2016. V. 424. P. 106–114.

12. Tabakova T., Avgouropoulos G., Papavasiliou J., Manzi M., Bokuzzi H., Tenchev K., Vinidigni F., Jaoannide T. CO-free hydrogen production over Au/CeO₂-Fe₂O₃ catalysts: Part 1. Impact of the support composition on the performance for the preferential CO oxidation reaction. *Applied. Catalysis B: Environmental*. 2011. V. 101. Issues 3–4. P. 256–265.
13. Mahmood Andache, Ali Nemati Kharat, Mehran Rezaei. Preparation of mesoporous nanocrystalline CuO–ZnO–Al₂O₃ catalysts for the H₂ purification using catalytic preferential oxidation of CO (COPROX). In. *J. Hydrogen Energy*. 2019. V. 44. Issue 50. P. 27401–27411.
14. Dey S., Dhal G.C. Deactivation and regeneration of hopcalite catalyst for carbon monoxide oxidation: a review. *Materials today chemistry*. 2019. V. 14. 100180.
15. Yafei Guo, Jin Lin, Jian Sun, Jubing Zhang, Changhai Li, Shouxiand Lu. Precursor Effects on Catalytic Behaviors of Copper–Manganese–Cerium Ternary Oxides Pellets for Low-Temperature CO Oxidation. *Catalysis Letters*. 2019. pp 1–13.
16. Dzhaferova S.T., Medzhidov A.A., Akhmedov M.M., Ialchin B., Fatullaeva P.A., Agaeva S.A., Abbasov M.G. Poluchenie nanorazmernykh poroshkov metodom gidrotermalnogo sovmestnogo razlozheniia nitratov Cu, Co i Al v poliino srede. *Azer. him. zhurn*. 2018. № 2. S. 20–26.
17. Kathleen Mingle, Jochen A. Lauterbach, Synthesis-Structure-Activity Relationships in Co₃O₄ Catalyzed CO Oxidation / *Front. Chem*. 25 May 2018 / <https://doi.org/10.3389/fchem.2018.00185>.

3.4.2.5. Synthesis and characterization of cobalt oxide nanostructures a brief review. /S.J. Mammadyarova/. *Azerbaijan Chemical Journal*. – 2021. – #2. – pp. 80-93. – eng.; abs.: eng., az., rus.

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The newest achievement in the synthesis of cobalt oxide nanoparticles are considered. Cobalt oxide nanoparticles have attracted a great attention due to their uncommon properties and application in a supercapacitor, optoelectronic device, Li-ion battery gas sensor and electrochromic devices. Recently, nanostructured transition metal oxides with valuable properties have become a new class of materials for many technological fields. Cobalt oxide nanoparticles obtained from various precursors show different size distribution as well as different optical, electrical, magnetic, and electrochemical properties. A reduction in particle size to nanometer-scale leads to changes in properties compared to bulk ones due to quantum size effects. Depending on the application area, the choice of an appropriate synthesis method for nanoparticles with desirable properties is a crucial factor. This work aims to provide additional information on the synthesis methods and properties of cobalt oxide nanoparticles. Fig. 4, Tab. 1, Ref. 96.

Keywords: cobalt oxide, crystallite size, supercapacitor, synthesis method

References::

1. Raghavender T., Ramesh K.G., Pravansu S.M. Nanostructured Co₃O₄ electrodes for supercapacitor applications from plasma spray technique. *J. Power Sources*. 2012. V. 209. P. 44–51.
2. Xu M., Wang F., Zhao M., Yang S., Song X. Molten hydroxides synthesis of hierarchical cobalt oxide nanostructure and its application as anode material for lithium ion batteries. *Electrochimica Acta*. 2011. V. 56. P. 4876–4881.
3. Vetter S., Haffer S., Wagner T., Tiemann M. Nanostructured Co₃O₄ as a CO gas sensor: Temperature-dependent behavior. *Sensors and Actuators B: Chemical*. 2015. V. 206. P. 133–138.
4. Ronan B., Gregory C., Sabine V. Sonochemical oxidation of vanillyl alcohol to vanillin in the presence of a cobalt oxide catalyst under mild conditions. *Ultrasonics Sonochemistry*. 2017. V. 36. P. 27–35.
5. Haldorai Y., Kim J.Y., Ezhil A.T., Vilian, Heo N.S., Huh Y.S., Han Y. An enzyme-free electrochemical sensor based on reduced graphene oxide Co₃O₄ nanospindle composite for sensitive detection of nitrite. *Sensors and Actuators B: Chemical*. 2016. V. 227. P. 92–99.

6. Moon J., Kim T.K., VanSaders B., Choi C., Liu Z., Jin S., Chen R. Black oxide nanoparticles as durable solar absorbing material for hightemperature concentrating solar power system. *Solar Energy Material s& Solar Cells*. 2015. V. 134. P. 417–424.
7. Wang L., Song X., Zheng Y. Electrochromic properties of nanoporous Co_3O_4 thin films prepared by electrodeposition method. *Micro & Nano Letters*. 2012. V. 7. P. 1026–1029.
8. Nethravathi C., Sen S., Ravishankar N., Rajamathi M., Pietzonka C., Harbrecht B. Ferrimagnetic Nanogranular Co_3O_4 through Solvothermal Decomposition of Colloidally Dispersed Monolayers of α -Cobalt Hydroxide. *J. Phys. Chem. B*. 2005. V. 109. P. 11468–11472.
9. Dong Q., Wang X., Willis W.S., Song D., Huang Y., Zhao J., Li B., Lei Y. Nitrogen-doped Hollow Co_3O_4 Nanofibers for both Solid-state pH Sensing and Improved Non-enzymatic Glucose Sensing. *Electroanalysis*. 2019. V. 31. P. 1– 11.
10. Mishra S., Yogi P., Sagdeo P.R., Kumar R. TiO_2 – Co_3O_4 core–shell nanorods: bifunctional role in better energy storage and electrochromism. *ACS Appl. Energy Mater*. 2018. V. 1. P. 790–798.
11. Kupfer B., Majhi K., Keller D.A. Thin film $\text{Co}_3\text{O}_4/\text{TiO}_2$ heterojunction solar cells. *Adv. Energy Mater*. 2015. V. 5. P. 1401007.
12. Shi X., Han S., Sanedrin R.J., Zhou F. and Selke M. Synthesis of Cobalt Oxide Nanotubes from Colloidal Particles Modified with a Co(III)- Cysteinato Precursor. *Chem. Mater*. 2002. V. 14. P. 1897–1902.
13. Ding Y., Wang Y., Su L., Bellagamba M., Zhang H., Electrospun Y. Lei. Co_3O_4 nanofibers for sensitive and selective glucose detection. *Biosensors and Bioelectronics*. 2010. V. 26. P. 542–548.
14. Vickers D., Archer L.A., Floyd-Smith T. Synthesis and characterization of cubic cobalt oxide nanocomposite fluids. *Colloids and Surfaces A: Physicochem. Eng. Aspects*. 2009. V. 348. P. 39–44.
15. Ozkaya T., Baykal A., Toprak M.S., Koseoğlu Y., Durmuş Z. Reflux synthesis of Co_3O_4 nanoparticles and its magnetic characterization. *Journal of Magnetism and Magnetic Materials*. 2009. V. 321. P. 2145–2149.
16. Bhatte K.D., Bhanage B.M. Synthesis of cobalt oxide nanowires using a glycerol thermal route. *Materials Letters*. 2013. V. 96. P. 60–62.
17. Zhang Y., Chen Y., Wang T., Zhou J., Zhao Y. Synthesis and magnetic properties of nanoporous Co_3O_4 nanoflowers. *Microporous and Mesoporous Materials*. 2008. V. 114. P. 257–261.
18. Wu J., Dai Y., Pan Z., Huo D., Wang, T. Zhang H., Hu J., Yan S. Co_3O_4 hollow microspheres on polypyrrole nanotubes network enabling long-term cyclability sulfur cathode. *Applied Surface Science*. 2020. V. 510. P. 145529.
19. Wadekar K.F., Nemade K.R. and Waghuley S.A. Chemical synthesis of cobalt oxide (Co_3O_4) nanoparticles by co-precipitation method. *Research J. Chemical Sciencies*. 2017. V. 7. P. 53–55.
20. Viljoen E.L., Moloto M.J., Thabede P.M. Impact of acetate ions on the shape of Co_3O_4 nanoparticles. *Digest Journal of Nanomaterials and Biostructures*. 2017. V. 12. P. 571–577.
21. Allaedini G. and Muhammad A. Study of influential factors in synthesis and characterization of cobalt oxide nanoparticles. *J. Nanostructure in Chemistry*. 2013. V. 3. P. 1–16.
22. Sharifi S.L., Shakur H.R., Mirzaei A. Characterization of Cobalt Oxide Co_3O_4 Nanoparticles Prepared by Various Methods: Effect of Calcination Temperatures on Size, Dimension and Catalytic Decomposition of Hydrogen Peroxide. *Int. J. Nanosci. Nanotechnol*. 2013. V. 9. P. 51–58.
23. Xu J.M., Zhang J., Wang B.B., Liu F. Shaperegulated synthesis of cobalt oxide and its gassensing property. *Journ. of Alloys and Compounds*. 2015. V. 619. P. 361–367.
24. Luisetto I., Pepe F., Bemporad E. Preparation and characterization of nano cobalt oxide. *J Nanopart Res*. 2008. V. 10. P. 59–67.
25. Abdelhak L., Bedhif B., Amar B., Cherifa D., Benhebal H. Tuning of the physical properties by various transition metal doping in Co_3O_4 : TM (TM = Ni, Mn, Cu) thin films: A comparative study. *Chinese J. Physics*. 2018. V. 56. P. 1845–1852.
26. Katalin Sinko, Geza Szabo, and Miklos Zrinyi. Liquid-Phase Synthesis of Cobalt Oxide Nanoparticles. *Journal of Nanoscience and Nanotechnology*. 2011. V. 11. P. 1–9.

27. Shadrokh S., Farahmandjou M. and Firozabadi T.P. Fabrication and Characterization of Nanoporous Co Oxide (Co_3O_4) Prepared by Simple Solgel Synthesis. *Phys. Chem. Res.* 2016. V. 4. P. 153–160.
28. Devi V.S., Athika M., Duraisamy E., Prasath A., Sharma A.S., Elumalai P. Facile sol-gel derived nanostructured spinel Co_3O_4 as electrode material for high-performance supercapattery and lithiumion storage. *J. Energy Storage.* 2019. V. 25. P. 100815.
29. Wang G., Shen X., Horvat J., Wang B., Liu H., Wexler D., Yao J. Hydrothermal Synthesis and Optical, Magnetic, and Supercapacitance Properties of Nanoporous Cobalt Oxide Nanorods. *J. Phys. Chem. C.* 2009. V. 113. P. 4357–4361.
30. Hashemi Amiri S.E., Vaezi M.R. and Kandjani A.E. A comparison between hydrothermally prepared Co_3O_4 via H_2O_2 assisted and calcination methods. *J. Ceramic Processing Res.* 2011. V. 12. P. 327–331.
31. Nassar M.Y. Size-controlled synthesis of CoCO_3 and Co_3O_4 nanoparticles by free-surfactant hydrothermal method. *Materials Letters.* 2013. V. 94. P. 112–115.
32. Nugroho A., Kim J. Effect of KOH on the continuous synthesis of cobalt oxide and manganese oxide nanoparticles in supercritical water. *J. Ind. Eng. Chem.* 2014. V. 20. P. 4443–4446.
33. Elhag S., Ibupoto Z.H., Nour O., Willander M. Synthesis of Co_3O_4 Cotton-Like Nanostructures for Cholesterol Biosensor. *Materials.* 2015. V. 8. P. 149–161.
34. Sun S., Zhao X., Yang M., Ma L. and Shen X. Facile and Eco-Friendly Synthesis of Finger-Like Co_3O_4 Nanorods for Electrochemical Energy Storage. *Nanomaterials.* 2015. V. 5. P. 2335–2347.
35. Ma J., Zhang S., Liu W., Zhao Y. Facile preparation of Co_3O_4 nanocrystals via a solvothermal process directly from common Co_2O_3 powder. *J. Alloys and Compounds.* 2010. V. 490. P. 647–651.
36. Nassar M.Y., Mohamed T.Y., Ahmed I.S. One-pot solvothermal synthesis of novel cobalt salicylaldimine-urea complexes: A new approach to Co_3O_4 nanoparticles. *J. Molecular Structure.* 2013. V. 1050. P. 81–87.
37. Pang M., Long G., Jiang S., Ji Y., Han W., Wang B., Liu X., Xi Y., Wang D., Xu F. Ethanolassisted solvothermal synthesis of porous nanostructured cobalt oxides ($\text{CoO}/\text{Co}_3\text{O}_4$) for high-performance supercapacitors. *Chem. Eng. J.* 2015. V. 280. P. 377–384.
38. Al-Qirby L.M., Radiman S., Siong C.W., Ali A.M. Sonochemical synthesis and characterization of Co_3O_4 nanocrystals in the presence of the ionic liquid $[\text{EMIM}][\text{BF}_4]$, *Ultrasonics Sonochem.* 2017. V. 38. P. 640–651.
39. Askarinejad A. and Morsali A. Direct ultrasonicassisted synthesis of sphere-like nanocrystals of spinel Co_3O_4 and Mn_3O_4 . *Ultrasonics Sonochem.* 2009. V. 16. P. 124–131.
40. Kamar E.M. and Reda S.M. Sonochemical method for synthesizing Co_3O_4 /graphene nanocomposites for use as counter electrode in dye-sensitized solar Cells. *Appl. Phys. A.* 2016. V. 122. P. 688.
41. Xia X.H., Tu J.P., Zhang J., Huang X.H., Wang X.L., Zhang W.K., Huang H. Enhanced electrochromics of nanoporous cobalt oxide thin film prepared by a facile chemical bath deposition. *Electrochemistry Communications.* 2008. V. 10. P. 1815–1818.
42. Li Y., Huang K., Yao Z., Liu S., Qing X. Co_3O_4 thin film prepared by a chemical bath deposition for electrochemical capacitors. *Electrochimica Acta.* 2011. V. 56. P. 2140–2144.
43. Kandalkar S.G., Gunjekar J.L., Lokhande C.D., Joo O. Synthesis of cobalt oxide interconnected flacks and nano-worms structures using low temperature chemical bath deposition. *J. Alloys and Compounds.* 2009. V. 478. P. 594–598.
44. Kandalkar S.G., Dhawale D.S., Chang-Koo Kim. Chemical synthesis of cobalt oxide thin film electrode for supercapacitor application. *Synthetic Metals.* 2010. V. 160. P. 1299–1302.
45. Kung C., Lina C., Vittal R., Ho K. Synthesis of cobalt oxide thin films in the presence of various anions and their application for the detection of acetaminophen. *Sensors and Actuators. B.* 2013. V. 182. P. 429–438.
46. Obodo R.M., Nwanya A.C., Ekwealor A.B.C., Ahmad I., Zhao T., Osuji R.U., Maaza M., Ezema F.I. Influence of pH and annealing on the optical and electrochemical properties of cobalt (III) oxide (Co_3O_4) thin films. *Surfaces and Interfaces.* 2019. V. 16. P. 114–119.
47. Martínez-Gil M., Cabrera-German D., PintorMonroy M.I., García-Valenzuela J.A., Cota-Leal M., De W. la Cruz, Quevedo-Lopez M.A., PérezSalas R., Sotelo-Lerma M. Effect of annealing temperature on the thermal

- transformation to cobalt oxide of thin films obtained via chemical solution deposition. *Materials Science in Semiconductor Processing*. 2020. V. 107. P. 104825.
48. Valanarasu S., Dhanasekaran V., Karunakaran M., Chandramohan R., Mahalingam T. Role of Solution pH on the Microstructural Properties of Spin Coated Cobalt Oxide Thin Films. *J. of Nanosci. Nanotechnol.* 2013. V. 13. P. 1–6.
 49. Abdelhak L., Bedhief B., Amar B., Cherifa D., Benhebal H. Tuning of the physical properties by various transition metal doping in Co_3O_4 : TM (TM = Ni, Mn, Cu) thin films: A comparative study. *Chin. J. Physics*. 2018. V. 56. P. 1845–1852.
 50. Patil V., Joshi P., Chougule M., S. Sen. Synthesis and Characterization of Co_3O_4 Thin Film. *Soft Nanosci. Letters*. 2012. V. 2. P. 1–7.
 51. Khansari A., Salavati-Niasari M., Gholamrezaei S. Solid State Synthesis of Cobalt Oxide Nanohexagonals. *Synthesis and Reactivity in Inorganic, Metal-Organic, and Nano-Metal Chemistry*. 2015. V. 45. P. 1063–1068.
 52. Khalaji A.D. Synthesis, Characterization and Optical Properties of Co_3O_4 Nanoparticles. *Asian J. Nanosci. Materials*. 2019. V. 2. P. 186–190.
 53. Farhadi S., Safabakhsh J. and Zaringhadam P. Synthesis, characterization, and investigation of optical and magnetic properties of cobalt oxide (Co_3O_4) nanoparticles. *J. Nanostruct in Chem.* 2013. V. 3. P. 1–9.
 54. Zhang Y., Chen Y., Wang T., Zhou J., Zhao Y. Synthesis and magnetic properties of nanoporous Co_3O_4 nanoflowers. *Micropor. Mesopor. Mater.* 2008. V. 114. P. 257–261.
 55. Feng C., Wang H., Zhang J., Hu W., Zou Z. & Deng Y. One-pot facile synthesis of cobalt oxide nanocubes and their magnetic properties. *J. Nanopart Res.* 2014. V. 16. P. 2413.
 56. Turan E., Zeybekoğlu E., Kul M. Effects of bath temperature and deposition time on Co_3O_4 thin films produced by chemical bath deposition. *Thin Solid Films*. 2019. V. 692. P. 137632.
 57. Elaakib H., Pierson J.F., Chaik M., Samba C., Ait Dads H., Narjis A., Outzourhit A. Evolution of the structural, morphological, optical and electrical properties of reactively RF-sputtered cobalt oxide thin films with oxygen pressure. *Vacuum*. 2019. V. 159. P. 346–352.
 58. Ambika S., Gopinath S., Saravanan K., Sivakumar K., Ragupathi C., Sukantha T.A. Structural, morphological and optical properties and solar cell applications of thioglycolic routed nano cobalt oxide material. 2019. V. 5. P. 305–309.
 59. Alla S.K., Duvuru H.B., Shaw S.K., Vara Prasad B.B.V.S., Kumar M.K., Meena S.S., Gupta N., Prasad N.K. Zr-substituted cobalt oxide nanoparticles: structural, magnetic and electrical properties. *Journal of Materials Science: Materials in Electronics*. 2019. V. 30. P. 20088–20098.
 60. Duvuru H.B., Alla S.K., Shaw S.K., Meena S.S., Gupta N., Vara Prasad B.B.V.S., Kothawale M.M., Kumar M.K., Prasad N.K. Magnetic and dielectric properties of Zn substituted cobalt oxide nanoparticles. *Ceramics Int.* 2019. V. 45. P. 16512–16520.
 61. Kalam K., Seemen H., Mikkor M., Jõgiaas T., Ritslaid P., Tamm A., Kukli K., Kasikov A., Link J., Stern R., Dueñas S., Castán H. Electrical and magnetic properties of atomic layer deposited cobalt oxide and zirconium oxide nanolaminates. 2019. V. 669. P. 294–300.
 62. El A.M. Sayed and El-Gamal S. Synthesis and investigation of the electrical and dielectric properties of Co_3O_4 /(CMC+PVA) nanocomposite films. *J. Polym Res.* 2015. V. 22. P. 97.
 63. Durano M.M., Tamboli A.H., Kim H. Cobalt oxide synthesized using urea precipitation method as catalyst for the hydrolysis of sodium borohydride. *Colloids and Surfaces A: Physicochem. Eng. Aspects*. 2017. V. 520. P. 355–360.
 64. Ashraf Janjua M.R.S. Synthesis of Co_3O_4 Nano Aggregates by Co-precipitation Method and its Catalytic and Fuel Additive Applications. *Open Chem.* 2019. V. 17. P. 865–873.
 65. Pudukudy M., Yaakob Z. Sol-gel synthesis, characterisation, and photocatalytic activity of porous spinel Co_3O_4 nanosheets *Chemical Papers*. 2014. V. 68. P. 1087–1096.

66. Farhadi S., Javanmard M. and Nadri G. Characterization of Cobalt Oxide Nanoparticles Prepared by the Thermal Decomposition of $[\text{Co}(\text{NH}_3)_5(\text{H}_2\text{O})](\text{NO}_3)_3$ Complex and Study of Their Photocatalytic Activity. *Acta Chim. Slov.* 2016. V. 63. P. 335–343.
67. Kang M. and Zhou H. Facile Synthesis and Structural Characterization of Co_3O_4 Nanocubes. *AIMS Mater. Science.* 2015. V. 2. P. 16–27.
68. Ren Q., Feng Z., Mo S., Huang C., Li S., Zhang W., Chen L., Fu M., Wu J., Ye D. 1D- Co_3O_4 , 2D- Co_3O_4 , 3D- Co_3O_4 for catalytic oxidation of toluene. *Catalysis Today.* 2019. V. 332. P. 160–167.
69. Kung C., Lin C., Li T., Vittal R., Ho K. Synthesis of Co_3O_4 thin films by chemical bath deposition in the presence of different anions and application to H_2O_2 sensing. *Procedia Eng.* 2011. V. 25. P. 847–850.
70. Behling R., Chatel G., Valange S. Sonochemical oxidation of vanillyl alcohol to vanillin in the presence of a cobalt oxide catalyst under mild conditions. *Ultrasonics Sonochem.* 2017. V. 36. P. 27–35.
71. Dong Q., Wang X., Willis W.S., Song D., Huang Y., Zhao J., Li B. and Lei Y. Nitrogen-doped Hollow Co_3O_4 Nanofibers for both Solid-state pH Sensing and Improved Non-enzymatic Glucose Sensing. *Electroanal.* 2019. V. 31. P. 1–11.
72. Dai G., Xie J., Li C., Liu S. Flower-like Co_3O_4 /graphitic carbon nitride nanocomposite based electrochemical sensor and its highly sensitive electrocatalysis of hydrazine. *J. Alloys and Compounds.* 2017. V. 727. P. 43–51.
73. Palani S. and Arumugam S. Nano Co_3O_4 as Anode Material for Li-Ion and Na-Ion Batteries: An Insight into Surface Morphology. *Chemistry Select.* 2018. V. 3. P. 5040–5049.
74. Fan L., Zhang W., Zhu S., and Lu Y. Enhanced Lithium Storage Capability in Li-Ion Batteries Using Porous 3D Co_3O_4 Nanofiber Anodes. *Ind. Eng. Chem. Res.* 2017. V. 56. P. 2046–2053.
75. Wang Y., Guo R., Liu W., Zhu L., Huang W., Wang W., Zheng H. Co_3O_4 nanospheres composed of highly interconnected nanoparticles for boosting Li-Ion storage. *J. Power Sources.* 2019. V. 444. P. 227260.
76. Wang D., Yu Y., He H., Wang J., Zhou W., Abruña H.D. Template-Free Synthesis of Hollow Structured Co_3O_4 Nanoparticles as High-Performance Anodes for Lithium-Ion Batteries. *ACS Nano.* 2015. V. 9. P. 1775–1781.
77. Zheng F., Wei L. Synthesis of ultrafine Co_3O_4 nanoparticles encapsulated in nitrogen-doped porous carbon matrix as anodes for stable and long-life lithium ion battery. *J. Alloys and Compounds.* 2019. V. 790. P. 955–962.
78. Park J.S., Shin D.O., Lee C.S., Lee Y., Kim J.Y., Kim K.M., Shin K. Mesoporous perforated Co_3O_4 nanoparticles with a thin carbon layer for high performance Li-ion battery anodes. *Electrochimica Acta.* 2018. V. 264. P. 376–385.
79. Keshmarzi M.K., Daryakenari A.A., Omidvar H., Javanbakht M., Ahmadi Z., Delaunay J., Badrnezhad R. Pulsed electrophoretic deposition of nanographitic flakenanostructured Co_3O_4 layers for efficient lithium-ion-battery anode. *J. Alloys and Compounds.* 2019. V. 805. P. 924–933.
80. Hu R., Zhang H., Bu Y., Zhang H., Zhao B., Yang C. Porous Co_3O_4 nanofibers surface-modified by reduced graphene oxide as a durable, high-rate anode for lithium ion battery. *Electrochim. Acta.* 2017. V. 228. P. 241–250.
81. Wang B., Wang S., Tang Y., Ji Y., Liu W., Lu X. Hydrothermal Synthesis of Mesoporous Co_3O_4 Nanorods as High Capacity Anode Materials for Lithium Ion Batteries. *Energy Procedia.* 2019. V. 158. P. 5293–5298.
82. Yang J., Gao M., Lei J., Jin X., Yu L., Ren F. Surfactant-assisted synthesis of ultrathin two dimensional Co_3O_4 nanosheets for applications in lithium-ion batteries and ultraviolet photodetector. *J. Solid State Chem.* 2019. V. 274. P. 124–133.
83. Dwivedi P.K., G. Parte, M. Thripuranthaka, M.V. Shelke. High efficiency lithium storage in 3D composite foam of Co_3O_4 nanoparticles integrated carbon nanohorns. *Mater. Sci. Eng.* 2021. V. 263. P. 114839.
84. Wang H., Zhu Y., Yuan C., Li Y., Duan Q. Cobalt-phthalocyanine-derived ultrafine Co_3O_4 nanoparticles as high-performance anode materials for lithium ion batteries. *Appl. Surface Sci.* 2017. V. 414. P. 398–404.
85. Liu Y., Wan H., Jiang N., Zhang W., Zhang H., Chang B., Wang Q., Zhang Y., Wang Z., Luo S., Sun H. Chemical reduction-induced oxygen deficiency in Co_3O_4 nanocubes as advanced anodes for lithium ion batteries. *Solid State Ionics.* 2019. V. 334. P. 117–124.

86. Deng J., Lv X., Zhong J., Sun X. Carbon coated porous Co₃O₄ nanosheets derived from cotton fibers as anodes for superior lithium ion batteries. *Appl. Surface Sci.* 2019. V. 475. P. 446–452.
87. Zhang L., Li H., Li K., Li L., Wei J., Feng L., Fu Q. Morphology-controlled fabrication of Co₃O₄ nanostructures and their comparative catalytic activity for oxygen evolution reaction. *J. Alloys and Compounds.* 2016. V. 680. P. 146–154.
88. Hou Y., Hou C., Zhai Y., Li H., Chen T., Fan Y., Wang H., Wang W. Enhancing the electrocatalytic activity of 2D micro-assembly Co₃O₄ nanosheets for Li–O₂ batteries by tuning oxygen vacancies and Co³⁺/Co²⁺ ratio. *Electrochim. Acta.* 2019. V. 324. P. 134884.
89. Dhas C.R., Venkatesh R., Kirubakaran D.D., Merlin J.P., Subramanian B., Ezhil Raj A.M. Electrochemical sensing of glucose and photocatalytic performance of porous Co₃O₄ films by nebulizer spray technique. 2017. V. 186. P. 561–573.
90. Philippot K., De Tovar J., Romero N., Denisov S.A. Light-driven water oxidation using hybrid photosensitizer-decorated Co₃O₄ nanoparticles. 2018. V. 9. P. 506–515.
91. Atique Ullah A.K.M., Amin F.B., Hossain A. Tailoring surface morphology and magnetic property by precipitants concentrations dependent synthesis of Co₃O₄ nanoparticles. *Ceramics Int.* 2020. V. 46. P. 27892–27896.
92. Yetim N.K. Hydrothermal synthesis of Co₃O₄ with different morphology: Investigation of magnetic and electrochemical properties. *J. Molecular Structure.* 2021. V. 1226. P. 129414.
93. Zdorovets M.V., Shumskaya A.E., Kozlovskiy A.L. Investigation of the effect of phase transformations on the magnetic and electrical properties of Co/Co₃O₄ nanowires. *J. Magnetism and Magnetic Mater.* 2020. V. 497. P. 166079.
94. Ramamoorthy C., Rajendran V. Effect of surfactants assisted Co₃O₄ nanoparticles and its structural, optical, magnetic and electrochemical properties. *Optik.* 2017. V. 145. P. 330–335.
95. Anandhababu G., Ravi G. Facile synthesis of quantum sized Co₃O₄ nanostructures and their magnetic properties. *Nano-Structures & Nano Objects.* 2018. V. 15. P. 1–9.
96. Yin K., Ji J., Shen Y., Xiong Y., Bi H., Sun J., Xu T., Zhu Z., Sun L. Magnetic properties of Co₃O₄ nanoparticles on graphene substrate. *J. Alloys and Compounds.* 2017. V. 720. P. 345–351.

4.3. Processing Technologies

3.4.3.1. Anionic zeolite nanomaterial – environmentally safe complex fertilizer with prolonged action. /G. Tsintskaladze, T. Sharashenidze, M. Zautashvili, M. Burdjanadze, G. Antia, N. Mumladze/. *Bulletin of the Georgian National Academy of Sciences.* – 2021. – v. 15. – #3. – pp. 59-64. – eng.; abs.: eng., geo.

The development and implementation of effective and cost-effective environmental technologies is one of the priority problems in Georgia for the rehabilitation of soil fertility and natural vegetation cover. The paper proposes a new method for nanomodification of natural zeolite - clinoptilolite, based on the introduction of the appropriate salt into the structure of the zeolite so that the resulting material does not lose its zeolite structure and acquires both cation-exchange and anionexchange properties. Some amount of ammonium dihydrogen phosphate (NH₄H₂PO₄), potassium nitrate (KNO₃) and cations mixed with them (Fe, Ca, Mn, Zn, Mg, Cu, Mo, Co, Sn) were introduced by fusion method into the clinoptilolite structure. Only the amount of ammonium dihydrogen phosphate changed, while the amount of potassium nitrate (KNO₃) and cations remained unchanged. Accordingly, zeolite nanomaterials of various composition, structure and properties were obtained, which were studied by the methods of chemical, IR spectroscopic and X-ray diffractometric analyses. The obtained zeolite nanomaterials as fertilizers of complex composition and long-term action were used to study their effect on wheat productivity both in the open field and in laboratory conditions. Zeolite nanomaterials of three different compositions were studied. Tab. 3, Ref. 14.

Keywords: natural zeolite, zeolite-anionic form, nanomodified, I.R. spectroscopic, agriculture

References:

1. Andronikashvili T., Urushadze T. (2008). Ispolzovanie tseolitsoderzhashchikh porod v rastenievodstve. J. Agrochemistry. 12: 63-79. M. (in Russian).
2. Sheudzhen A. (2005). Soderzhanie mikroelementov v pochvakh i dostupnostikh rasteniyam. Materials of the regional scientific-practical conference: Fertilizers and harvest. 238-269. Maykop (in Russian).
3. Marshania I. (1991). Agrochemistry. 741 p. Tbilisi (in Georgian).
4. Zavalin A., Blagoveshchenskaya G., Chernova L., Shmyreva N. (2012) Upravlenie azotnym pitaniem rastenii v pochve. J. Agrochemical Bulletin. 4:38-40. Moscow (in Russian).
5. Volynkin V., Volynkin O. (2013). Deistvie sostava udobreniya i doz azota pri sistematicheskom primenenii v sevooborote i na monokulture pshenitsy. J. Fertility. 2:20-22. M. (in Russian).
6. Kudashkin M. (2011). Vliianie azotnykh i mikroudobrenii na urozhainostozimoi pshenitsy razlichnykh srokov seva v sevooborotakh agrolandshaftov iuga Nechernozemya. J. Agrochemistry. 5:26-34. M. (in Russian).
7. Tsitsishvili G., Tsintskaladze G., Tsitsishvili V., Chipashvili D., Tsintskaladze Z. (2006). New form of phosphorous containing clinoptilolite. Azerbaijan Chemical Journal. 3:100-102. Baku.
8. Tsintskaladze G., Eprikashvili L., Zautashvili M., Sharashenidze T., Burdjanadze M., Burkiashvili N., Gabunia V. (2013). New biotechnological material based on natural zeolite acts as a fertilizer of a prolonged action and prospects of its application. 559-562. II International Scientific-Practical Conference: Bioeconomy and Sustainable Development of Agriculture. Tbilisi (in Georgian).
9. Flanigen E., Khatami H., Szymanski H. (1974). Infrared structural studies of zeolites frameworks. Chapter 16:201-229. In: Flanigen, E.M. and Sand, L.B., Eds., molecular sieve zeolites. Advances in Chemistry. 101:460-488, American Chemical Society. Washington DC.
10. Tsitsishvili G., Andronikashvili T., Kardava M. (1993). Prirodnye tseolity v zemledelii, 128p., Tbilisi (in Russian).
11. Andronikashvili T., Urushadze T., Eprikashvili L. (2009). Tseolitoderzhashchie substraty – novyi put ot rastenievodstva k rasteniyeirozvodstvu. J. Annals of Agrarian Science. 7. 4:14-45. Tbilisi (in Russian).
12. Tsintskaladze G., Tsitsishvili V., Sharashenidze T., Zautashvili M., Iluridze G., Tsintskaladze P. (2016) Possibilities of agricultural application of zeolite nanomaterials enriched by nitrate ions. Proceedings of the Georgian National Academy of Sciences. Chemical Series. 42. 1:78-80. Tbilisi (in Georgian).
13. Tsintskaladze G., Eprikashvili L., Urushadze T., Kordzakhia T., Sharashenidze T., Zautashvili M., Burjanadze M. (2016) Nanomodified natural zeolite as a fertilizer of prolog acting. J. Annals of Agrarian Science. 14:163-168. Tbilisi.
14. Tsintskaladze G., Eprikashvili L., Mumladze N., Gabunia V., Sharashenidze T., Zautashvili M., Kordzakhia T., Shatakishvili T. (2017). Nitrogenous zeolite nanomaterial and the possibility of its application in agriculture. J. Annals of Agrarian Science. 15:365-369. Tbilisi.

4.4. Nanobiotechnology

3.4.4.1. Growing technology for soybeans with nanoherbicides. /A. Korakhashvili, T. Kacharava, L. Korakhashvili/. Annals of Agrarian Science. – 2021. – v. 19. - #3. – pp. 199-203. – eng.; abs.: eng.

Modern herbicide market in agriculture is about 2 billion tons and about 73 billion dollars industry with sophisticated multi-impact problems with food safety and human health, with increasing of weed resistance with every passing year at the topmost. Nanoherbicides under development in the current decade of our century could be a new strategy to address all the issues caused by the conventional non-nanoherbicides. From the beginning of 21 century group of Georgian scientists with farmers associations have begun development nanoherbicides (experimental name “Nanocooper 076”, which is under registration) in soybean experimental pilot plots and farmer’s fields, which will allow farmers to clear

their soybean plantings from weeds without using toxic chemicals, like Glyphosate. As the potential use of nanostructured nanomaterials enables the use of nanoherbicides effectively and rules out the emergence of various weed-resistant population at an early stage of growing agricultural crops (first weeks after sowing), these very desirable nano technological methods and practices in general agriculture are reviewed by this article. Fig. 2, Tab. 1, Ref. 18.

Keywords: soya seed pilling, Nanocooper 076, soil contamination, friendly nanotechnologies, nanoherbicides, agriculture

References:

1. FAO. Country Programming Framework for Georgia. 2016 to 2020. Italy. 2015.
2. Gullner G., Komivec N, Rennenberg H. Detoxification of Chloroacetinilide by Transgenic Poplars. In: Phytoremediation: environmental and molecular biological aspects. OECD workshop. Hungary. Abstr. 2004. 24 pp.
3. Korakhashvili A. Soybean Seed Inoculation Method. Georgia State Patent # 1180. Tbilisi. Georgia. 1996. 5 pp. (in Georgian).
4. Korakhashvili A. New Growing Technologies of Grain Legumes and Their role in Farmers Economics. "Caravan". Aleppo. Syria. 2001. pp 23-29.
5. Korakhashvili A., Annual Management Plan for Farming by Computer Program BARMEX, Third European Conference on Precision Agriculture. Montpellier. France. 2001. pp 47-51.
6. Agladze G., Korakhashvili A. Grass landraces of Georgian arid pastures. Report of a Working Group on Forages. Elvas. Portugal. 199. 97 pp.
7. Korakhashvili A., Teo Urushadze. Growing of Oldest Legumes by Advance Technologies in Georgia. "Grain Production". # 3. Moscow. Russia. 2002. pp. 34-35 (in Russian).
8. Korakhashvili A., D. Kirvalidze, T. Kvrivishvili, R. Vaismiller, E. Sanadze. Research of Cinnamonic Calcareous Soil Fertilizing Systems for Pastures of Akhaltsikhe District. Communications in Soil Sciences and Plant Analysis. Taylor and Francis. USA. vol. 42. #7. (2011) 767-786.
9. Korakhashvili A. Regeneration and Conservation of Chickpea Genetic Resources of Georgia. International Conference on Enhanced. Genepool Utilization. Cambridge. United Kingdom. 2014. pp. 43-44.
10. Korakhashvili A. Seed registration, development and certification, in Enabling the Business of Agriculture, WB/EBRD. Washington. USA. 2016. pp. 126-131.
11. Korakhashvili A., T. Urushadze, D. Kirvalidze. Endemic and Released Legume Crops Sustainable Production in Georgia. Lam LAMBERT Academic Publication. Germany. USA. UK. 2018. 55 pp.
12. Mahendra Shah, Strong Maurice. Food in the 21st Century: from Science to Sustainable Agriculture. Washington. USA. 1999. 72 pp.
13. Njoroge W. John. Indicators of Sustainable Farming. IFOAM. Imsbac. Germany. 1997. 124 pp.
14. Ronald D. Knutson, J.B. Penn, Barry L., Flinch Baugh. Agricultural and Food Policy. New Jersey. USA. 1998. 521 pp.
15. Zaalishvili G., Khatiashvili G., Ugrekheldze D., Gordeziani M., Kvesitadze G. Plant potential for detoxification (Review). Appl. Biochem Microbial 36. 2000. pp. 443- 451.
16. Korakhashvili A., Kirvalidze D. Chickpea Genetic Resources Regeneration and Safety Duplication in Georgia, Universal J. of Agricultural Research, USA, Vol. 4(3) (2016) 67-70.
17. Korakhashvili A., Chickpea Genetic Resources Regeneration and Safety Duplication in Georgia. CABI Oxfordshire-Boston. UK-USA. 2016. pp. 210-221.
18. Korakhashvili A., T. Sanikidze, L. Korakhashvili. Adaptation of Food Safety Communication Systems RASFF and INFOSAN in Georgian Cheese Production. Workshop of AASSA comity. Academies of Sciences. New Delhi. India. 2017. pp. 18-21.