

Georgian Technical University
TECHINFORMI

E ISSN 2667-9221

<https://doi.org/10.36073/2667-9221>

Caucasus Abstracts Journal of
Nanoscience *and*
Nanotechnology

N2, 2020

Tbilisi, Georgia

Georgian Technical University
TECHINFORMI

E ISSN 2667-9221

<https://doi.org/10.36073/2667-9221>

Caucasus Abstracts Journal of
Nanoscience *and*
Nanotechnology

N2, 2020

Tbilisi, Georgia

Editors-in-Chief:

Levan Chkhartishvili – GTU, Department of Engineering Physics, Georgia

Madona Kopaleishvili – GTU, Institute Techninformi, Georgia

Editorial Board:

Irina Bedinashvili - GTU, Institute Techninformi, Georgia

Giorgi Bibileishvili - GTU, Membrane Technologies Engineering Institute, Georgia

Nikoloz Chikhradze – GTU, Grigol Tsulukidze Mining Institute, Georgia

Teimuraz Chubinishvili - GTU, Institute Techninformi, Georgia

Mzia Ghogheliani - GTU, Institute Techninformi, Georgia

Nelly Makhviladze - GTU, Institute Techninformi, Georgia

Text Editor:

Valeri Sarjveladze - GTU, Institute Techninformi, Georgia

Editorial Office: TECHINFORMI, 47 Kostava St., Tbilisi

Tel.: 233-53-15; 233-59-03

E-mail: tech@techninformi.ge

<https://techninformi.ge>

© GTU, TECHINFORMI 2020

<http://techninformi.ge/>

All contents in this website including website layout and design, images, text, programs and all other information are the property of American Scientific Publishers and is protected by copyright and other intellectual property laws. All materials including names, trademarks, and logos posted on the website are subject to copyrights owned by American Scientific Publishers. The contents of any American Scientific Publishers website pages should not be reproduced, copied, displayed, distributed, modified, published, printed, created derivative works from or sell or license all or any part of the contents, products or services obtained from this website in any form or medium to anyone, created or compiled as database, collection or compilation, stored in a retrieval system, transmitted, translated into any foreign languages or otherwise used in form whatsoever by any means electronic, mechanical, photocopying, scanning, recording, or otherwise without the written permission of the American Scientific Publishers. All registered trade marks, names, logo and similar related materials used on any American Scientific Publishers website pages, even when not marked as such, are not to be considered unprotected by law. All rights are reserved.

Contents

1. NANOPHYSICS	5
1.1. Phenomena and Effects.....	5
1.2. Properties of Materials and Structures	9
3. NANOBIOLOGY	14
3.2. Biophysics	14
3.3. Biochemistry	16
4. NANOTECHNOLOGY	19
4.1. Materials and Structures.....	19
4.2. Obtaining Technologies	22
4.3. Processing Technologies	27
5. NANOENGINEERING	29
5.1. Devices and Sensors	29
6. NANOMEDICINE	34
6.2. Medical Chemistry	34

1. NANOPHYSICS

1.1. Phenomena and Effects

2.1.1.1. Electron transport mechanism in composites based on polybenzimidazole matrix with graphite nanoparticles. /V.A. Kuznetsov, A.N. Lavrov, B.Ch. Kholkhoev, V.G. Makotchenko, E.N. Tkachev, V.F. Burdukovskii, A.I. Romanenko/. Proceedings of NAS RA. Physics. – 2020. – vol. 55. – #1. – pp. 78-85. – rus.; abs.: rus., arm., eng.

The paper presents an experimental study of the electron transport in composite samples based on an insulating matrix of polybenzimidazole with graphite nanoparticles as a conducting filler. Based on a qualitative analysis of the temperature dependences of electrical resistance obtained for the samples with different filler concentrations, it was established that the electron transport occurred by tunneling between conducting filler particles, with the variable-range hopping conduction taking place at low temperatures. Fig. 4, Ref. 17.

Keywords: electron transport, composite samples, insulating matrix, graphite nanoparticles, electrical resistance

References:

1. A.B. Kaiser, V. Skakalova. Chem. Soc. Rev, 40, 3786 (2011).
2. S. Ravi, A.B. Kaiser, C.W. Bumby. Chemical Physics Letters, 496, 80 (2010).
3. M. Shiraishi, M. Ata. Synthetic Metals, 128, 235 (2002).
4. T.M. Barnes, J.L. Blackburn, J. van de Lagemaat, T.J. Coutts, M.J. Heben. ACSNANO, 2, 1968 (2008).
5. V. Skakalova, A.B. Kaiser, U. Dettlaff, K. Arstila, A.V. Krashennikov, J. Keinonen, S. Roth. Physica Status Solidi B, 245, 2280 (2008).
6. B.Ch. Kholkhoev, E.N. Gorenskaya, S.A. Bal'zhinov, I.A. Farion, G.N. Batorova, A.V. Nomoev, P.S. Timashev, B.R. Radnaev, R.K. Chailakhyan, V.E. Fedorov, V.F. Burdukovskii. Russian Journal of Applied Chemistry, 89, 780 (2016).
7. V.G. Makotchenko, E.D. Grayfer, A.S. Nazarov, S.-J. Kim, V.E. Fedorov. Carbon, 49, 3233 (2011).
8. V.A. Kuznetsov, B.Ch. Kholkhoev, V.G. Makotchenko, A.N. Lavrov, Ye.N. Gorenskaya, A.S. Berdinsky, V.F. Burdukovskii, A.I. Romanenko, V.Ye. Fedorov. Nanoindustry, 12, 48 (2019).
9. E.N. Tkachev, T.I. Buryakov, V.L. Kuznetsov, S.I. Moseenkov, I.N. Mazov, S.I. Popkov, K.A. Shaikhutdinov. J. Exp. Theor. Phys., 116, 860 (2013).
10. H. Gerischer, R. McIntyre, D. Scherson, W. Storck. Journal of Physical Chemistry, 91, 1930 (1987).
11. J.W. McClure. Physical Review, 108, 612 (1957).
12. R. Ahuja, S. Auluck, O. Eriksson, B. Johansson. Journal of Physics-Condensed Matter, 9, 9845 (1997).
13. J.R. Dahn, J.N. Reimers, A.K. Sleight, T. Tiedje. Physical Rev. B, 45, 3773 (1992).
14. R.C. Tatar, S. Rabi. Physical Review B, 25, 4126 (1982).
15. J.W. McClure. IBM Journal of Research and Development, 8, 255 (1964).
16. E.J. Mele, J.J. Ritsko. Physical Review Letters, 43, 68 (1979).
17. B.I. Shklovskii, A.L. Efros. Electronic Properties of Doped Semiconductors. Berlin, Heidelberg. Springer-Verlag Berlin Heidelberg, 1984.

2.1.1.2. Quantum model of a trapezoidal limiting potential profile in a spherical nanocrystal. /H.S. Nikoghosyan, S.L. Harutyunyan, V.F. Manukyan, G.H. Nikoghosyan/. Proceedings of NAS RA. Physics. – 2019. – vol. 54. – #4. – pp. 471-477. – rus.; abs.: rus., arm., eng.

Considered is a limiting potential model for a spherical quantum dot with three variation parameters - well depth, external and internal radii. The stationary s-states are calculated by the exact solution and in the WKB approximation. For states with $l \neq 0$, an approximate consideration is applied, subject to the conditions of semiclassicality. As a result, the energy values of the lower bound states are presented at fixed external and varying internal radii in wells of various depths. The dependence of the energy gap between the levels in the well on the variation parameters is demonstrated. Fig. 1, Ref. 14.

Keywords: limiting potential model, spherical quantum dot, external and internal radii, lower bound states, energy gap

References:

1. P.A. Maksym, T. Chakraborty. Phys. Rev. Lett., 65, 108 (1990).
2. А.О. Говоров, А. В. Чаплик. Письма в ЖЭТФ, 52, 681 (1990).
3. F.M. Peeters. Phys. Rev. B, 42, 1486 (1990).
4. W. Kohn. Phys. Rev., 123, 1242 (1961).
5. R.J. Boyd. Nature (London), 310, 480 (1984).
6. В.Д. Кривчик, А.Б. Грунин, Р. В. Зайцев. ФТП, 36, 1225 (2002).
7. В. Boyacıoğlu, A. Chatterjee. J. Appl. Phys., 112, 083514 (2012).
8. Д.А. Багдасарян, Д.Б. Айрапетян, А.А. Саркисян, Э.М. Казарян, А. Медвид. Изв. НАН Армении, Физика, 52, 177 (2017).
9. E. Kazaryan, L. Petrosyan, V. Shahnazaryan, H. Sarkisyan. Commun. Theoretical Physics, 63, 255 (2015).
10. Д. Багдасарян, Д. Айрапетян, Э. Казарян. Изв. НАН Армении, Физика, 51, 211 (2016).
11. Л.Д. Ландау, Е.М. Лифшиц. Квантовая механика, том 3. Москва, Наука, 1989.
12. Э. Вихман. Квантовая физика, БКФ, том 4. Москва, Наука, 1986.
13. Э.М. Казарян, С.Г. Петросян. Физические основы полупроводниковой наноэлектроники, Ереван, РАУ, 2005.
14. А.С. Давыдов. Квантовая механика, Москва, ФМ, 1963.

2.1.1.3. Nucleation mechanism and nanostructures' total energy calculation in CdTe-ZnTe-HgTe material system. /А.К. Simonyan, К.М. Gambaryan, V.M. Aroutiounian, М.К. Gambaryan, G.A. Avetisyan/. Proceedings of NAS RA. Physics. – 2019. – vol. 54. – #4. – pp. 478-484. – rus.; abs.: rus., arm., eng.

The continuum elasticity model is applied to quantitatively investigate the growth features and nucleation mechanism of quantum dots, nanopits, and joint QDs-nanopits structures in CdZnHgTe quasiternary material system. It is shown that for the CdZnHgTe solid solution deposited on CdTe substrate, at the critical strain of $\varepsilon^* = 0.006$ and $\varepsilon^* = 0.009$, the sign of island's critical energy and volume is changed. It is assumed that at $\varepsilon = \varepsilon^*$ the mechanism of the nucleation is changed from the growth of quantum dots to the nucleation of nanopits. Obviously, at small misfits ($\varepsilon < \varepsilon^*$) the bulk nucleation mechanism dominates. However, at $\varepsilon > \varepsilon^*$ when the energy barrier becomes negative as well as a larger misfit provides a low-barrier path for the formation of dislocations, the nucleation of pits becomes energetically preferable. Fig. 5, Ref. 15.

Keywords: continuum elasticity model, nucleation mechanism, quantum dots, critical strain, energy barrier

References:

1. К.М. Gambaryan. Nanoscale Res. Lett., 5, 587 (2010).
2. J. Tersoff, F.K. Le Goues. Phys. Rev. Lett., 72, 3570 (1994).
3. К.М. Gambaryan, V.M. Aroutiounian, V.G. Harutyunyan. Infrared Phys. & Techn., 54, 114 (2011).
4. V.M. Aroutiounian, К.М. Gambaryan, P.G. Soukiassian. Surf. Sci., 604, 1127 (2010).
5. S. Kasap, P. Capper. Springer Handbook of electronics and Photonics Materials., Part B., 1 (2007).
6. T. Potlog, N. Maticiuc, A. Mirzac, P. Dumitriu, D. Scortescu. Structural and optical properties of ZnTe thin film, International Semiconductor Conference SMICND–2012, 2012.
7. I. Duz, S. Erdem, S. Ozdemir Kart, V. Kuzucu. Archives of Materials Science and Engineering, 79, 5 (2016).
8. T. Wojtowicz, E. Janiket, et al. Journal of the Korean Phys. Society, 53, 3055 (2008).
9. G.I. Rusu, P. Prepelita, N. Apetroaei, G. Popa. Journal of Optoelect. & Advan. Materials, 17, 829 (2005).
10. M.A.M. Seyam, A. Elfalaky. Vacuum, 57, 31 (2000).
11. W.D. Lawson, S. Nielson, E.H. Putley, A.S. Young. J. Phys. Chem. Solids, 9, 325 (1959).
12. A. Rogalski. Optoelectronics Review, 18, 284 (2010).
13. J. Tersoff, R.M. Tromp. Phys. Rev. Letters, 70, 2782 (1993).
14. M. Żenkiewicz. J. Achieve. Mater. Manuf. Eng., 24, 137 (2007).
15. M. Biehl, F. Much, C. Vey. Int. Series of Numerical Mathematics, 149, 41 (2005).

2.1.1.4. Rashba spin-orbit interaction in semiconductor nanostructures (review). /B.G. Ibragimov/. Azerbaijan Journal of Physics. – 2020. – vol. 26. – #2. – pp. 3-9. – eng.; abs.: eng.

The work reviews the theoretical and experimental issue related to the Rashba spin-orbit interaction [1] in semiconductor nanostructures. The Rashba spin-orbit interaction has been a promising candidate for controlling the spin of electrons in the field of semiconductor spintronics. The work focuses on the study of the electrons spin and holes in isolated semiconductor quantum dots and rings in the presence of magnetic fields. Spin-dependent thermodynamic properties with strong spin-orbit coupling inside their band structure in systems are investigated. Additionally, specific heat and magnetization in two-dimensional, one-dimensional ring and quantum dot nanostructures with spin-orbit interaction are discussed. Fig. 4, Ref. 68.

Keywords: spin-orbit interaction, Rashba effect, two-dimensional electron gas, one-dimensional ring, quantum wire, quantum dot, semiconductor nanostructures

References:

1. E. Rashba. *Fiz. Tverd. Tela (Leningrad)* 2, 1224 (1960), [*Sov. Phys. Solid State* 2, 1109 (1960)].
2. W. Greiner. *Relativistic quantum mechanics* (Springer, 1987).
3. R. Winkler. *Spin-orbit Coupling Effects in TwoDimensional Electron and Hole Systems* (Springer Berlin).
4. G. Dresselhaus. *Spin-Orbit Coupling Effects in Zinc Blende Structures* *Phys. Rev.* 100, 580, 1955.
5. Y.A. Bychkov and E. I. Rashba. *Journal of Physics C: Solid State Physics* 17, 6039, 1984.
6. D. Stein, K.V. Klitzing, and G. Weimann. *Electron Spin Resonance on GaAs-Al_xGa_{1-x}As Heterostructures*. *Phys. Rev. Lett.* 51, 130, 1983.
7. H.L. Stormer, Z. Schlesinger, A. Chang, D.C. Tsui, A.C. Gossard, and W. Wiegmann. *Energy Structure and Quantized Hall Effect of Two-Dimensional Holes*. *Phys. Rev. Lett.* 51, 126, 1983.
8. S. Datta and B. Das. *Electronic analog of the electro-optic modulator*. *Appl. Phys. Lett.* 56, 665, 1990.
9. J.M. Kikkawa, I.P. Smorchkova, N. Samarth, and D.D. Awschalom. *Room-Temperature Spin Memory in Two-Dimensional Electron Gases*. *Science* 277, 1284–1287, 1997.
10. E.A. de Andrada e Silva, G.C. La Rocca, and F. Bassani. *Spin-orbit splitting of electronic states in semiconductor asymmetric quantum wells* *Phys. Rev. B* 55, 16293, 1997.
11. O. Voskoboynikov, C.P. Lee. *Spin-Orbit Interaction and All-Semiconductor Spintronics* *Journal of Superconductivity* volume 16, pages 361–363, 2003.
12. D.V. Khomitsky. *Scattering on the lateral onedimensional superlattice with spin-orbit coupling* *Phys. Rev. B* 76, 033404, 2007.
13. M.I. Dyakonov, V.I. Perel. *Spin relaxation of conduction electrons in noncentrosymmetric semiconductors*. *Soviet Physics Solid State*, 13 (12), 3023-3026, 1972.
14. M.I. Dyakonov, V.Y. Kachorovskii. "Spin Relaxation of Two Dimensional Electrons in Noncentrosymmetric Semiconductors," *Soviet Physics: Semiconductors*, *Sov. Phys. Semicond.* 20, 110, 1986.
15. J.D. Koralek, C.P. Weber, J. Orenstein, B.A. Bernevig, S.-C. Zhang, S. Mack, D.D. Awschalom. *Emergence of the persistent spin helix in semiconductor quantum wells*. *Nature (London)*, 458, 610, 2009.
16. P. Kleinert and V.V. Bryksin. *Electric-field induced long-lived spin excitations in two-dimensional spin-orbit coupled systems*, *Phys. Rev. B* 79, 045317, 2009.
17. M. Duckheim, D.L. Maslov, and D. Loss. *Dynamic spin-Hall effect and driven spin helix for linear spin-orbit interactions.*, *Phys. Rev. B* 80, 235327, 2009.
18. I.V. Tokatly, E.Ya. Sherman. *Ann. Phys.* 325, 1104, 2010.
19. Y.V. Pershin. *Long-Lived Spin Coherence States in Semiconductor Heterostructures*. *Phys. Rev. B* 71, 155317, 2005.
20. J.M. Kikkawa, D.D. Awschalom. *Lateral drag of spin coherence in gallium arsenide*. *Nature*, 1999, 397, 139.
21. S. Sanvito, G. Theurich, N.A. Hill. *Density functional calculations for III-V diluted ferromagnetic semi-conductors: a review*. *J. Supercond.*, 15, 85, 2002.
22. M. Peshkin, A. Tonomura. *The Aharonov-Bohm Effect*, in *Lecture Notes in Physics*, Vol. 340 (Springer-Verlag, Berlin), 1989.
23. H. Fai Cheung, Y. Gefen, E.K. Riedel, W.-H. Shih. *Persistent currents in small onedimensional metal rings*. *Phys. Rev. B* 37, 6050, 1988.
24. D. Loss, P. Goldbart. *Period and amplitude halving in mesoscopic rings with spin*. *Phys. Rev. B* 43, 13762, 1991.
25. M.V. Moskalets. *Oscillations of the thermodynamic properties of a one-dimensional mesoscopic ring caused by Zeeman splitting*. *JETP Lett.* 70, 602, 1999.

26. V.A. Margulis, V.A. Mironov. Magnetic moment of a 2D electron gas with spin-orbit interaction. *JETP*, 108, 656, 2009.
27. A.G. Aronov, Y.B. Lyanda-Geller. Spin-orbit Berry phase in conducting rings. *Phys. Rev. Lett.* 70, 343, 1993.
28. X.F. Wang, P. Vasilopoulos. Spin-dependent magnetotransport through a mesoscopic ring in the presence of spin-orbit interaction. *Phys. Rev. B*, 72, 165336, 2005.
29. B. Molnar, F.M. Peeters, P. Vasilopoulos. Spin-independent magnetotransport through a ring due to spin-orbit interaction. *Phys. Rev. B*, 69, 155335, 2004.
30. J. Splettstoesser, M. Governale, U. Zulicke. Persistent current in ballistic mesoscopic rings with Rashba spin-orbit coupling. *Phys. Rev. B*, 68, 165341, 2003.
31. J.S. Sheng, K. Chang. Spin states and persistent currents in mesoscopic rings: Spin-orbit interactions. *Phys. Rev. B*, 74, 235315, 2006.
32. X.-E. Yang, Y.-C. Zhou. Magnetism and structural distortion in the $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$ metallic ferromagnet. *Phys. Rev. B*, 53, 10167, 1996.
33. H.-Y. Chen, P. Pietilainen, T. Chakraborty. Electronic optical and magnetic properties of quantum ring in Novel systems, *Phys. Rev. B* 78, 2008, 073407.
34. V.A. Margulis, V.A. Mironov. Magnetic moment of an one-dimensional ring with spin-orbit interaction. *Physica E*, 43, 905–908, 2011.
35. S.A. Wolf, et al. Spintronics: a spin-based electronics vision for the future. *Science*, 294, 1488, 2001.
36. P.A. Wolff. Semiconductors and semimetals, in: J.K. Furdyna, J. Kossut (Eds.), *Diluted Magnetic Semiconductors*, Academic, New York, 1988.
37. A.V. Chaplik, L.I. Magarill. Effect of the spin-orbit interaction on persistent currents in quantum rings. *Superlattice. Microstr.*, 18, 4, 1995.
38. B.H. Mehdiyev, A.M. Babayev, S. Cakmak, E. Artunc. Rashba spin-orbit coupling effect on a diluted magnetic semiconductor cylinder surface and ballistic transport. *Superlattice Microstr.*, 46, 4, 2009.
39. A.M. Babanlı, B.G. Ibragimov. Specific heat in diluted magnetic semiconductor quantum ring. *Superlattices and Microstr.*, 111, 574–578, 2017.
40. A.M. Babanlı, B.G. Ibragimov. *AJP Fizika*, XXV, 01, 35–38, 2019.
41. A.M. Babanlı, B.G. Ibragimov. Magnetic moment of the lattice of non-interacting diluted magnetic semiconductor ring. *Modern Trends in Physics International Conference 01-03 May 2019 Baku*.
42. A.M. Babanlı, B.G. Ibragimov. "Aharonov-Bohm paramagnetism" and compensation points in noninteracting diluted magnetic semiconductor quantum ring. *J. Magn. Magn. Mater.*, 495, 165882, 2020.
43. D. Loss, D.P. Di Vincenzo. Quantum computation with quantum dots. *Phys. Rev. A*, 57, 120, 1998.
44. L.M. K. Vandersypen, et al. in *Quantum Computing and Quantum Bits in Mesoscopic Systems* (Kluwer Academic, New York), quant-ph/0207059, 2003.
45. A.V. Khaetskii, Y.V. Nazarov. Spin relaxation in semiconductor quantum dots. *Phys. Rev. B*, 61, 12639, 2000.
46. C.-M. Hu, J. Nitta, T. Akazaki, J. Osaka, P. Pfeffer, W. Zawadzki. Polaronic and bound polaronic effects in the energy states of an electron in a two-dimensional parabolic quantum dot in the presence of Rashba spin-orbit interaction. *Phys. Rev. B*, 60, 7736, 1999.
47. E.A. de Andrada e Silva, G.C. La Rocca, F. Bassani. Spin-orbit splitting of electronic states in semiconductor asymmetric quantum wells. *Phys. Rev. B*, 55, 16293, 1997.
48. N.F. Johnson. *J. Phys.: Condens. Matter* 7 965. Quantum dots: few-body, low-dimensional systems 1995.
49. S. Datta, B. Das. Electronic analog of the electro-optic modulator. *Appl. Phys. Lett.* 56, 665, 1990, <https://doi.org/10.1063/1.102730>
50. B.E. Kane. A silicon-based nuclear spin quantum computer. *Nature (London)*, 393, 133, 1998.
51. K. Bhattacharyya, A. Chatterjee. Polaronic and Bound Polaronic. Effects in The Energy States of An Electron in A TwoDimensional Parabolic Quantum Dot in The Presence of Rashba Spin-Orbit interaction. *AIP Conference Proceedings* 2142, 090008, 2019. <https://doi.org/10.1063/1.5122452>.
52. D. Bimberg. Quantum dots: Paradigm changes in semiconductor physics. *Semiconductors* 33, 951. 1999.
53. O. Steffens, M. Suhrke, U. Rössler. *Physica B*, 256–258, 147, 1998.
54. S. Tarucha et al. Spin effects in semiconductor quantum dot structures. *Physica E (Amsterdam)*, 3, 112, 1998.
55. H. Tamura. Magneto-photoluminescence in high magnetic fields from InGaAs/GaAs quantum dots formed in tetrahedral-shaped recesses. *Physica B*, 249–251, 210, 1998.
56. G.A. Prinz. Magnetoelectronics. *Science*, 282, 1660, 1998.
57. B.E. Kane. A silicon-based nuclear spin quantum computer. *Nature (London)*, 393, 133, 1998.
58. D.D. Awschalom, J.M. Kikkawa. Electron Spin and Optical Coherence in Semiconductors. *Phys. Today*, 52, 33, 1999.
59. S. Das Sarma, J. Fabian, X. Hu, I. Žutić. Spin Polarized Bipolar Transport and its applications. *Solid State Commun.*, 119, 207, 2001.

60. D. Loss and D.P. Di Vincenzo. Quantum computation with quantum dots. *Phys. Rev. A*, 57, 120, 1998.
61. O. Voskoboynikova. Magnetic properties of parabolic quantum dots in the presence of the spin-orbit interaction.
62. B. Boyacioglu, A. Chatterjee. Heat capacity and entropy of a GaAs quantum dot with Gaussian confinement. *J. Appl. Phys.*, 112, 083514, 2012. doi: 10.1063/1.4759350.
63. B. Boyacioglu, A. Chatterjee. Dia- and paramagnetism and total susceptibility of GaAs quantum dots with Gaussian confinement. *Phys. E*, 44, 1826, 2012. doi: 10.1016/j.physe.2012.05.001.
64. J.I. Climente, J. Planelles, J.L. Movilla. Magnetization of nanoscopic quantum rings and dots. *Phys. Rev. B*, 70, 081301, 2004. doi:10.1103/PhysRevB.70.081301.
65. S. Gumber, M. Kumar, M. Gamblur, M. Mohan, P. Kumar Jha, *Can. J. Phys.* 93, 1–5, 2015.
66. Sukirti Gumber, Manoj Kumar, Pradip Kumar Jha, Man Mohan. *Chin. Phys. B* vol. 25(5), 056502, 2016.
67. S. Gumber. *Metal*, *Can. J. Phys.*, 93(11), 1264–1268, 2015.
68. A.M. Babanlı, B.G. Ibragimov. Heat capacity of electrons in diluted magnetic semiconductor quantum dot. In *The 6th International conference on Control and Optimization with Industrial Applications*, July 11–13 2018. Baku, 2, 89–91, 2018.

1.2. Properties of Materials and Structures

2.1.2.1. Rheological properties of nanocomposites based on bifunctional clinoptilolite and ethylene/hexene copolymer. /I.V. Bayramova/. *Azerbaijan Chemical Journal*. – 2020. – #2. – pp. 83–89. – eng.; abs.: eng., az., rus.

The results of a study of the influence of temperature and shear stress on the rheological properties of ethylene/hexene copolymer and its clinoptilolite-filled nanocomposites are presented. Rheological measurements were carried out in the temperature range of 190–250 °C. The dependence of shear rate on shear stress, effective melt viscosity on shear rate, and melt viscosity on temperature in Arrhenius coordinates is studied. Using the universal temperature-invariant viscosity characteristics of nanocomposites allows make approximate calculations of effective viscosity close to the conditions of their processing by extrusion and injection molding by extrapolation to the region of high shear rates. Fig. 4, Tab. 1, Ref. 12.

Keywords: viscosity, shear rate, shear stress, nanocomposites, polymer melt, clinoptilolite

References:

1. Berlin A.A., Volfson S.A., Oshman V.G. *Principy sozdaniya kompozitsionnykh materialov*. M.: Himija, 1990. 240 s.
2. Simonov-Emeljanov I.D., Kuleznev V.N., Trofimicheva L.Z. Obobshchennye parametry dispersnoy struktury napolnennykh polimerov. *Plast. massy*. 1989. No 1. S. 19–22.
3. Kakhramanov N.T., Ismailzade A.D., Arzumanova N.B., Mammadli U.M., Martynova Q.S. Filled composites based on polyolefins and clinoptilolite. *Am. Sci. J.* 2016. V. 4. No 4. P. 60–65.
4. Bessonova N.P., Krashennikov S.V., Korobko A.P. i dr. Struktura i svojstva nizkokristallicheskih poliolefinov, modifitsirovannykh nanoalmazami. *Vysokomol. soed.* 2015. T. 57. No 6. S. 544–554.
5. Osama Al Helo, Osipchik V.S., Petuhova A.V., Kravchenko T.P., Kovalenko V.A. Modifikacija na polnennogo poli propilena. *Plast. massy*. 2009. № 1. S. 43–46.
6. Osipchik V.S., Nesterenkova A.I. Talkonapolnennye kompozicii na osnove polipropilena. *Plast. massy*. 2007. № 6. S. 44–46.
7. Ermakov S.N., Kerber M.L., Kravchenko T.P. Himicheskaja modifikacija i smeshenie polimerov pri reakcionnoj jekstruzii. *Plast. massy*. 2007. № 10. S. 32–41.
8. Peseckij S.S., Bogdanovich S.P. Nanokompozity poluchaemye dispergirovaniem glin v rasplavah polimerov. *Tez. Dokl. Mezhdun. Nauchno-tehnich. Konfer. «Polimernye kompozity i tribologija»*. Gomel, 2015. S. 5.
9. Kakhramanov N.T., Mustafayeva F.A., Arzumanova N.B., Hamedova L.H., Ljaljaeva R.N. Reologicheskie svojstva poli mernykh smesey na osnove polijetilena nizkoj i vysokoj plotnosti. *Kompozity i nanostrukury*. 2018. T. 10. № 4(40), S. 166–170.
10. Cherdynceva S.V., Belousov S.I., Krashennikov S.V. i dr. Vlijanie vida organicheskogo modifikatora montmorillonita na fiziko-himicheskie svojstva nanokompozitov na osnove poliamida-6, poluchennykh smesheniem v rasplave. *Plast. massy*. 2013. № 5. S. 39–43.

11. Kahramanov N.T., Bayramova I.V., Mamedli U.M. Deformacionno-prochnostnye svoystva napolnennykh polimernykh materialov. Elmi mecmueller, Milli Aviasiya Akademiyasi. 2017. No 3. S. 47–54.
12. Kakhramanov N.T., Huseynova Z.N., Osipchik V.S., Kurbanova R.V., Arzumanova N.B. Reaction extrusion of dynamic elastoplasts on the basis of polyolefines and butadien-nitril rubber. Azerb. chem. journ. 2019. № 1. S. 65–71.

2.1.2.2. Zinc-containing nanocomposites based on high-pressure polyethylene. /T.M. Guliyeva/. Azerbaijan Chemical Journal. – 2020. – #2. – pp. 34-38. – eng.; abs.: eng., az., rus.

The effect of nanofiller additives containing nanoparticles of zinc oxide stabilized by a polymer matrix of maleinated polyethylene, obtained by a mechanochemical method, on the properties of composites based on high-pressure polyethylene was studied by X-ray phase and thermographic analyzes. The improvement of the strength, deformation and rheological parameters, as well as the thermo-oxidative stability of the obtained nanocomposites was revealed, which is apparently due to the synergistic effect of the interaction of zinc-containing nanoparticles with maleic maleinated polyethylene groups. It is shown that polyethylene based nanocomposites can be processed both by pressing and by injection molding and extrusion, which expands its application field. Fig. 2, Tab. 2, Ref. 15.

Keywords: polyethylene, zinc-containing nanofillers, maleinized polyethylene, physicol - mechanical properties, X-ray phase and thermographic analyzes

References:

1. Joseph H. Koo. Polymer nanocomposites. Processing, characterization and applications. New York: McGraw-Hill. Nanoscience and Technology Series. 2006. 289 p.
2. Suzdalev I.P., Suzdalev P.I. Nanoclasteriy nanoclasternyye sistemy. Uspehi himii. 2001. T. 70. No 3. S. 203–240.
3. Pomogailo A.D. Giбриdnye polimer – neorganicheskie nanokompozity. Uspehi himii. 2000. T. 6. No 1. S. 60–89.
4. Pomogailo A.D., Rozenberg A.S., Ufli and I.E. Nanochastitcy metallov v polimerakh. M.: Himiia, 2000. 672 s.
5. Tretiakov A.O. Polimernye nanokompozity – materialy XXI veka. Oborudovanie i instrumenty dlia professionalov. 2003. V. 37. No 2. S. 18–20.
6. Mihailin Iu.A. Polimernye nanokompozitcionnye materialy. Polimernye materialy. 2009. No 7. S. 10–13.
7. Gubin S.P., Yurkov G.Yu., Kosobudsky I.D. Nanomaterials based on metal-containing nanoparticles in polyethylene and other carbon-chain polymers. International Journal of Materials and Product Technology. 2005. V. 23. No 1–2. P. 2–25.
8. Entsi clopediia polimerov. M.: Sovet. Entsi clopediia. 1974. T. 2. S. 328.
9. Foster L. Nanotekhnologii. Nauka, innovatsii i vozmozhnosti. M.: Tekhnosfera. 2008. 352 s.
10. Antipov E.M., Guseva M.A., Gerasin V.A., Korolyov Iu.M., Rebrov A.V., Fischer H.R., Razumovskaia I.V. Struktura i deformatsionnoe povedenie nanokompozitov na osnove PE i modifitsirovannykh gliin. Vysokomol. soed. 2003. A. T. 45. № 11. S. 1874–1884.
11. Savinova M.E., Semenova E.S., Sokolova M.D. Issledovanie fiziko-mehani cheskikh svoistv PE80B, modifitsirovannogo nanoshpineliu magniia i tceolitami. Elektr. na uchn. zhurn. Neftegazovoe delo. 2011. No 6. S. 328–333.
12. Kurbanova N.I., Aliyev A.T., Guliyeva T.M., Ragimova C.K., Axmadbekova C.F., Is henko N.Y., Nurullayeva D.R. Metal-containing nanoparticles in maleinized polyethylene matrix. PolyChar 26 World Forum on Advanced Materials. Georgiya. 2018. Tbilisi. P. 59.
13. Guliyeva T.M., Kurbanova N.I. Obtaining and study of the structure and properties of metal containing nanoparticles in the matrix of maleinized polyethylene. Genc tedqiqatchi. 2019. № 4. P. 32–37.
14. Praktikum po himii i fizike polimerov. Pod red. V.F. Kurenkova. M.: Himiia, 1990. 299 s.
15. Pomogailo A.D. Molekuliarnye polimer-polimernye kompozitsii. Sinteticheskie aspekty. Uspehi himii. 2002. T. 71. No 1. S. 5–38.

2.1.2.3. Effect of bentonite concentration on properties and regularity of crystallization of nanocomposite materials based on the mixtures of high and low-density polyethylene. /F.A. Mustafayeva, N.T. Kakhramanov, N.B. Arzumanova, N.Ya. Is henko, I.A. Ismayilov/. Azerbaijan Chemical Journal. – 2020. – #1. – pp. 53-58. – eng.; abs.: eng., az., rus.

The results of research of the effect of bentonite concentration on the regularity of crystallization and the nature of changes of ultimate tensile strength, tensile yield strength and elongation at break of nanocomposite materials based on the mixtures of high and low density polyethylene are presented. Fig. 2, Tab. 2, Ref. 9.

Keywords: crystallization, dilatometry, specific volume, polymer blend, high-density polyethylene, low-density polyethylene, bentonite

References:

1. Agwuncha S.C., Ibrahim I.D., Sadiku E.R. 14 Improving the thermal and flame resistance properties of polyolefins. Polyolefin fibres: Structure, properties and industrial applications. Edited by: Ugbohue S.C.I. Elsevier. 2017. P. 421–448.
2. Oliveira S.V., Araujo E.M., Pereira C.M.C., Leite. Polyethylene/bentonite clay nanocomposite with flame retardant properties. *Polymeros-Ciencia e Tecnologia. Special issue.* V. 27. 2017. P. 91–98.
3. Beyer G. Nanocomposites: a new class of flame retardants for polymers. *Plastics, additives and compounding.* V. 4. No 10. 2002. P. 22–28.
4. Ahmed L., Zhang B., Hatanaka L.C., Mannan. Application of polymer nanocomposites in the flame retardancy study. *Journal of loss prevention in the process industries.* V. 55. 2018. P. 381–391.
5. Kalgaonkar R.A., Jog J.P. 13 Polyolefin/clay nanocomposites. *Nanofibers and nanotechnology in textiles.* Edited by: Brown P.J., Stevens K. Woodhead. Publishing Limited. 2007. P. 351–385.
6. Mostal'gina L.V., Elizarova S.N., Kostin A.V. *Bentonitovye gliny Zaural'ya: ekologiya i zdorov'e cheloveka.* Kurgan: Izd-vo Kurganskogo gosudarstvennogo universiteta. 2010. 148 s.
7. Teh J.W., Blom H.P., Rudin A. A study on the crystallization behaviour of polypropylene, polyethylene and their blends by dynamic mechanical and thermal methods. *Polymer.* V. 35. No 8. 1994. P. 1680–1687.
8. Fikhtner R.R., Volkov T.I., Shalatskaya S.A., Trizno M.S. Study of crystallization of industrial polyethylene and polyethylene mixture. *Polymer Science U.S.S.R.* V. 21. No 10. 1979. P. 2596–2603.
9. Peneva Y., Minkova L. Non-isothermal and isothermal crystallization of nanocomposites based on functionalized polyethylenes. *Polymer Testing.* V. 23. No 3. 2006. P. 366–376.

2.1.2.4. Metal-containing nanocomposites based on isotactic polypropylene and ethylene-propylene-diene rubber. /N.A. Alimirzayeva/. *Azerbaijan Chemical Journal.* – 2020. – #1. – pp. 41-45. – eng.; abs.: eng., az., rus.

The work summarizes the data of studies on the effect of nanofiller additives containing copper oxide nanoparticles stabilized by a high-pressure polyethylene matrix obtained by the mechanochemical method on the physico-mechanical, rheological properties and crystallization of thermoplastic mixed elastomers on the basis of isotactic polypropylene and ethylene propylene diene rubber. The work presents the prospects of using these additives to elastomers that provides for producing a fine-spherical layered structure of the composition characterized by improved melt flow rates, rheological, physicomachanical properties, and thereby expand the scope of application obtained nanocomposites. Fig. 3, Tab. 1, Ref. 12.

Keywords: metal-containing nanocomposites, isotactic polypropylene, ethylene propylene diene rubber, copper oxide nanoparticles, physico-mechanical, rheological properties, crystallization

References:

1. Joseph H. Koo. *Polymer nanocomposites. Processing, characterization and applications.* New York: McGraw-Hill. Nanoscience and Technology Series. 2006. 289 p.
2. Suzdalev I.P., Suzdalev P.I. Nanoclusters and nanocluster systems. *Advances in Chemistry.* 2001. V. 70. No 3. P. 203–240.
3. Pomogailo A.D. Hybrid polymer – inorganic nanocomposites. *Advances in chemistry.* 2000. V. 6. No 1. P. 60–89.
4. Pomogailo A.D., Rosenberg A.S., Uflyand I.E. *Metal nanoparticles in polymers.* M.: Chemistry, 2000. 672 p.
5. Tretyakov A.O. *Polymer nanocomposites materials of the 21st century. Equipment and tools for professionals.* 2003. (37). No 2. P. 18–20.
6. Mikhaylin Yu.A. *Polymer nanocomposite materials.* Polymer materials. 2009. No 7. P. 10–13.

7. Gubin S.P., Yurkov G.Yu., Kosobudsky I.D. Nanomaterials based on metal-containing nanoparticles in polyethylene and other carbon-chain polymers. *International Journal of Materials and Product Technology*. 2005. V. 23. No 1–2. P. 2–25.
8. Prut E.V., Erina N.A., Karger-Kocsis J., Medintseva T.I. Effects of Blend Composition and Dynamic Vulcanization on the Morphology and Dynamic Viscoelastic Properties of PP/EPDM Blends. *J. Appl. Polym. Sci.* 2008. V. 109. P. 1212–1220.
9. Kurbanova N.I., Alimirzoeva N.A., Guseinova Z.N., Nurullayeva D.R. Ecological Method of Preparation of Metal-Containing Nanoparticles in Polyethylene Matrix I TWCCST. 2017. Baku. Azerbaijan. 10-13 Sept. Book of Proceedings. P. 24–26.
10. Kurbanova N.I., Alimirzoeva N.A., Guseinova Z.N., Kuliyeu A.M., Kakhramanov N.T., Gasanova A.A. Preparation and investigation of properties of metal-containing nanocomposites on the basis of isotactic polypropylene and ethylene propylene diene rubber. *Processes of Petrochemistry and Oil Refining*. 2018. V.19. No 3. P. 274–281.
11. Kurbanova N.I., Alimirzoyeva N.A., Kuliyeu A.M., Guseinova Z.N., Ischenko N.Ya. Metal-containing Nanocomposites on the Basis of Isotactic Polypropylene. *Inorganic materials: applied research*. 2019. V. 10. No 2. P. 411–415.
12. Kurbanova N.I., Alimirzoeva N.A., Arzumanova N.B., Kachramanov N.T. Influence of metal containing nanofiller on rheological properties of mixed thermoelastoplasts on the basis of isotactic polypropylene and ternary ethylene-propylenediene elastomer. *Processes of Petrochemistry and Oil Refining*. 2019. V. 20. No 4. P. 254–260.

2.1.2.5. Influence of single-walled carbon nanotubes on dielectric relaxation and electric conductivity of a smectic A liquid crystal with positive dielectric anisotropy. /T.D. Ibragimov, A.R. Imamaliyev, G.F. Ganizade/. *Azerbaijan Journal of Physics*. – 2020. – vol. 26. – #3. – pp. 3-6. – eng.; abs.: eng.

The effect of single-walled carbon nanotubes (SWCNTs) on the dielectric and conductivity properties of a smectic A liquid crystal 4-nitrophenyl-4'-decyloxybenzoic acid has been studied. It is shown that the additive of SWCNTs with concentration of 0.5% leads to a decrease in the order parameter of 5CB. In this case, the clearing point is raised, the longitudinal component of the dielectric permittivity decreases while the transverse component increases. The incipient percolation effect promotes to the dominance of hopping electron conductivity over ionic conductivity, leading to an increase in specific conductance. Fig. 3, Ref. 9.

Keywords: smectic A liquid crystal; single-walled carbon nanotubes, dielectric permittivity; electric conductivity

References:

1. N.R. Jber, A.A. Rashad, M.S. Shihab. *J. Molecular Structure*, 2013. 1043, 28–36.
2. S.P. Yadav, S. Singh. *Progress in Materials Science*, 2016. 80, 38–76.
3. D. Singh, U.B. Singh, M.B. Pandey, R. Dabrowski, R. Dhar. *Optical Materials*, 2018. 84, 16–21.
4. D. Singh, U.B. Singh, M.B. Pandey, R. Dhar. *Liquid Crystals*, 2019. 46 (9), 1389-1395.
5. T. Vimal, S. Pandey, S.K. Gupta, D.P. Singh, R. Manohar. 2015. *Journal of Molecular Liquids*, 204, 21–26.
6. G.V. Varshini, D.S. Rao, P.K. Mukherjee, S.K. Prasad. *J. Phys.* 2018. Chem. B, 122 (47), 10774–10781.
7. R. Verma, M. Mishra, R. Dhar, R. Dabrowski. *Liquid Crystals*, 2016. 44 (3), 544–556.
8. R. Verma, M. Mishra, R. Dhar, R. Dabrowski. *Journal of Molecular Liquids*, 2016. 221, 190–196.
9. L. Blinov. *Structure and Properties of Liquid Crystal*, 2011. Springer: New York.

2.1.2.6. Dielectric, conductivity, and electro-optic properties of liquid crystal 5cb doped by single-walled carbon nanotubes. /T.D. Ibragimov, A.R. Imamaliyev, G.F. Ganizade/. *Azerbaijan Journal of Physics*. – 2020. – vol. 26. – #2. – pp. 10-14. – eng.; abs.: eng.

The effect of single-walled carbon nanotubes (SWCNTs) on the dielectric, conductive, and electro-optic properties of nematic liquid crystal 4-cyano-4'-pentylbiphenyl (5CB) has been studied. It is shown that the additive of SWCNTs with concentration of 0.5% leads to strong interaction between SWCNTs and molecules, which increases the order parameter of 5CB. As a result, the clearing point is raised, the longitudinal component of the dielectric permittivity increases while the transverse component decreases. The incipient percolation effect promotes to the dominance of hopping electron conductivity over ionic conductivity, leading to an increase in specific conductivity. In this case, the elastic splay constant of 5CB is

enhanced and, accordingly, the threshold voltage of the S-effect increases. A decrease in the Van-der-Waals interaction between molecules decreases viscosity. As a result, the flip-flop motion of molecules becomes easier and the switching time reduces. Fig. 9, Tab. 1, Ref. 8.

Keywords: liquid crystal, single-walled carbon nanotubes, dielectric relaxation, electric conductivity, threshold voltage

References:

1. N. R. Jber, A.A. Rashad. M. S. Shihab. Effects of carbon nanotubes on the physical properties of a nematic liquid crystal N-(4-methoxybenzylidene)-4-butylaniline. *Journal of Molecular Structure*, v. 1043, pp. 28–36, 2013.
2. S. V. Burylov, Yu. L. Raikher. Orientation of a solid particle embedded in a monodomain nematic liquid crystal. *Phys. Rev. E*, v. 50, pp. 358–367, 1994.
3. D. Singh, U. Bahadur, S.M. Brushan, R. Dabrowski, R. Dhar. Improvement of orientational order and display parameters of liquid crystalline material dispersed with singlewall carbon nanotubes. *Material Letters*, v. 216, pp. 5–7, 2018.
4. K.A. Park, S.M. Lee, S.H. Lee, Y.H. Lee. Anchoring a liquid crystal molecule on a singlewalled carbon nanotube. *J. Phys. Chem. C*, v. 111, pp. 1620–1624, 2007.
5. I. Dierking, G. Scalia, P. Morales. Liquid crystal–carbon nanotube dispersions. *Journal of Applied Physics*, v. 97, pp. 044309-1–5, 2005.
6. M.V. Gorkunov, M.A. Osipov. Mean-field theory of a nematic liquid crystal doped with anisotropic nanoparticles. *Soft Matter*. v. 7, pp. 4348–4356, 2011.
7. A.V. Koval'chuk. Low-frequency dielectric relaxation at the tunnel charge transfer across the liquid/electrode interface. *Functional Materials*. v. 8, No. 4, pp. 690–693, 2001.
8. L. Blinov. *Structure and properties of liquid crystal*, New York: Springer, p. 439, 2011.

2.1.2.7. EPR investigations of γ -irradiated polytetrafluoroethylene/CdS nanocomposites. /E.G. Hajieva/. *Azerbaijan Journal of Physics*. – 2019. – vol. 25. – #4. – pp. 22-25. – eng.; abs.: eng.

It is shown that g-factor values for PTFE/CdS nanocomposites at the dose 5kQr correspond to free electrons. The nonlinear dependence of signal intensity on craze number is observed with an increase of craze cycle in EPR spectra. Fig. 4, Tab. 1, Ref. 23.

Keywords: γ -irradiation, nanocomposites, polytetrafluoroethylene, magnetic field, EPR-spectra, dielectric properties, dielectric loss, dielectric constant, crazing, g-factor

References:

1. A.D. Pomogailo, A.S. Rozenberg, I.E. Ufland. *The metal nano-particles in polymers*. M., Chemistry, 2000.
2. A.L. Buchachenko. Nano-chemistry is the direct way to high technologies. *Uspehi himii* | 2003, vol. 53, № 5, pp. 419–421.
3. Z.P. Hedwig. *Radiation chemistry of molecules (E.E. Finkel): Polymer radiation electric conduction*, M., Atomizdat, 1978, pp. 121–134.
4. P. Keyzer, K. Tzui, F. Williams. *The investigation of stabilized electrons in low-molecular organic glasses and polymers by methods of optical spectroscopy and EPR (E.E. Finkel)*, M., Atomizdat, 1978, pp. 135–175.
5. A.V. Vannikov, V.K. Matveyev, V.P. Sichkar, A.P. Tutnev. *Electric properties: Radiation effects in polymers*. M., Science, 1982, p. 272.
6. A.M. Magerramov, M.K. Dashdamirov. *Chemistry of high energies*. 2005, vol. 39, № 3, pp. 176–182.
7. A.L. Volinskii, A.E. Mikushev, L.M. Yarisheva, N.F. Bakeyeva. *Russian Chemical Journal (Journal of Russian Chemical Community named after D.M. Mendeleev)*, 2005, vol. XLIX, № 6, pp. 118–128.
8. A.M. Magerramov, M.A. Nuriyev. The influence of γ -radiation on charge state of polytetrafluoroethylene/CdS nanocomposites, *Journal of Radiation Research. ANAS*, 2015, v. 2, № 1, p. 18–25.
9. A.P. Tutnev, V.S. Sayenko, E.D. Pojidayev, N.S. Kostyukov. *Polymer dielectric properties in the ionizing radiation fields*, M., Science, 2005, p. 453.
10. M.U. Yablokov, V.G. Shevchenko, A.B. Gilman, A.A. Kuznecova. *Chemistry of High Energies*, 2014, vol. 48, № 4, pp. 326–330.

11. M.A. Nuriyev, A.M. Magerramov, A.A. Shukurova. Influence of gamma irradiation on charge state of nanocomposites PTFE/CdS. VII Euroasian Conf. "Nuclear Sci. and ITS Appl." Baku, 2014, pp. 244–245.
12. M.K. Kerimov, A.M. Magerramov, E.G. Hajieva. Electrothermo-luminescence of polymer compositions. 8th International Symposium "Polymer for Advanced Technology", 2005, Budapest, p. 50.
13. A.M. Magerramov, M.A. Nuriyev, E.G. Gadjiyeva. Electron treatment of materials: The influence of γ -radiation on polypropylene/CdS composite photothermoluminescence, 2017, 53(5), pp. 21–25.
14. A.L. Buchachenko. Spectroscopic methods of polymer investigations: The polymer investigation by the method of electron paramagnetic resonance, M., "Znanye", 1975, p. 82.
15. V.K. Mlinchuk, E.R. Klinshont, S.Y. Pshejckii. Macroradicals, M., Chemistry, 1980, p. 264.
16. V.Y. Kabanov, V.I. Feldman. Chemistry of high polymers. 2009, vol. 43, № 1, pp. 5–21.
17. M.A. Bruk. Chemistry of High Energies. 2006, vol. 40, № 6, p. 403.
18. V.S. Sayenko, V.I. Feldman, A.P. Tutnev. Chemistry of High Energies. 2011, vol. 45, № 1, pp. 51–54.
19. A. Oshima, S. Ikeda, E. Katoh, Y. Tabata. Radiation physics and chemistry, 2001, v. 62, № 1, pp. 39–45.
20. M.A. Bruk, A.V. Spirin, et al. II International Conf. "Micro- and nanoelectronics–2005", Moscow, Zvenigorod, Abstracts, 2005, p. 1.
21. A.M. Magerramov, M.A. Nuriyev, A.A. Shukurova, E.A. Alahyarov. Nanocomposites on the base of nonaxial oriented polymers. Materials of 1st International Scientific Conference "Nanotechnologies and its applications in technique", Baku, APU, 2010, pp. 98–101.
22. E.I. Kapius. Journal of Physical Chemistry, 2011, vol. 85, № 4, pp. 748–752.
23. I.V. Kuleshov. Polymer radiothermoluminescence, M., Chemistry, 1991, p. 128.

3. NANOBIOLOGY

3.2. Biophysics

2.3.2.1. Structural and magnetic properties of silver oleic acid multifunctional nanohybrids. /S. Khutsishvili, P. Toidze, M. Donadze, M. Gabrichidze, T. Agladze, N. Makhaldiani/. Annals of Agrarian Science. – 2019. – vol. 17. – #2. – pp. 242-250. – eng.; abs.: eng.

Sols of core-shell silver NPs are synthesized by an electrochemical method. The method provides for the ability to adjust the particle size by changing both the concentration of oleic acid and the residence time W_0 in the organic phase. We synthesized silver nanoparticles with oleic acid concentration of 0.25% (Ag&0.25%OA) and 0.75% (Ag&0.75%OA). The silver nanoparticles have been studied using modern physical–chemical methods: Transmission Electron Microscopy (TEM); Fourier Transform Infrared Spectroscopy (FT-IR); Dynamic Light Scattering (DLS); Thermogravimetric and Differential Thermal Analysis (TGA and DTA); Electron Paramagnetic Resonance (EPR). DTA curves indicate the chemical nature of bond ligand in the secondary shell. This conclusion is supported by quantum chemical simulation using the quantum-chemical software HyperChem-8 and semi-empirical calculation method ZINDO. In the EPR spectra of silver-containing sols Ag&0.25%OA and Ag&0.75%OA a complex wide asymmetric signal with several resonant lines is recorded, which is consistent with a wide-size distribution of nanoparticles. It is important to note that a change in the oleic acid layers of the nanoparticles seems to affect the dimension of the nanocrystallites that are being formed. The presence of the FMR resonance line in Ag&0.75%OA may indicate the presence of Ag-cubic cells in nanoparticles with internal magnetic fields significantly larger than the Zeeman field, the available EPR in the X-band range. Fig. 5, Tab. 2, Ref. 30.

Keywords: Core-shell, nanoparticles, oleic acid, ligand, charge, activation energy

References:

- 1 M. Donadze, M. Gabrichidze, S. Calvache, T. Agladze, Novel method of preparation of the hybrid metal (I)–metal (II) oxide nanoparticles, Int. J. Transactions of the IMF 94 (1) (2016)16-23.
- 2 J.H. Flynn, The Isoconversional method for determination energy of activation at constant heating rate, J. Therm. Anal. 27 (1) (1983) 95-102
- 3 T. A. Ozawa New method of analyzing thermo- gravimetric data, Bull. Chem. Soc. Japan 38 (1965) 1881-1886.

- 4 T. Agladze, M. Donadze, P. Toidze et al. Synthesis and Size Tuning of Metal Nanoparticles, *Z. Phys. Chem.* 227 (2013) 1187-1198.
- 5 D.H. Lee FTIR spectral characterization of thin film coatings of oleic acid on glasses, *J. Mat. Sci.* 34 (1999) 139-146.
- 6 C. Doyle, Kinetic analysis of thermogravimetric data, *J. Appl. Polym. Sci.* 5 (15) (1961) 285-292.
- 7 K. Yang, H. Peng, Y. Wen, N. Li, Re-examination of characteristic FTIR spectrum of secondary layer in bilayer oleic acid-coated Fe₃O₄ nanoparticles, *Appl. Surf. Sci.* 256 (2010) 3093-3097.
- 8 Q. Lan, C. Liu, F. Yang et al., Synthesis of bilayer oleic acid-coated Fe₃O₄ and interface nanoparticles and their application in pH-responsive Pickering emulsions, *J. Coll. Sci.* 310 (2007) 260-269.
- 9 L. Shen, P.E. Laibinis and T.A. Hatton, Bilayer surfactant stabilized magnetic fluids: synthesis and interactions at interfaces, *Langmuir* 15 (1999) 447-453.
- 10 S. Nellutla, S. Nori, S.R. Singamaneni, J.T. Prater, J. Narayan, A.I. Smirnov, Multi-frequency ferromagnetic resonance investigation of nickel nanocubes encapsulated in diamagnetic magnesium oxide matrix, *J. Appl. Phys.* 120 (22) (2016) 1-9.
- 11 V. Angelov, H. Velichkova, E. Ivanov, R. Kotsilkova, M.H. Delville, M. Cangiotti, A. Fattori, M.F. Ottaviani, EPR and rheological study of hybrid interfaces in gold-clay-epoxy nanocomposites, *Langmuir* 30 (44) (2014) 13411-13421.
- 12 M.V. Lesnichaya, B.G. Sukhov, E. Gasilova, G. Aleksandrova, T. Vakul'skaya, S. Khutsishvili, A. Sapozhnikov, I. Klimenkov, B. Trofimov, Chiroplasmonic magnetic gold nanocomposites produced by one-step aqueous method using κ-carrageenan, *Carbohydrate Polymers* 175 (2017) 18-26.
- 13 A. Smirnov, EPR studies of nanomaterials. In: (Ed.) S. Misra, *Multifrequency* Wiley-VCH, Verlag (2011) 825-843.
- 14 M. Schlott, H. Schaeffer, B. Elschner, Gd³⁺-ESR in the intermediate valent cerium compounds Ce_xLa_{1-x}O₂, *Zeitschrift für Physik B Condensed Matter* 63 (4) (1986) 427-436.
- 15 J. Stöhr, H. Siegmann, *Magnetism: From Fundamentals to Nanoscale Dynamics*, Springer-Verlag, Berlin Heidelberg, 2006.
- 16 P. Venegas, P. Netto, Exchange narrowing effects in the EPR linewidth of Gd diluted in Ce compounds, *J. Appl. Phys.* 83 (11) (1998) 6958-6968.
- 17 P. Shin, S. Wu, Magnetic anisotropic energy gap and strain effect in Au nanoparticles, *Nanoscale Research Letters* 5 (2010) 25-30.
- 18 M. Kakazey, N. Ivanova, G. Sokolsky, J. Gonzalez-Rodriguez, Electron paramagnetic resonance of MnO₂ powders, *Electrochemical and Solid-State Letters* 4 (5) (2001) 1-4.
- 19 F. Blatter, K. Blazey, Conduction electron spin resonance of silver in zeolite AgY, *Z. Phys. D - Atoms, Molecules and Clusters* 18 (1991) 427-429.
- 20 S. Sako Kimura, K. Size, Effect in CESR of magnesium and calcium small particles, *Surface Sci.* 156 (1985) 511-515.
- 21 X. Li, A. Vannice, ESR studies of well-dispersed Ag crystallites on SiO₂, *J. Catalysis* 151 (1995) 87-95.
- 22 M. Ali, A. Shames, S. Gangopadhyay, B. Saha, D. Meyerstein, Silver(II) complexes of tetrazamacrocycles: studies on e.p.r. and electron transfer kinetics with thiosulfate ion, *Transition Metal Chemistry (Dordrecht, Neth.)* 29 (2004) 463-470.
- 23 M. Kester, A. Allred, Ligand-induced disproportionation of silver (I), *J. American Chemical Society* 94 (1972) 7189-7189.
- 24 S. Khutsishvili, T. Vakul'skaya, N. Kuznetsova, T. Ermakova, A. Pozdnyakov, G. Prozorova, Formation of stable paramagnetic nano-composites containing zero-valence silver and copper in a polymeric matrix, *J. Phys. Chem. C* 118 (33) (2014) 19338-1701
- 25 J. McMillan, B. Smaler, Paramagnetic resonance of some silver(II) compounds. *J. Chem. Phys.* 35 (1961) 1698-1701.
- 26 H. Moon, J. Kim, M. Suh, Redox-active porous organic framework Chemie producing silver nanoparticles from Ag^I ions at room temperature, *Angew. Chem. Int. Ed.* 44 (2005) 1261-1265.
- 27 G. Deligiannakis, Y. Trapalis, C. Boukos, N. Kordas, CW and pulsed EPR study of silver nanoparticles in SiO₂ matrix, *J. Sol-Gel Science Technology* 13 (1998) 503-508.
- 28 S. Khutsishvili, T. Vakul'skaya, G. Aleksandrova, B. Sukhov, Stabilized silver nanoparticles and clusters Ag_n of humic-based bioactive nanocomposites, *J. Cluster Sci.* 28 (2017) 3067-3074.
- 29 V. Timoshenko, T. Shabatina, Yu. Morozov, G. Sergeev, Complexation and chemical transformations in the ternary system silver-carbon tetrachloride- mesogenic cyanobiphenyl at low temperatures, *J. Struc. Chem.* 47 (1) (2006) 145-150.
- 30 J. Michalik, H. Yamada, D. Brown, L. Kevan, Small silver clusters in smectic clay interlayers, *J. Phys. Chem.* 100 (1996) 4213-4218.250.

3.3. Biochemistry

2.3.3.1. Revealing a Nonergodic Mechanistic Pattern for Electron Exchange between Azurin and Electrodes Coated by Nanofilms under the Glassy Environmental Conditions. /T. Dolidze, R. van Eldik, D. Khoshtariya/. *Bulletin of the Georgian National Academy of Sciences.* – 2019. – vol. 13. – #4. – pp. 97-103. – eng.; abs.: eng., geo.

Fast-scan protein-film voltammetry was applied to explore interfacial biomimetic electron exchange under the environmental glass forming conditions. Gold electrodes were coated with 1-pentanethiol SAM–azurin (Az, blue cupredoxin) assemblies and placed in contact with a water-doped and buffered protic ionic melts of choline dihydrogen phosphate ([ch][dhp]), served as electrolyte media, allowing for a necessary cell conductivity under the virtually solid, semi-solid and liquid electrolyte conditions over 273–353 K, within which the electron exchange rate was studied as a function of the water amount and temperature. Exposure of the Az films to the semi-solid electrolyte greatly affected the protein's conformational dynamics, hence the ET rate, via the mechanism occurring in the extra complicated dynamically-controlled regime. Results are compared to the earlier studies on the reference system with a conventional electrolyte, allowing for the disclosure of mutually-entangled mechanistic motifs. Under the “standard” condition (with no [ch][dhp] added), the Az biomolecule may reside in an apparently ergodic state, whereas upon adding of [ch][dhp] to allow water content ranging between 6 to 15 waters per [ch][dhp], system displays anomalous temperature dependences, suggesting that the reactive system crosses a broad, well-manifested nonergodic zone which arises from the continuous interplay (freezing/unfreezing) of ET-coupled highly cooperative conformational modes of the Az protein, inherently linked to the electrolyte's slowest collective relaxation(s). Above this [ch][dhp] concentration, allowing the water content between 1.65 to 3.7 waters per ion pair, the system returns to a series of new, quasi-ergodic states, with each displaying virtually linear Arrhenius patterns yet with distinct parameters. Fig. 2, Tab. 3, Ref. 22.

Keywords: redox protein, electron exchange, interphase, self-assembly, nonergodicity

References:

1. Ellis R.J. (2001) Macromolecular crowding: an important but neglected aspect of the intracellular environment. *Curr. Opin Struct. Biol.* 11:114–119.
2. Zhou H.-X., Rivas G., Minton A.P. (2008) Macromolecular crowding and confinement: biochemical, biophysical and potential physiological consequences. *Annual Rev. Biophys.*, 37:375–397.
3. Mittal S., Chowhan R.K., Singh L.R. (2015) Macromolecular crowding: macromolecules' friend or foe. *Biochim. Biophys. Acta* 1850:1822–1831.
4. Frauenfelder H., et al. (2009) A unified model of protein dynamics. *Proc. Natl. Acad. Sci. USA* 106:5129–5134.
5. Jansson H., Bergman R., Swenson J. (2011) Role of solvent for the dynamics and the glass transition of proteins. *J. Phys. Chem. B* 115:4099–4109.
6. Lee A.G. (2004) How lipids affect the activities of integral membrane proteins. *Biochim. Biophys. Acta* 1666:62–87.
7. Bondar A.-N., White S.H. (2004) Hydrogen bond dynamics in membrane protein function. *Biochim. Biophys. Acta* 1818:942–950.
8. Belieres J.-P., Angell C.A. (2007) Protic ionic liquids: preparation, characterization and proton free energy level representation. *J. Phys. Chem. B* 111:4926–4937.
9. Rana U.A., et al. (2010) Proton transport in choline dihydrogen phosphate/H₃PO₄ mixtures. *Phys. Chem. Chem. Phys.* 12:11291–11298.
10. Fujita K., Ohno H. (2010) Enzymatic activity and thermal stability of metalloproteins in hydrated ionic liquids. *Biopolymers* 93:1093–1099.
11. Khoshtariya D.E., et al. (2006) Kinetic, thermodynamic, and mechanistic patterns for free (unbound) cytochrome C at Au/SAM junctions: impact of electronic coupling, hydrostatic pressure, and stabilizing/denaturing additives. *Chemistry—A European J.* 12:7041–7056.
12. Khoshtariya D.E., et al. (2010) Fundamental signatures of short and long-range electron transfer for the blue copper protein azurin at Au/SAM junctions. *Proc. Natl. Acad. Sci. USA* 107:2757–2762.
13. Khoshtariya D.E. et al. (2014) Long-range electron transfer with myoglobin immobilized at Au/Mixed–SAM junctions: mechanistic impact of the strong protein confinement. *J. Phys. Chem. B* 118:692–706.
14. Weber K., Hockett L., Creager S. (1997) Long-range electronic coupling between ferrocene and gold in alkanethiolate-based monolayers on electrodes. *J. Phys. Chem. B* 101:8286–8291.

15. Palmer R.G. (1982) Broken ergodicity. *Adv. Phys.* 31:669–735.
16. Mauro J.C., Gupta P.K., Loucks R.J. (2007) Continuously broken ergodicity. *J. Chem. Phys.* 126:184511 (11 p.).
17. Mallamace F., et al. (2011) The role of the dynamic crossover temperature and the arrest in glass-forming fluids. *Eur. Phys. J. E* 34:94 (94 p).
18. Lebard D.N., Matyushov D.V. (2010) Protein-water electrostatics and principles of bioenergetics. *Phys. Chem. Chem. Phys.* 12:15335–15348.
19. Matyushov D.V. (2011) Nanosecond Stokes shift dynamics, dynamical transition, and gigantic reorganization energy of hydrated heme proteins. *J. Phys. Chem. B*, 115:10715–10724.
20. Matyushov, D.V. (2013) Protein electron transfer: dynamics and statistics. *J. Chem. Phys.* 139:025102 (12 p.).
21. Zusman L.D. (1994) Dynamics solvent effects in electron transfer reactions. *Z. Phys. Chem.* 186:1–29.
22. Bixon M., Jortner J. (1999) Electron transfer – from isolated molecules to biomolecules. *Adv. Chem. Phys.* 106:35–202.

2.3.3.2. A new generation of biocompatible nanoparticles made of resorbable poly(ester amide)s. /T. Kantaria, T. Kantaria, G. Titvinidze, S. Kobauri, M. Ksovreli, T. Kachlishvili, N. Kulikova, D. Tugushi and R. Katsarava/. *Annals of Agrarian Science.* – 2019. – vol. 17. – #1. – pp. 49-58. – eng.; abs.: eng.

A new generation of resorbable nanoparticles (NPs) were prepared on the basis of amino acid-based biodegradable (AABB) poly (ester amide)s (PEAs) for drug delivery application. The NPs were fabricated by cost-effective polymer deposition/solvent displacement (nanoprecipitation) method on the basis of three different AABB PEAs recently developed by our group: (i) PEA composed of amino acid leucine as a basic component, (ii) cationic PEA composed of amino acid arginine for imparting positive charge, and (iii) functional PEA composed of amino acid leucine and lateral poly(ethylene glycol) groups acting as surfactant as well as PEGylating agent. The mean particle diameter (MPD), polydispersity index (PDI) and zeta-potential (ZP) were determined by Dynamic Light Scattering (DLS). Moreover, the stability (resuspendability) of the NPs over time at low temperature was investigated. The NPs were studied for *in vitro* cell compatibility using four different stable cell lines: A549 (human), U937 (human), RAW264.7 (murine), Hepa 1-6 (murine). The produced nanoparticles exhibit high stability and cell compatibility and have potential for the application as drug delivery devices. Fig. 2, Ref. 37.

Keywords: Biodegradable polymers; nanoprecipitation; nanoparticles; biodegradable surfactant; PEGylation; *in vitro* cell compatibility

References:

- 1 R. Bisht, A. Mandal, J.K. Jaiswal, I.D. Rupenthal, Nanocarrier mediated retinal drug delivery: overcoming ocular barriers to treat posterior eye diseases, *Advanced review* 10 (2018) e1473. doi: 10.1002/wnan.1473.
- 2 L. Zhang, F.X. Gu, J.M. Chan, A.Z. Wang, R.S. Langer, O.C. Farokhzad, Nanoparticles in medicine: therapeutic applications and developments, *Clin. Pharmacol. Ther.* 83 (2018) 761–769.
- 3 S. Mallakpour, V. Behranvand, Polymeric nanoparticles: recent development in synthesis and application, *Express Polym. Lett.* 10 (2016) 895–913.
- 4 S. Schubert, J.T. Delaney Jr, U.S. Schubert, Nanoprecipitation and nanoformulation of polymers: from history to powerful possibilities beyond poly(lactic acid), *Soft Matter* 7 (2011) 1581–1588.
- 5 R. Liang, L. Dong, R. Deng, J. Wang, K. Wang, M. Sullivan, J. Tao, Surfactant-free biodegradable polymeric nanoparticles generated from self-organized precipitation route: Cellular uptake and cytotoxicity, *Eur. Polym. J.* 57 (2014) 187–201.
- 6 L.A. Dailey, E. Kleemann, M. Wittmar, T. Gessler, T. Schmehl, C. Roberts, W. Seeger, T. Kissel, Surfactant-free, biodegradable nanoparticles for aerosol therapy based on the branched polyesters, DEAPA-PVAL-g-PLGA. *Pharm. Res.* 20 (2003) 2011–2020.
- 7 K. Ulbrich, K. Hola, V. Subr, A. Bakandritsos, J. Tucek, B. Zboril, Targeted drug delivery with polymers and magnetic nanoparticles: covalent and noncovalent approaches, release control, and clinical studies, *Chem. Rev.* 116 (2016) 5338–5431.
- 8 E. Marin, M.I. Briceno, C. Caballero-George, Critical evaluation of biodegradable polymers used in nanodrugs, *Int. J. Nanomedicine.* 8 (2013) 3071–3091.
- 9 D.K. Knight, E.R. Gillies, K. Mequanint, Strategies in functional poly(ester amide) syntheses to study human coronary artery smooth muscle cell interactions, *Biomacromolecules.* 12 (2011) 2475–2487.
- 10 M. Jacoby, Custom-made biomaterials. *Chem. Eng. News.* 79 (2001) 30–35.

- 11 N. Arabuli, G. Tsitlanadze, L. Edilashvili, D. Kharadze, T. Gogvadze, V. Beridze, Z. Gomurashvili, R. Katsarava, Heterochain polymers based on natural α -amino acids. Synthesis and enzymatic hydrolysis of regular poly(ester amide)s based on bis(L-phenylalanine) α,ω -alkylene diesters and adipic acid, *Macromol. Chem. Phys.* 195 (1994) 2279–2289.
- 12 T. Kartvelishvili, G. Tsitlanadze, L. Edilashvili, N. Japaridze, R. Katsarava, Amino acid based bioanalogous polymers. Regular poly(ester urethane)s and poly(ester urea)s based on bis(phenylalanine)- α,ω -alkylene diesters, *Macromol. Chem. Phys.* 198 (1997) 1921–1932.
- 13 R. Katsarava, D. Tugushi, Z. Gomurashvili, Poly(ester urea) polymers and methods of use. US Patent No. 8,765,164 (accessed on 1 July 2014) <https://www.google.ch/patents/US8765164>.
- 14 R. Katsarava, Active polycondensation: from peptide chemistry to amino acid based biodegradable polymers, *Macromol. Symp.* 199 (2003) 419–429.
- 15 R. Katsarava, Z. Gomurashvili, Biodegradable polymers composed of naturally occurring α -amino acids, in: A. Lendlein, A. Sisson (Eds.), *Handbook of Biodegradable Polymers—Isolation, Synthesis, Characterization and Applications*, Wiley-VCH Verlag GmbH & Co. KGaA: Weinheim, Germany, 2011, pp. 107–131.
- 16 K. Defife, K. Grako, G. Cruz-Aranda, S. Price, R. Chantung, K. Pacpherson, R. Koshabe, S. Gopalan, W.G. Turnell, Poly(ester amide) co-polymers promote blood and tissue compatibility, *J. Biomater. Sci.* 20 (2009) 1495–1511.
- 17 H. Sun, F.M. Meng, A.A. Dias, M. Hendriks, J. Feijen, Z. Zhong, α -Amino acid containing degradable polymers as functional biomaterials: rational design, synthetic pathway, and biomedical applications, *Biomacromolecules*. 12 (2011) 1937–1955.
- 18 A. Ghaffar, G.J.J. Draaisma, G. Mihov, A.A. Dias, P.J. Schoenmakers, S.J. van der Val, Monitoring the in vitro enzyme-mediated degradation of degradable poly(ester amide) for controlled drug delivery by LC-ToF-MS, *Biomacromolecules*. 12 (2011) 3243–3251.
- 19 M. Trollsas, B. Maslanka, N. Pham, Q. Lin, S. Hossainy, H.L. Hsu, M.H. Ngo, Polyesteramide coatings for drug eluting stents: controlling drug release by polymer engineering, *Stud. Mechanobiol. Tissue Eng. Biomater.* 8 (2011) 127–143.
- 20 R. Katsarava, N. Kulikova, J. Puiggali, Amino acid based biodegradable polymers—promising materials for the applications in regenerative medicine, *J. J. Regener. Med.* 1 (2016) 012.
- 21 K. Markosishvili, G. Tsitlanadze, R. Katsarava, J.G. Jr. Morris, A. Sulakvelidze, Novel sustained-release matrix based on biodegradable poly(ester amide)s and impregnated with bacteriophages and an antibiotic shows promise in management of infected venous stasis ulcers and other poorly healing wounds, *Int. J. Dermatol.* 41 (2002) 453–458.
- 22 D. Jikia, N. Chkhaidze, E. Imedashvili, I. Mgaloblishvili, G. Tsitlanadze, R. Katsarava, J.Jr. Glenn Morris, A. Sulakvelidze, The use of a novel biodegradable preparation capable of the sustained release of bacteriophages and ciprofloxacin, in the complex treatment of multidrug-resistant staphylococcus aureus-infected local radiation injuries caused by exposure to Sr90, *Clin. Exp. Dermatol.* 30 (2005) 23–26.
- 23 C.C. Chu, R. Katsarava. Elastomeric functional-biodegradable copolyester amides and copolyester urethanes (accessed on 5 August 2008) <http://www.google.tl/patents/US7408018>.
- 24 S.H. Lee, I. Zinai, K. Carpenter, R. Katsarava, G. Jokhadze, C.C. Chu, Y. Huang, E. Verbeken, O. Bramwell, I. De Scheerder, M.K. Hong, In vivo biocompatibility evaluation of stents coated with a new biodegradable elastomeric and functional polymer, *Coron. Artery Dis.* 13 (2002) 237–241.
- 25 Z. Gomurashvili, H. Zhang, J. Da, T.D. Jenkins, J. Hughes, M. Wu, L. Lambert, K.A. Grako, K.M. DeFife, K. MacPherson, V. Vassilev, R. Katsarava, V.G. Turnell, From drug-eluting stents to biopharmaceuticals: poly(ester amide) a versatile new bioabsorbable biopolymer. in: A. Mahapatro, A.S. Kulshrestha (Eds.), *ACS Symposium Series 977: Polymers for Biomedical Applications*, Oxford University Press: Oxford, UK, 2008, pp. 10–26.
- 26 M. Kropp, K.-M. Morawa, G. Mihov, A.K. Salz, N. Harmening, A. Franken, A. Kemp, Dias, J. Thies, S. Johnen, G. Thumann, Biocompatibility of poly(ester amide) (PEA) microfibrils in ocular tissues, *Polymers*. 6 (2014) 243–260.
- 27 V. Andrés-Guerrero, M. Zongc, E. Ramsay, R. Rojas, S. Sarkhel, B. Gallego, R. de Hoz, A.I. Ramirez, J.J. Salazar, A. Trivino, J.M. Ramirez, E.M. Del Amo, N. Cameron, B. de-Las Heras, A. Urtti, G. Mihov, A. Dias, R. Herrero-Vanrell, Novel biodegradable polyesteramide microspheres for controlled drug delivery in ophthalmology, *J. Control. Release*. 211 (2015) 105–117.
- 28 S. Laurent, L.H. Yahia, Protein corona: applications and challenges, in: B. Martinac (Ed.), *Protein-Nanoparticle Interactions*, Springer-Verlag Berlin Heidelberg, 2013, <https://doi:10.1007/978-3-642-37555-2>.
- 29 B. Sahoo, M. Goswami, S. Nag, S. Maiti, Spontaneous formation of a protein corona prevents the loss of quantum dot fluorescence in physiological buffers, *Chem Phys Lett.* 445 (2007) 217–220.
- 30 C. Le Boursais, L. Acar, H. Zia, P.A. Sado, T. Needham, R. Leverge, Ophthalmic drug delivery systems—recent advances, *Prog. Retin. Eye Res.* 17 (1998) 33–58.
- 31 M. Mudgil, N. Gupta, M. Nagpal, P. Pawar, Nanotechnology: a new approach for ocular drug delivery system, *Int. J. Pharm. Pharm. Sci.* 4 (2012) 105–112.

- 32 P. Dandagi, S. Kerur, V. Mastiholimath, A. Gadad, A. Kulkarni, Polymeric ocular nanosuspension for controlled release of acyclovir: in vitro release and ocular distribution, *Iranian J. Pharm. Res.* 8 (2009) 79–86.
- 33 C. Giannavola, C. Bucolo, A. Maltese, D. Paolino, M.A. Vandelli, G. Puglisi, V.H.L. Lee, M. Fresta, Influence of preparation conditions on acyclovir-loaded poly-D,L-lactic acid nanospheres and effect of PEG coating on ocular drug bioavailability, *Pharmaceutical Research*. 20 (2003) 584–590.
- 34 Tem. Kantaria, Teng. Kantaria, S. Kobauri, M. Ksovreli, T. Kachlishvili, N. Kulikova, D. Tugushi, R. Katsarava, Biodegradable nanoparticles made of an amino acid based ester polymers: preparation, characterization, and in vitro biocompatibility study, *Appl. Sci.* 6 (2016) [https://doi:10.3390/app6120444](https://doi.org/10.3390/app6120444).
- 35 T. Memanishvili, N. Zavrashvili, N. Kupatadze, D. Tugushi, M. Gverdsiteli, V.P. Torchilin, C. Vandrey, L. Baldi, S.S. Manoli, R. Katsarava, Arginine-based biodegradable ether-ester polymers of low cytotoxicity as potential gene carriers, *Biomacromolecules*. 15 (2014) 2839–2848.
- 36 N. Zavrashvili, G. Jokhadze, M. Gverdsiteli, G. Otinashvili, N. Kupatadze, Z. Gomurashvili, D. Tugushi, R. Katsarava, Amino acid based epoxy-poly(ester amide)s – a new class of functional biodegradable polymers: synthesis and chemical transformations, *J. Macromol. Sci., Part A, Pure & Appl. Chem.* 50 (2013) 449–465.
- 37 T. Mosmann, Rapid colorimetric assay for cellular growth and survival: Application to proliferation and cytotoxicity assays, *J. Immunol. Methods*. 65 (1983) 55–63.

4. NANOTECHNOLOGY

4.1. Materials and Structures

2.4.1.1. The Inelastic/Elastic and Tribological Properties of PTFE-Based Nanocomposites Filled with Co Cluster-Doped CNTs. /E. Kutelia, G. Darsavelidze, T. Dzigrashvili, D. Gventsadze, O. Tsurtsunia, L. Gventsadze, T. Kukava, L. Rukhadze, L. Nadaraia, I. Kurashvili, S. Bakhtiyarov/. *Bulletin of the Georgian National Academy of Sciences*. – 2020. – vol. 14. – #1. – pp. 57-63. – eng.; abs.: eng., geo.

The elastic/inelastic behavior and tribological properties of new PTFE-based nanocomposite materials filled with 5wt% and 10wt% Co atomic cluster-doped carbon nanotubes (CNTs) were investigated using low-frequency amplitude-independent (AIIF) and amplitude-dependent (ADIF) internal friction measurements, compressive deformation and tribological test methods. It is shown that the Co atom cluster-doped CNTs filler provides a considerable positive effect on the physicomechanical characteristics of the respective PTFE-based nanocomposite materials used for tribological applications. The obvious effectiveness of the externally applied gradient magnetic field in the process of mixture preparation has been established for sintering of the PTFE-based nanocomposite materials modified by carbon nanotubes doped with the ferromagnetic atom (Co) clusters, finally resulting in the improvement of wear- and creep resistance of the obtained nanocomposites. Fig. 2, Tab. 2, Ref. 13.

Keywords: PTFE, co cluster-doped CNTs, nanocomposite, internal friction, wear

References:

1. Flom D.G., Porile N.T. (1955) Friction of teflon sliding on teflon. *Journal of Applied Physics*, 26(9):1088–1092.
2. Ohzawa Y., Wada Y. (1964) Mechanical relaxations and transitions in polytetrafluoroethylene. *Japanese Journal of Applied Physics*, 3(8):436–447.
3. Biswas S.K. (1992) Friction and wear of PTFE, review. *Wear*, 158:193–211.
4. Moore D.F. (1975) Principles and applications of tribonics, 488, Pergamon Press, New York.
5. Ayman A.A., El-Sharafei B.Z., Alshennawy A.A., El-Masry A.A., Wasel W.A. (2012) Friction and wear of polymer composites filled by nano-particles: a review. *World Journal of Nano Science and Engineering*, 2:32–39.
6. Ye J., Burriss D.L., Xie T. (2016) A review of transfer film and their role in ultra-low-wear sliding of polymers. *Lubricants*, 4(4):1–15.
7. Gandotra H., Mahajan S., Jandival S., Gupta S. (2018) Effect of fillers on tribological properties of PTFE: a review. *International Journal of Scientific and Technical Advancements*, 4(1):147–150.
8. Chen W.X., Li F., Han G., Xia J.B., Wang L.Y., Tu J.P., Xu Z.D. (2003) Tribological behavior of carbon nanotube-filled PTFE composites. *Tribology Letters*, 15(3):275–278.
9. King V.B. (Ed.) (2007) Nanotechnology (research advance), Nova Publishers, Inc.: New York.

10. Rukhadze L.N., Kutelia E.R., Maisuradze N.I., Eristavi B.G., Bakhtyarov S.I. (2010) Magnetic carbon nanopowders. *International Journal of Manufacturing and Technology*, 4(2):75–80.
11. Kutelia E., Rukhadze L., Jalabadze N., Dzigrashvili T., Tsurtsumia O., Gventsadze D. (2018) Nucleation and growth of carbon nanoforms on the surface of metallic plate-substrates and the mechanism of their doping with the clusters of ferromagnetic atoms. *Advanced Materials Letters*, 9(12):867–871.
12. Kutelia E., Gventsadze D., Tsurtsumia O., Rukhadze L., Jalabadze N., Kukava T., Dzigrashvili T. (2018) Investigation of new antifrictional/frictional nanocomposites based on PTFE matrix filled with Fe-doped carbon nanoparticles. *Advanced Materials Letters*, 9(5):320–325.
13. Kutelia E., Darsavelidze G., Dzigrashvili T., Kukava T., Rukhadze L., Gventsadze L., Tsurtsumia O., Nadaraia L., Kurashvili I., Bakhtyarov S. (2018) Internal friction in PTFE-based nanocomposite materials filled with Fe cluster-doped CNTs. *Georgian Engineering News*, 87(3):5–13.

2.4.1.2. The Influence of Cycling Deformation and Annealing on the Elastic/Inelastic Properties of PTFE-Based Nanocomposite Filled with 7.5wt% Fe Cluster-Doped CNTs. /E. Kutelia, G. Darsavelidze, T. Dzigrashvili, L. Rukhadze, D. Gventsadze, I. Kurashvili, L. Nadaraia, O. Tsurtsumia, L. Gventsadze, I. Losaberidze, S. Bakhtyarov/. *Bulletin of the Georgian National Academy of Sciences*. – 2020. – vol. 14. – #2. – pp. 36-41. – eng.; abs.: eng., geo.

For the development of PTFE-based nanocomposites with the regulated technological mechanical and parameters, it is necessary to reveal a correlation between their structural and dynamical mechanical properties. The researches in this direction were performed using a low-frequency internal friction technique. The behavior of the elastic/inelastic properties of PTFE-based nanocomposite material filled with the optimal (7.5 wt%) concentration of Fe atom cluster-doped carbon nanotubes (CNTs), depending on high amplitude cycling deformation and post-deformation annealing was investigated using amplitude-independent (AIIF) and amplitude-dependent (ADIF) internal friction measurements. The characteristics of dynamical-mechanical strengthening of the Fe cluster doped PTFE-based polymeric materials were determined for the first time, and the possible mechanisms of strengthening have been analyzed. It was shown that high-amplitude cyclic deformation leads to a considerable reduction in activation energy (H , kcal/mole) of β (crystalline) and α (amorphous) relaxation processes, the magnitude of critical amplitudes (ϵ_c) of microplastic deformation beginning and shear modulus ($G \sim f^2$) in comparison to those for the initial sample before cyclic deformation. It was also found that the post-deformation annealing of the cyclically deformed sample at 150°C/30 min ensures a complete restoration of the above parameters to the values exceeding those for the initial sample. Fig. 2, Tab. 1, Ref. 16.

Keywords: PTFE, nanocomposite, Fe cluster-doped CNT, internal friction, shear modulus

References:

1. Moore D.F. (1975) *Principals and applications of tribonics*, 488, Pergamon Press, New York.
2. Flom D.G., Porile N.T. (1955) Friction of teflon sliding on teflon. *Journal of Applied Physics*, 26(9):1088–1092.
3. McCrum N.G. (1959) An internal friction study of polytetrafluoroethylene. *Journal of Polymer Science*, 34:355–369.
4. Bunn C.W., Cobbold A.J., Palmer R.P. (1958) The fine structure of polytetrafluoroethylene. *Journal of Polymer Science*, 28:363–376.
5. Speerschneider C.J., Li C.H. (1962) Some observation of structure of polytetrafluoroethylene. *Journal of Applied Physics*, 33: 1871-1875.
6. Weeks J.J., Sanchez I.C., Eby R.K., Poser C.I. (1980) Order-disorder transitions in polytetrafluoroethylene. *Polymer*, 21:325–331.
7. Clark E.S. (1999) The molecular confirmations of polytetrafluoroethylene – forms II and IV. *Polymer*, 40:4659–4665.
8. Aly A. A., Zeidan E.-S.B., Alshennawy A.A., El-Masry A.A., Wasel W.A. (2012) Friction and wear of polymer composites filled by nano-particles. *World Journal of Nano Science and Engineering*, 2:32–39.
9. Gandotra H., Mahajan S., Jandaval S., Gupta S. (2018) Effect of fillers on tribological properties of PTFE. *International Journal of Scientific and Technical Advancements*, 4(1):147–150.
10. Ye J., Burriss D.L., Ti. Xie (2016) A review of transfer film and their role in ultra-low-wear sliding of polymers. *Lubricants*, 4(4):1–15.
11. Chen W.X., Li F., Han G., Xia J.B., Wang L.Y., Tu J.P., Xu Z.D. (2003) Tribological behavior of carbon nanotube-filled PTFE composites. *Tribology Letters*, 1593:275–278.

12. Rukhadze L.N., Kutelia E.R., Maisuradze N.I., Eristavi B.G., Bakhtyarov S.I. (2010) Magnetic carbon nanopowders. *International Journal of Manufacturing and Technology*, 4(2):75–80.
13. Kutelia E., Rukhadze L., Jalabadze N., Dzigrashvili T., Tsurtsumia O., Gventsadze D. (2018) Nucleation and growth of carbon nanoforms on the surface of metallic plate-substrates and the mechanism of their doping with the clusters of ferromagnetic atoms. *Advanced Materials Letters*, 9(12):867–871.
14. Kutelia E., Gventsadze D., Tsurtsumia O., Rukhadze L., Jalabadze N., Kukava T., Dzigrashvili T. (2018) Investigation of new antifrictional/frictional nanocomposites based on PTFE matrix filled with Fe-doped carbon nanoparticles. *Advanced Materials Letters*, 9(5):320–325.
15. Kutelia E., Darsavelidze G., Dzigrashvili T., Kukava T., Rukhadze L., Gventsadze L., Tsurtsumia O., et al. (2018) Internal friction in PTFE-based nanocomposite materials filled with Fe cluster-doped CNTs. *Georgian Engineering News*, 87(3):5–13.
16. Blanter M.S., Golovin I. S., Neuhauser H., Sining H. R. (2007) *Internal friction in metallic materials*. Springer-Verlag, 250, Berlin, Heidelberg.

2.4.1.3. Study of the physico-chemical properties of silver nanoparticles stabilized with oleic acid using theoretical calculation. /P. Toidze, M. Gabrichidze/. *Ceramics and Advanced Technologies*. – 2019. – vol. 21. – #1(41). – pp. 14-20. – geo.; abs.: geo., eng.

Inorganic-organic core-shell nanoparticles are considered to be common building blocks for synthesis of multifunctional hybrid nanocomposites, which are promising materials for biomedical and catalytic application. Knowledge of the metal nanoparticle-ligand interaction mechanism is crucial for design strategy of such materials. In the present study silver-oleic acid capped nanoparticles are used as a model for mono- and bilayer ligand chemisorption. HyperChem software generates molecules (the builder), perform structural optimizations, and analyze molecular orbitals and its relation to functionality. Molecular modeling involves the development of mathematical models of molecules that can be used to predict and interpret their properties. A quantum mechanical model of the electronic structure of a molecule, which involves solving the Schrödinger equation. Quantum mechanics can be used to predict electronic properties of molecules, such as dipole moments and spectroscopy. Quantum chemical simulation leads farther insight into the mode of bonding and structure of adsorbed layer. OA interaction with Ag atoms results in charge density increase at metal surface and creation of negative electrostatic potential at carboxyl group owing to covalent bonding. Formation of secondary layer accompanied by redistribution of charge density: slight decrease in metal surface charge density, double decrease of charge density at C=C bond and strong increase in negative charge of carboxyl group of secondary layer. The absence of a double bond in the molecule of stearic acid affects the quality of stabilization of the surface of silver nanoparticles. Controlled release of biologically active silver from nanosilver can be regulated by the surface ligands. The capabilities of nanosilver in inhibiting bacteria were ascribed to the surface ligand-mediated silver ion release from both extracellular process and intracellular manner. The studies AgNPs showed that internalized AgNPs caused cell damage through binding with chain-related proteins and interrupting the electron transfer process. The HyperChem program allows quantum-chemical calculations to explain the role of oleic acid in the formation of mono- and bilayers, the catalytic effect of nano-silver in the oxidation of oleic acid with permanganate and conformational changes in the peptide fragment. Quantum-mechanical calculations allow one to establish the bond lengths in the molecule, the values of the effective charges, and the distribution of the electrostatic potential. Fig. 6, Ref. 17.

Keywords: effective charges, electrostatic potential, oleic acid, stearic acid, lipid-II, nanocomposite

References:

1. www.hyper.com
2. Comotti M., Della Pina C., Matarrese R., Rossi M. The Catalytic Activity of “Naked” Gold Particles. *Angew. Chem. Int. Ed.* (2004), 43, 5812–5815.
3. Rashid M.H., Mandal T.K. Templateless Synthesis of Polygonal Gold Nanoparticles: An Unsupported and Reusable Catalyst with Superior Activity. *Adv. Funct. Mater.* (2008), 18, 2261–2271.

4. Wang L., Yamauchi Y. Strategic Synthesis of Tri metallic Au@Pd@Pt Core–Shell Nanoparticles from Poly (vinylpyrrolidone)-Based Aqueous Solution toward Highly Active Electrocatalysts. *Chem. Mater.* (2011), 23, 2457–2465.
5. Polshettiwar V., Varma R.S. Green chemistry by nano-catalysis. *Green Chem.* (2010), 12, 743–754.
6. Li D., Sun C.Y., Huang Y.G., Li J.H., Chen S.W. Surface effects of monolayer-protected gold nanoparticles on the redox reactions between ferricyanide and thiosulfate. *Sci. Chin. Ser. B* (2005), 48, 424–430.
7. Lu Y., Lu X., Mayers B.T. et al. Synthesis and characterization of magnetic Co nanoparticles: A comparison study of three different capping surfactants. *Journal of Solid State Chemistry* (2008), 181, 1530–1538.
8. Murray J.S., Lane P., Politzer P. Relationships between impact sensitivities and molecular surface electrostatic potentials of nitroaromatic and nitroheterocyclic molecules. *Mol. Phys.* (1995), 85, 1–8.
9. Owens F.J., Jayasuriya K., Abrahamsen L., Politzer P. Computational analysis of some properties associated with the nitro groups in polynitroaromatic molecules. *Chem. Phys. Lett.* (1985), 116, 434–438.
10. Marambio–Jones C., Hoek E.M.V. A review of the anti bacterial effects of silver nanomaterials and potential implications for human health and the environment. *J. Nano Res.* (2010), 12(5), 1531–1551.
11. Liu J, Hurt R.H. Ion release kinetics and particle persistence in aqueous nano-silver colloids. *Environ. Sci. Technol.* (2010), 44(6), 2169–2175.
12. Ma R., Levard C., Marinakos S.M., et al. Size-controlled dissolution of organic coated silver nanoparticles. *Environ. Sci. Technol.* (2012), 46(2), 752–759.
13. Sotiriou G.A., Pratsinis S.E. Anti bacterial activity of nanosilver ions and particles. *Environ. Sci. Technol.* (2010), 44(14), 5649–5654.
14. Liu J., Sonshine D.A, Shervani S., Hurt R.H. Controlled release of biologically active silver from nanosilver surfaces. *ACS Nano* (2010), 4(11), 6903–6913.
15. Kim J.S., Kuk E., Yu K.N., et al. Antimicrobial effects of silver nanoparticles. *Nanomed. Nanotechnol.* (2007), 3(1), 95–101.
16. Long Y.-M., Hu L.-G., Yan X.-T., et al. Surface ligand controls silver ion release of nanosilver and its antibacterial activity against *Escherichia coli*. *Int. J. Nanomed.* (2017), 12, 3193–3206.
17. Donadze M., Gabrichidze M., Calvache S., Agladze T. Novel method for preparation of the hybrid metal (I)-metal (II) oxide nanoparticles. *Int. J. Sur. Eng. Coat.* (2016), 94(1), 16-23.

4.2. Obtaining Technologies

2.4.2.1. Obtaining of bionanoceramic super paramagnetic materials for the creation of local controlled hyperthermia for malignant cancer therapy. /Z. Kovziridze, N. Nijaradze, N. Darakhvelidze/. *Ceramics and Advanced Technologies.* – 2019. – Vol. 21. – #1(41). – pp. 21-37. – geo.; abs.: geo., eng.

The article deals with such matters as a comparative study of anticancer properties of hyperthermia induced by hematite nanoparticles and the mechanisms of their impact; creation of principally new drug of high anticancer effect; preparation-concentration of a drug containing hematite nano-particles, control of activity; a comparative study of anticancer activity of the drug; the determination-development of optimal regime and schemes; an analysis of powder of the obtained hematite nanoparticles showed homogeneous spreading of particles according to their dimensions and correspondingly – good stability. By further treatment of ferric ions obtained in Zeta potential device above Curie (769oC) temperature (800oC) the hematite nanoparticles of 80 nm size were obtained in oxidizing medium at the regime 4-5C/min. Tab. 1, Fig. 12, Ref. 24.

Keywords: nanoparticle, hematite, hyperthermia, stability

References:

1. P. Wust, B. Hildebrandt, G. Sreenivasa, B. Rau, J. Gellermann, H. Riess, R. Felix, P.M. Schlag. *Lancet Oncol.*, 3, 487489 (2002).
2. P. Moroz, S.K. Jones, B.N. Gray. *J. Surg. Oncol.*, 77, 259269 (2001).
3. R.K. Gilchrist, R. Medal, W.D. Shorey, R.C. Hanselman, J.C. Parrott, C.B. Taylor. *Ann. Surg.*, 146, 596606 (1957).
4. H. Matsuki, T. Yanada, T. Sato, K. Murakami, S. Minakawa. *Mater. Sci. Eng. A*, 181–182, 13661368 (1994).
5. R. Hergt, W. Andra, C. d’Ambly, I. Hilger, W. Kaiser, U. Richter, H. Schmidt. *IEEE Trans. Magn.*, 34, 37453754 (1998).

6. M. Shinkai, M. Yanase, M. Suzuki, H. Honda, T. Wakabayashi, J. Yoshida, T. Kobayashi. *J. Magn. Magn. Mater.*, 194, 176184 (1999).
7. A. Jordan, R. Scholz, P. Wust, H.H. Föhling, R. Felix, *J. Magn. Magn. Mater.*, 201, 13419 (1999).
8. A. Jordan, R. Scholz, K. Maier–Hauff, M. Johannsen, P. Wust, J. Nadobny, H. Schirra, H. Schmidt, S. Deger, S. Loening, W. Lanksch, R. Felix. *J. Magn. Magn. Mater.*, 225, 118126 (2001).
9. R. Müller, R. Hergt, M. Zeisberger, W. Gawalek. *J. Magn. Magn. Mater.*, 289, 1316 (2005).
10. T. Atsumi, B. Jeyadevan, Y. Sato, K. Tohji. *J. Magn. Soc. Jpn.*, 30, 555560 (2006).
11. G.F. Goya, R. Fernandez–Pacheco, M. Arruebo, N. Cassinelli, M. R. Ibarra. *J. Magn. Magn. Mater.*, 316, 132135 (2007).
12. T. Atsumi, B. Jeyadevan, Y. Sato and K. Tohji. *J. Magn. Magn. Mater.*, 310, 28412843 (2007).
13. L.-Y. Zhang, H.-C. Gu, X.-M. Wang. *J. Magn. Magn. Mater.*, 311, 228233 (2007).
14. J.-P. Fortin, C. Wilhelm, J. Servais, C. Ménager, J.-C. Bacri, F. Gazeau. *J. Am. Chem. Soc.*, 129, 26282635 (2007).
15. C. Dennis, A.J. Jackson, J.A. Borchers, R. Ivkov, A.R. Foreman, J.W. Lau, E. Goernitz, C. Gruettner. In: 52nd Annual Conference on Magnetism and Magnetic Materials, No. 59, Tampa, Florida, (2007) p. 29.
16. R.E. Rosensweig. *J. Magn. Magn. Mater.*, 252, 370374 (2002).
17. M. Suto, Y. Hirota, M. Mamiya, A. Fujita, R. Kasuya, K. Tohji, B. Jeyadevan. *J. Magn. Magn. Mater.*, 321, 14931496 (2009).
18. M. Suto, Y. Hirota, M. Mamiya, R. Kasuya, A. Fujita, K. Tohji, B. Jeyadevan. *J. Magn. Soc. Jpn.*, 33, 391395 (2009).
19. J.-P. Fortin, F. Gazeau, C. Wilhelm. *Eur. Biophys. J.*, 37, 223228 (2008).
20. T. Sugimoto, E. Matijević. *J. Colloid Interface Sci.*, 74, 227243 (1980).
21. M. Tada, S. Hatanaka, H. Sanbonsugi, N. Matsushita, M. Abe. *J. Appl. Phys.*, 93, 75667568 (2003).
22. B. Jeyadevan. *J. Cream. Soc. Jpn.*, 118(6), 391–401 (2010).
23. J. Bullivant. Stable Superparamagnetic Ferrofluids for the Treatment of Secondary Liver Cancer by Hiperthermia, Dissertation, Univ. Florida, 2008.
24. R.D. Zysler, D. Fiorani, A.M. Testa. *J. Magn. Magn. Mater.*, 224, 5–11 (2001).
25. R.D. Zysler, et al. *J. Magn. Magn. Mater.*, 224, 39–48 (2001).

2.4.2.2. Electrosynthesis and application of nanomagnetite for purification of water previously contaminated by phenol. /M. Donadze, N. Makhaldiani/. *Ceramics and Advanced Technologies*. – 2020. - vol. 22. – #1(43). – pp. 15-22. – geo.; abs.: geo., eng.

The aim of the study is the electrosynthesis of Fe_3O_4 nanomagnetite and the purification of precontaminated water from phenol using a filter containing nanomagnetite. The main component of the filter is magnetite nanoparticles stabilized with oleic acid, obtained by electrosynthesis in a two-layer bath. An aluminum arc was used as a rotating cathode and optimal electrolysis parameters were determined. A porous filter was obtained after impregnation of bohemite with magnetic nanoparticles and its subsequent burning at 450°C. In a two-layer bath, a monodispersed sol of magnetite in hexane was obtained. The optimal parameters of electrolysis are determined. The resulting nanomagnetite was characterized by X-ray analysis (XRD), infrared spectroscopy (FT-IR), elementary analysis and scanning microscopy (SEM-EDS). Particle size determined by dynamic light scattering (DLS Malvern). A filter based on nanomagnetite shows a significant effect in the process of purifying drinking water from phenol. Monodisperse organosol of nanomagnetite was obtained by electrolysis in a two-layer bath. A porous filter containing nanomagnetite can be used to purify water contaminated with phenol at the place of consumption. Fig. 8, Ref. 17.

Keywords: nanomagnetite, electrosynthesis, bohemite, Fenton mechanism, phenol

References:

1. Sh. Zhang, X. Zhao, H. Niu, Y. Shi, Y. Cai, G. Jiang, Superparamagnetic Fe_3O_4 nanoparticles as catalysts for the catalytic oxidation of phenolic and aniline compounds. *Journal of Hazardous Materials* 167 (2009) 560–566.
2. C.L. Hsueh, Y.H. Huang, C.C. Wang, C.Y. Chen, Degradation of azo dyes using low iron concentration of Fenton and Fenton-like system. *J. Chemosphere* 58 (2005) 1409–1414.
3. S. Pereira da Silva, D. Costa de Moraes, D. Samios. Iron Oxide Nanoparticles Coated with Polymer Derived from Epoxidized Oleic Acid and Cis-1,2-Cyclohexanedicarboxylic Anhydride: Synthesis and Characterization. *J. Mater. Sci. Eng.*, 2016, 5:3. DOI: 10.4172/2169-0022.1000247.

4. A.A. Baharuddin, B.Ch. Ang, N.A. Abu Hussein, A. Andriyana, Y.H. Wong. Mechanisms of highly stabilized ex-situ oleic acid-modified iron oxide nanoparticles functionalized with 4-pentynoic acid. *Materials Chemistry and Physics*, 203 (2018) 212-222.
5. L. Zhang, R. He, H.-Ch. Gu. Oleic acid coating on the monodisperse magnetite nanoparticles. *Applied Surface Science*, j.apsusc.2006.05.023.
6. F. Fajaroh, H. Setyawan, W. Widiyastuti, S. Winardi. Synthesis of magnetite nanoparticles by surfactant-free electrochemical method in an aqueous system. *Advanced Powder Technology*, 23 (2012) 328–333.
7. R.F.C. Marques, C. Garcia, P. Lecante, J.L. Ribeiro, L. Noe, N.J.O. Silva, V.S. Amaral, A. Millan, M. Verelst, Electro-precipitation of Fe₃O₄ nanoparticles in ethanol, *J. Magn. Magn. Mater.* 320 (2008) 2311–2315.
8. L. Cabrera, S. Gutierrez, N. Menendes, M.P. Morales, P. Herrasti, Magnetite nanoparticles: Electrochemical synthesis and characterization. *Electrochem. Acta*, 53 (2008) 3436–3441.
9. S. Franger, P. Berthet, J. Berthon, Electrochemical synthesis of Fe₃O₄ nanoparticles in alkaline aqueous solutions containing complexing agents, *J. Solid State Electrochem.*, 8 (2004) 218–223.
10. S. Franger, P. Berthet, O. Dragos. Large influence of the synthesis conditions on the physico-chemical properties of nanostructured Fe₃O₄. *J. Nanopart. Res.*, 9 (2007) 389–402.
11. M. Doandze, T. Agladze. Strategy for Nanohybridized Synthesis of M_aM_bO_x System, In the Book: Chemical Engineering and polymers Production of Functional and Flexible Materials, Apple Academic Press, Chapter 13, Canada.
12. T. Agladze, M. Donadze, M. Gabrichidze, P. Toidze, J. Shengelia, N. Boshkov, T.N. Tsvetkova. *Z. Phys. Chem.*, 227 (2013) 1187–1198.
13. M. Donadze, M. Gabrichidze, T. Agladze. Novel method of Fabrication of Hybrid Metal(I)/Metal(II) Oxides Nanoparticles. *Transaction of the IMF: The International Journal of Surface Engineering and Coatings*, 94(1), (2016) 16-23.
14. S. Khutsishvili, P. Toidze, M. Donadze, M. Gabrichidze, T. Agladze, N. Makhdiani, Structural and Magnetic Properties of Silver Oleic Acid Multifunctional Nanohybrids, *Annals of Agrarian Science*, 17 (2019) 153–157.
15. თ. აგლაძე, მ. დონაძე. ვერცხლის მონოდისპერსიული ნანონაწილაკების მიღების ხერხი. *GE P 2019 7022 B*, 2019.
16. თ. აგლაძე, მ. ბაციკაძე, მ. დონაძე, ბაქტერიოციდული თვისებების მქონე ვერცხლის ნანონაწილაკების ელექტროსინთეზი ორფენიან აბაზანაში. *GE P 2011 5254 B*, 2011.
17. K. Yang, H. Peng, Y. Wen, N. Li. Reexamination of characteristic FTIR spectrum of secondary layer in bilayer oleic acid-coated Fe₃O₄ nanoparticles. *Applied Surface Science*, 256 (2010) 3093–3097.

2.4.2.3. Synthesis and application of the hybrid nanocomposite - Ag@MnO_x for purification with bacteria (e.coli) and heavy metals contaminated water. /N. Makhdiani, M. Donadze/. *Ceramics and Advanced Technologies*. – 2020. - vol. 22. - #1(43). – pp. 23-36. – geo.; abs.: geo., eng.

The aim of the study was the synthesis of the Ag@MnO_x nanocomposite and the purification of water from bacteria and heavy metal ions using filter containing a hybrid nanocomposite. The main filter component is silver nanoparticles stabilized with oleic acid, obtained by electrosynthesis in a two-layer bath. An aluminum arc was used as a rotating cathode and optimal electrolysis parameters were determined. Hybrid nanocomposite obtained by oxidation of oleic acid with potassium permanganate. A porous filter was obtained by coating of honeycomb structure cordierite with a primary layer-washcoat (γ-Al₂O₃) and its subsequent impregnation with a hybrid nanocomposite. The use of an arc instead of a disk-shaped cathode in a two-layer bath reduces the size of silver particles and increases the degree of monodispersity. A filter based on a hybrid nanocomposite shows a good antibacterial effect in the process of purification of drinking water from E.coli bacteria; good sorption effect for copper ions and sorption and oxidative effect for manganese ions. A porous honeycomb structure filter containing a nanohybrid composite Ag @MnO_x can be used to purify water contaminated with bacteria and heavy metals at the place of consumption (well water, exotic tourism zone, etc.). Tab. 1, Fig. 14, Ref. 16.

Auth.

Keywords: nanosilver, hybrid nanocomposite, cordierite, heavy metals, coli index

References:

1. საქართველოს მთავრობის დადგენილება № 58, სასამართლო წყლის ტექნიკური რეგლამენტის დამტკიცების შესახებ.

2. World Health Organization (WHO) and the United Nations Children's Fund (UNICEF). Progress on Drinking Water, Sanitation and Hygiene: 2017 Update and SDG Baselines; WHO: Geneva, Switzerland, 2017.
3. B.G. Ershov, E. Janata, A. Henglein. Growth of silver particles in aqueous solution: long-lived "magic" clusters and ionic strength effects. *J. Phys. Chem.*, 97, 2 (1993) 339–343.
4. B. Domènech, M. Muñoz, D.N. Muraviev, J. Macanás. Polymer-Silver Nanocomposites as Antibacterial materials, Microbial pathogens and strategies for combating them: science, technology and education (A. Méndez-Vilas, Ed.) (2013) 630–640.
5. L. Abhishek, R. Abhishek karthik, K. Deepak Kumar, G. Sivakumar. Advanced Water Treatment Using Nano-Materials, *Int. J. of Innovative Research in Science, Engineering and Technology* (An ISO 3297:2007 Certified Organization), Vol. 3, Iss. 11 (2014) 17130–17138.
6. M. Doandze, T. Agladze. Strategy for Nanohybridized Synthesis of $M_aM_bO_x$ System. In the Book: Chemical Engineering and Polymers Production of Functional and Flexible Materials, Apple Academic Press, Chapter 13, Canada.
7. T. Agladze, M. Donadze, M. Gabrichidze, P. Toidze, J. Shengelia, N. Boshkov, T.N. Tsvetkova. *Z. Phys. Chem.*, 227 (2013) 1187–1198.
8. M. Donadze, S. Calvache, M. Gabrichidze, T. Agladze. Novel method of Fabrication of Hybrid Metal(I)/Metal(II) Oxides Nanoparticles. *Transaction of the IMF: The International Journal of Surface Engineering and Coatings*, 94(1), (2016) 16–23.
9. S. Khutsishvili, P. Toidze, M. Donadze, M. Gabrichidze, T. Agladze, N. Makhaldiani. Structural and Magnetic Properties of Silver Oleic Acid Multifunctional Nanohybrids. *Annals of Agrarian Science*, 17 (2019) 153–157.
10. თავაძე, მ. დანაძე. ვერცხლის მონოლსპერსიული ნანონაწილაკების მიღების ხერხი, *GE P 2019 7022 B*, 2019.
11. N. Garti, E. Avni. Permanganate Oxidation of Oleic Acid Using Emulsion Tehnology, *JAOCS*, 58(8) (1981) 840–841.
12. A. Stefanescu, A.C. van Veen, C. Mirodatos, J.C. Beziat, E. Duval–Brunel. Wall coating optimization for microchannel reactors. *Catalysis Today*, 125, (2007) 16–23.
13. M. Bakhtiari, F. Khorasheh, A. Zamanian, A. Nakhaeipour, M. Irani. Preparation, Evaluation and Characterization of monolithic Catalysts for fisher Tropsh. *Petroleum & Coal*, 50(3), (2008) 56–61.
14. T. Zhou, L. Li, J. Cheng, Z. Hao. Preparation of binary washcoat deposited on cordierite substrate for catalytic applications. *Ceramics International*, 36 (2010) 529–534.
15. C. Agrafiotis, A. Tsetsekou. Deposition of mesoporous γ -alumina coatings on ceramic honeycombs by sol–gel methods. *J. Eur. Ceram. Soc.*, 22 (2002) 423–434.
16. D. Gregory, K. Carlson. Effect of soluble Mn concentration on oxidation kinetics Source. *Journal of American Water Works Association*, 95(1) (2003), 98–108.

2.4.2.4. Obtaining metal-containing nanoparticles in polyethylene matrix by mechano-chemical method and study of their properties. /S.K. Ragimova/. *Azerbaijan Chemical Journal*. – 2020. – #2. – pp. 20-25. – eng.; abs.: eng., az., rus.

Metal-containing nanoparticles in the matrix of high-pressure polyethylene are obtained by the mechanochemical method without the use of organic solvents by high-speed thermal decomposition of salts of organic acids under conditions of high shear deformations. The phase composition and structure of the obtained nanocomposites were studied by X-ray phase, scanning electron microscope, and thermogravimetric analyzes. It is shown that the formation of nanoparticles of metal oxides in the polymer matrix, contribute to the stabilization of the composite, raising the temperature of the onset of its thermaloxidative degradation. Micrographs of the obtained nanocomposites indicate the formation of layered structures that possess high fracture toughness. Fig. 8, Tab. 1, Ref. 12.

Keywords: metal-containing nanoparticles, high pressure polyethylene, mechano-chemical method, Xray phase, scanning electron microscope and thermogravimetric analyzes

References:

1. Gubin S.P., Yurkov G.Yu., Kosobudsky I.D. Nanomaterials based on metal-containing nanoparticles in polyethylene and other carbon-chain polymers. *Int. J. Materials and Product Technology*. 2005. V. 23. No 1–2. P. 2–25.
2. Pomogailo A.D. *Molekuliarnye polimer-polimernye kompozitcii. Sinteticheskie aspekty. Uspehi himii*. 2002. T. 71. No 1. S. 5–38.
3. Mihailin Iu.A. *Polimernye nanokompotcionnye materialy. Polimernye materialy*. 2009. No 7. S. 10–13.

4. Gubin S.P. Chto takoe nanochastitsy? Tendentsii razvitiia nanohimii i nanotekhnologii. Ros. him. zhurn. 2000. XLIV. No 6. S. 23–31.
5. Kosobudskii I.D., Gubin S.P. Novyi tip metallopolimerov – metallicheskie clastery v polimernykh matritcakh. Vysokomolek. soedineniia. 1985. A. T. 27. No 3. S. 689–695.
6. Iurkov G. Iu., Kozi nkin A.V., Nedoseikina T.I., Shuvaev A.T., Vlasenko V.G., Gubin S.P., Kosobudskii I.D. Nanochastitsy medi v polietilenovoi matritce. Neorgan. materialy. 2001. T. 37. No 10. S. 1175–1179.
7. Tarasova N.P., Nefedov O.M., Lunin V.V. Himiia i problemy ustoychivogo razvitiia i sokhraneniia okruzhaiushchei sredy. Uspehi himii. 2010. T. 79. No 6. S. 491–492.
8. Kurbanova N.I., Alimirzoeva N.A., Guseynova Z.N., Nurullayeva D.R. Ecological Method of Preparation of Metal-Containing Nanoparticles in Polyethylene Matrix. 3 st International Turkic World Conference on Chemical Sciences and Technologies. Azerbaijan, Baku. 2017. 10–13 Sept. Book of Proceedings. P. 24–26.
9. Kurbanova N.I., Kuliye v A.M., Alimirzoeva N.A., Aliyev A.T., Ichenko N.Ya., Nurullayeva D.R. Preparation of copper-containing nanoparticles in polyethylene matrix. 5th Int. Caucasian Symposium on Polymer & Advanced Materials. Georgia, Tbilisi 2017. 2–5 July. P. 91.
10. Berlin A.I. Nekotorye perspektivy razvitiia polimernykh konstruktsionnykh materialov. Vysokomolek. soed. 2010. A. T. 52. No 9. S. 1541–1550.
11. Chvalun S.N. Polimernye nanokompozity. Priroda. 2000. No 7. S. 1–12.
12. Praktikum po himii fizike polimerov. Pod red. V.F. Kurenkova. M.: Himiia, 1990. 299 s.

2.4.2.5. Synthesis, conversion and antimicrobial activity of derivatives of 2-hydroxy-1-haloidphenoxyethers of nonanol-2. /S.A. Mammadov, A.A. Mahmudova, G.G. Mammadova, V.S. Hasanov, N.P. Ladokhina, L.F. Zeynalova/. Azerbaijan Chemical Journal. – 2020. – #1. – pp. 59-65. – eng.; abs.: eng., az., rus.

Reactivity of hydroxyl group 1-haloid-phenoxy-2-hydroxynonanes to nucleophilic substitution with acetoxymethylchloride was studied. It was found that regardless of the nature of haloids and their position in aryloxy radical, the yields of acetoxymethyl ethers make nearly 70% that proves high reactivity of hydroxyl group. Indeed, during the reaction with isocyanates phenoxynonanol-2 forms urethanes with yield 69–70%. Initial synthon was prepared by the reaction of 1-bromine-nonanol-2 with substituted phenols. Study of synthesized polyethers and urethanes as antimicrobial additives to lubricating oils gave positive results. It was determined that urethanes exhibit stronger antimicrobial effect than polyethers. Fig. 2, Table. 3, Ref. 12.

Keywords: phenoxyethers, urethanes, polyethers, acetoxymethyl, antimicrobial additives

References:

1. Balazadə P.Ş., Mahmudova Ə.Ə., Həsənov V.S. 1-(o-Bromfenoksi)-3-noniltio-2-propanolun bəzi törəmələrinin sintezi və xassələrinin tədqiqi // Azerb. Chem. Journ. 2009. № 1. S. 182–185.
2. Həsənov V.S., Mahmudova Ə.Ə., Babayeva G.V., Rəhimov İ.S. 1-Fenoksi-3-propiltio-2-propanolun bəzi törəmələrinin sintezi və xassələrinin tədqiqi // Azerb. Chem. Journ. 2014. № 1. S. 104–108
3. Həsənov V.S., Mahmudova Ə.Ə., Babayeva G.V., Zeynalova L.F. N,N-dimetilditiokarbamin turşusunun α -hidroksipropil efirinin sintezi və bəzi cəvrimləəri // Azerb. Chem. Journ. 2013. № 3. S. 121–124.
4. Həsənov V.S., Mahmudova Ə.Ə., Babayeva G.V., Baxşiyeva Ü.Ş. 1-(m-bromfenoksi)-2-dekanolun sintezi, cəvrimləəri və cəvrimlə məhsullarının mikrob əleyhinə xassələrinin tədqiqi // Azerb. Chem. Journ. 2013. № 4. S. 101–105.
5. Həsənov V.S., Mahmudova Ə.Ə., Babayeva G.V., Baxşiyeva Ü.Ş., Əkbərli H.N. 1-(m-Toliloksi)-2-oktanolun sintezi cəvrimləəri və cəvrimlə məhsullarının mikrob əleyhinə xassələrinin tədqiqi // Azerb. Chem. Journ. 2014. №3. S.110-116.
6. Chen Xingkuan, Jaqueleine Zi Mei, Xu Jianfeng, Mou Chengli. Dynamic kinetic identification of ethers of carboxylic acids catalyzed with carbenes. J. Amer. Chem. Soc. 2016. V. 138. No 23. P. 7212–7215.
7. Liu Heng, Eisen Moris S. Selective, catalyzed with actinides etherification between aldehydes and alcohols with proton transfer for obtaining of asymmetric complex ethers. Organometallics. 2017. V. 36. No 8. P. 1461–1464.
8. Wu Xia, Geng Xiao, Jhao Peng, Zhang Zungjing. J2-promoted formal [3+2] cycleaddition of α metyl enyl isocyanides with metyl ketons: a route to 2,5-disubstituted oxazoles. Chem. Commun. 2017. V. 53. No 4. P. 3439–3441.
9. Young Jae Choi, Yoo Chang Kim, Sook Jin Park, Jun Min Jung, Young Hoon Jung. Total synthesis of (–)-codonopsinine via regioselective and diastereoselective amination using chlorosulfonyl isocyanate. Tetrahedron. 2017. V. 73. No 30. P. 4458–4463.

10. Wi Xiao, Masen Jess, Norh L. Formation of isocyanurates in synthesis process of oxazolidinones from epoxides and isocyanates catalyzed with chromium complex (Chrom Salphen). *J. Chem. Eur J.* 2018. V. 23. No 52. P. 12937–12943.
11. Tamula Nagarajo, J. Nagesh, P.R., Krishna S. Balasubramanian S. Catalyzed Oxidative N,S-Bond Formation: Metal-Free Regioselective Synthesis of N-Substituted and 3,3-Disubstituted 5-amino-1,2,4-thiadiazoles. *J. Org. Chem.* 2012. V. 82. No 10. P. 5310–5316.
12. Guo Wei, Zhen Mingming, Tan Wen, Zheng Leyin. Three-component tandem annealing promoted with visible light in the synthesis of 2-iminothiazolin-4-ones. *J. Org. Chem.* 2018. No 3. P. 1402–1413.

2.4.2.6. Deposition of nanodrop phase from emitter tip on nearby mobile surface. /I.S. Gasanov, S.A. Aliyev, I.I. Gurbanov, E.M. Akberov, F.E. Mamedov, A.H. Kerimova/. *Azerbaijan Journal of Physics.* – 2020. – vol. 26. – #1. – pp. 40-43. – eng.; abs.: eng.

The formation processes of low-sized structures by means of a fine-dispersed phase of liquid metal ion source (LMIS) are considered. The emitting tip is located in close distance from moved surface with the aim of deposition of narrow stripes. At distance tip – surface near 80 μm on the axis of thin and wide traces of (In^+ , Sn^+) ions the massive continuous paths by width of several microns are obtained. The structure of deposited stripes by the length more than 10 mm is the grain structure. At further approach of tip to surface, the path melts because of high density of ion current and heterogeneous profile of its cross-section become smooth. For deposition of narrower structures, the effective cooling of conducting mobile substrate is necessary. Fig. 5, Ref. 8.

Keywords: liquid metal ion source, field emission, nanoparticle

References:

1. V.V. Badan, I.S. Gasanov. The finely dispersed phase and instability of the emission of liquid metal ion sources. *Technical Physics Letters*, v. 15, No. 17, p. 49–52, 1989.
2. I.S. Gasanov, I.I. Gurbanov. Formation of charged nanoparticles at capillary instability of the liquid emitter. *JJAP*, 47, No. 10, p. 8226–8229, 2008.
3. *Charged Particle Optics*. Edited by J. Orloff (CRC Press, London, New York, 2009), p. 665.
4. V.E. Badan, V.V. Vladimirov, V.P. Gorshkov, I.A. Soloshenko. Instability of Rayleigh and Faraday in liquid metal ion sources. Drop emission and the phenomenon of microdroplet chaos. *Technical Physics*, v. 63, No. 6, p. 47–65, 1993.
5. I.S. Gasanov, I.I. Gurbanov. Nanostructure operations by means of the liquid metal ion sources. *Rev. Sci. Instr.*, 83, 02B906, 2012.
6. I.S. Gasanov, I.I. Gurbanov, E.M. Akbarov. Losses of ion energy in the multicomponent beam. *Eur. Phys. J. D.*, 2015. DOI: 10.1140/epjd/e2015-50531-0
7. I.S. Gasanov, I.I. Gurbanov, E.M. Akbarov. Ions passage through nanodroplets in multicomponent beam. *Acta Physica Polonica A*, 134, No. 1, p. 119-121, 2018. DOI: 10.12693/APhysPolA.134.119
8. C. Akhamedliyev, L. Bischoff, G.L.R. Mair, C.J. Aidinis, Th. Ganetsos. Investigation of emission stabilities of liquid metal ion sources, *Microelectron. Eng.*, 73-74, p. 120-125, 2006.

4.3. Processing Technologies

2.4.3.1. Nanotechnologies or technological advances of the future. /I. Phutkaradze/. *Chemistry Accounts.* – 2019. – vol. 3. – #1. – pp. 16-19. – geo.; abs.: geo.

Nanotechnologies are the technology of the future. It is a multidisciplinary field, which includes chemistry, biology, physics, computer science, medicine and engineering. Its main task is to manage processes at the molecular level for which purpose it uses the smallest nanoparticles sized from 1 to 100 nanometers. In the field of technologies nanomaterial chips, processors and other means of communication are noteworthy, which according to scientists, will make a nanorevolution in the world. Fig. 3, Ref. 5.

Keywords: nanotechnology, nanoparticle, nanosystem, nanomolecule, nanomedicine, nanorevolution

References

1. ა. ბიბილაშვილი. თანამედროვე ნანოტექნოლოგიები. თბილისი, თბილისის სახ. უნივ. გამომც., 2014.
2. ბლოგი: „ნანოტექნოლოგია“. ი. ჩიქოვანი. 24 მარტი, 2015. <http://iorami.blogspot.com/>
3. ბლოგი: „ნანოტექნოლოგიებზე დაფუძნებული სასუქები“, 26 ივლისი, 2013. <http://www.tabula.ge/ge/story/73252-siaxle-nanoteqnologiebze-dafudznebulisasuqebi>.
4. ბლოგი: “How nanotechnology works” – K. Bonsor, J. Strickland. <https://science.howstuffworks.com/nanotechnology.htm>
5. ბლოგი: “Nanotechnology”, February 2018, The Guardian. <https://www.theguardian.com/science/nanotechnology>.

2.4.3.2. Results of laboratory and industrial tests of IKHLAS-1 nanodemulsifier on the Akkulka Oil Field and the new mechanism of destruction of oil emulsions. /T.K. Dashdiyeva/. Azerbaijan Chemical Journal. – 2020. – #3. – pp. 34-45. – eng.; abs.: eng., az., rus.

The article presents the results of laboratory and industrial tests of the IKHLAS-1 nanodemulsifier for the Akkulka Oil Field of the LLC Tetisaralgaz of the Republic of Kazakhstan. According to the test results, it was found that the IKHLAS-1 nanodemulsifier under all technological conditions of primary oil preparation shows significant advantages compared to the basic DMO-86520 demulsifier. Therefore, IKHLAS-1 was recommended for widespread introduction of the Akkulka Oil Field at the Group Installation of primary preparation of oil. Implementation results (since October 2017) also confirms the high efficiency of the IKHLAS-1 nanodemulsifier. The article sets out also a new mechanism for the destruction of oil emulsions. Tab. 9, Ref. 15.

Keywords: nanodemulsifier IKHLAS-1, oil field nanotechnology, oil production, nanotechnology in oil and water preparation, new mechanism, destruction of oil emulsions

References:

1. Evdokimov I.N., Fesan A.A. Multi-step formation of asphaltene colloids in dilutes solutions Colloid and Surfaces A: Physicochemical and Engineering Aspects. 2016. V. 492. P. 170–180.
2. Miraləmov H.F., İsmayılov Q.Q. Neftin, qazın boru kəmərləri ilə nəqli. Bakı: NQETLİ, 2010. 505 s.
3. Levchenko D.N., Bergshtein N.V., Hudakova A.D., Nicolaeva N.M. Emulsii nefi s vodoi i metody ikh razrusheniia. M.: Energoizdat, 1987. 464 s.
4. Pozdnyshev G.N. Stabilizatsiia i razrushenie nefiianykh emulsii. M.: Nedra, 1982. 224 s.
5. Nugmanov A.K., Gasanov A.A., Dashdiyeva T.K. New hybrid aggregate state of some organic substances and their prospects in oil-field nanotechnology. Int. J. ultidisciplinary Res. Modern Education. 2019. V. 5. ISSUE 1. P. 149–161.
6. Hutorianskii F.M., Akhmedi S., Dosso U., Soltani Bekhnaz. Issledovaniia protsessa obezvozhivaniia i obessolivaniia ochen ti azheloi vysokoviazkoi nefi Verbluzhego mestorozhdeniia Astrahanskoi oblasti. Mir nefteproduktov. 2015. № 3. C. 10–16.
7. Nugmanov A.K., Dashdiyeva T.K. Surface pressure is one of the main criteria for evaluating the effectiveness of nano demulsifiers for destruction of reverse and direct, emulsions. Int. J. Innovative Res. in Sci. Engineering and Technology. 2019. V. 8. Issue 1. P. 234-242.
8. Borisov S.I., Kateev M.V., Kalinin E.S., Kalinina O.S., Meloshenko N.P., Sorokin V.V. Mehanizm deistviia PAV kak deemulgatorov nefiianykh emulsii. <http://naukarus.com/mehanizm-deystviyapav-kak-deemulgatorov-nefiyanyh-emulsiy>.
9. Evdokimov I.N., Losev A.P. Eksperimentalnye dokazatelstva otsutstviia inversii v promyslovykh vodoneftiianykh emulsiikh. Burenie i nef. 2010. № 5. S. 26–27.
10. Loskunova Iu.V., Iudina N.V., Volkova G.I., Anufrieva R.V. Izuchenie viazkostno-temperaturnogo povedeniia vodoneftiianykh emulsii v tochke inversii faz. Mezhdunar. zhurn. pricl. i fundament. Issledovaniia. 2017. № 10. S. 221–225. <https://applied-research.ru/ru/article/view?id=11892>
11. Grimes B. Population balance model for batch gravity separation of crude oil and water emulsions. Part i: Model formulation. J. Dispersion Science and Technology. 2012. V. 33. No 4. P. 578–590.
12. Gumbatov G.G., Dashdiev R.A. Primenenie PAV dlia likvidatsii avariinykh razlivov nefi na vodnoi poverkhnosti. Baku: Elm, 1998. 210 s.
13. Abramzon A.A., Gaevoi G.M. Poverkhnostnoaktivnye veshchestva. Spravochnik. L.: Himiia, 1979. 376 s.
14. Rusanov A.I., Shchekin A.K., Volkov N.A. Diffuziia v mitcelliiarnykh sistemakh: teoriia i molekuliarno modelirovanie. Uspehi himii. 2017. T. 86. № 7. C. 567–588.

15. Dashdiyeva T.K. Selection the nanoemulsions oilcollecting reagents based on adsorbtion studies on the boundary water-air interface. M.Nağiyev adına Kataliz və qeyri-üzvi kimya institutu. Konf. mater. Bakı: 15–16 noyabr 2016. S. 214–215.

5. NANOENGINEERING

5.1. Devices and Sensors

2.5.1.1. The effect of surface recombination on the open circuit voltage of a solar cell based on a single nanowire with a radial p-n-junction. /S.G. Petrosyan, V.A. Khachatryan, S.R. Nersesyan/. Proceedings of NAS RA. Physics. – 2020. – vol. 55. – #3. – pp. 343-357. – rus.; abs.: rus., arm., eng.

An analytical model is proposed for studying the effect of surface recombination on the characteristics of a solar cell based on a nanowire with a radial p-n junction formed between its ‘core’ and ‘shell’ of different types of conductivity. When varying over a wide range of shell widths, the effect of surface recombination on such important parameters of a solar cell as short circuit current, open circuit voltage, and the efficiency of solar energy conversation into electrical energy is considered. It is shown that the relatively low open circuit voltage, often observed experimentally in such solar cells, can be caused by significant surface recombination on the sidewall of the nanowire, the role of which increases with decreasing nanowire diameter and increasing surface to volume ratio. Fig. 6, Ref. 27.

Keywords: analytical model, solar cell, conductivity, short circuit current, open circuit voltage, solar energy

References:

1. В.Г. Дубровский, Г.Е. Цырлин. В.М. Устинов. ФТП, 43, 1585 (2009).
2. C. Soci, A. Zhang, X.-Y. Bao, H. Kim, Y. Lo, D. Wang. J. Nanosci. Nanotechnol., 10, 1 (2010).
3. B. Tian, J. Kempa, Ch. Lieber. Chem. Soc. Rev., 38, 16 (2009).
4. R.R. La Pierre, A.A.E. Chia, S.J. Gibson, et al. Phys. Stat. Solidi RRL, 7, 815 (2013).
5. Н.В. Сибирев, К.П. Котляр, А.А. Корякин, и др. ФТП, 52, 1464 (2018).
6. V. Schmidt, H. Rien, S. Senz, et al. Small, 2, 85 (2006).
7. E. Patolsky, G. Zheng, O. Hayden, et al. Proc. Natl. Acad. Sci. USA (PNAS), 101, 14017(2004).
8. B.T. Tian, X. Zheng, T.J. Kempa, et al. Nature, 449, 885 (2007).
9. T.J. Kempa, J.F. Cahoon, S.-K. Kim, et al. Proc. Natl. Acad. Sci. USA (PNAS), 109, 1407 (2012).
10. H.J. Fan, P. Werner, M. Zacharias. Small, 2, 700 (2006).
11. L. Tsakalakos, J. Balch, J. Fronheiser, et al. Appl. Phys. Lett., 91, 233117 (2007).
12. E. Garnett, P. Yang. J. Am. Chem. Soc. 130, 9224 (2008).
13. M.D. Kelzenberg, D.B. Turner–Evans, B. M. Kayes, et al. NanoLett., 8, 710 (2008).
14. C. Colombo, M. Heib, M. Gratzel, A. Fontcuberta I Morral. Appl. Phys. Lett., 94, 173108 (2009).
15. P. Krogstrup, H.I. Jorgensen, M. Heiss, et al. Nature Photonics, 7, 306 (2013).
16. B. M. Kayes, H. Atwater. J. Appl. Phys., 97, 114302 (2005).
17. R. R. La Pierre. J. Appl. Phys., 109, 034311 (2011).
18. A. Boukai, P. Haney, A. Katzenmeyer, et al. Chem. Phys. Lett., 501, 153 (2011).
19. J.D. Christesen, X. Zhang, Ch.W. Pinion, et al. NanoLett., 12, 6024 (2012).
20. S.G. Petrosyan, A.E. Yesayan, S.R. Nersesyan, V.A. Khachatryan. Semiconductors, 52, 2022 (2018).
21. F. Voigt, T. Stelzner, S. Christianser. Mater. Sci. Eng. B, 177, 1558 (2012).
22. J. Kupec, B. Witzigmann. Opt. Express, 17, 10399 (2009).
23. Г. Пикус. Основы теории полупроводниковых приборов, М., Наука, 1965.
24. S. Petrosyan, A. Yesayan, S. Nersesyan. World Academy of Science, Engineering and Technology, 71, 1065 (2012).
25. H. Wang, X. Liu, Zh.M. Zhang. Int. J. Thermophys, 34, 213 (2013).
26. M.A. Green. Sol. Energy Mater. Sol. Cells, 92, 1305 (2008).
27. J.V. Holm, H.I. Jorgensen, P. Krogstrup, et al. Nat. Commun., 4, 1498 (2013).

2.5.1.2. Detection of iron nanoparticles in aqueous solutions by microwave sensor. /L. Odabashyan, N. Margaryan, G. Ohanyan, M. Manvelyan, D. Hambaryan, T. Abrahamyan, R. Khachatryan, A. Babajanyan/. Proceedings of NAS RA. Physics. – 2020. – vol. 55. – #2. – pp. 251-258. – rus.; abs.: rus., arm., eng.

The aqueous solution with iron nanoparticles is investigated by a microwave stripline sensor based on the optimized double quadratic-shape design. Due to real-time near-field electromagnetic interaction between microwaves and sample S11, the reflection coefficient of the sensor changed depending on iron nanoparticles concentration in the aqueous solution at resonant frequency. This work examined the iron nanoparticles concentration in the 0–20 µg/l concentration range at an operating frequency at about 1.7 GHz. The measured minimum detectable signal was 0.035 dB/(µg/l) or and 0.25 MHz/(µg/l) and the measured minimum detectable concentration was 1.4 µg/l and 0.2 µg/l, respectively. The microwave response of sensor systems can be explained by the additional structural changes of water clusters due to the metal nanoparticles ablation. This implemented method has approachable development process and the accuracy of measurement is high, thus it can be applied as a physicochemical sensor for non-invasive monitoring of metal nanoparticles in complex liquids. Fig. 4, Ref. 14.

Keywords: iron nanoparticles, microwave stripline sensor, minimum detectable signal, ablation, physicochemical sensor

References:

1. A.I. Adejumo, O.A. Babatunde, P.O. Abimbola, A.A. Tabitha, O.D. Adewumi, O. Toyin. Water Challenges of an Urbanizing World, BoD, 2018.
2. F.D. Owa. Mediterranean Journal of Social Sciences, 4, 2039 (2013).
3. S. Sharma, A. Bhattacharya. Applied Water Science, 7, 1043 (2017).
4. V. Rashmi, D. Pratima. Recent Research in Science and Technology, 52, 98 (2013).
5. T.J. Arif, M. Azam, K. Siddiqui, A. Ali, I. Choi, Q.M. Rizwanul Haq. Int. J. Mol. Sci., 16, 29592 (2015).
6. Sh. Rasalingam, R. Peng, R.T. Koodali. Journal of Nanomaterials, 2014, 617405 (2014).
7. J. Yang, B. Hou, J. Wang, B. Tian, J. Bi, N. Wang, X. Li, X. Huang. Nanomaterials (Basel), 9, 424 (2019).
8. G. Gennarelli, S. Romeo, M.R. Scarfi, F. Soldovieri. IEEE Sensors J., 13, 1857 (2013).
9. L. Odabashyan, A. Babajanyan, Zh. Baghdasaryan, S. Kim, J. Kim, B. Friedman, J.-H. Lee, K. Lee. MDPI Sensors, 19, 5525 (2019).
10. B. Hovhannisyan, D. Hambaryan, L. Odabashyan, A. Babajanyan. Proceedings of the Yerevan State University: Physical and Mathematical Sciences, 53, 132 (2019).
11. B.J. Minasyan, L.A. Odabashyan, Zh.A. Baghdasaryan, A.Zh. Babajanyan, K. Lee. Proceedings of the Yerevan State University: Physical and Mathematical Sciences, 53, 60 (2019).
12. T. Abrahamyan, R. Khachatryan, D. Hambaryan, B. Hovhannisyan, B. Minasyan, L. Odabashyan, A. Babajanyan. J. Contemp. Phys. (Armenian Acad. Sci.), 54, 196 (2019).
13. A.V. Markel. J. Opt. Soc. Am. A, 33, 1244 (2016).
14. D.M. Pozar. Microwave Engineering, 4th ed., John Wiley & Sons, 2012, pp. 28-35.

2.5.1.3. Gas nanosensors made from semiconductor metaloxides. /V.M. Aroutiounian/. Proceedings of NAS RA. Physics. – 2019. – vol. 54. – #4. – pp. 485-501. – rus.; abs.: rus., arm., eng.

Usually, semiconductor gas sensors made from metal oxides require high pre-heating of the work body. Advantages for nanoscale sensors are the possibility to work at remarkable lower than 300°C temperature of its work body up to room temperature (practically without preheating of the sensor). Today's experimental results obtained for gas sensors made from metal oxides are reported in this review. Fig. 1, Ref. 100.

Keywords: semiconductor, gas sensor, metal oxide, work body, nanoscale sensors, preheating

References:

1. Semiconductor Gas Sensors, R. Jaaniso, O.K. Tan (ed.), Woodhead Publishing, 2013.
2. Encyclopedia of Nanoscience and Nanotechnology, S.E. Lyshevski (ed.), CRC Press, 2014.
3. F.-G. Banika. Chemical and biological sensors. Technosphaera Press, 2014.

4. G. Korotcenkov, I. Blinov, M. Ivanov, J.R. Stetter. *Sensors and Actuators B: Chemical*, 120, 679 (2007).
5. K. Khojier, N.Z. Dehnavi. *J. Theor. Appl. Phys.*, 11, 157 (2017).
6. M.R. Alenezi, S.J. Henley, N.G. Emerson, S.R.P. Silva. *Nanoscale*, 6, 235 (2014).
7. N. Nasiri, R. Bo, F. Wang, L. Fu, A. Tricoli. *Adv. Mater.*, 28, 4336 (2015).
8. A. Tricoli, M. Righettoni, A. Teleki. *Angew. Chem. Int. Ed.*, 49, 7632 (2010).
9. H. Chen, R. Bo, T. Tran-Phu, G. Liu, A. Tricoli. *Chem Plus Chem*, 83, 569 (2018).
10. S.-J. Choi, F. Fuchs, R. Demadrille, B. Grévin, B.-H. Jang, S.-J. Lee, J.-H. Lee, H. L. Tuller, I.-D. Kim. *ACS Appl. Mater. Interfaces*, 6, 9061 (2014).
11. H.G. Moon, Y. Jung, S.D. Han, Y.-S. Shim, B. Shin, T. Lee, J.-S. Kim, S. Lee, S.C. Jun, H.-H. Park, C. Kim, C.-Y. Kang. *ACS Appl. Mater. Interfaces*, 8, 20969 (2016).
12. G. Martinelli, M.C. Carotta, M. Ferroni, Y. Sadaoka, E. Traversa. *Sens. Actuator B Chem.*, 55, 99 (1999).
13. H. Chen, R. Bo, A. Shrestha, B. Xin, N. Nasiri, J. Zhou, I. Di Bernardo, A. Dodd, M. Saunders, J. Lipton–Duffin, T. White, T. Tsuzuki, A. Tricoli. *Adv. Opt. Mater.*, 6, 1800677 (2018).
14. N. Nasiri, R. Bo, L. Fu, A. Tricoli. *Nanoscale*, 9, 2059 (2017).
15. J. Pan, J. Li, Z. Yan, B. Zhou, H. Wu, X. Xiong. *Nanoscale*, 5, 3022 (2013).
16. Y.H. Choi, D.H. Kim, S.H. Hong. *ACS Appl. Mater. Interfaces*, 10, 14901 (2018).
17. R. Zhang, T. Zhou, L. Wang, Z. Lou, J. Deng, T. Zhang. *New J. Chem.*, 40, 6796 (2016).
18. Q. Wang, N. Yao, C. Liu, D. An, Y. Li, Y. Zou, X. Tong. *J. Nanomater*, 2016, 1 (2016).
19. D. An, N. Mao, G. Deng, Y. Zou, Y. Li, T. Wei. *Ceram. Int.*, 42, 3535 (2016).
20. X. Jia, M. Tian, R. Dai, D. Lian, S. Han, X. Wu. *Sens. Actuators B Chem.*, 240, 376 (2017).
21. X. Wang, B. Ding, Y. Liu, X. Zhu, H. Li, M. Xia, H. Fu, M. Li. *Sens. Actuators B Chem.*, 264, 119, (2018).
22. P. Gunawan, L. Mei, J. Teo, J. Ma, J. Highfield, Q. Li, Z. Zhong. *Langmuir*, 28, 14090, (2012).
23. H. Liu, M. Li, O. Voznyy, L. Hu, Q. Fu, D. Zhou, Z. Xia, E.H. Sargent, J. Tang. *Adv. Mater.*, 26, 2718 (2014).
24. L. Li, L. Gu, Z. Lou, Z. Fan, G. Shen. *ACS Nano*, 11, 4067 (2017).
25. Q. Huang, D. Zeng, H. Li, C. Xie. *Nanoscale*, 4, 5651 (2012).
26. S.M. Sedghi, Y. Mortazavi, A. Khodadadi. *Sens. Actuator B Chem.*, 145, 7 (2010).
27. F. Shao, M.W.G. Hoffmann, J. D. Prades, R. Zamani, J. Arbiol, J. R. Morante, E. Varechkina, M. Rumyantseva, A. Gaskov, I. Giebelhaus, T. Fischer, S. Mathur, F. Hernández–Ramírez. *Sens. Actuator B Chem.*, 181, 130 (2013).
28. M. Righettoni, A. Tricoli, S.E. Pratsinis. *Chem. Mater.*, 22, 3152 (2010).
29. J. Deng, Q. Fu, W. Luo, X. Tong, J. Xiong, Y. Hu, Z. Zheng. *Sens. Actuator B Chem.*, 224, 153 (2016).
30. E. Lhuillier, M. Scarafagio, P. Hease, B. Nadal, H. Aubin, X.Z. Xu, N. Lequeux, G. Patriarche, S. Ithurria, B. Dubertret. *Nano Letters*, 16, 1282 (2016).
31. Z. Song, Z. Wei, B. Wang, Z. Luo, S. Xu, W. Zhang, H. Yu, M. Li, Z. Huang, J. Zang, F. Yi, H. Liu. *Chem. Mater.*, 28, 1205 (2016).
32. X. Chen, Z. Guo, W.-H. Xu, H.-B. Yao, M.-Q. Li, J.-H. Liu, X.-J. Huang, S.-H. Yu. *Adv. Funct. Mater.* 21, 2049 (2011).
33. J. Tan, M. Dun, L. Li, J. Zhao, W. Tan, Z. Lin, X. Huang. *Sens. Actuators B Chem.*, 249, 44 (2017).
34. M. Napi, S.M. Sultan, R. Ismail, M. Ahmad, G. Chai. *Journal of Nanomaterials*, 2019, Article ID 4574507 (2019).
35. J. Gong, Y. Li, Z. Hu, Z. Zhou, Y. Deng. *J. Phys. Chem. C*, 114, 9970 (2010).
36. N. Du, H. Zhang, B.D. Chen, X.Y. Ma, Z.H. Liu, J.B. Wu, D. R. Yang. *Adv. Mater.*, 19, 1641 (2007).
37. Q. Qi, P.-P. Wang, J. Zhao, L.-L. Feng, L.-J. Zhou, R.-F. Xuan, Y.-P. Liu, G.-D. Li. *Sens. Actuator B Chem.*, 194, 440 (2014).
38. C. Turner, P. Španěl, D. Smith. *Physiological Measurement*, 27, 321 (2006).
39. S. Davies, P. Spanel, D. Smith. *Kidney International*, 52, 223 (1997).
40. S.-J. Kim, S.-J. Choi, J.-S. Jang, H.-J. Cho, W.-T. Koo, H. L. Tuller, I.-D. Kim. *Advanced Materials*, 29, 1700737 (2017).
41. L. Zainovia. *Advances and material science and engineering*, Boca Raton, FL: CRC Press/Taylor & Francis Group, 2018.
42. X. Xu, H. Zheng, Ch. He, Ch. Pu, Y. Leng, G. Li, Sh. Hou, Y. Zhu, L. Fu, G. Li. *The Royal Soc. Chem.*, 6, 47083 (2016).
43. Y. Li, H. Ban, M. Yang. *Sens. Actuator B Chem.*, 224, 449 (2016).
44. V. Galstyan, *Sensors*, 17, 2947 (2017).
45. X. Li, Z. Li, J. Wang, J. Zhang. *Sens. Actuator B Chem.*, 240, 273 (2017).
46. Q. He, Z. Zeng, Z. Yin, H. Li, S. Wu, X. Huang, H. Zhang. *Small*, 8, 2994 (2012).
47. L. Mädler, A. Roessler, S.E. Pratsinis, T. Sahm, A. Gurlo, N. Barsan, U. Weimar. *Sens. Actuators B Chem.*, 114, 283 (2006).
48. V.M. Aroutiounian. *Graphene Science Handbook, Applications and Industrialization*. CRC Press Taylor & Francis Group, USA, Fl., Boca Raton, Chapter 20, 299 (2016).
49. O. Leenaerts, B. Partoens, F.M. Peeter. *Physical Review B*, 77, 125416 (2008).
50. M. Gautam, A.H. Jayatissa. *Journal of Applied Physics*, 111, 094317 (2012).
51. F. Schedin, A.K. Geim, S.V. Morozov, E.W. Hill, P. Blake, M.I. Katsnelson, et al. *Nature Materials*, 6, 652 (2007).

-
52. J.D. Fowler, M.J. Allen, V.C. Tung, Y. Yang, R.B. Kaner, B.H. Weiller. *ACS Nano*, 3, 301 (2009).
 53. H.J. Yoon, D.H. Jun, J.H. Yang, Zh. Zhou, S.S. Yang, M. Ming-Cheng. *Sensors and Actuators B*, 157, 310 (2011).
 54. G. Ko, Y. Jung, K.Y. Lee, K. Lee, J. Kim. *Journal of Crystal Growth*, 326, 208 (2011).
 55. M.G. Chung, D.H. Kim, H.M. Lee, T. Kim, J.H. Choi, D.K. Seo, et al. *Sensors and Actuators B*, 166–167, 172 (2012).
 56. M.G. Chung, D.-H. Kim, D.K. Seo, T. Kim, H. U. Im, H.M. Lee, et al. *Sensors and Actuators B*, 169, 387 (2012).
 57. R. Rapola, J.M. Kalaw, F.B. Sevilla. *Applied Mechanics and Materials*, 492, 321(2014).
 58. Z. Song, Z. Wei, B. Wang, Z. Luo, S. Xu, W. Zhang, H. Yu, M. Li, Z. Huang, J.Zang, F. Yi, H. Liu. *Chem. Mater.*, 28, 1205 (2016).
 59. Sh. Srivastava, K. Jain, V.N. Singh, S. Singh, N. Vijayan, N. Dilawar, G. Gupta, T.D. Senguttuvan. *Nanotechnology*, 23, 205501 (2012).
 60. Q.A. Drmosh, Z.H. Yamani, A.H.Y. Hendi, M.A. Gondal, R.A. Moqbel. 17th International Meeting on Chemical Sensors – IMCS, 525 (2018).
 61. J.-H. Lee, A. Katoch, S.-W. Choi, J.-H. Kim, H.W. Kim, S.S. Kim. *ACS Appl. Mater. Interfaces*, 7, 3101 (2015).
 62. P.A. Russo, N. Donato, S.G. Leonardi, S. Baek, D.E. Conte, G. Neri, N. Nicola Pinna. *Angew. Chem. Int. Ed.*, 51, 11053 (2012).
 63. R. Ghosh, A.K. Nayak, S. Santra, D. Pradhan, K. Prasanta Guha. *RSC Advance*, 5, 50165 (2015).
 64. S. Romyantsev, G. Liu, M.S. Shur, R.A. Potyrailo, A.A. Balandin. *Nano Letters*, 12, 2294 (2012).
 65. V. Aroutiounian, Z. Mkhitarian, A. Adamian, C.-G. Granqvist, L. Kish. *IEEE Sensor Journal*, 8, 786 (2008).
 66. V. Aroutiounian, Z. Mkhitarian, A. Adamian, C.-G. Granqvist, L. Kish. *Procedia Chemistry*, 1, 216 (2009).
 67. J. Wang, B. Singh, J.-H. Park, et al. *Sensors and Actuators B*, 194, 296 (2014).
 68. S. Hafiz, R. Ritikos, T. Witcher, et al. *Sensors and Actuators B*, 193, 692 (2014).
 69. G. Lu, L.E. Ocola, J. Chen. *Nanotechnology*, 20, 445502 (2009).
 70. J.T. Robinson, E.K. Perkins, E.S. Snow, Z. Wei, P.E. Sheehan. *Nano Letters*, 8, 3137 (2008).
 71. Y. Zhu, Ah. Murali, W. Cai, X. Li, J.W. Suk, J.R. Potts, et al. *Advanced Materials*, 22, 3906 (2010).
 72. V. Singh, D. Joung, I. Zhai, S. Das, S.I. Khondaker, S. Seal. *Progress in Materials Science*, 56, 1178 (2011).
 73. T.V. Cuong, V.H. Pham, J.S. Chung, A.G. Joshi, N. Singh, S. Singh. *Materials Letters*, 64, 2479 (2010).
 74. Q. He, Z. Zeng, Z. Yin, H. Li, S. Wu, X. Huang, H. Zhang. *Small*, 8, 2994 (2012).
 75. X. Li, X. Li, Z. Li, J. Wang, J. Zhang. *Sens. Actuator B Chem.*, 240, 273 (2017).
 76. H. Tan, Y. Fan, Y. Zhou, Q. Chen, W. Xu, J.H. Warner. *ACS Nano*, 10, 7866 (2016).
 77. H.G. Moon, Y.R. Choi, Y.-S. Shim, K.-I. Choi, J.-H. Lee, J.-S. Kim, S.-J. Yoon, H.-H. Park, C.-Y. Kang, H.W. Jang. *ACS Appl. Mater. Interfaces*, 5, 10591 (2013).
 78. H.G. Moon, Y. Jung, S.D. Han, Y.-S. Shim, W.-S. Jung, T. Lee, S. Lee, J.H. Park, S.-H. Baek, J.-S. Kim. *Sens. Actuator B Chem.*, 257, 295 (2018).
 79. S. Yang S., Ch. Jiang, S. Wei. *Appl. Phys. Rev.*, 4, 021304 (2017).
 80. M. Righettoni, A. Tricoli, S. Gass, A. Schmid, A. Amann, S.E. Pratsinis. *Anal. Chim. Acta*, 738, 69 (2012).
 81. L. Mädler, A. Roessler, S.E. Pratsinis, T. Sahm, A. Gurlo, N. Barsan, U. Weimar. *Sens. Actuators B Chem.*, 114, 283 (2006).
 82. Q.A. Drmosh, Z.H. Yamani, A.H. Hendi, M.A. Gondal, R.A. Moqbel, T.A. Saleh, M.Y. Khan. *Applied Surface Science*, 464, 616 (2019).
 83. B. Grabowska–Polanowska, J. Faber, M. Skowron, P. Miarka, A. Pietrzycka, I. Śliwka, A. Amann. *J. Chromatography A*, 1301, 179 (2013).
 84. K. Kostikas, A. Koutsokera, S. Papiris, K. Gourgoulianis, S. Loukides. *Clin. Exp. Allergy*, 38, 557 (2008).
 85. Z. Dai, C.-S. Lee, B.-Y. Kim, C.-H. Kwak, J.-W. Yoon, H.-M. Jeong, J.-H. Lee. *ACS Appl. Mater. Interfaces*, 6, 16217 (2014).
 86. W.C. Chan, D.J. Maxwell, X. Gao, R.E. Bailey, M. Han, S. Nie. *Current Opinions in Biotechnology*, 13, 40 (2002).
 87. N. Barsan, U. Weimar. *J. Phys. Condensed Matter*, 15, R813 (2003).
 88. H.R. Kim, A. Haensch, I.D. Kim, N. Barsan, U. Weimar, J.H. Lee. *Adv. Funct. Mater.*, 21, 4456 (2011).
 89. K.-I. Choi, H.-J. Kim, Y.C. Kang, J.-H. Lee. *Sens. Actuator B Chem.*, 194, 371 (2014).
 90. N. Nasiri, R. Bo, H. Chen, T.P. White, L. Fu, A. Tricoli. *Adv. Opt. Mater.*, 4, 1787 (2016).
 91. N. Nasiri, R. Bo, L. Fu, A. Tricoli. *Nanoscale*, 9, 2059 (2017).
 92. A.T. Güntner, M. Righettoni, S.E.J.S. Pratsinis, A.B. Chemical. *Sens. Actuator B Chem*, 223, 266 (2016).
 93. J.-S. Jang, S.-J. Choi, S.-J. Kim, M. Hakim, I.-D. Kim. *Adv. Funct. Mater.*, 26, 4740 (2016).
 94. N. Nasiri, Ch. Clarke. *Biosensors*, 9, 43 (2019).
 95. A. Forleo, L. Francioso, S. Capone, P. Siciliano, P. Lommens, Z. Hens. *Sensors and Actuators B Chem*, 146, 111 (2010).
 96. J. Kim, K. Yong. *J. Phys. C*, 115, 7218 (2011).
 97. H. Fan, X. Jia. *Solid State Ionics*, 192, 688 (2011).
 98. J. Huang, L. Wang, C. Gu, Z. Wang, Y. Sun, J.-J. Shim. *Sens. Actuator B Chem*, 207, 782 (2015).

99. V. Aroutiounian. *Sensors & Transducers*, 228, 1 (2018).
100. V. Aroutiounian. *Reports of National Academy of Sciences*, 119, 3 (2019).

2.5.1.4. Electronic and transport properties of boron nitride nanodevice (BNNT). /I.M. Danglyan, E.M. Kazaryan, D.B. Hayrapetyan/. *Armenian Journal of Physics*. – 2019. – vol. 12. – #4. – pp. 344-348. – eng.; abs.: eng.

Boron nitride nanotubes are valued due to their physical and chemical properties. They can be applied in the field of design and developing of optoelectronic devices of new generation. In this paper, the transport and electronic properties of both pure boron nitride and boron nitride nanotube with embedded carbon atoms have been calculated in the framework of the Density Functional Theory (DFT). The results show that a nanodevice with embedded carbon atoms has wider transmission spectrum than the pure one. Transmission eigenvalues for both nanodevices were computed. A nanodevice with impurity has higher transmission eigenvalues than the pure one. Fig. 4, Tab. 1, Ref. 6.

Keywords: Boron Nitride nanotube, transmission spectrum, eigenvalue, eigenstate

References:

1. M. L. Cohen, A. Zettl. *Phys. Today* 2010, 63, 34–38.
2. J.-F Jia, H.-S. Wu, H. Jiao. The structure and electronic property of BN nanotube, *Physica B*, 2006, 381(1), 90-95.
3. I. M. Danglyan, E. M. Kazaryan, D. B. Hayrapetyan. The Impact of Carbon Atoms on Boron Nitride Nanotubes, *Journal of Physics: Conference Series*, 2019.
4. J. Taylor, H. Guo, J. Wang. Ab initio modeling of open systems: Charge transfer, electron conduction, and molecular switching of a C60 device. *Phys. Rev. B*. 2001, 63, 121104.
5. J.P. Perdew, A. Zunger. Self-interaction correction to density-functional approximations for manyelectron systems. *Phys. Rev. B*. 1981, 23, 5048–5079.
6. M. Büttiker, Y. Imry, R. Landauer, S. Pinhas. Generalized many-channel conductance formula with application to small rings. *Phys. Rev. B*. 1985, 31, 6207–6215.

2.5.1.5. Semiconductor gas sensors using Arduino nano. /V. Aroutiounian, A. Hovhannisyan/. *Armenian Journal of Physics*. – 2019. – vol. 12. – #4. – pp. 325–328. – eng.; abs.: eng.

A programmable board with its own processor and Arduino Nano memory was used. The board has a couple of dozen contacts, to which all kinds of components (displays, light emitting devices (LEDs), sensors, motors, routers, magnetic locks, etc.) can be connected. A gas detector using Arduino Nano circuit was proposed. Fig. 3, Ref. 9.

Keywords: gas detector, semiconductor, Arduino Nano

References:

1. Aroutiounian V. M. *Dekker Encyclopedia of Nano-science and Nanotechnology*, Second Edition, Taylor and Francis: New York, pp. 1–10, 2012.
2. Aroutiounian V. M. *Semiconductor gas sensors*, Woodhead Publishing Series in Electronic and Optical. Materials N 38, chapter 12, pp. 408–430, 2013.
3. Aroutiounian V. M. *Advanced Sensors for Safety and Security*, NATO Science for Peace and Security Series B: Physics and Biophysics, chapter 9, pp. 105–124, 2013.
4. Aroutiounian V. M. *Graphene Science Handbook. Applications and Industrialization*. CRC Press Taylor&Francis Group, USA, Fl., Boca Raton, chapter 20, pp. 299–310, 2016.
5. Aroutiounian V. M. *Sensors & Transducers*, v. 223, N 7, pp. 9–21, 2018.
6. Aroutiounian V. M. *International Scientific Journal for Alternative Energy and Ecology*, N 01–03 (249–251), pp. 38–48, 2018.
7. Aroutiounian V. M. *Sensors & Transducers*, v. 228, N 12, pp. 1–16, 2018.
8. Aroutiounian V. M., Zakaryan H. *Sensors & Transducers*, v. 212, N 5, pp. 50–56, 2017.
9. Aroutiounian V., Kirakosyan V. *Armenian Journal of Physics*, v. 11, N 3, pp. 160–165, 2018.

2.5.1.6. Numerical study of Josephson nanostructures using parallel computing. /I.R. Rahmonov, E.V. Zemlyanaya, M.V. Bashashin, P. Atanasova, A.R. Rahmonova, Yu.M. Shukrinov/. Armenian Journal of Physics. – 2019. – vol. 12. – #3. – pp. 233-239. – eng.; abs.: eng.

The phase dynamics of the stack of long JJs, the length of which exceeds the Josephson penetration depth λ_J , taking into account the inductive and capacitive couplings between junctions and diffusion current is investigated. Numerical simulation of current-voltage characteristics of the stack is based on numerical solution of a system of nonlinear partial differential equations by the fourth order Runge-Kutta method and finite-difference approximation. The calculations are performed using the MPI technique for parallel implementation. The methodical calculations on multi-processor cluster (LIT JINR) with a different number of parallel MPI-processes are carried out. It is shown that the developed parallel algorithm provides about 7-fold acceleration in comparison with serial simulation. Fig. 2, Tab. 1, Ref. 22.

Keywords: Josephson junction, inductive coupling, capacitive coupling

References:

1. R. Kleiner, F. Steinmeyer, G. Kunkel, P. Müuller, Phys. Rev. Lett. 68 (1992) 2394.
2. Yu. M. Shukrinov and F. Mahfouzi, Phys. Rev. Lett. 98 (2007) 157001.
3. Yu. M. Shukrinov, F. Mahfouzi, and M. Suzuki, Phys. Rev. B 78 (2008) 134521.
4. I.R. Rahmonov, Y.M. Shukrinov, A. Irie, JETP Letters 99 (2014) 632.
5. Yu. M. Shukrinov, A. E. Botha, S. Yu. Medvedeva, M. R. Kolahchi and A. Irie, Chaos 24 (2014) 033115.
6. T.A. Fulton and R.C. Dynes, Solid St. Commun. 12 (1972) 57.
7. N. F. Pedersen and D. Welner, Phys. Rev. B 29 (1984) 2551.
8. D.W. McLaughlin and A. C. Scott, Phys. Rev. A 18, (1978) 1652.
9. S. Lin, X. Hu, Phys. Rev. Lett. 100 (2008) 247006.
10. R. Kleiner, T. Gaber, G. Hechtfisher, Phys. Rev. B 62 (2000) 4086.
11. Y. Matsuda, M.B. Gaifullin, K. Kumagai, K. Kadowaki, T. Mochiku, Phys. Rev. Lett. 75, (1995) 4512.
12. A.A. Yurgens, Supercond.Sci.Technol. 13, (2000) R85.
13. T.M. Benseman, A.E. Koshelev, K.E. Gray, W.K. Kwok, U. Welp, K. Kadowaki, M.Tachiki, T. Yamamoto, Phys. Rev. B 84 (2011) 064523.
14. L. Ozyuzer, A.E. Koshelev, C. Kurter et al., Science 318 (2007) 1291.
15. I.R. Rahmonov, Yu.M. Shukrinov, P.Kh. Atanasova, E.V. Zemlyanaya, M.V. Bashashin, JETP 124131 (2017).
16. I. R. Rahmonov, Yu. Shukrinov, P. Atanasova, E.V. Zemlyanaya, O. I. Streltsova, M. Zuev, A. Plecenik, and A. Irie, EPJ Web of Conferences 173, 06011 (2018).
17. S. Sakai, P. Bodin, N.F. Pedersen, J. Appl. Phys. 73 (1993) 2411.
18. M. Machida, S. Sakai, Phys. Rev. B 70 (2004) 144520.
19. Yu.M. Shukrinov, F. Mahfouzi, Physica C 434 (2006) 6.
20. Y.M. Shukrinov, I. R. Rahmonov, JETP 115 (2012) 289.
21. Y.M. Shukrinov, F. Mahfouzi, Phys. Rev. Lett. 98 (2007) 157001.
22. Y.M. Shukrinov, M. Hamdipour, JETP Letters 95 (2012) 307.

6. NANOMEDICINE

6.2. Medical Chemistry

2.6.2.1. Impact of Ag nanoparticles on microorganisms, causative agents of purulent-inflammatory processes. /M. Mishina, A. Syrova, V. Abramenko, V. Makarov, O. Hopta/. Georgian Medical News (GMN). – 2019. – #4. – pp. 139-143. – eng.; abs.: eng., rus., geo.

Determination of the Ag nanoparticles' impact on microorganisms causative agents of purulent-inflammatory processes was carried out. It was stated that the greatest significance of growth inhibition zone was found in Staphylococcus aureus and Streptococcus pyogenes with sample length from 1 to 6 mm and Escherichia coli with 5–6 mm sample length. The investigated strains in an amount 10^4 – 10^6 CFU/ml were sensitive to Ag nanoparticles activity, but at concentration 108 CFU/ml and more all strains were

found persistent to samples of various length. The ability to form biofilms with planktonic cells of microorganisms under Ag nanoparticles activity sufficiently reduced from 3.4 (*Candida albicans*) to 5.5 (*Klebsiella pneumonia*) in investigated strains. The disorganization of daily biofilms was found in determining of Ag nanoparticles impact on formed biofilms of microorganisms. Fig. 7, Ref. 17.

Keywords: Ag nanoparticles, silver nanoparticles, antimicrobial effects, catheter-associated infections, biofilms, reference strains of microorganisms

References:

1. Винник Ю.С., Перьянова О.В., Онзуль Е.В., Теплякова О.В. Микробные биопленки в хирургии: механизмы образования, лекарственная устойчивость, пути решения проблемы // *Новости хирургии*. – Т. 8, № 6. – 2010. – С. 115–125.
2. Герхардт Ф. Методи загальної бактеріології: Пер. з англ. Т. 1 – М.: Світ, 1983. – 536 с. Режим доступу до публікації: <http://medical.co.nf>
3. Лапач С.Н., Чубенко А.В., Бабич П.Н. Статистические методы в медико-биологических исследованиях с использованием Excel. – К.: МОРИОН, 2000. – 320 с.
4. Методичні вказівки щодо застосування уніфікованих мікробіологічних (бактеріологічних) методів дослідження в клініко-діагностичних лабораторіях. Додаток I до Наказу Міністерства охорони здоров'я № 535 від 22 квітня 1985. – 123 с.
5. Мусаелян А.Г. Биопленочные инфекции // *Бюллетень медицинских Интернет-конференций* (ISSN 2224-610). – 2017. – Том 7. № 1. – С. 269–271.
6. Осипов В.П., Лукьянова Е.М., Антипкин Ю.Г. и др. Методика статистической обработки медицинской информации в научных исследованиях. К.: Планета людей; 2002:200.
7. Патент на корисну модель 47944 Україна, МПК G09B23/00. Спосіб відтворення біоплівок мікроорганізмів *in vitro*. А.Я.Циганенко, М.М.Мишина, Р.А. Курбанов(UA); Харк. націон. мед. ун-т. – № u200910353; Заявл. 12.10.2009; Опубл. 25.02.2010, Бюл. № 4.
8. Синетар Е.О., Брич О.І., Лоскутова М.М., Ткачик І.П. Антибіотикостійкість та адгезивні властивості збудників катетер-асоційованих інфекцій сечовивідних шляхів // *Мікробіологічний журнал* 2014; 76(3):41-46.
9. Синетар Е. О. Вплив на частинки срібла на формування біоплівки бактеріями *Enterococcus faecalis* // *Вісник проблем біології і медицини* 2015; Вип. 4, Том 1(124):201–205.
10. Соколова Т.Н. Микробные биопленки и способы их обнаружения // *Журнал Гродненского государственного медицинского университета* 2014; 4:12–14.
11. Хренов П.А., Честнова Т.В. Обзор методов борьбы с микробными биопленками при воспалительных заболеваниях // *Вестник новых медицинских технологий* – 2013 – Том 7, № 1. Электронное издание <http://www.medtsu.tula.ru/VNMT/Bulletin/E2013-1/4102.pdf>
12. Franci G., Falanga A., Galdiero S., Palomba L., Rai M., Morelli G., Galdiero M.. Silver Nanoparticles as Potential Antibacterial Agents. *Molecules* 2015; 20:8856–8874.
13. Kim J.S., et al. Antimicrobial effects of silver nanoparticles // *Nanomedicine: Nanotechnology, Biology and Medicine* 2007; 3(1):95–101.
14. Hatipoglu M.K., Keleştemur S., Altunbek M., Culha M. Source of cytotoxicity in a colloidal silver nanoparticle suspension // *Nanotechnology* 2015; 26(19):195- 103.
15. Rabin N., Zheng Y., Opoku-Temeng C., Du Y., Bonsu E., Sintim H.O.. Biofilm formation mechanisms and targets for developing antibiofilm agents // *Future Medicinal Chemistry* 2015; 7(4):493–512.
16. O'Toole G.A., Kaplan H.B., Kolter R. Biofilm formation as microbial development // *Ann. Rev. Microbiol.* 2000; 54:49–79.
17. Eckhardt S., Brunetto P.S., Gagnon J., Pribe M., Giese B., Nanobio K.M.. Silver: Its Interactions with Peptides and Bacteria, and Its Uses in Medicine // *Chemical Reviews* 2013; 113(7):4708–4754.

2.6.2.2. Formulation of biodegradable polymeric nanoparticles containing cytotoxic substance of plant origin. /L. Ebralidze, A. Tsertsvadze, E. Sanaia, D. Berashvili, A. Bakuridze/. *Georgian Medical News (GMN)*. – 2020. – #2. – pp. 137-142. – eng.; abs.: eng., rus., geo.

Formulation of novel drug delivery system is one of the approaches for improvement of pharmacological activity of drugs. This implies encapsulation of the API into the biocompatible polymeric material. Objective of the research was the formulation of biodegradable amino acid-based polyesteramide nanoparticles composing cytotoxic substance of plant origin. The research materials and methods included:

biodegradable polyesteramide (PEA), alkaloids from *Vinca Minor*, surfactants (Tween 80, polyvinyl pyrrolidone, polyvinyl alcohol, Poloxamer 188). NPs size (mean particle diameter) and size distribution (polydispersity index, PDI), and zeta-potential (ZP) were measured by dynamic light scattering (DLS) using a Zetasizer Nano ZS (Malvern Instruments, U.K.) at 25°C, UV spectrophotometer was used for %EE study. Amino acid-based PEA particles were fabricated by the modified emulsification method. Based on the studies optimal composition and fabrication condition of PEA NPs was determined. The conditions of the NPs fabrication were as follows: the O/W ratio: 1:10; the solvent: DMSO; polymer concentration in the organic phase: 50.0 mg/mL; surfactants (PVA) concentration in aqueous phase 0.5%, the stirring rate: 1000 rpm. The influence of the various factors such as organic solvents, surfactants, as well as a polymer concentration in the organic phase, surfactant concentration in the aqueous phase, the organic/water phase ratio on the NPs fabrication was studied. The NPs were characterized by size (mean particle diameter & size distribution (polydispersity index, PDI), and zeta-potential (ZP). Increase concentration of the surfactant (polyvinyl alcohol) from 0.1% to 0.5% decrease average particle size from 568±63 to 169±1.66 respectively. EE% was obtained to be around 50%. Tab. 7, Fig. 3, Ref. 8.

Keywords: Nanoparticle, entrapment, polymer, biodegradable, alkaloids

References:

1. Chen X., et al. Significant suppression of non-small-cell lung cancer by hydrophobic poly(ester amide) nanoparticles with high docetaxel loading. *Front. Pharmacol.* 2018; 9(2):1–11.
2. Crucho C.I. Synthesis of Polymeric Nanoparticles for Biomedical Delivery Applications 2015; 7.
3. Crucho C.I., Barros M.T. Polymeric nanoparticles: A study on the preparation variables and characterization methods. *Mater. Sci. Eng. C.* 2017; 80:771–784.
4. He Z., et al. Scalable fabrication of size-controlled chitosan nanoparticles for oral delivery of insulin. *Biomater.* 2017; 130:28–41.
5. Heinz H., et al. Nanoparticle decoration with surfactants: Molecular interactions, assembly, and applications. *Surf. Sci. Rep.* 2017; 72(1):1–58.
6. Hossen S., Hossain M.K., Basher M.K., Mia M.H., Rahman M.T., Uddin M.J. Smart nanocarrier-based drug delivery systems for cancer therapy and toxicity studies: A review. *J. Adv. Res.* 2018.
7. Rizvi S.A., Saleh A.M. Applications of nanoparticle systems in drug delivery technology. *Saudi Pharm. J.* 2018; 26(1):64–70.
8. Wurm F.R., Weiss C.K. Nanoparticles from renewable polymers. *Front. Chem.* 2014; 2:1–13.